

## **Acentric structure (*P3*) of bechererite, $\text{Zn}_7\text{Cu}(\text{OH})_{13}[\text{SiO}(\text{OH})_3\text{SO}_4]$**

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### **ABSTRACT**

The crystal structure of bechererite from the Tonopah-Belmont mine, Arizona, was reinvestigated using a single-crystal X-ray diffractometer (MoK $\alpha$  radiation) equipped with a CCD area detector. The structure was refined in space group *P3* ( $a = 8.319(2)$ ,  $c = 7.377(1)$  Å) with 1247 unique reflections up to  $(\sin\theta)/\lambda = 0.65$  leading to  $R1 = 2.7\%$ ,  $wR2 = 6.4\%$ . In agreement with the previously reported centric structure ( $P\bar{3}$ ), bechererite is composed of (001) brucite-like sheets formed by edge-sharing  $(\text{Zn,Cu})\text{O}_6$  octahedra and  $1/7$  ordered octahedral vacancies. The layers are connected parallel to  $c$  by ditetrahedral  $(\text{Zn,Cu})_2(\text{OH})_7$  units. Due to the attractive force of a weak hydrogen bond, the bridging O atom of the ditetrahedral unit is displaced from the threefold axis. A characteristic feature of the structure is isolated tetrahedra, which connect only with one apex to the octahedral sheet. The acentric structure (*P3*) reveals ordering between  $\text{SiO}(\text{OH})_3$  and  $\text{SO}_4$  tetrahedra yielding characteristic distortions of the neighboring  $(\text{Zn,Cu})\text{O}_6$  octahedra caused by bond-valence requirements of the shared O atoms. The  $[\text{SiO}(\text{OH})_3]^{1-}$  tetrahedron is only 70(1)% occupied and is partly substituted by tetrahedral  $[\text{O}_4\text{H}_7]^{1-}$  clusters. The existence of  $\text{H}_2\text{O}$  molecules in the  $[\text{O}_4\text{H}_7]^{1-}$  cluster is supported by FTIR spectroscopy.