

# Magnetic Contributions to Corundum-Eskolaite and Corundum-Hematite Phase Equilibria: a DFT Cluster Expansion Study

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## Abstract

Magnetic contributions have the potential to significantly influence predicted phase stability within alloy and mineral mixing phase diagrams, yet have been historically challenging to incorporate due to a significant increase to phase space sampling. In this work, we employ a computational protocol that includes spin orientation as an additional configurational component within multi-component cluster expansions between magnetic and non-magnetic metal oxide alloys (calculated using density functional theory (DFT) and the generalized gradient approximation). This approach was used to determine the effect of magnetic contributions to corundum-eskolaite and corundum-hematite phase equilibria from first principles.

Two-component cluster expansions of the magnetic components of eskolaite and hematite were first performed showing the ability of this method to properly calculate their respective magnetic properties. Two-component cluster expansions were then performed for non-magnetic Al(III) and ferromagnetic Cr(III) and Fe(III), and phase diagrams were calculated for later comparison. Finally, a non-magnetic Al(III) and “up” and “down” magnetic configuration for anti-ferromagnetic Cr(III) and Fe(III) were performed. Magnetic contributions to the calculated phase diagram for the corundum-eskolaite system was shown to be inconsequential, but is absolutely vital for accurate determination of the corundum-hematite solvus.

## Introduction

24

25 Corundum-eskolaite,  $\alpha$ -(Al,Cr)<sub>2</sub>O<sub>3</sub> and corundum-hematite,  $\alpha$ -(Al,Fe)<sub>2</sub>O<sub>3</sub> phase equilibria have been exten-  
26 sively researched in experimental studies(Jacob (1978); Sitte (1985); Chatterjee et al. (1982); Atlas & Sumida  
27 (1958); Turnock & Eugster (1962); Feenstra et al. (2005)), but little work has been done to describe these  
28 solid-solutions from first principle calculations.(Pinney et al. (2009); Chatterjee et al. (2016); Eremin et al.  
29 (2008)) This is perhaps due to additional complexity introduced by magnetic degrees of freedom present in  
30 both the Cr(III) and Fe(III) cations and the difficulty required to accurately calculate these contributions. More  
31 accurate modeling of these magnetic properties and their influence on phase equilibria would allow for better  
32 understanding of the underlying phenomena governing these systems and make predictive finite temperature  
33 structures more accessible for further study.

34 Corundum, hematite, and eskolaite are isostructural R-3c space group minerals that are important com-  
35 pounds in the production of many industrial materials. Additionally, hematite and eskolaite both exhibit unique  
36 magnetic properties making them of high interest for a multitude of spintronics applications.(Pattanayak et al.  
37 (2021); Khan et al. (2015); Borisov et al. (2005); He et al. (2010)) Alloy composition is often a guiding factor  
38 in material design of these systems, and accurate simulations of alloy compositions are highly desirable.

39 Prior computational work for both systems has been confined to ab initio energy calculations of dilute limit  
40 substitutions(Chatterjee et al. (2016)) and pair configurations(Pinney et al. (2009)) within moderately sized  
41 supercells or from classical molecular dynamics free energy calculations yielding associated phase equilibria  
42 diagrams.(Eremin et al. (2008)) While the MD simulations provide complete solvus line calculations, they rely  
43 on interaction potentials using fitted parameters to better model thermodynamics. Presently, to the authors'  
44 knowledge, there are currently no complete first-principle solvus line calculations for these systems.

45 Accurate determination of phase equilibria in alloys is typically performed through Cluster Expansion  
46 (CE) calculations(Sanchez et al. (1984)), where ensemble average free energies can be determined rapidly and  
47 used to create phase diagrams. This method normally examines atomic configurations only, e.g. A B site con-  
48 figurations in binary alloys, but can be extended to capture magnetic effects as well. This is achieved through  
49 a multi-component cluster expansion,(Van De Walle (2009)) where "up" and "down" magnetic moments at  
50 each spin polarized cation site is included as an additional configurational component. This approach may be  
51 essential for the eskolaite and hematite systems since they are both anti-ferromagnetic at low temperatures and  
52 capturing these low energy states are essential for determining mixing energies.

53 This approach, however, can result in missing intermediate energy microstates, lowering the magnetic en-  
54 tropy. As a result, while this approach improves phase diagram calculations through the inclusion favorable  
55 anti-ferromagnetic pair interactions, it overestimates the stability of these phases resulting in greatly increased

56 magnetic disordering temperatures. These missing states can be included through methods in which magnetic  
57 moments are treated with constant magnitude and variable direction during Monte Carlo finite temperature cal-  
58 culations allowing for estimation of Heisenberg-like spin-spin interaction energies.(Lavrentiev et al. (2010);  
59 Garrity (2019)), but this is not readily available in current cluster expansion computational packages. Addi-  
60 tionally, due to this, finer features of the phase diagram due to interaction between magnetic and chemical  
61 ordering(Burton (1991)) would not be accurately modeled in these calculations.

62 It is important to note that CE requires a series of ab initio calculations of the alloy configurations in order  
63 to obtain interaction energies. Thus they are somewhat limited in that more computationally expensive hybrid  
64 functionals are impractical to implement. For transition metal oxides and magnetic systems, use of other  
65 functionals, e.g. GGA's, often require careful implementation of Hubbard U parameters to correctly describe  
66 magnetic interactions.(Shi et al. (2009); Rollmann et al. (2004)) However tuning this parameter does provide  
67 a means in which to lower magnetic pairing energies, thereby artificially matching the expected magnetic  
68 disordering properties. This will also affect atomic interactions as well though, so the effect of tuning this  
69 parameter on the resulting phase diagrams typically needs to be examined.

70 Toward this end, we implement this multi-component cluster expansion by examining the mixing prop-  
71 erties of corundum-eskolaite and corundum-hematite. As we demonstrate, the relative impact of including  
72 the magnetic contributions to the corundum-eskolaite and corundum-hematite alloy to the calculated phase  
73 diagrams may be significant. The computational protocol is outlined in the Theory and Methodology sec-  
74 tion and is first verified by determination of the known lattice and magnetic properties of the pure systems  
75 of hematite and eskolaite. To formulate a basis for comparison, corundum-hematite and corundum-eskolaite  
76 phase diagrams were first calculated assuming a ferromagnetic orientations only. These systems were then re-  
77 calculated now including the additional magnetic degrees of freedom and the resulting shifts in the calculated  
78 phase diagrams are presented. While a more complete picture of the calculated phase diagram would also  
79 include vibrational contributions, inclusion of these effects and their possible interactions with the magnetic  
80 contributions is beyond the scope of this work.

## 81 **Theory and Methodology**

82 The corundum-eskolaite and corundum-hematite systems represent an alloy mixture of a non-magnetic (NM)  
83 Al and magnetic (M) Cr or Fe metal ion. Treatment of the magnetic contributions as a configurational state  
84 of either an “up” or “down” collinear magnetic moment results in a ternary phase diagram comprised of NM,  
85  $M\uparrow$ , and  $M\downarrow$  components as represented in Fig 1.

86 Solid solution phase diagrams can be calculated given that the energy of any random alloy configuration is  
87 known. This is achieved through the thermodynamic integration of the semi-grand-canonical ensemble of  
88 alloy configurations with respect to temperature and chemical potential.(van de Walle & Asta (2002)) In order  
89 to simulate an ensemble of configurations requires rapid estimation of any given alloy configuration energy.  
90 This can be achieved through multi-component cluster expansions in the form of Eq. 1, where E is the energy  
91 of some configuration  $\sigma$ ,  $\alpha$  is a cluster,  $m_\alpha$  are multiplicities of clusters with equivalent symmetry, and  $J_\alpha$   
92 is the Effective Cluster Interaction (ECI) energy. The terms inside the angle brackets represent the cluster  
93 correlation functions, which are averaged over all clusters symmetry equivalent to  $\alpha$ . Details of this equation  
94 are given more completely in other work.(Van De Walle (2009))

$$E(\sigma) = \sum_{\alpha} m_{\alpha} J_{\alpha} \langle \prod_i \gamma_{\alpha_i, M_i}(\sigma_i) \rangle_{\alpha} \quad (1)$$

95 The primary task in a cluster expansion calculation is to truncate the complete cluster expansion basis to some  
96 finite sum that is predictive of new configurations. This is achieved by fitting the Effective Cluster Interac-  
97 tion term,  $J_\alpha$ , to a series of energies of different configurations calculated by ab initio methods. New alloy  
98 configurations can now be rapidly calculated and used for statistical sampling and generation of alloy phase  
99 diagrams. In this study, cluster expansion calculations are separated into three configurational categories;  
100 purely magnetic, atomic, and atomic plus magnetic. For each category, cluster expansions were performed. In  
101 **Sec. Properties of Eskolaite and Hematite**, we first examine the individual magnetic properties of eskolaite  
102 and hematite using CE. The magnetism of eskolaite and hematite are well known, and the initial goal is to  
103 ensure that cluster expansion methods will accurately capture their magnetic properties validating further use  
104 within multi-component systems. These were performed using  $M\uparrow$  and  $M\downarrow$  lattice site configurations. The  
105 purely magnetic configurations were also calculated over a series of DFT+U values. Adjusting this parameter  
106 will affect the strength of magnetic interactions as well as the cohesive energy of the crystal. Both of these ef-  
107 fects will ultimately influence the final calculation of alloy phase diagrams, and would ideally be set to values  
108 satisfying accuracy in both instances.

109 In **Sec. Configurational Phase Boundary of Corundum-Eskolaite and Corundum-Hematite**, cluster  
110 expansions are performed using NM and  $M\uparrow$  lattice site configurations only, and binary phase diagrams were  
111 calculated. Both eskolaite and hematite are normally anti-ferromagnetic, but this methodology restricts the  
112 magnetic species to higher energy ferromagnetic states only. This provides a simplified first approximation  
113 with which to first compare to experimental data, and to examine how introducing magnetic degrees of freedom

114 affects the resulting phase diagram calculations.

115 In Sec. **Configurational and Magnetic Contributions to Phase Boundary of Corundum-Eskolaite**  
116 **and Corundum-Hematite**, the configurational plus magnetic system is determined by the construction of a  
117 4-component cluster expansion. For example, the corundum-eskolaite system consists of a series of labeled  
118 chemical components Cr1, Cr2, Al3, Al4. Here the Cr1 is a  $M\uparrow$  component and the Cr2 is a  $M\downarrow$  component.  
119 The aluminum components lack a magnetic moment, but are separated into two identical cluster expansion  
120 components in order to yield the correct compositional high temperature statistics. For example, at the high  
121 temperature limit for equiatomic composition, there should be an equal number of Cr and Al atoms. Without  
122 including a second, but identical, aluminum component, the high temperature limit would give an equal prob-  
123 ability of occupying a lattice site by either a Cr1, Cr2, or a single Al, leading to over representation of Cr. This  
124 ultimately leads to a quaternary, instead of the ternary phase diagram picture in Fig. 1, but can be projected  
125 into a final binary phase diagram between the total Cr and Al components.

126 This approach allows for the correction of the ground state energy of an anti-ferromagnetic system, which  
127 is improperly modeled as ferromagnetic in the binary configurational system. It also allows for the inclusion  
128 of the magnetic disordering transition in the free energy landscape, which may shift the purely configurational  
129 phase boundary in unpredictable ways. The disordering temperature, can be represented by Eq. 2, and demon-  
130 strates that it can be shifted up or down depending on the strength of the change in magnetic energy or entropy  
131 upon going from an ordered phase  $a$ , to a disordered phase  $b$ .

$$T_{config+mag}^{a \rightarrow b} = \frac{\Delta U_{config}^{a \rightarrow b} + \Delta U_{mag}^{a \rightarrow b}}{\Delta S_{config}^{a \rightarrow b} + \Delta S_{mag}^{a \rightarrow b}} \quad (2)$$

132 The major drawback of this method is the inclusion of only the two magnetic states, which artificially  
133 heightens the magnetic disordering temperature. However, the magnitude of the difference between configu-  
134 rational plus magnetic system vs. just the binary configurational system would at least be informative of the  
135 possible importance of this effect.

## 136 **Computational Details**

137 The crystal structure used is isomorphic between corundum, eskolaite, and hematite and is described using the  
138 rhombohedral primitive unit cell of space group  $R\bar{3}c$ . Electronic structure calculations were performed using  
139 the Vienna ab-initio simulation package (VASP)(Kresse & Furthmüller (1996)) within the framework of plane-  
140 wave density functional theory in periodic boundary conditions. Projector Augmented Wave (PAW)(Kresse  
141 & Joubert (1999)) methods were employed to represent the core electron energies using VASP POTCAR v5.4

142 PAWs. Due to the large number of calculations required for determination of ECIs, the exchange correlation  
143 is represented by the computationally inexpensive generalized gradient approximation (GGA+U) using the  
144 Perdew, Burke, and Ernzerhof (PBE)(Perdew et al. (1996)) variant.

145 DFT+U was added using the VASP implementation of the Dudarev method(Dudarev et al. (1998)). The  
146 Hubbard J value was kept constant at 0 eV, and the Hubbard U value was adjusted relative to J. A value of U  
147 equal to 0 eV represents the absence of the DFT+U correction. U-J Hubbard correction values of J = 0 and  
148 of U = 0 through 5 eV were used to compare lattice properties of the eskolaite and hematite systems. From  
149 these, it was found that a U = 0 eV value for eskolaite and U = 2 eV value for hematite best described overall  
150 properties which are detailed further in the results section.

151 The plane wave basis energy cutoff for all structures was converged to energy differences of less than 1  
152 meV per 50 eV increase. For all systems, the maximum required energy cutoff, 900 eV, was used across all  
153 systems to ensure accurate energy comparisons. To avoid Pulay stress error,(Francis & Payne (1990)) the  
154 energy cutoff was increased by 30% for all geometry optimizations. A static calculation at the original cutoff  
155 was then performed to obtain the total energy. Brillouin zone sampling was performed with gamma-centered  
156 k-points using the Monkhorst-Pack scheme(Pack & Monkhorst (1977)) and again converged to energy differ-  
157 ences within 1 meV for all unit cell crystal structures with a k-point mesh of  $4 \times 4 \times 4$ . All supercell structures  
158 were sampled inversely proportional to their size in each cell dimension. Electronic minimization energy  
159 cutoffs were set to  $1 \times 10^{-5}$  eV. Atom positions, cell shape, and cell size were all allowed to change during  
160 geometry minimization. Geometry was minimized with respect to cell stress and was set to a cutoff value of  
161  $-1 \times 10^{-4}$  eV in VASP.

162 Configurations used to calculate cluster expansions were generated until cross-validation scores of less  
163 than 0.25 eV or a minimum of 25 configurations were calculated.

164 Magnetic disordering temperatures were determined from identifying peaks in energy variance vs. temper-  
165 ature plots using the Alloy Theoretical Automated Toolkit (ATAT) Monte Carlo implementation 'emc2'.(van de  
166 Walle & Asta (2002)) All simulations were performed with an enclosed radius of 40 Å, resulting in system  
167 sizes of  $\approx 37000$  metal ions. Systems were initiated in random configurations, and stepped down in tempera-  
168 ture steps of 1 K.

169 Phase boundary tracing was performed using the ATAT Monte Carlo implementation 'phb'.(van de Walle  
170 & Asta (2002)) All systems were again performed with a enclosed radius of 40 Å. Phase boundaries were  
171 traced between stable phases identified from cluster expansions at temperature increments of 10K.

172 The quaternary solid-solution phase boundary cannot be determined through the phase boundary tracer, as  
173 this method is implemented for binary systems only. Instead, the system was scanned over regions of chemical

174 potentials and temperatures, beginning from the high temperature disordered state for a given potential. The  
175 ATAT multi-component Monte Carlo implementation ‘memc2’(van de Walle & Asta (2002)) was used to  
176 perform the determination of the composition as a function of decreasing temperature at a constant chemical  
177 potential. Again, an enclosed radius of 40 Å was used, and temperature steps of 10K. Phase boundaries were  
178 identified through “memc2” phase boundary detection algorithms.

## 179 **Results and Discussion**

### 180 **Properties of Eskolaite and Hematite**

181 Both eskolaite and hematite are experimentally known to be high spin and anti-ferromagnetic at low temper-  
182 atures.(Foner (1963); Searle & Dean (1970)) Hematite is known to have a Morin transition(Morin (1950)) at  
183 250 K, where canting of the magnetic moments causes a net ferromagnetic moment. However, the magnetic  
184 moments overall retain their AF alternation within the 4 cation primitive cell, and remain orientated along the  
185 easy axis until disordering at hematite’s Nèel temperature at 953 K.

186 To confirm the correct ground state magneto-crystalline configurations can be accurately captured by clus-  
187 ter expansion methods, a series of cluster expansions were performed for both systems for a range of U values.  
188 The properties of the lowest energy magnetic configurations from these Cluster Expansions were then exam-  
189 ined and tabulated for the different values of U.

190 The eskolaite system magnetic structure is well behaved throughout all U values and maintains the initial  
191 magnetic configuration after geometry relaxation for all tested structures. The resulting cross-validation scores  
192 are all  $< 0.005$  eV, well below the recommended value of 0.25 eV,(van de Walle (2019)) ensuring good  
193 predicted energies of new configurations. The cross-validation scores and figures of the predicted energies vs.  
194 calculated energies can be found in the Supplementary Information for all systems. In contrast, the hematite  
195 magnetic structure is highly sensitive to the U parameter. As noted in prior computational work,(Rollmann  
196 et al. (2004)) the U parameter for hematite greatly affects the cell volume, causing changes in the balance  
197 between the Crystal Field Energy (CFE) and the pairing energy. With no correction, the lattice volume is  
198 highly reduced, raising the CFE, making the low spin configuration more favorable. For  $U = 0$  eV this system  
199 is highly unstable upon application of a Cluster Expansion, and often does not retain its magnetic structure  
200 upon geometry relaxation. While a CE fit was ultimately achieved for  $U = 0$  eV, the overall CV score was poor.  
201 For  $U = 1$  eV, not enough stable configurations could be made to produce a valid cluster expansion. Starting  
202 at U equal to 2 eV however, the high spin configuration is adopted, and magnetic configurations become  
203 much more stable. Cluster Expansions for systems with  $U \geq 2$  eV retain their magnetic configurations upon

204 geometry relaxations and result in low cross-validation scores,  $< 0.005$  eV, in line with those from the eskolaite  
205 systems.

206 For both the eskolaite and hematite systems, the major contributions to the total energy are given by  
207 identical sets of pair cluster ECIs as illustrated in Fig. 2. The multiplicity and distances of these pairs on the  
208 initial lattice geometry are given in Table 1. The ECI pair energies were plotted in the top figures of Fig. 3  
209 as function of  $U$  and indicate decreasing interaction strengths with increasing  $U$ . The strongest interactions  
210 are positive, indicating energetic favorability of anti-ferromagnetism for both systems. As the interaction  
211 strengths are decreasing with  $U$ , it may be predicted that this parameter will ultimately affect the magnetic  
212 interaction strengths of these metals within a mixture of aluminum. This effect can be seen by examining  
213 the shifts in the predicted magnetic disordering temperature of these systems across  $U$ . As  $U$  increases, the  
214 calculated magnetic disordering temperature decreases. This may provide a method to “tune”  $U$  to give the  
215 correct disordering temperature.

216 For eskolaite the two primary interactions are from ECI1 and ECI2 (Figure 2). These correspond to a direct  
217 exchange interaction between  $t_{2g}$  orbitals of the shorter metal-metal bonds across edge sharing octahedra. For  
218 hematite, the two primary interactions are instead ECI3 and ECI4. These correspond to a superexchange inter-  
219 action between  $e_g$  orbitals of the metal ions across the  $p$  orbitals of oxygens across corner sharing octahedra.

220 The bottom figures of Fig. 3 show the magnetic interaction parameters  $J$  corresponding to each ECI.  
221 Details of this conversion are found in the SI. Note that while for this case, ECIs were able to be converted  
222 to  $J$  parameters, this is only possible for 2-component cluster expansions with a maximum of 2-body clusters,  
223 and uniform spin values at each magnetic atom site. While useful for comparison to experimental data, and  
224 determination of good  $U$  parameters, better energetic predictions of magnetic configurations may rely on  
225 inclusion of  $n$ -body clusters. ECIs derived from the full multi-component,  $n$ -body cluster expansion would  
226 be more accurate in final construction of phase diagrams.

227 These parameters were compared to experimental values derived from neutron scattering data Samuelsen  
228 et al. (1970); Samuelsen & Shirane (1970) and those from other computational work Shi et al. (2009); Nabi  
229 et al. (2010) that fit these parameters to sets of calculated magnetic configuration energies. Data for these  
230 figures are listed in Table 2. As ECIs and experimental  $J$  values were ordered by increasing distance, the  
231 labeling is equivalent. The experimental data has been scaled by a factor of two to account for the fitted  
232 Heisenberg Hamiltonian used in these models not including a common leading  $1/2$  term. For eskolaite, the  $U$   
233 = 1 eV gave reasonable agreement with experiment, though it shows more significant contributions from the  
234  $J_3$  and  $J_4$ . For hematite, the  $U = 5$  eV matches very closely with experiment, though identifies  $J_1$  as being  
235 negative. The computational models are not directly comparable, but should give reasonably similar results.



236 For comparison, two prior works were cited using values generated from Hubbard parameters similar to those  
237 used in this study. The eskolaite prior computational work was fit to a cluster expansion of 12 configurations  
238 only, and the reported ECI values were converted to J values here showing good agreement. The hematite  
239 study was fit directly to the Heisenberg Hamiltonian, and its values are reported here without conversion. Note  
240 that in this study, J3 was reported as two degenerate sets of J values, J3 and J4, thereby shifting the higher J  
241 parameters up in index by one. They have been renumbered here to match the order from the experimental  
242 data, and the resulting comparison shows good agreement with these calculations.

243 For all values of U for eskolaite, and for  $U \geq 2$  for hematite, the same lowest energy AFM magnetic  
244 configurations were found, and are pictured in Fig. 4. These lowest energy magneto-crystalline structures  
245 were found to be of the same size as the primitive cell, with no symmetry breaking magnetic interactions. The  
246 magnetic ordering difference between each system is a result of the primarily direct exchange interactions of  
247 eskolaite and the primarily superexchange interactions of hematite resulting in + - + - and + + - - configurations  
248 respectively. These results are in line with prior computational work identifying these as the lowest energy  
249 primitive cell configurations.(Shi et al. (2009); Rollmann et al. (2004)) Additionally, based on the size of the  
250 energy gaps between the FM and AFM configurations, we predict that magnetic contributions will have more  
251 significant effect on the corundum:hematite phase diagram.

252 The crystal properties of these systems as well as the aluminum corundum system were tabulated for the  
253 lowest energy configurations for different values of U in Table 3. The most notable result is that for both  
254 systems, an increasing U parameter results in moving further from the experimental cohesive energy value.  
255 For eskolaite, this presents little issue, as all calculated parameters trend towards the experimental value with  
256 decreasing U and are close to these values with DFT+U correction. But this does present an issue for hematite,  
257 as the magnetic properties are only properly captured at  $U \geq 2$ .

## 258 **Configurational Phase Boundary of Corundum-Eskolaite and Corundum-Hematite**

259 In the simplest approximation of a metal-metal solid oxide solution, the magnetic metal is approximated with  
260 zero magnetic degrees of freedom, and is fixed to a single orientation. This requires a new set of Cluster  
261 Expansions to be performed for the  $xM_1_2(1-x)M_2_2O_3$  systems. For eskolaite, the two metal components  
262 are Al and Cr, and for hematite, Al and Fe. In these binary systems, the phase diagram can be determined  
263 through matching of the chemical potentials for each ground state using the Monte Carlo “phb” method from  
264 ATAT. The resulting phase diagram, shown in Fig. 5, for the corundum eskolaite system well approximates  
265 the experimental phase diagram, but the boundary becomes more reduced in temperature with increasing U.  
266 Notably, there is an initial large reduction in the phase boundary, with the shift seeming to converge with

267 increasing U. It should also be commented that the disconnected portions of the calculated phase boundary are  
268 due to the increasingly smaller temperature steps required to sample this region, and are not due to an absence  
269 of a phase boundary.

270 The phase diagram for the corundum hematite system is only slightly more complex than the eskolaite  
271 mixture with inclusion of an additional ground state identified from DFT calculations of configurations used  
272 to create the cluster expansion. It should first be noted that experimental phase diagrams have not reported  
273 the existence of this state, but also have never reported a closed solvus in the literature that the authors are  
274 aware of. The result of these calculations is the prediction of a new stable phase composed of equal amounts  
275 Fe(III) and Al(III), pictured in the phase diagram in Fig. 5, and having ferromagnetic ordering. This prediction  
276 is based off *ab-initio* calculations of ground state configurations only, and do not include zero point energy  
277 vibrational contributions.

278 The higher values of U follow a similar pattern to those in the Al:Cr system, where increasing U leads  
279 to decreased gaps in the mixing energy, and a lowered phase boundary, though the shift is less drastic. The  
280 intermediate stable  $\text{Al}_2\text{Fe}_2\text{O}_6$  phase is identified as a ground state for all values of U. Overall the calculated  
281 phase boundary is much further from the experimental values when only accounting for the configurational  
282 energies and is likely due to the missing magnetic contributions. Additionally, as increasing U decreases the  
283 cohesive energy of hematite, starting at  $U = 2$  eV may be leading to lower phase boundary temperatures.

## 284 **Configurational and Magnetic Contributions to Phase Boundary of Corundum-Eskolaite** 285 **and Corundum-Hematite**

286 To determine the magnetic contributions to the phase boundary, each system was modeled using a 4-component  
287 cluster expansion in order to include configurational and magnetic terms simultaneously. For the corun-  
288 dum:eskolaite system, this consisted of a series of labeled chemical components Cr1,Cr2,Al3,Al4. Here the  
289 Cr1 is a chromium(III) with an ‘up’ magnetic moment, and the Cr2 is a ‘down’ magnetic moment. The  
290 aluminum components lack a magnetic moment, but are separated into two identical Cluster Expansion com-  
291 ponents in order to yield the correct compositional high temperature statistics.

292 The phase diagrams generated from this are quaternary, but can be simplified by combining the aluminum  
293 components and the magnetic components to form a binary phase diagram. This will mean that the magneti-  
294 zation axis is “hidden”. While it is possible to map out the magnetic phase diagram in conjunction with the  
295 atomic configuration diagram, the focus of this work is on shifts to the latter upon the inclusion of magnetism.  
296 The following diagrams represent the anti-ferromagnetic and disordered region of the magnetization axis, i.e.  
297 equal parts up and down magnetic moments.

298 Within the corundum-eskolaite system, Fig. 6, it can be seen that there is very little change in the phase dia-  
299 gram in the aluminum rich region. On the chromium rich side, for  $U = 0$  eV, there is a noticeable shift upwards  
300 in the phase boundary, and slight shifts downwards at  $U > 0$ . This is likely due to the presences of the artifi-  
301 cially high magnetic disordering temperature occurring right at the region for the expected corundum:eskolaite  
302 phase transition. Overall however, there is very little difference upon inclusion of magnetic degrees of freedom  
303 into this system, and a ferromagnetic treatment of chromium ions gives good agreement with experiment. This  
304 might be predicted as the magnetic disordering temperature of eskolaite is 309K, intersecting the miscibility  
305 gap with corundum at a comparatively low temperature.

306 For the corundum:hematite system, the  $\text{Fe}_2\text{Al}_2\text{O}_6$  was again determined to be a ground state for all values  
307 of  $U$ . As it is ferromagnetic, it was omitted from the final binary phase diagram. The inclusion of the magnetic  
308 terms for this system resulted in a drastic shift to the phase diagram. For  $U = 2$  eV, the miscibility gap has  
309 greatly increased, and the phase boundaries are very well aligned with the experimental data. For increasing  
310  $U$  the phase boundary decreases, but even at it's lowest value at  $U = 5$  eV, it is still closer to experiment than  
311 the purely configurational calculated phase boundary. For all values of  $U$ , the calculated magnetic disordering  
312 temperature is well above the boundary for the calculated miscibility gap between Al and Fe. As the magnetic  
313 disordering temperature for hematite occurs at 905K, it should normally intersect the miscibility gap with  
314 corundum. The presented model therefore does not represent the physical system completely, but is clearly  
315 able to capture more accurate mixing properties.

316 Ultimately, the magnetic interactions for some systems are strong enough to require consideration when  
317 constructing phase diagrams, though it may not be clear from the outset how significant the changes to critical  
318 disordering temperatures may be.

## 319 Implications

320 Magnetic contributions may have a significant impact on material phase boundaries, but present an exceptional  
321 challenge to incorporate accurately into an already immense configurational space. By inclusion of an Ising-  
322 like model of magnetism into the configurational portion of a multi-component cluster expansion, it is at least  
323 possible to simultaneously capture this simplified magnetic model in conjunction with atomic configurations.  
324 Future work would benefit from inclusion of Heisenberg-like interactions as part of the Monte Carlo finite  
325 temperature calculations which allow for rotation of magnetic moments in addition to swapping of atomic  
326 sites.

327 While for the corundum-eskolaite system, where it may be expected that the weaker magnetic pair interac-

328 tions would result in little noticeable effect and negligible changes to phase diagram calculation were present,  
329 for corundum-hematite, inclusion of magnetic degrees of freedom is absolutely essential for reproducing ex-  
330 perimental results. These results show the necessity of incorporating magnetism into phase equilibria for  
331 systems in which strong magnetic pair interactions are present.

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## 339 Tables

Table 1: Effective cluster interaction (ECI) distances with respect to cluster expansion parent lattice dimen-  
sions.

ECI	multiplicity	distance (Å)
1	2	2.65
2	6	2.78
3	6	3.50
4	12	3.84
5	2	4.76

Table 2: Calculated magnetic exchange energies (J) as compared to experiment and prior computational work.

J (K)	1	2	3	4	5
<b>Cr<sub>2</sub>O<sub>3</sub></b>					
U = 0 eV	-202	-167	21.9	35.6	-12.8
U = 1 eV	-125	-103	31.8	40.0	2.73
U = 2 eV	-68.1	-66.8	30.5	40.0	-11.4
U = 3 eV	-53.0	-48.6	35.1	38.9	-0.885
Exp. <sup>a</sup>	-174.6	-75.6	1.4	0.4	-4.4
Comp. <sup>b</sup>	-147	-107	7.80	14.6	-23.0
<b>Fe<sub>2</sub>O<sub>3</sub></b>					
U = 2 eV	0.127	-22.2	-162	-94.7	-41.9
U = 3 eV	-13.2	-22.0	-127	-80.6	6.47
U = 4 eV	-4.27	-6.76	-88.6	-67.6	-4.38
U = 5 eV	-9.65	-3.02	-71.1	-57.8	-3.69
Exp. <sup>c</sup>	12	3.2	-59.4	-46.4	-2.
Comp. <sup>d</sup>	-15.9	-6.3	-72.1	-50.1	-9.

<sup>a</sup> Samuelsen et al. (1970)

<sup>b</sup> Shi et al. (2009) LDSA+U U-J = 2.42 eV

<sup>c</sup> Samuelsen & Shirane (1970)

<sup>d</sup> Nabi et al. (2010) GGA+U U-J = 5 eV

Table 3: Lattice properties of corundum, eskolaite, and hematite. Properties of magnetic structures were derived from their ground-state AFM magneto-crystalline configurations

	<i>a</i>	$\alpha$	<i>V</i>	mag. mom.	CohE	$\Delta E$ FM/AFM	Nèel/Curie
	(Å)	(deg)	(Å <sup>3</sup> )	( $\mu_B$ )	(kJ/mol/f.u.)	(meV)	(K)
<b>Al<sub>2</sub>O<sub>3</sub></b>							
Exp. <sup>a</sup>	5.12	55.28	84.5				
	5.18	55.30	87.6				
<b>Cr<sub>2</sub>O<sub>3</sub></b>							
Exp.	5.35 <sup>b</sup>	55.0 <sup>b</sup>	95.7 <sup>b</sup>	2.48 <sup>c</sup> , 2.76 <sup>d</sup>	2679 <sup>e</sup>		309 <sup>f</sup>
U = 0 eV	5.42	54.2	97.3	2.69	2582	106	1090
U = 1 eV	5.36	56.3	99.8	2.81	2505	56.1	790
U = 2 eV	5.39	56.2	100.8	2.88	2433	32.1	600
U = 3 eV	5.41	56.1	101.9	2.94	2365	15.0	410
<b>Fe<sub>2</sub>O<sub>3</sub></b>							
Exp.	5.43 <sup>a</sup>	55.3 <sup>a</sup>	100.8 <sup>a</sup>		2403 <sup>e</sup>		953 <sup>g</sup>
U = 0 eV	5.42	54.3	97.56	0.98	2427		
U = 2 eV	5.47	55.0	102.5	3.95	2359	397	3130
U = 3 eV	5.48	55.0	102.9	4.06	2312	323	2520
U = 4 eV	5.47	55.1	102.9	4.14		265	2230
U = 5 eV	5.47	55.1	102.7	4.43		219	1822

<sup>a</sup> Pauling & Hendricks (1925)

<sup>b</sup> Zachariasen (1928)

<sup>c</sup> Brown et al. (2002)

<sup>d</sup> Corliss et al. (1965)

<sup>e</sup> Glasser & Sheppard (2016)

<sup>f</sup> Brockhouse (1953)

<sup>g</sup> De Boer & Dekkers (1998)

## Figures

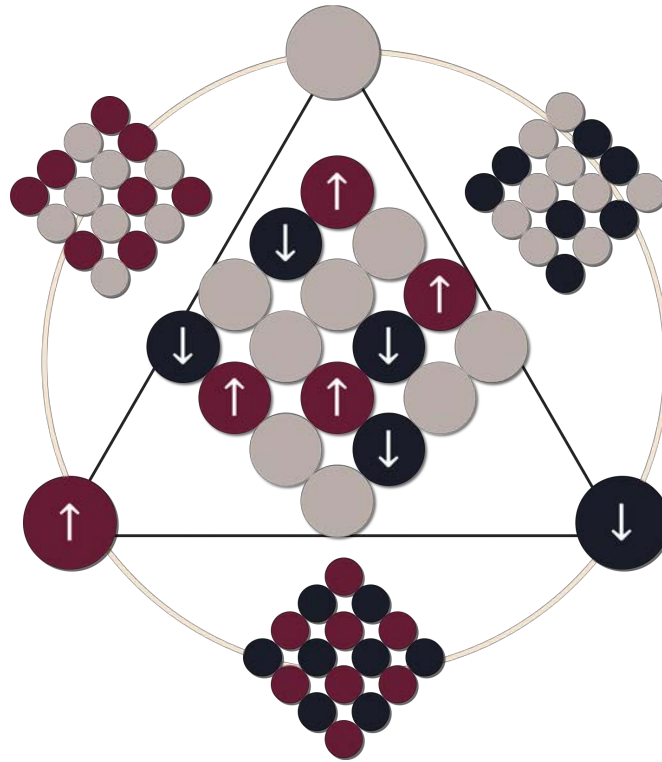


Figure 1: Ternary phase diagram of atomic and magnetic configurational components.

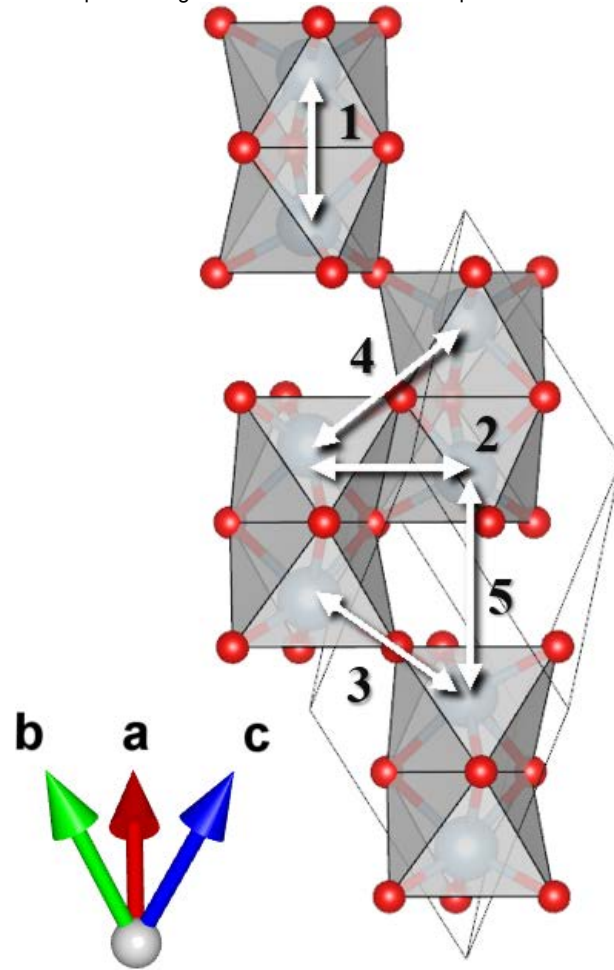


Figure 2: Effective cluster interactions of eskolaite and hematite.

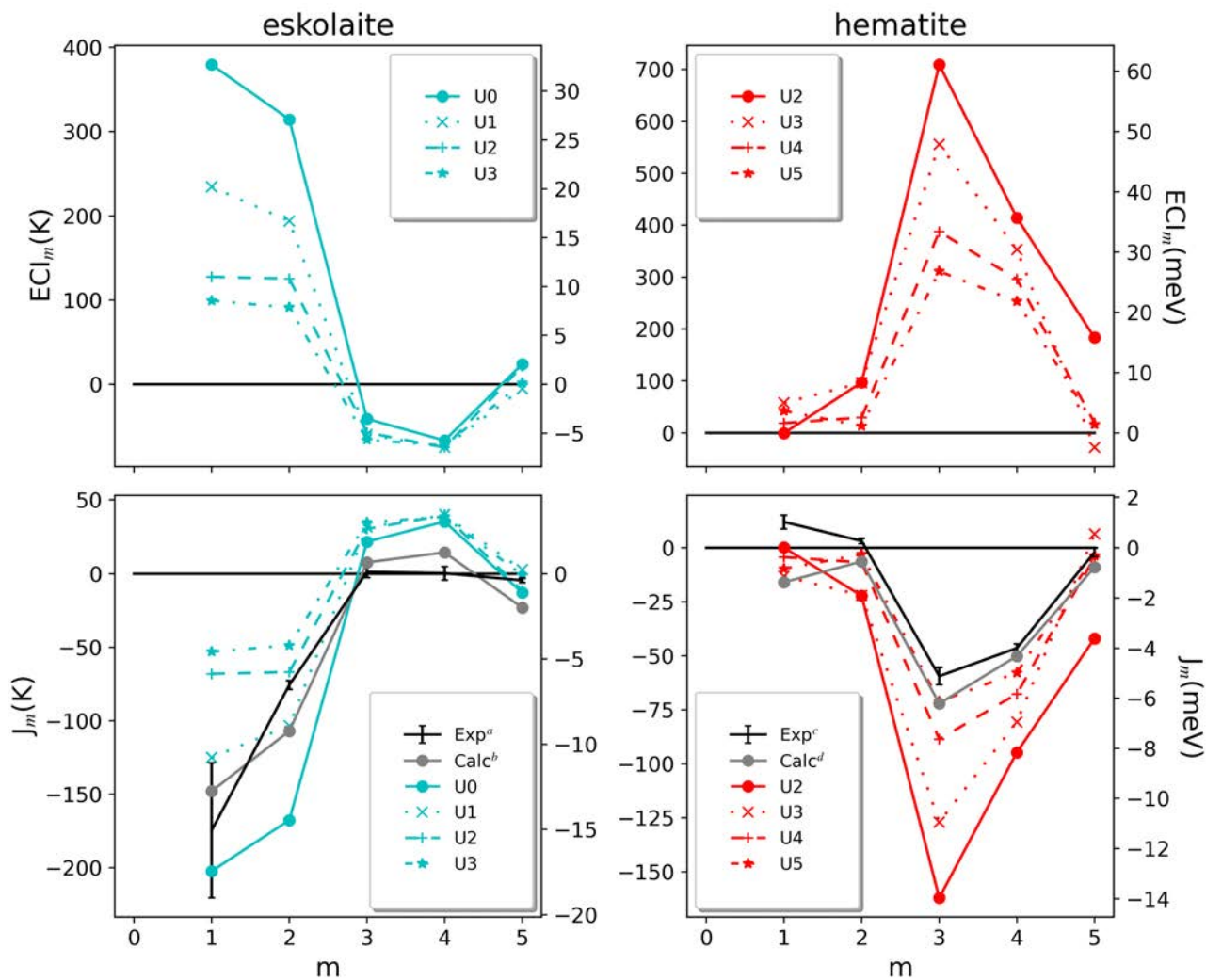


Figure 3: Top: Variation in the magnitude of effective cluster interaction (ECI) values of eskolaite and hematite calculated over a range of GGA+U values. The ECI values here correspond with those shown in Figure 2. Bottom: ECI's have been converted to magnetic exchange terms J for comparison to experimental estimates and values obtained from prior computational work.

<sup>a</sup> Samuelsen et al. (1970)

<sup>b</sup> Shi et al. (2009) LSDA+U U-J = 2.42 eV

<sup>c</sup> Samuelsen & Shirane (1970)

<sup>d</sup> Nabi et al. (2010) GGA+U U-J = 5 eV



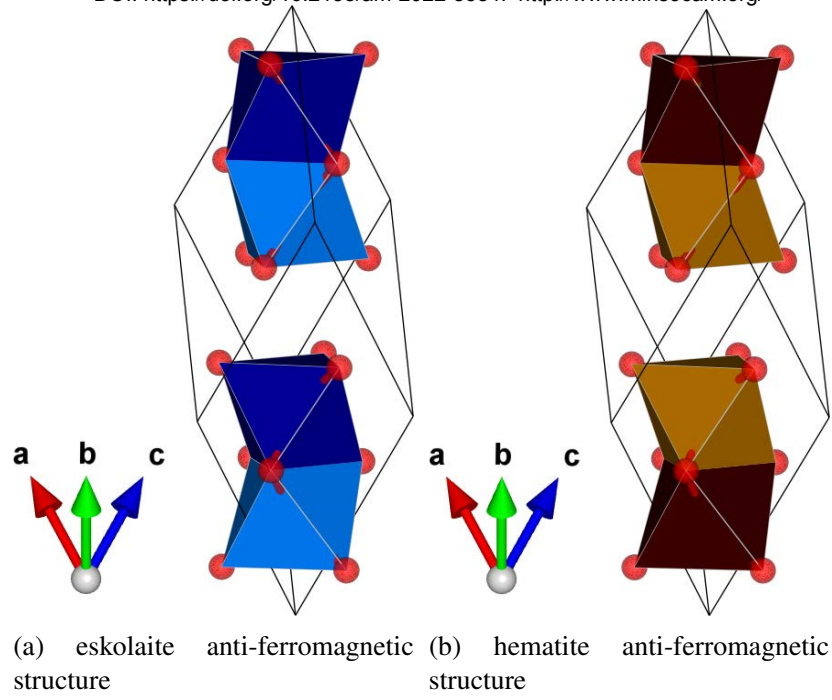


Figure 4: Magneto-crystalline structures computed for eskolaite and hematite. The lighter and darker color polyhedra represent the opposing directions of the magnetic moments.

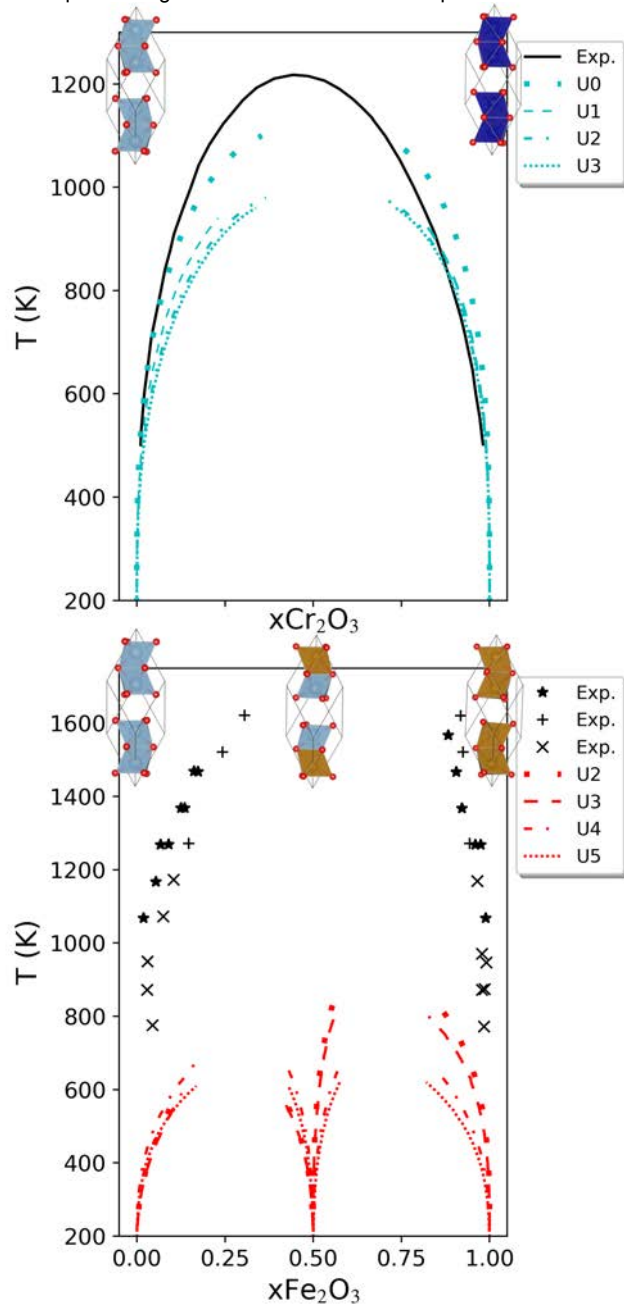


Figure 5: Calculated phase boundary of (Top) Al<sub>2</sub>O<sub>3</sub> and ferromagnetic Cr<sub>2</sub>O<sub>3</sub> and (Bottom) Al<sub>2</sub>O<sub>3</sub> and ferromagnetic Fe<sub>2</sub>O<sub>3</sub> across values of U vs. the experimental curve.(Chatterjee et al. (1982)) Ground state structures are pictured at the top of each plot.

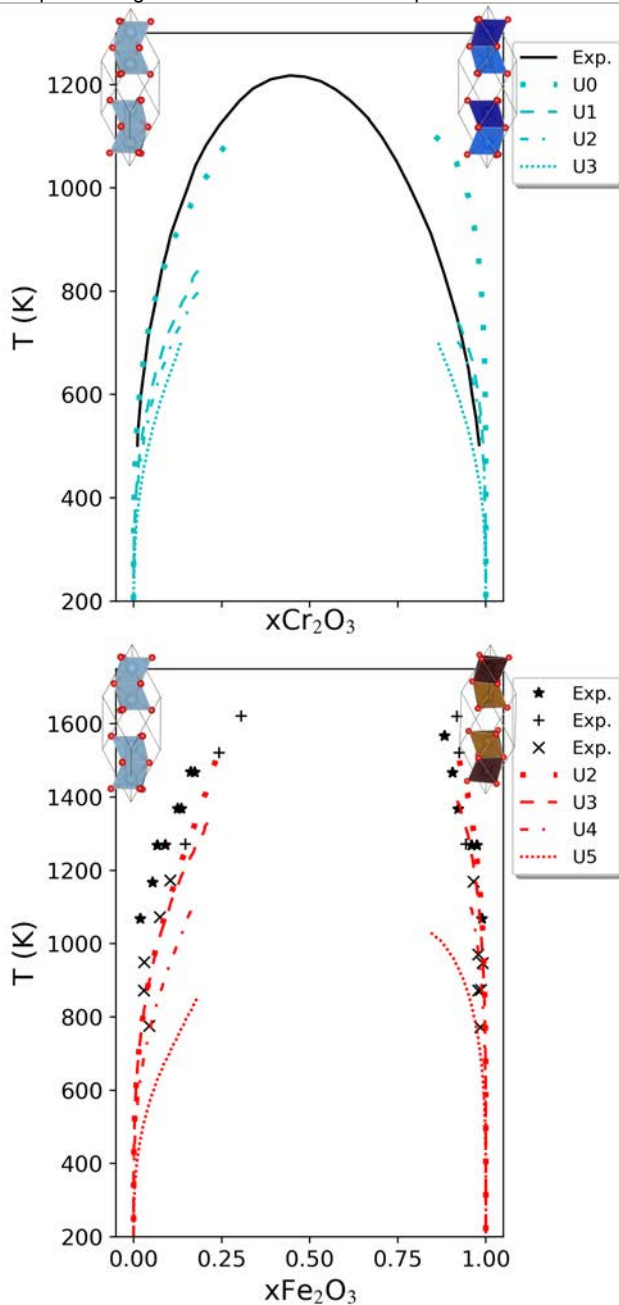


Figure 6: Calculated phase boundary of (top) Al<sub>2</sub>O<sub>3</sub> and anti-ferromagnetic Cr<sub>2</sub>O<sub>3</sub> and (bottom) Al<sub>2</sub>O<sub>3</sub> and anti-ferromagnetic Fe<sub>2</sub>O<sub>3</sub> across values of U vs. the experimental curve. Ground state structures are pictured at the top of each plot.

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