

Comparison of the crystal chemistry of selected $\text{MSi}_6\text{O}_{15}$ -based silicates

S.M. HAILE¹ AND B.J. WUENSCH²

¹Materials Science Department, California Institute of Technology, Pasadena, California 91125, U.S.A.

²Department of Materials Science and Engineering, Massachusetts Institute of Technology,
Cambridge, Massachusetts 02139, U.S.A.

ABSTRACT

The structures of four $\text{A}_3\text{MSi}_6\text{O}_{15} \cdot (n\text{H}_2\text{O})$ silicates ($\text{A} = \text{Na}$ or K , $\text{M} = \text{Nd}$ or Y) recently determined by the authors are compared with one another and with the structures of related silicates. The 2:5 Si:O ratio in these compounds (silicate tetrahedra linked by sharing three O atoms per tetrahedron plus one unshared O atom) permits layer, double-chain, or double-ring configurations. In $\alpha\text{-K}_3\text{NdSi}_6\text{O}_{15} \cdot 2\text{H}_2\text{O}$, $\beta\text{-K}_3\text{NdSi}_6\text{O}_{15}$, and $\text{Na}_3\text{NdSi}_6\text{O}_{15} \cdot 2\text{H}_2\text{O}$, the linkage is found to result in a corrugated layered structure, in $\beta\text{-Na}_3\text{YSi}_6\text{O}_{15}$ a double-chain structure, and in $\alpha\text{-Na}_3\text{YSi}_6\text{O}_{15}$ a unique double-ring structure. Although the factors that govern the stabilities of ring vs. chain vs. layered structures remain to be completely elucidated, it is apparent in the layered structures that larger M cations (such as Nd and Ce) produce a greater degree of corrugation than do Zr and Ti cations. The more open structures of neodymium and cerium silicates contain large channels that may serve as pathways for fast alkali ion transport.