

Crystal-chemistry of sulfates from the Apuan Alps (Tuscany, Italy). VI. Tl-bearing alum-(K) and voltaite from the Fornovolasco mining complex

CRISTIAN BIAGIONI^{1,*}, DANIELA MAURO¹, MARCO PASERO¹, ELENA BONACCORSI¹,
GIOVANNI ORAZIO LEPORE^{2,†}, FEDERICA ZACCARINI³, AND HENRIK SKOGBY⁴

¹Dipartimento di Scienze della Terra, Università di Pisa, Via S. Maria 53, I-56126 Pisa, Italy

²CNR-IOM-OGG c/o ESRF, 71 Avenue des Martyrs CS 40220 F-38043 Grenoble Cedex 9, Grenoble, France

³Department of Applied Geological Sciences and Geophysics, University of Leoben, Peter Tunner Str. 5, A-8700 Leoben, Austria

⁴Department of Geosciences, Swedish Museum of Natural History, Box 50007, SE-10405 Stockholm, Sweden

ABSTRACT

Thallium-bearing samples of alum-(K) and voltaite from the Fornovolasco mining complex (Apuan Alps, Tuscany, Italy) have been characterized through X-ray diffraction, chemical analyses, micro-Raman, infrared (FTIR), Mössbauer, and X-ray absorption spectroscopy (XAS). Alum-(K) occurs as anhedral colorless grains or rarely as octahedral crystals, up to 5 mm. Electron-microprobe analysis points to the chemical formula $(\text{K}_{0.74}\text{Tl}_{0.10})_{\Sigma 0.84}(\text{Al}_{0.84}\text{Fe}_{0.14})_{\Sigma 0.98}\text{S}_{2.05}\text{O}_8 \cdot 12\text{H}_2\text{O}$. The occurrence of minor NH_4^+ was detected through FTIR spectroscopy. Its unit-cell parameter is $a = 12.2030(2) \text{ \AA}$, $V = 1817.19(9) \text{ \AA}^3$, space group $P\bar{a}3$. Its crystal structure has been refined down to $R_1 = 0.0351$ for 648 reflections with $F_o > 4\sigma(F_o)$ and 61 refined parameters. The crystal structure refinement agrees with the partial substitution of K by 12 mol% Tl. This substitution is confirmed by XAS data, showing the presence of Tl^+ having a first coordination shell mainly formed by 6 O atoms at $2.84(2) \text{ \AA}$. Voltaite occurs as dark green cubic crystals, up to 1 mm in size. Voltaite is chemically zoned, with distinct domains having chemical formula $(\text{K}_{1.94}\text{Tl}_{0.28})_{\Sigma 2.22}(\text{Fe}_{3.57}^{3+}\text{Mg}_{0.94}\text{Mn}_{0.55})_{\Sigma 5.06}\text{Fe}_{3.06}^{3+}\text{Al}_{0.98}\text{S}_{11.92}\text{O}_{48} \cdot 18\text{H}_2\text{O}$ and $(\text{K}_{2.04}\text{Tl}_{0.32})_{\Sigma 2.36}(\text{Fe}_{3.83}^{2+}\text{Mg}_{0.91}\text{Mn}_{0.29})_{\Sigma 5.03}\text{Fe}_{3.05}^{3+}\text{Al}_{0.97}\text{S}_{11.92}\text{O}_{48} \cdot 18\text{H}_2\text{O}$, respectively. Infrared spectroscopy confirmed the occurrence of minor NH_4^+ also in voltaite. Its unit-cell parameter is $a = 27.2635 \text{ \AA}$, $V = 20265(4) \text{ \AA}^3$, space group $Fd\bar{3}c$. The crystal structure was refined down to $R_1 = 0.0434$ for 817 reflections with $F_o > 4\sigma(F_o)$ and 87 refined parameters. The partial replacement of K by Tl is confirmed by the structural refinement. XAS spectroscopy showed that Tl^+ is bonded to six O atoms, at $2.89(2) \text{ \AA}$. The multi-technique characterization of thallium-bearing alum-(K) and voltaite improves our understanding of the role of K-bearing sulfates in immobilizing Tl in acid mine drainage systems, temporarily avoiding its dispersion in the environment.

Keywords: Alum-(K), voltaite, thallium, XAS, crystal structure, Fornovolasco, Apuan Alps, Tuscany, Italy