

## Experimental observations of TiO<sub>2</sub> activity in rutile-undersaturated melts

MICHAEL R. ACKERSON<sup>1,2,\*</sup> AND BJØRN O. MYSEN<sup>2</sup>

<sup>1</sup>National Museum of Natural History, Smithsonian Institution, 10th and Constitution Avenue NW, Washington, D.C. 20560, U.S.A.

<sup>2</sup>Geophysical Laboratory, Carnegie Institution for Science, 5251 Broad Branch Road NW, Washington, D.C. 20015-1305, U.S.A.

### ABSTRACT

An estimate of TiO<sub>2</sub> activity ( $a_{\text{TiO}_2}^{\text{melt-sat}}$ ) is necessary for the application of trace-element thermobarometry of magmatic systems where melts are typically undersaturated with respect to rutile/anatase. Experiments were performed in the system SiO<sub>2</sub>-Na<sub>2</sub>O-TiO<sub>2</sub> to develop two independent methods of estimating  $a_{\text{TiO}_2}^{\text{melt-sat}}$ —one based on the commonly applied rutile-saturation technique and another utilizing a novel Ti-in-tridymite thermometer. It is demonstrated that the rutile-saturation model can lead to an overestimate of  $a_{\text{TiO}_2}^{\text{melt-sat}}$  relative to TiO<sub>2</sub> activity calculated using the solubility of Ti in tridymite (SiO<sub>2</sub>) coexisting with rutile. Overestimation via the rutile-saturation technique is due to variations in the solubility mechanisms of Ti in the melt phase as a function of Ti content. In natural systems, overestimates of  $a_{\text{TiO}_2}^{\text{melt-sat}}$  will lead to an underestimation of crystallization temperatures by Ti-based trace-element thermobarometers. Although this study is not directly applicable to natural systems, it lays the groundwork for future research on natural composition magmas to constrain TiO<sub>2</sub> activity in melts.

**Keywords:** Thermobarometry, experimental petrology, Raman, Ti activity