## The systematics of Cr<sup>3+</sup> and Cr<sup>2+</sup> partitioning between olivine and liquid in the presence of spinel

## BEN HANSON\* AND JOHN H. JONES<sup>†</sup>

Planetary Science Branch, SN2, Lyndon B. Johnson Space Center, National Aeronautics and Space Administration, Houston, Texas 77058, U.S.A.

## Abstract

The partitioning behavior of Cr into olivine in basaltic systems has been parameterized and can now be modeled over a wide range of redox conditions and liquid compositions. The Cr<sup>2+</sup>/Cr<sup>3+</sup> in spinel-saturated experimental systems can be estimated based on a simple model of Cr solubility in basalt. Fe<sup>3+</sup> appears to suppress the presence of Cr<sup>2+</sup> in basaltic systems. We predict that, in Fe-free systems, all Cr is trivalent at log  $f_{O_2} = -3$  (i.e., QFM+3 to QFM+4), whereas all Cr is trivalent at approximately Ni-NiO(QFM+1) in Febearing systems. Cr<sup>2+</sup> predominates under redox conditions <IW-1 in both Fe-bearing and Fe-free systems.

 $D_{\text{Cr}^{2+}}$  and  $D_{\text{Cr}^{3+}}$  (olivine/liquid) have been determined in various liquid compositions and temperatures.  $D_{\text{Cr}^{3+}}$  (i.e.,  $f_{o_2} \ge QFM$ , appropriate for most terrestrial or martian basalts) strongly covaries with the ratio of non-bridging oxygens to tetrahedrally coordinated cations (NBO/T) (Mysen 1983) and can be estimated using the equation

$$D_{\rm Cr^{3+}}^{\rm (ol/liq)} = -0.39 \cdot \frac{\rm NBO}{\rm T} + 1.29.$$

This relationship appears to be valid over the entire pressure range of olivine stability, from 1 atm to 15 GPa.

 $D_{c,2^+}$  (i.e.,  $\leq$  IW-1, appropriate for lunar and some asteroidal basalts) is sensitive to liquid composition and temperature and can be estimated using either

$$D_{\rm Cr^{2+}}^{\rm (ol/liq)} = 0.24 \cdot D_{\rm Mr}^{\rm (ol/liq)} - 0.07$$

or

$$D_{\rm Cr^{2+}}^{\rm (ol/liq)} = 0.66 \cdot \left[ \frac{10,000}{\rm T(K)} \right] - 4.48.$$

The 1/T equation is probably only valid at 1 atm pressure, but the  $D_{Mg}$  equation may be useful at higher pressures as well. The Cr content of spinel-saturated liquids is a function of temperature, composition, and  $f_{O_2}$ . However, the Cr content of spinel-saturated liquids is buffered by spinel and is insensitive to the bulk Cr content of the system (e.g., Roeder and Reynolds 1991). Therefore, the Cr content of a crystallizing, spinel-saturated basalt cannot be modeled using Raleigh fractionation models.