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Tables Prepare each table in Microsoft Word or Excel with its proper (and informative) heading in place (see example below). Do not use font sizes smaller than 11 point.

Pyroxenes	Title
James. A. Author and Carol C. Coauthor	Author List
Department of Scientific Studies State University City, State 12345-6789, U.S.A.	Address
j.author@SU.edu; CC.Coauthor@SU.edu	Email addresses
ROCK-FORMING PYROXENES	1st order
Assemblages of natural pyroxenes [no period, no caps]	2nd order
Subsolidius assemblages. (An indented, italicized, bold heading is	the form for a third-

order heading and that the text should follow after it exactly as this text does.) 3rd order

Table 23. The composition [center heading,	
Constituent	Weight %
Milk	XX
Cocoa	уу
Sugar	ZZ

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Use Equation (yy) and (Eqn. yy) when referring to equations in the text.

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$$\sum t_1 z = 6.589[\{T_{1z} - O\} - 1.605]$$
(32)

Use the following style in both the equation <u>and</u> surrounding text:

NUMBERS, FUNCTIONS, and CONSTANTS are roman (not italic, not bold) SCALAR VARIABLES are *italicized* VECTORS and MATRICES are **bold faced** Abbreviations and symbols should be as below. There is a space between all numbers and units:

Common abbreviations (note Roman type)

cf. = compare e.g. = for example et al. = and others etc. = and other things i.e. = that is p. = page(s) vs. = versus

Length (l)

$$\begin{split} & m = meter(s) \\ & cm = centimeter(s) \ (1 \ cm = 10^{-2} \ m) \\ & mm = millimeter(s) \ (1 \ mm = 10^{-3} \ m) \\ & \mu m = micrometer(s) \ (1 \ \mu m = 10^{-6} \ m) \ NOT \ micron \ or \ \mu \\ & nm = nanometer(s) \ (1 \ nm = 10^{-9} \ m) \\ & Å = angstrom(s) \ (1 \ Å = 10^{-10} \ m) \\ & pm = picometer(s) \ (1 \ pm = 10^{-12} \ m) \end{split}$$

Mass (m)

g = gram(s) kg = kilogram(s) mg = milligram(s) µg = microgram(s)

Volume (V)

L = liter(s)mL = milliliter(s) cm³ = cubic centimeter(s) Å³ = cubic angstrom(s)

Pressure (P)

Pa = pascal(s) kPa = kilopascal(s) MPa = megapascal(s) GPa = gigapascal(s) bar or bars (no abbreviation) kbar = kilobar(s) atm = atmosphere(s)

Temperature (T)

K = Kelvin(s) °C = degrees Celsius [space before degree sign. c.f. angles where there is no space between digit and degree sign]

Energy:

J = joule(s) kJ = kilojoule(s) (preferred) cal = calorie(s) kcal = kilocalorie(s) **Density (D or \rho)** g/cm³ = gram(s) per cubic centimeter

Concentration (use square brackets [] around chemical symbol, e.g., [Ca⁻²]) % = percent [no space prior] % = per thousand [no space prior] wt% = weight percent mol% = mole percent vol% = volume percent M = molar concentration (mol/L) m = molal concentration (mol/kg) ppm = parts per million ppb = parts per billion

Time (t)

s = second(s) min = minute(s) h = hour(s) d = day(s) yr = year(s) Ma = million years Ga = billion years (ago or date) m.y. = million years b.y. = billion years (duration)

Frequency (f or v)

Hz = hertz kHz = kilohertz MHz = megahertz

Miller indices--note that letter symbols are italicized but numeric values are not, e.g., [hkl] but [100] hkl = diffraction symbol (hkl) = face symbol {hkl} = form symbol [hkl] = edge or zone symbol

Unit-cell measurements: *a*, *b*, *c* = edge lengths α , β , γ = angles **a**, **b**, **c** = vectors--note boldface

Optical measurements: dispersion (r, v), optic axial angle $(2V, 2V_x, 2V_z)$; principal vibration directions (E, O; X, Y, Z); refractive indices $(n; \varepsilon, \omega; \alpha, \beta, \gamma)$

REFERENCES

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Hawthorne FC, Oberti R (2007) Classification of the amphiboles. Rev Mineral Geochem 67:55-88

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Article published only on WWW:

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Book:

Tossell JA, Vaughn DJ (1992) Theoretical Geochemistry: Application of Quantum Mechanics in the Earth and Mineral Sciences. Oxford University Press, New York

Article in Book:

Keil K (1969) Meteorite composition. In: Handbook of Geochemistry. Vol 1. Wedepohl KH (ed) Springer, Berlin, p 78-115

Edited Book:

Schulze DG, Stuicki JW, Bertsch PM (eds) (1999) Synchroton X-ray Methods in Clay Science. The Clay Minerals Society, Boulder Colorado

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Dissertation/Thesis:

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Reference List

FIGURES

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