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3	Nb and Ta intracrustal differentiation during granulite-facies metamorphism: evidence from
4	geochemical data of natural rocks and thermodynamic modeling
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ABSTRACT

Both continental crust and depleted mantle are characterized by subchondritic Nb/Ta, leading to 15 a mass imbalance when compared to the bulk Earth. Even though several potential high Nb/Ta 16 reservoirs in Earth's core and undepleted mantle have been proposed, little attention has been 17 given to those in the crust. Here we present bulk-rock and rutile geochemical data for samples 18 from a lower crustal pelitic granulite, North China Craton, which exhibit systematic variation in 19 their Nb and Ta contents. High-temperature (HT) and ultra-high temperature (UHT) granulite 20 21 residues exhibit Nb/Ta ratios that are close to chondritic and subchondritic, respectively, whereas leucosomes from UHT granulite mostly have suprachondritic Nb/Ta. These variations are best 22 explained via competition for Nb and Ta between biotite and rutile during metamorphism, 23 although initial bulk-rock Nb/Ta values also have an effect. As biotite preferentially incorporates 24 Nb over Ta, the early stages of biotite dehydration melting produce a high-Nb/Ta residue and a 25 low-Nb/Ta melt; however, geochemical modeling suggests that once biotite is depleted, the 26 Nb/Ta ratio of the system is instead controlled by rutile growth, which promotes formation of a 27 lower Nb/Ta residue and a higher Nb/Ta melt, even though the volume of melt produced at this 28 29 stage may be small. We propose that *in-situ* and *in-source* leucosomes and leucocractic veins in UHT terranes may retain a high Nb/Ta geochemical signature. However, residual crustal-derived 30 A2-type granites that experience significant fractionation of Nb- or Ta-bearing minerals during 31 crystallization or contamination from other low-Nb/Ta sources cannot retain this high Nb/Ta 32 ratio, even though these ratios are generally higher than that of S-type granites. Anhydrous 33 partial melting of metapelite can generate Nb-rich melts, such that high temperature leucosomes 34 35 in addition to related A2-type granites may represent significant Nb deposits.

36 Keywords: Nb/Ta; granulite; partial melting; forward modeling; ultra-high temperature; rutile

37

INTRODUCTION

38	Although the continental crust constitutes only $\sim 0.6\%$ mass of the silicate Earth, it
39	contains a very large proportion of incompatible elements, which makes it a critical reservoir for
40	mass-balance calculations of the Earth as a whole (Sclater et al., 1980). Previous studies have
41	shown that both continental crust and depleted mantle have subchondritic Nb/Ta (Rudnick, 1995;
42	Kelemen, 1995; Rudnick et al., 2000), which is problematic for chondritic Earth models, since
43	no complementary suprachondritic Nb/Ta reservoirs are known. Although intracrustal Nb/Ta
44	differentiation has received relatively little attention, several high Nb/Ta domains have been
45	proposed to exist within the Earth to address this issue. These include the core, the lower mantle,
46	the carbonatite-metasomatized lithospheric mantle, refractory eclogite formed during slab
47	melting, and deeply crystalized high-Nb cumulates (Aulbach et al., 2008; Cartier et al., 2014;
48	Münker et al., 2003; Pfänder et al., 2012; Rudnick et al., 2000; Tang et al., 2019; Wade and
49	Wood, 2001; Willbold and Stracke, 2006). Recently, Stepanov and Hermann (2013) suggested
50	that significant partial melting of restitic lower crust could produce a potential high-Nb and high
51	Nb/Ta reservoir. In their model, incipient partial melting of mica-rich rocks during prograde
52	metamorphism after muscovite incongruent melting ends, but before biotite dehydration melting
53	begins, will produce a biotite-rich residue. Given that biotite incorporates Nb over Ta, the
54	residue will evolve towards high Nb/Ta values during melt extraction. Progressive biotite-
55	dehydration melting would then produce high-Nb/Ta rutile and ilmenite, since the $D_{\text{Nb}}/D_{\text{Ta}}$ of
56	rutile and ilmenite are close to unity. Nonetheless, experimental work performed by Xiong et al.
57	(2011) showed that D_{Nb}/D_{Ta} of rutile and ilmenite at conditions relevant to biotite dehydration
58	melting (850 °C) is approximately 0.8, and would be even lower during fluid-absent melting
59	(Klemme et al., 2005; Xiong et al., 2011). Partition coefficients for Nb and Ta in different

minerals are highly variable, and each is sensitive to pressure (*P*), temperature (*T*), and melt
composition (Klemme et al., 2005; Xiong et al., 2011). Thus, understanding the behavior of these
elements in the crust can place critical constraints on the mechanisms of intracrustal
differentiation.

64 High-grade metamorphism and anatexis are ubiquitous in the lower crust, and melts that subsequently ascend through the overlying rock column and crystallize as granite at shallower 65 depths represent a key mass-transfer process that leads to geochemical differentiation (Brown et 66 al., 2011). During isobaric heating at pressures greater than ~ 0.5 GPa, such as during burial or 67 crustal thickening, metapelites experience continuous partial melting as temperature increases, 68 and biotite undergoes incongruent melting to form peritectic garnet and rutile (Sawyer et al., 69 2011; Patiño-Douce et al., 1991; Stevens et al., 2007). As a restitic phase, rutile is a major host 70 for high field strength elements (HFSE) and act as a reservoir for the vast majority of Nb and Ta 71 72 in a rock (Meinhold, 2010). Therefore, a detailed study of bulk-rock and rutile geochemistry in pelitic granulites is critical to decipher the evolution of the continental crust and develop small-73 and large-scale reservoir models. 74

In this study, a series of high-temperature (HT) to ultrahigh temperature (UHT) pelitic granulites from the northwestern part of the North China Craton were selected for a bulk-rock and rutile geochemical study. These data are compared to the results from petrological forward modeling for a typical metapelite that combines phase diagram analysis, accessory mineral solubility modeling, and trace element modeling. Integration of geochemical data from the study area with data for other granulites preserved worldwide allow Nb-Ta intracrustal geochemical fractionation processes during granulite-facies metamorphism to be further understood.

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GEOLOGICAL SETTING

83	To explore Nb and Ta intracrustal geochemical fractionation processes during
84	progressive heating, we studied several samples of granulite that share a common protolith, but
85	record different thermal evolutions. These comprised pelitic granulites from the northern part of
86	the Trans-North China Orogen and eastern part of the Khondalite Belt, North China Craton (Fig.
87	1a). The HT-HP granulites from the Trans-North China Orogen have been well-studied since the
88	2000s and UHT granulites in the Khondalite Belt contain well-preserved mineral assemblages
89	diagnostic of peak metamorphic temperatures above 900 °C (Guo et al., 2002; Santosh et al.,
90	2012). The Trans-North China Orogen is a north-south trending Paleoproterozoic orogenic belt
91	(ca. 1.93–1.80 Ga) that separates the Eastern Block and Western Block of the North China
92	Craton (Zhao et al., 2005). The northern part of the Trans-North China Orogen is dominated by
93	tonalite-trondhjemite-granodiorite (TTG) and diorite gneisses that were emplaced at ca. 2.5 Ga,
94	and that were metamorphosed at ca. 1.85 Ga. (Liu et al., 2012). Mafic granulite occurs as
95	lenticular enclaves within the TTG gneisses, and metapelitic layers adjacent to the TTG gneisses
96	record peak metamorphism at $1.92-1.85$ Ga and $P-T$ conditions of $1.1-1.2$ GPa at temperatures
97	< 900 °C, followed by decompression and cooling to the solidus (Huang et al., 2018; Wu et al.,
98	2016). Additionally, HT-HP pelitic granulites have also been recognized in the Manjinggou and
99	Huangtuyao areas (Wang et al., 2016; Wu et al., 2016; Zhang et al., 2016) (Fig. 1b).
100	Based on the subdivision by Zhao et al. (2005), the Western Block is divided into the
101	Yinshan Block in the north and the Ordos Block in the south (Fig. 1a). The Khondalite Belt is an
102	east-west trending Paleoproterozoic orogen, formed by the collision of the Yinshan Block and
103	Ordos Block at ca. 1.95 Ga (Zhao et al., 2005). The eastern Khondalite Belt is dominated by
104	aluminous gneiss, garnet-bearing orthogneiss, and minor gabbro and norite (Fig. 1b). UHT

105	granulites with the diagnostic mineral assemblage sapphirine + quartz, initially recognized at the
106	Tuguiwula area, record an initial stage of isobaric cooling from peak $P-T$ conditions of 0.8–0.9
107	GPa at temperatures > 960 °C, followed by decompression and cooling to the solidus (Huang et
108	al., 2019; Li and Wei, 2018; Santosh et al., 2012). UHT metamorphism has since been
109	recognized from other localities, such as Heling'er, Xiaoshizi, and Zhaojiayao, indicating that
110	heating was widespread in the eastern part of the Khondalite Belt (Jiao et al., 2013; Li et al.,
111	2016; Liu et al., 2012). The garnet-bearing orthogneiss is peraluminous in composition and was
112	emplaced at 1.95–1.94 Ga (Huang et al., 2019; Peng et al., 2012; Wang et al., 2018). This garnet-
113	bearing orthogneiss and associated aluminous gneiss then experienced UHT metamorphism at ca.
114	1.92 Ga, which was likely induced by emplacement of nearby mafic magmas (Huang et al., 2019;
115	Peng et al., 2010; Santosh et al., 2012). Given the belt-like distribution with small-scale and arc-
116	like geochemical affinities, Peng et al. (2010) suggested that the mafic magmas are associated
117	with a northwestward ridge-subduction-related mantle upwelling event.

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SAMPLE DESCRIPTION

119 HT–HP granulite from the Trans-North China Orogen

Pelitic granulites from the Manjinggou area (Fig. 2a), northern Trans-North China Orogen, occur as layers adjacent to the TTG gneiss (Wu et al., 2016). *In-situ* to *in-source* leucosomes recognized in this area are interpreted as products of the crystallization of anatectic melts that have not migrated away from their source (Sawyer, 2008) (Fig. 2b). Minor synmetamorphic S-type granite has been reported in this area. Given the relatively hot conditions of peak metamorphism (~ 850 °C) recorded by the metapelitic rocks, a significant amount of melt (up to 22 vol.%) should have been generated during heating (Huang et al., 2021); however, this

127 amount is not observed in the field. Melts generated during metapelite partial melting are expected to dominantly ascend and collect in the upper crust to form granitic plutons (Brown, 128 2013). Therefore, the paucity of syn-metamorphic granite could be explained by the erosion of 129 130 the overlying upper crustal levels during exhumation. The leucosomes have a simple mineralogy 131 dominated by felsic minerals (i.e. plagioclase, K-feldspar and quartz), and were likely derived from biotite dehydration melting (Wu et al., 2016). Ten granulite samples (restite), collected 132 133 from the Manjinggou area (Fig. 1b), contain K-feldspar, quartz, plagioclase, garnet, sillimanite, biotite, rutile, and minor kyanite (Fig. 3a, b). Accessory minerals include zircon, monazite, and 134 apatite. In this study, detailed petrological work was conducted on sample 18MJ09 (GPS: 135 40°22'43.266"N, 114°27'56.404"E). 136 Garnet in sample 18MJ09 usually occurs as coarse-grained poikiloblasts that contain 137 138 many small inclusions of quartz and/or biotite in the core domains (Fig. 3c). These inclusions, 139 along with the garnet core itself, represent part of the prograde mineral assemblage (M_0). Kyanite 140 is occasionally present as inclusions in the garnet rims (Fig. 3a, b), but not in the matrix. Biotite, 141 K-feldspar, plagioclase, quartz, and rutile, that occur as inclusions with kyanite in garnet rims, 142 define the peak pressure mineral assemblage (M₁). A key melt-forming prograde metamorphic reaction $(M_0 \rightarrow M_1)$ is thus proposed to be biotite + quartz + plagioclase \rightarrow garnet + kyanite + K-143 144 feldspar + melt (Wu et al., 2016). Rutile inclusions within garnet are either isolated or occur in spatial association with quartz and/or biotite (Fig. 3c, d). In the matrix, K-feldspar, plagioclase, 145 quartz, sillimanite, garnet, biotite, and rutile represent a retrograde (decompressed) mineral 146 assemblage (M₂). Thus, the polymorphic transition kyanite \rightarrow sillimanite likely occurred during 147 decompression (Wu et al., 2016). Rutile in the matrix occurs in contact with quartz, feldspar, 148 garnet, and sillimanite. 149

150 UHT granulite from the eastern Khondalite Belt

Compared with HT granulite from northern Trans-North China Orogen, UHT granulite 151 from the Tuguiwula area (Fig. 2c) in the eastern Khondalite Belt experienced more extensive 152 partial melting (Fig. 2d). Alongside *in-situ* to *in-source* leucosomes, leucocratic veins were also 153 recognized in the field and are interpreted as the product of crystallization of anatectic melts that 154 migrated out of their source domain and were injected into adjacent rocks (Sawyer, 2008). 155 Meanwhile, voluminous syn-metamorphic S-type granites, which compose 40 vol. % of the 156 157 eastern Khondalite Belt, have been reported by many workers (e.g. Huang et al., 2019; Peng et al., 2012; Wang et al., 2018). The leucosomes mainly contain felsic minerals with minor garnet, 158 sillimanite, and spinel. Given that few hydrous minerals remain in the residues and leucosomes, 159 these leucosomes most likely crystallized from anhydrous melts. Fourteen felsic granulite 160 samples (restites) and nine layered leucosomes in the granulite were collected from the 161 162 Tuguiwula area. The sedimentary protoliths of these granulite samples varied lithologically 163 between graywacke and pelite. Here, we focus on felsic paragneiss (restite) sample 16TG53 164 (GPS: 40°46′26.57″N, 113°11′4.05″E), which is a meta-graywacke (Fig. 2c), and contains 165 plagioclase, quartz, K-feldspar, garnet, sillimanite, biotite, and rutile (Fig. 4). Accessory minerals 166 include zircon, monazite, and apatite. In sample 16TG53, sillimanite occurs only as inclusions in garnet – not in the matrix 167

(Fig. 4a, c) – and so it is considered part of the prograde mineral assemblage (M_0). This assemblage additionally included plagioclase, K-feldspar, quartz, biotite, and rutile, as recorded by inclusions in garnet. Rutile inclusions in garnet occurs either as isolated grains or in association with quartz and/or biotite (Fig. 4). Coarse-grained garnet (>3 mm diameter), plagioclase, K-feldspar, quartz, and rutile in the matrix form the peak assemblage (M_1),

173	indicating progression of the prograde metamorphic reaction biotie + sillimanite \rightarrow garnet + K-
174	feldspar + rutile + melt. Rutile grains occur both in contact with felsic minerals and garnet (Fig.
175	4b, c, d). Coarse-grained matrix garnet and K-feldspar are locally separated by fine- to medium-
176	grained biotite, sillimanite, and rutile (Fig. 4c, d), suggesting that K-feldspar, quartz, garnet,
177	sillimanite, biotite, and rutile represent a retrograde mineral assemblage (M2) that formed due to
178	the reaction garnet + K-feldspar + rutile + melt \rightarrow biotite + sillimanite.
170	METHODS
1/9	METHODS
180	Analytical methods
181	Whole-rock major and trace elements.
182	Whole-rock major element compositions were determined by X-Ray fluorescence (XRF)
183	at the State Key Laboratory of Lithospheric Evolution, Chinese Academy of Sciences, Beijing,
184	China. The samples were crushed and powdered to 200 mesh using an agate mortar. Afterwards,
185	approximately 0.50 g of rock powder was ignited at 1000 °C for ~1 hr to obtain loss on ignition
186	(LOI) and then fused with three to four drops of lithium tetraborate using a CLAISSE M4
187	Fluxer. Uncertainties depend on the concentrations in the sample, but are generally better than
188	±1%.
189	Whole-rock trace element analyses were obtained at the China University of Geosciences
190	(Wuhan). Rock powders were digested using a HF+HNO ₃ solution in high-pressure Teflon
191	bombs. Trace element concentrations were analyzed by inductively coupled plasma mass
192	spectrometry (ICP-MS) using an Agilent 7700e system. The analytical precision and accuracy
193	for trace element concentrations are described by Liu et al. (2008). The precision for most
194	elements was typically better than $\pm 5\%$ RSD (relative standard deviation) calculated by analyses

of repeated unknown samples and certified USGS reference materials (BHVO-2, BCR-2 and
RGM-2) (USGS, 1996, 2004).

197 Electron probe microanalysis (EPMA).

198 Rutile compositions were analyzed via EPMA at the State Key Laboratory of

- 199 Lithospheric Evolution, Institute of Geology and Geophysics, Chinese Academy of Sciences,
- 200 Beijing, China (IGGCAS) using a CAMECA SX51. The microprobe employed an accelerating
- voltage of 20 kV, a beam current of 200 nA, and a beam diameter of 5 µm. The following
- 202 elements were analyzed: Ti, V, Zr, Nb, Ta, Cr, Fe, Al, and Si. The counting time on the peak was
- 203 10 s, 120 s, 240 s, 120 s, 240 s, 120 s, 120 s, 120 s, and 240 s, respectively (Wang et al., 2017).
- The detection limits are 706 ppm for Ti, 84 ppm for V, 118 ppm for Zr, 144 ppm for Nb, 210

ppm for Ta, 154 ppm for Cr, 126 ppm for Fe, 60 ppm for Al, and 36 ppm for Si. Rutile R10 was

used as a standard (Luvizotto et al., 2009b). Silicon concentrations were used as a quality control

to detect and avoid contamination associated with submicroscopic zircon inclusions (Zack et al.,

208 2004; Luvizotto et al., 2009a). Rutile measurements with Si contents higher than 300 ppm and

209 those showing abnormal Ti concentrations were eliminated from the data set.

Laser-ablation quadrupole inductively coupled plasma mass spectrometry (LA-Q ICP-MS).

Trace element analysis of rutile was conducted by LA-ICP-MS at the Wuhan

213 SampleSolution Analytical Technology Co., Ltd., Wuhan, China. Detailed operating conditions

- for the laser ablation system and the ICP-MS instrument and data reduction are the same as those
- described by Zong et al. (2017). Laser sampling was performed using a GeolasPro laser ablation
- system that consists of a COMPexPro 102 ArF excimer laser (wavelength of 193 nm and

217	maximum energy of 200 mJ) and a MicroLas optical system. An Agilent 7900 ICP-MS
218	instrument was used to acquire ion-signal intensities. Helium was used as a carrier gas. Argon
219	was used as the make-up gas and mixed with the carrier gas via a T-connector before entering
220	the ICP. A "wire" signal smoothing device is included in this laser ablation system (Hu et al.,
221	2015). The spot size and frequency of the laser were set to 44 μm and 5 Hz, respectively. Trace
222	element compositions of minerals were calibrated against various reference materials (NIST 610,
223	BHVO-2G, BCR-2G, and BIR-1G) without using an internal standard (Jochum et al., 2007, 2011;
224	Liu et al., 2008; USGS, 1996, 2004). Each analysis incorporated a background acquisition of
225	approximately 20-30 s followed by 50 s of data acquisition from the sample. The Excel-based
226	software ICPMSDataCal was used to perform off-line selection and integration of background
227	and analyzed signals, time-drift correction and quantitative calibration for trace element analysis
228	(Liu et al., 2008). The mean values of uncertainties (2σ) and detection limits are 256 ppm and
229	1.09 ppm for V, 158 ppm and 1.24 ppm for Zr, 5.95 ppm, and 0.196 ppm for Hf, 94.6 ppm and
230	1.86 ppm for Nb, 6.57 ppm and 0.101 ppm for Ta, and 68.3 ppm and 12.8 ppm for Cr.
231	

- 232 Modeling of partial melting of metapelite
- 233 Phase equilibria modeling.

234 Phase equilibria modeling was conducted on a typical metapelite (Ague, 1991) using

THERMOCALC v. 3.40 (Powell and Holland, 1988) with internally consistent data set (ds62) of

Holland and Powell (2011). The calculations were performed in the Na₂O-CaO-K₂O-FeO-MgO-

237 Al_2O_3 -SiO₂-H₂O-TiO₂-Fe₂O₃ chemical system using *a*-*x* models from White et al. (2014).

238 Quartz, rutile, and H₂O were considered as pure phases. The H₂O content was assigned so that

239	the modeled composition was minimally saturated at the solidus to ensure fluid-absent melting
240	(cf. Palin et al., 2016a). The Fe ³⁺ /Fe ^{total} was constrained so that rutile was stable in the $P-T$
241	range. Uncertainty on the position of the assemblage field boundaries is commonly less than ± 50
242	°C and ± 1 kbar at 2 σ . (Palin et al., 2016b).
243	Progressive partial melting in a rock would induce an interconnected network of grain-
244	boundary melt-filled channels that allow its escape at a critical threshold (7 vol. %) in the
245	absence of syn-anatectic deformation (Rosenberg and Handy, 2005). This threshold would be
246	even lower in the presence of syn-anatectic deformation (Brown, 2010). We modeled melt loss
247	events as realistically as possible by conducting thermodynamic modeling in a conditional open
248	system (e.g. Yakymchuk and Brown, 2014). The critical melt connectivity threshold of 7 vol. %
249	was used to model open-system processes, whereby once that proportion was reached along the
250	proposed $P-T$ paths of interest, six-sevenths of the melt present were removed as a 'melt loss
251	event', based on the common observation of small proportions (1 vol.%) remaining of melts
252	occurring as films on grain boundaries in rapid chilled and drained migmatites (Yakymchuk and
253	Brown, 2014). In reality, this melt drainage process would represent a melt fraction that leaves
254	the local system and ascends through the overlying crustal column towards the Earth's surface.
255	This was performed during calculation by manipulation of the 'rbi' script, as outlined by

256 Yakymchuk and Brown (2014). The compositions of extracted melt are listed in Table S4. Three

thermal gradients including isobaric heating at 0.8 GPa, 775 °C/GPa, and 550 °C/GPa were

selected to represent the overriding plate of a subduction zone, burial, and crustal thickening

260 good approximation of the starting point of partial melting (Yakymchuk and Brown, 2014), was

settings, respectively. A temperature of 640 °C, which is close to the wet-granite solidus and a

selected as the lower boundary of the phase diagram.

259

262 **Trace element modeling.**

Trace element modeling was carried out using the batch melting equation $C_{melt}/C_{source} =$ 263 $1/[D + F \times (1 - D)]$ (Shaw, 1970), where C_{source} and C_{melt} represent concentrations of a trace 264 element in the source rock and the resultant melt, respectively; D (= $\sum Kd \times X$) is the bulk 265 partition coefficient, where Kd is the mineral/melt partition coefficient and X is the proportion in 266 mol. % of the mineral. F is the degree of melting. Initial bulk concentrations of 0.19 wt. % P₂O₅, 267 17 ppm Nb, 1.5 ppm Ta, 150 ppm Zr, and 150 ppm LREE were used for the metapelite (Taylor 268 269 and McLennan, 1985; Ague, 1991; Yakymchuk et al., 2018). The proportions of accessory minerals in the residue were calculated using the refined zircon solubility model from Boehnke 270 et al. (2013), the monazite solubility model from Stepanov et al. (2012), and the apatite solubility 271 model from Harrison and Watson (1984). The Kd values used in the modeling are mostly mean 272 values taken from the literature and tabulated in Table 1, except for rutile. As a peritectic phase 273 from biotite dehydration melting, rutile will always be stable during granulite-facies 274 metamorphism. However, the D_{Nb}/D_{Ta} of rutile decreases with temperature (Xiong et al., 2011), 275 and would be around 0.8 at the conditions relevant to biotite dehydration melting (i.e. 850 °C) 276 277 and approximately 0.45 in dry material at UHT conditions (Xiong et al., 2011; Klemme et al., 2005). In such cases, specific Kd values of rutile, used in the modeling, are listed in Table 1. The 278 calculated Nb and Ta concentrations in the melts are listed in Table S4, and masses of accessory 279 minerals left in the residue are listed in Table S5. 280

281

RESULTS

282 Bulk-rock composition of granulites and leucosomes

283	Granulites collected from the northern Trans-North China Orogen have mostly similar
284	bulk-rock compositions, with $SiO_2 = 53.2-68.4$ wt.% (mean 62.7 wt.%), $Al_2O_3 = 14.6-23.2$ wt.%
285	(mean 18.5 wt.%), and MgO = $2.1-3.4$ wt.% (mean 2.6 wt.%). Heat-producing elements (e.g. K,
286	Th, U) have consistent and relatively high concentrations ($K_2O = 2.3-4.4$ wt.%, U = 0.6-2.1
287	ppm, and Th = $10.4-18.6$ ppm). Niobium concentrations range from 12.8 to 15.6 ppm, and
288	Nb/Ta ratios vary from 16.9 to 24.8, with a mean of 20.5, which is slightly higher than the
289	chondritic value of 19.9 (Fig. 5a) (Münker et al., 2003). By comparison, restites collected from
290	the eastern Khondalite Belt have more variable SiO ₂ (51.8–70.4 wt.%, mean 59.3 wt.%), higher
291	Al ₂ O ₃ (13.6–25.1 wt.%, mean 20.2 wt.%), and higher MgO (2.2–6.1 wt.%, mean 3.5 wt.%). The
292	lower SiO ₂ and higher Al ₂ O ₃ and MgO concentrations likely resulted from significant melt loss
293	(Lal et al., 1978; Clifford et al., 1981; Droop and Bucher-Nurminen, 1984; Raith et al., 1997;
294	Baba, 2003; Brandt et al., 2007; Huang et al., 2021). Heat-producing elements usually show
295	variable concentrations, but are more depleted than those in HT granulites ($K_2O = 0.5-4.6$ wt.%,
296	U = 0.2-1.3 ppm, Th = 0.6-11.3 ppm). As discussed by Ewing et al. (2014), UHT
297	metamorphism is an important process in driving the differentiation of the continental crust into
298	an upper crust rich in heat-producing elements and a relatively depleted lower crust. Variability
299	in heat-producing element concentrations in restites from the eastern Khondalite Belt likely
300	relate to various degrees of partial melting. These UHT restites have Nb contents in the range
301	3.7–26.8 ppm and Nb/Ta ratios of 12.5–21.1, with a mean value of 16.8, which is lower than the
302	chondritic value, but higher than that for a bulk continental crust (BCC) of 12.4 (Rudnick and
303	Gao, 2014) (Fig. 5b). Conversely, leucosomes within these granulites, which represent

304	anhydrous melts generated in-situ during UHT metamorphism, have a distinct bulk-rock
305	composition from the restites. All the leucosomes are granitic and peraluminous, having SiO_2
306	contents in the range of 66.3–72.9 wt. %, and A/CNK [molar $Al_2O_3/(CaO + Na_2O + K_2O)$]
307	values in the range of 1.3–2.4. These leucosomes have Nb contents of 0.4 ppm to 10.1 ppm, and
308	Nb/Ta ratios of 17.1 to 44.1, with a mean of 24.1 (Fig. 6a).
309	Rutile trace elements
310	Rutile grains in HT-HP granulite sample 18MJ09 have Zr concentrations mostly ranging
311	from 644 ppm to 2920 ppm (Fig. 7a). Based on the results of pressure-corrected Zr-in-rutile
312	thermometry (Tomkins et al., 2007), sample 18MJ09 registered a peak temperature of 883 °C at
313	a pressure of 1.1 GPa suggested by Wu et al. (2016). Specifically, Zr concentrations of isolated
314	rutile inclusions in the matrix and rutile inclusions associated with quartz in garnet
315	porphyroblasts cluster into two groups at 644–929 ppm and 2400–2920 ppm, of which the
316	former group represents prograde growth and the latter represents growth at peak metamorphism.
317	In the matrix, rutile in contact with felsic minerals or sillimanite usually has a high and restricted
318	range of Zr concentrations (2560–2870 ppm), but those in contact with garnet have more
319	variable Zr concentrations (1120–2620 ppm). Niobium concentrations in rutile are mostly 1320–
320	2260 ppm, and Nb/Ta ratios are 16.3–32.6 with a mean value of 20.7 (Fig. 7a). Among those
321	rutile grains, isolated rutile inclusions in garnet with high Zr concentrations have variable Nb/Ta
322	ratios of 16.3–32.6, whereas others show limited variation (18.8–24.4). Figure 7a shows that
323	rutile grains with high Nb/Ta usually have high Zr concentrations, indicating that they formed at
324	higher temperatures, and so are inferred to have formed from a biotite-breakdown peritectic
325	reaction (Fig. 7a).

326	Rutile in UHT granulite sample 16TG53 has Zr concentrations ranging from 314 to 6710
327	ppm (Fig. 7b), corresponding to temperatures of 645–980 °C at 0.80 GPa (cf. Huang et al.,
328	2019). Among the various habits of rutile, isolated rutile inclusions and those associated with
329	quartz in garnet porphyroblasts have the highest Zr concentrations. Most other rutile grains have
330	relatively lower Zr concentrations, except for several that occur in contact with felsic minerals.
331	Rutile in the matrix contains 607–3960 ppm Nb, with a mean of 2780 ppm, and Nb/Ta ratios
332	range from 10.3 to 35.0 with a mean value of 17.5. Some individual rutile inclusions within
333	garnet have much lower Nb concentrations (607–637 ppm) but higher Nb/Ta ratios (34.1–35.0).
334	Some retrograde rutile grains around garnet have lower Zr concentrations (884–1160 ppm) but
335	higher Nb/Ta ratios (38.6–64.7) (Fig. 7b).

336 Modeling results

Since melt produced during partial melting will leave its host rock when it reaches a 337 critical melt connectivity transition (MCT) of 7 vol. % (Rosenberg and Handy, 2005), 338 339 petrological modeling of equilibrium phase assemblages must take into account repeated melt loss and the production of an increasingly refractory residuum. Isobaric heating at 0.8 GPa from 340 subsolidus conditions (700 °C) to UHT conditions (1000 °C) was used to simulate thermal 341 342 metamorphism of the lower continental crust. At these conditions, five 'melt loss events' are predicted to occur at 754 °C, 780 °C, 808 °C, 828 °C, and 913 °C (Fig. 6b, 8). The first melt 343 344 produced during metamorphism forms via muscovite-dehydration melting reactions, whereas 345 melt generated between melt loss events 2 and 4 is produced by biotite-dehydration melting 346 reactions. Melt lost during melt loss event 5 is generated via anhydrous breakdown of feldspar 347 and quartz. All melt fractions show a progressive increase in A/CNK (1.12-1.21) with increasing temperature. Melts expelled during melt loss events 1-4 show increasing (but subchondritic) 348

Nb/Ta ratios of 4.91–15.0. The final melts to be generated and expelled have suprachondritic
Nb/Ta ratios of 26.8, which is consistent with bulk compositions recorded by leucosomes in
natural granulites (Fig. 6).

352

DISCUSSION

353 Intracrustal differentiation during granulite-facies metamorphism

Several episodes of partial melting are known to occur in felsic rock types during prograde metamorphism, which are driven by the breakdown of various hydrous and anhydrous minerals (Brown, 2013; Sawyer et al., 2011). Loss of melt from these host rocks and its vertical ascent to shallower levels induces intracrustal differentiation, which is the main form of mass transfer that controls the evolution of the continental crust through geological time (Brown et al., 2011; Palin and Santosh, 2021).

Fluid-present melting may occur in pelitic rocks at temperatures as low as ~650 °C due to 360 simultaneous consumption of grain-boundary fluid and anhydrous felsic minerals (Sawyer et al., 361 2011). Thus, Nb-Ta fractionation may occur between the melt and residue during this process. 362 However, as the main mineral hosts for Nb and Ta (i.e. muscovite, biotite, and oxide minerals) 363 364 are not involved in these reactions, the Nb-Ta content and Nb/Ta of the residues will likely not 365 change and the expelled melts will be Nb- and Ta-poor. Once free fluids are exhausted during 366 heating, incongruent melting of hydrous minerals will occur, with muscovite- and biotite-367 breakdown reactions considered as the primary mechanism for generating crustal (S-type) granites (Brown, 2013; White et al., 2017). Partial melt in the source rock may generate 368 substantial overpressure, leading to fracturing and so aiding subsequent melt loss (Clemens and 369 Droop, 1998). Our modeling of isobaric heating at 0.8 GPa indicates that metapelites can lose up 370

to 27 vol. % of their original mass via extraction of five discrete melt fractions (Fig. 8). The heat-371 producing element (U, Th and K) budget of the source will be progressively depleted, as the 372 minerals hosting these elements, e.g. zircon, monazite and apatite, will progressively dissolve 373 374 during partial melting and these elements will readily enter the silicate melt prior to its extraction (Rudnick and Gao, 2014; Ewing et al., 2014; Huang et al., 2021). The HT granulites studied here 375 show limited depletion in heat-producing element concentrations, which indicates only a 376 377 relatively minor degree of partial melting and melt loss compared to those that reached UHT conditions. The bulk rock Nb/Ta ratio at this stage is thus mainly controlled by muscovite and 378 biotite stability, since mica is ubiquitous, but peritectic rutile is scarce. As biotite preferentially 379 incorporates Nb over Ta (Stepanov and Hermann, 2013), anatexis would produce a restite with a 380 high Nb/Ta ratio and melt with a low Nb/Ta ratio (Fig. 5a, 6b). However, the amount of biotite in 381 the restite will decrease during heating, as it is consumed during melt-producing reactions. As 382 such, the Nb/Ta budget of the residue will instead be controlled by rutile and ilmenite once 383 biotite is scarce. After biotite is totally consumed at about 840 °C, the restites show a strong 384 385 depletion in heat-producing elements, since U and Th host minerals (e.g. zircon, monazite and apatite) are mostly dissolved along with biotite (e.g. Ewing et al., 2014; Huang et al., 2021). The 386 UHT residues in this study usually show clear depletion in heat-producing elements, indicating 387 388 substantial partial melting and melt loss. During anhydrous melting, the rate of melt production is predicted to dramatically decrease (Sawyer et al., 2011), but this process will generate a 389 relatively low Nb/Ta restite and high Nb/Ta melt (Fig. 5b, 6), since rutile is the main host 390 mineral and preferentially incorporates Ta over Nb under anhydrous melting conditions 391 392 (Klemme et al., 2005, Xiong et al., 2011).

393 Rutile in both samples shows a bimodal distribution of Zr concentrations, which is consistent with data reported previously by Luvizotto et al. (2009a). In that study, they proposed 394 that the highest Zr contents recorded peak temperatures; as such, peak metamorphism in the two 395 samples studied here would be 880 °C for 18MJ09 and 980 °C for 16TG53, which is consistent 396 with phase equilibria modeling results (Wu et al., 2016; Huang et al., 2019). Biotite was stable 397 during each metamorphic stage in sample 18MJ09 and rutile preserves variable Nb/Ta values. 398 However, most rutile in UHT granulite sample 16TG53 has a narrow range of subchondritic 399 Nb/Ta values and controls the bulk-rock Nb/Ta ratio (Fig. 8). From these patterns, it can be 400 deduced that biotite breakdown during HT granulite-facies metamorphism caused variability in 401 402 rutile Nb/Ta ratios. However, continued metamorphism and melting at UHT conditions, where biotite is no longer stable, leads to the continuous dynamic recrystallization of rutile, which is the 403 only Nb- and Ta- carrier in the residue. This in turn causes the Nb/Ta ratio to evolve towards an 404 equilibrium value (Luzizotto et al., 2009a). However, several rutile inclusions within garnet and 405 retrograde rutile grains around garnet have much higher Nb/Ta values than the majority. Among 406 407 these, high Nb/Ta isolated rutile inclusions in garnet have Cr/Nb > 1, which indicates that they may be rutile sourced from a metamafic rock (Meinhold, 2010). Since Nb and Ta are strongly 408 incompatible in garnet, the Nb and Ta concentrations of these isolated rutile inclusions in garnet 409 410 could be preserved even during UHT metamorphism (Fulmer et al., 2010). In such cases, these 411 isolated rutile inclusions in garnet are most likely inherited rutile derived from eroded mafic rocks (Meinhold, 2010). Retrograde rutile grains with high Nb/Ta ratios are predicted to have 412 413 formed alongside garnet, biotite, K-feldspar, and high Nb/Ta melts.

414 Nb/Ta of extracted melt along various geothermal gradients

415	To further explore Nb/Ta geochemical fractionation during granulite-facies
416	metamorphism in different tectonic settings, geochemical modeling was also performed along
417	two additional geothermal gradients for comparison with the isobaric heating path discussed
418	above. These gradients were 775 °C/GPa and 550 °C/GPa, with the former considered as the
419	lower threshold of the high dT/dP field representing UHT metamorphism (e.g. Brown and
420	Johnson, 2018), and the latter representing a typical intermediate dT/dP thermal gradient passing
421	through the high-pressure granulite facies field (O'Brien and Rötzler, 2003; Palin et al., 2020).
422	P-T pseudosections and changing mineral proportions during heating are shown in the
423	Supplementary Files. The modeling results for Nb/Ta from extracted melt against temperature
424	are shown in Figure 9.

Four melt loss events are predicted to occur before 1000 °C along a thermal gradient of 425 775 °C/GPa (Fig. 9b). Given that five similar melt loss events are predicted during isobaric 426 427 heating at 0.8 GPa, collisional tectonic processes appear to reduce the fertility of rocks more effectively than static heating (e.g. Yakymchuk and Brown, 2014). The extracted melts generated 428 during melt loss events 1 to 4 for 775 °C/GPa have subchondritic Nb/Ta values of 6.75–14.2, 429 430 which are slightly higher than those for corresponding melts generated at melt loss events 1 to 4 during isobaric heating at 0.8 GPa (Fig. 9a, b), as more rutile is stable during metamorphism at a 431 432 thermal gradient of 775 °C/GPa. It is notable that the melt produced at UHT conditions would 433 also have a suprachondritic Nb/Ta ratio (24.1), even though no melt loss events would occur.

Along a thermal gradient of 550 °C/GPa, eight melt loss events would occur up to 1000 °C (Fig. 9c), caused by a much higher initial H_2O content in the metapelite needed to ensure minimal saturation at the solidus (cf. section 4.2.1). Specifically, melt generated at melt loss

437	events 4 to 8 (in the short interval 988–992 °C) is formed via muscovite-dehydration melting.
438	The Nb/Ta ratios of these melts show a systematic increase upon heating. However, no
439	suprachondritic melt is generated along such a lower thermal gradient (7.90–13.7), since
440	muscovite preferentially incorporates Nb over Ta (Raimbault and Burnol, 1998; Stepanov and
441	Hermann, 2013), and remains stable up to 992 °C.
442	These data suggest that melt generated and lost sequentially during metamorphism will
443	progressively evolve toward high Nb/Ta values upon heating, irrespective of the thermal
444	gradients. However, we predict that the range of Nb/Ta values of extracted melts will be greater
445	at higher thermal gradients (i.e. 4.9-26.8 during isobaric heating at 0.8 GPa) compared to low
446	thermal gradients (i.e. 7.9–13.7 along 550 °C/GPa). Higher thermal gradient thus leads the crust
447	to become more geochemically differentiated in terms of Nb-Ta systematics.

448

IMPLICATIONS

449 Figure 10 summarizes measured Nb/Ta values for typical HT (peak temperature < 900 °C) 450 and UHT granulites, including well-studied examples from the Ivrea Verbano zone, northern Italy, and the Napier complex, Antarctic (Ewing et al., 2014; Grew et al., 2006). Although both 451 452 HT and UHT granulite extend over wide Nb/Ta ranges, HT granulites usually have higher Nb/Ta 453 values than UHT granulites, with average Nb/Ta values in HT granulites being close to chondritic, and those in UHT granulites being subchondritic. Since most granulites have Nb 454 contents clustering in the range 0-40 ppm, the difference of Nb/Ta values comes from having 455 higher Ta contents within UHT granulites than those of HT granulites (Fig. 5, S5). Further, as 456 most low-grade argillaceous sediments have low Nb/Ta values (mean = 11.0, Taylor and 457 458 McLennan, 1985), we suggest that the Nb/Ta value of a metapelite would show a bell-shaped evolution during prograde heating, including progressive increase and decrease of Nb/Ta ratios 459

during low and high temperature melting, respectively. The modeling results in this study 460 qualitatively show this trend, but no suprachonditic Nb/Ta values were reproduced in the 461 residues of partial melting (Fig. 9). This is inconsistent with natural samples, as HT granulites 462 463 have previously been proposed as a high Nb/Ta reservoir (Stepanov and Hermann, 2013). One possible reason for this mismatch is that protoliths have different TiO₂ contents; for example, all 464 the granulites considered here have less than 1.0 wt.% TiO₂, and all HT granulites have an 465 466 average TiO₂ content of 0.79 wt.%. However, the TiO₂ content of the protolith used for modeling was 1.06 wt.%, and melt extraction lead to an increase in residual TiO_2 contents of up to 1.30-467 1.63 wt.% (i.e., two-time higher than those of HT granulites) (Table S3, S4). Higher bulk-rock 468 TiO₂ contents can stabilize more rutile in the residue, potentially leading to lower Nb/Ta after 469 melt loss. 470

Leucosomes in UHT terranes are thought to have crystallized from melt derived from *in*-471 situ anatexis of their host rocks, and usually have a high Nb/Ta ratio (up to 44). By using the Nb 472 and Ta contents of rutile in granulites and rutile/melt partition coefficient, the calculated melts in 473 equilibrium with UHT-stage rutile would have Nb contents up to 71.5 ppm, and Nb/Ta values up 474 475 to 42 (Fig. 11). For the same UHT sample (16TG53), calculated melts in equilibrium with prograde-stage rutile (M_0) would have lower Nb contents and Nb/Ta (up to 22 ppm and 22, 476 respectively). For the HT granulite (18TG09), calculated melts would have a Nb content below 477 14 ppm, but suprachondritic Nb/Ta value in the range of 20-40 (Fig. 11). However, muscovite-478 bearing peraluminous granite samples compiled by Ballouard et al. (2020), that were interpreted 479 to be derived from segregation, ascent, and emplacement of relatively low-temperature melts 480 generated by muscovite dehydration, have low Nb/Ta values in the range 5-15 (mean = 6.86) 481 (Fig. 6a). Highly differentiated muscovite-bearing peraluminous granites, that experienced 482

significant magmatic-hydrothermal alteration, have Nb/Ta < 5 (Ballouard et al., 2016), and were
excluded from the compilation. Our modeling results for partial melting of an average pelite also
show a monotonous Nb/Ta increase in melt extracted during heating, which is similar to that of
natural samples (Fig. 6b, 9).

Given that the Nb/Ta ratio of natural leucosomes in the UHT terrane and calculated 487 anhydrous melts at UHT condition are both suprachondritic, A2-type granite, which is derived 488 from anatexis of dry refractory granulite residue (Collins et al., 1982; Eby, 1990, 1992; 489 490 Ballouard et al., 2020), are expected to have a suprachondritic Nb/Ta ratio and potentially represent a relatively high Nb/Ta reservoir within the continental crust. Nonetheless, most A2-491 type granites are characterized by subchondritic Nb/Ta, even though they generally have higher 492 Nb/Ta than highly peraluminous granites at given Nb and Ta contents (Ballouard et al., 2020) 493 (Fig. 11). As discussed above, the melts generated by anhydrous melting can have 494 suprachondritc Nb/Ta values, although natural processes associated with granite generation are 495 often more complicated than the relatively simplistic model treatment applied above. Such 496 natural complexity involves episodic melt production, melt extraction, mixing, differentiation, 497 498 and emplacement (Clemens, 2012; Brown, 2013). High Nb/Ta minerals, such as biotite, may fractionate during any of these processes, leading to a decrease of Nb/Ta in the melt (Stepanov et 499 al., 2014). In addition, magma mixing and contamination from other low Nb/Ta sources, such as 500 501 a UHT residual mush, would also induce a chemical change in the melt during its ascent (e.g. Korhonen et al., 2015). Then, anhydrous melt, if unfractionated and lacking contamination, 502 might retain a high Nb/Ta ratio, leading to *in-situ* leucosome, *in-source* leucosome, and some 503 504 leucocractic veins in UHT terranes being potential high-Nb/Ta reservoirs.

505 We note that Nb and Ta intracrustal geochemical fractionation is important to both the "missing Nb paradox" that has implications for understanding whole-Earth geochemistry, but 506 also to economic geology research into critical metal-bearing deposits. As important rare-metal 507 508 elements, economic enrichments of Nb and Ta can be promoted by extreme degrees of fractional crystallization of assemblages comprising feldspar (over 90 %) from low-temperature S-type 509 granite melts (Ballouard et al., 2016, 2020; Stepanov et al., 2014). Our petrological modeling 510 511 results indicate that anhydrous melts produced in UHT terranes would have Nb contents mostly over 20 ppm, but potentially up to 71.5 ppm. Nonetheless, the Ta contents in these anhydrous 512 melts would not be economically valuable, as rutile in the residual host retains too much Ta, 513 514 leading to a subchondritic Nb/Ta in UHT granulites. A2-type granites appear to have a greater potential for retaining elevated Nb contents (Fig. 11) (Linnen and Cuney, 2005). We therefore 515 propose that UHT melting of metapelite/metagraywacke may indeed represent an efficient 516 process to produce high Nb and high Nb/Ta melts that may evolve and collect in the upper crust 517 to form A2-type granites. Some of these granites may have the potential to form economic Nb 518 519 deposits if they are significantly evolved and reach columbite saturation. High-T leucosomes from granulite terranes, representing frozen melts that did not migrated toward the upper crust, 520 521 could theoretically represent a new, still undiscovered, type of Nb deposits.

522

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851	48.
852	FIGURE CAPTIONS
853	FIGURE 1. (a) Tectonic subdivision of the North China Craton showing Precambrian units
854	(Zhao et al., 2005). The red square represents location of Figure 1b. (b) Geological sketch map of
855	the study area (modified after Guo et al., 2001). The ages of different groups and igneous rocks
856	are labeled, of which igneous ages are in black and metamorphic ages are in red.
857	
858	FIGURE 2. Field photographs of the HT–HP and UHT granulite from the study area. (a-b) HT-
859	HP granulite from the Manjingou area and (c-d) UHT granulite from the Tuguiwula area. (a)
860	Typical HT-HP pelitic granulite (sample 18MJ09) collected from the Manjingou area, northern
861	Trans-North China Orogen. (b) Minor degrees of partial melting in the HT–HP granulite. (c)
862	Typical UHT pelitic granulite (sample 16TG53) from the Tuguiwula area, east Khondalite Belt.
863	(d) Extensive partial melting in the UHT granulite.
864	

865	FIGURE 3. Photomicrographs of HT-HP granulite sample 18MJ09, shown in plane-polarized
866	light. Panels (a) and (b) show that kyanite occurs as inclusions within garnet rims but is absent in
867	the matrix. Sillimanite is ubiquitous in the matrix. (c) Inclusions of quartz and/or biotite within
868	poikiloblastic garnet cores represent a prograde mineral assemblage (M ₀), and rutile occurs as
869	isolated inclusions in garnet (M_1) and in the matrix (M_2) . (d) Rutile and biotite inclusions within
870	garnet (M_1) and rutile in the matrix.

871

FIGURE 4. Sample photomicrographs of UHT granulite, 16TG53, shown in plane-polarized light. (a) Isolated rutile, quartz, and sillimanite inclusions in garnet, representing part of a prograde assemblage (M_0). (b) Isolated rutile inclusion within garnet (M_0) and rutile in the matrix in contact with felsic minerals (M_1). (c) Rutile inclusion in garnet associated with biotite (M_0), rutile in the matrix in contact with felsic minerals, and garnet (M_1). (d) Rutile in the matrix in contact with felsic minerals (M_1) and tiny retrograde rutile along with biotite surrounding garnet (M_2).

879

FIGURE 5. Bulk-rock composition plots of Nb/Ta vs Nb contents of granulite samples from the

study area and other granulite terranes worldwide. Geochemical data summarized from:

882 Trivandrum Block, South India (Nandakumar and Harley, 2019); Ivrea Verbano Zone, north

Italy (Ewing et al., 2014); Ulashan-Daqingshan terrane, Khondalite Belt (Cai, 2014);

884 Bhopalpatnam granulite belt, central India (Vansutre et al., 2013); Napier complex, Antarctic

(Grew et al., 2006); Altay, northwestern China (Yang et al., 2015); and the Highland complex,

886 Sri Lanka (Table S3). Nb/Ta values of 12.4 in bulk continental crust (BCC) and 19.9 in chondrite

are from Rudnick and Gao (2014) and Münker et al. (2003), respectively.

889	FIGURE 6. (a) Plots of bulk-rock Nb/Ta vs A/CNK (molar Al ₂ O ₃ /(Na ₂ O+K ₂ O+CaO)) from
890	muscovite-bearing peraluminous granites worldwide (Ballouard et al., 2020) and leucosomes of
891	UHT granulites from the Khondalite Belt. Highly differentiated muscovite-bearing peraluminous
892	granites with Nb/Ta < 5 suspected to have experienced significant magmatic-hydrothermal
893	alteration were ruled out from the compilation. (b) Plots of Nb/Ta vs A/CNK for calculated
894	melts.
895	
896	FIGURE 7. Plots of Nb/Ta vs Zr concentration of rutile (LA-ICP-MS data) in (a) HT-HP
897	granulite sample 18MJ09 from Trans-North China Orogen and (b) UHT samples 16TG53 from
898	Khondalite Belt.
899	
900	FIGURE 8. Mineral proportion vs. metapelite temperature during isobaric heating at 0.8 GPa.
901	Mineral abbreviations: Qz: quartz; Sill: sillimanite; Ky: kyanite; Rt: rutile; Ilm: ilmenite; Bt:
902	biotite; Kfs: K-feldspar; Ms: muscovite; Pl: plagioclase; Grt: garnet.
903	
904	FIGURE 9. Calculated Nb/Ta ratios vs. temperature for petrological modeling. (a) Nb/Ta
905	evolution of extracted melts and residue during isobaric heating at 0.8 GPa. (b) Nb/Ta evolution
906	of extracted melts and residue along a 775 °C/GPa geotherm. (c) Nb/Ta evolution of extracted
907	melts and residues along a 550 °C/GPa geotherm.
908	

- 909 FIGURE 10. Nb/Ta frequency plots for typical pelitic granulites worldwide. Geochemical data
- 910 are the same as shown in Fig. 5.
- 911
- 912 FIGURE 11. Niobium and Ta concentrations in S-type granite, A2-type granite, leucosome in
- 913 UHT terrane (this study), and calculated melts in equilibrium with rutile. Granite data are from
- 914 Ballouard et al. (2020).

915	FIGURE S1. (a) $P-T$ mosaic pseudosection panels of a metapelite in an open system along
916	isobaric heating at 0.8 GPa (green arrow), contoured with isopleths of mol. % melt (black dash
917	line and number within square symbols). (b) Mineral proportion evolution vs. temperature.
918	
919	FIGURE S2. (a) $P-T$ mosaic pseudosection panels for a metapelite experiencing open-system
920	metamorphism along a 775 °C/GPa geotherm (green arrow), contoured with isopleths of mol. $\%$
921	melt. (b) Mineral proportion evolution vs. temperature.
922	
923	FIGURE S3. (a) <i>P</i> – <i>T</i> mosaic pseudosection panels for a metapelite experiencing open-system
924	metamorphism along a 550 °C/GPa geotherm (green arrow), contoured with isopleths of mol. %
925	melt. (b) Mineral proportion evolution vs. temperature.
926	
927	FIGURE S4. Tectonic subdivision of Sri Lanka metamorphic basement modified after Cooray
928	(1994). Samples of the studied UHT metamorphic rocks (18HC05, 18HC06, 18KC21, 18KC22-2,
929	18KC23, 18KC96-2) are shown as a red star. Am: amphibolite facies; Gn: granulite facies.
930	
931	FIGURE S5. Bulk-rock composition plots of Nb/Ta vs Ta contents of granulite samples from
932	the study area and other granulite terranes worldwide. Geochemical data summarized from:
933	Trivandrum Block, South India (Nandakumar and Harley, 2019); Ivrea Verbano Zone, north
934	Italy (Ewing et al., 2014); Ulashan-Daqingshan terrane, Khondalite Belt (Cai, 2014);
935	Bhopalpatnam granulite belt, central India (Vansutre et al., 2013); Napier complex, Antarctic
936	(Grew et al., 2006); Altay, northwestern China (Yang et al., 2015); and the Highland complex,

- 937 Sri Lanka (Table S3). Nb/Ta values of 12.4 in bulk continental crust (BCC) and 19.9 in chondrite
- are from Rudnick and Gao (2014) and Münker et al. (2003), respectively.

	D _{Nb}	D_Ta	References
Mineral	Mean	Mean	
Kfs	0.03	0.04	1-5
PI	0.03	0.04	2-3
Grt	0.02	0.04	2,6-8
Bt	3.89	1.53	1-3,9
Ms	3.5	0.4	10
Rt	159.5(51.1)	196(113)	11-12
llm	0.715	1.115	13
Zrn	0.46	0.64	2
Ар	0.012	0.012	2
Mnz	0.03	0.06	2

 Table 1 Partition coefficients used in trace element modeling

1: Nash and Crecraft (1985); 2: Acosta-Vigil et al. (2010); 3: Fedele et al. (2015); 4: Sweeney et al. (1995); 5: White et al. (2003); 6: Klemme et al. (2002); 7: Qian and Hermann (2013); 8: Fulmer et al. (2010); 9: Stepanov and Hermann (2013); 10: Raimbault and Burnol (1998); 11: Xiong et al. (2011); 12: Klemme et al. (2005); 13: Klemme et al. (2006). Note that D_{Nb} and D_{Ta} of rutile from Xiong et al. (2011)¹¹ were averaged values at 850 °C used for modeling of biotite dehydration melting, whereas those from Klemme (2005)¹² were used for modeling at anhydrous condition. Mineral abbreviations from Whitney and Evans (2010)









Figure 5





Figure 7











Figure 11

