## Single-crystal compression and crystal structure of clinopyroxene up to 10 GPa

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## Abstract

The hydrostatic compression of synthetic single crystals of diopside, CaMgSi<sub>2</sub>O<sub>6</sub>, and hedenbergite, CaFeSi<sub>2</sub>O<sub>6</sub>, was studied at 33 pressures up to 10 GPa by X-ray diffraction. In addition, intensity data for hedenbergite were collected at 12 pressures up to 10 GPa. For determination of the elasticity two crystals were loaded together in a diamond cell. The axial compressibilities  $\beta_a$ ,  $\beta_b$ , and  $\beta_c$  of diopside and hedenbergite are 2.36(4), 3.17(4), and 2.50(4) × 10<sup>-3</sup> GPa<sup>-1</sup>, and 1.93(5), 3.38(6), and 2.42(8) × 10<sup>-3</sup> GPa<sup>-1</sup>, respectively. The bulk moduli ( $K_{T_0}$ ) and their pressure derivatives ( $K'_{T_0}$ ) were determined simultaneously from a weighted linear fit of a third order Birch-Murnaghan equation of state to the volume data at elevated pressures.  $K_{T_0}$  and  $K'_{T_0}$  are 104.1(9) GPa and 6.2(3) for diopside and 117(1) GPa and 4.3(4) for hedenbergite, respectively.

The unit-cell parameters decrease continuously with pressure. The larger polyhedra show more compression than the smaller ones. Between 0.1 MPa and 10 GPa the polyhedral volumes of CaO<sub>8</sub>, FeO<sub>6</sub>, and SiO<sub>4</sub> decrease by 8.4, 6.6, and 2.9%, respectively. The longest bonds of CaO<sub>8</sub> and FeO<sub>6</sub> show most compression. Significant compression in the two shortest Si-O1 and Si-O2 bond lengths of the SiO<sub>4</sub> tetrahedra was observed at relatively low pressures, resulting in a tetrahedral volume compression of 1.6% between 0.1 GPa and 4 GPa and 1.3% between 4 and 10 GPa. The compression of the unit cell can be described by the volume compression of the individual CaO<sub>8</sub> and FeO<sub>6</sub> polyhedra, with the SiO<sub>4</sub> tetrahedron playing a minor role. Diopside is more compressible than hedenbergite as shown by their axial and volume compressibilities because the FeO<sub>6</sub> octahedron is significantly more rigid than MgO<sub>6</sub> at high pressures. This observation implies that octahedrally coordinated Fe<sup>2+</sup> behaves differently from Mg at high pressures, in contrast to their behavior at ambient conditions.