

2. Revision

A fast open data reduction workflow for the electron microprobe flank method to determine Fe³⁺/ΣFe contents in minerals

Dominik C. Hezel¹, Heidi E. Höfer¹, Andreas Fichtner²

¹Goethe-Universität Frankfurt, Altenhöferallee 1, 60438 Frankfurt am Main, Germany

dominik.hezel@em.uni-frankfurt.de

²ETH Zürich, Institut für Geophysik, Sonneggstrasse 5, 8092 Zürich, Schweiz

**corresponding author:* Dominik Hezel- dominik.hezel@em.uni-frankfurt.de

keywords: flank method, garnet, ferrous/ferric iron, data reduction, python

2. revision to *American Mineralogist*

November, 2023

Formatting:	Times New Roman, 12 pt, double spacing
Pages (main text):	13
Words (main text):	4019
Words (abstract):	204
Figures:	7
Tables:	0
Spelling:	British English

Abstract

32
33

34 Knowing the $\text{Fe}^{3+}/\Sigma\text{Fe}$ ratio in minerals is important to understand the formation and evolution
35 of minerals and their host rocks. Variable $\text{Fe}^{3+}/\Sigma\text{Fe}$ in *e.g.*, garnet is accompanied by a change
36 of their characteristic $\text{FeL}\alpha$ and $\text{FeL}\beta$ X-ray emission lines, which can be quantified with
37 electron microprobe measurements using the flank method. The required data reduction process
38 to determine the $\text{Fe}^{3+}/\Sigma\text{Fe}$ remained complex. Here we present a new data reduction tool taking
39 garnet data as an example. This new *Flank Reduction* app is a freely-available, graphical user
40 interface-driven, web-based application to reduce flank method data quickly and easily. The
41 entire data reduction process is achieved in minutes compared to hours or days, as it was before.
42 Flank Reduction provides comprehensive insights into *e.g.*, the employed flank method
43 standards, obtained data, or errors, through a wide array of controls and visualisation tools.
44 Documentation with comprehensive information on the flank method, data reduction, as well
45 as training material such as video tutorials or sample datasets are available on a dedicated
46 webpage. Flank Reduction emphasises the high value of FAIR (Findable, Accessible,
47 Interoperable, Reproducible) and open research software and demonstrates how current
48 developments in coding and app implementation can facilitate the development of powerful
49 and expandable research software.

50

1. Introduction

51

52

53 The iron oxidation state in minerals such as metamorphic garnet, pyroxenes or
54 amphiboles provides important information on the redox state of these minerals and their host
55 rocks at the time of last equilibration. The ambient oxygen fugacity prevailing in a rock when
56 a mineral equilibrates with its surroundings is recorded in its $\text{Fe}^{3+}/\Sigma\text{Fe}$ ratio (throughout this
57 work this ratio refers to a wt%-ratio). The *in-situ* determination of $\text{Fe}^{3+}/\Sigma\text{Fe}$ on a μm -scale in a
58 mineral by electronprobe microanalysis (EPMA) is possible with the flank method (Höfer and
59 Brey 2007). This method has been developed over the past decades in a series of papers by
60 Höfer et al. (1994), Höfer et al. (1996), Höfer et al. (2000), Höfer (2002), and brought to its
61 final state in Höfer and Brey (2007). The flank method has been applied to garnets (*e.g.*, Wang
62 et al. 2022; Tang et al. 2019, Gudelius et al. 2019, Tao et al. 2018, Malaspina et al. 2009, 2010),
63 sodic amphiboles (Enders et al. 2000), wuestite (Höfer et al. 2000; Longo et al. 2011), and
64 glasses (Zhang 2018). It can potentially be applied to other minerals of interest, such as
65 pyroxenes (Höfer et al. 2004).

66 In brief, the flank method works as follows: The wavelengths and intensities of the
67 characteristic $\text{FeL}\alpha$ and $\text{FeL}\beta$ X-ray emission lines of a mineral such as garnet depend on its
68 $\text{Fe}^{3+}/\Sigma\text{Fe}$ as well as its ΣFe . Figure 1 shows this dependency exemplified in spectra of an Fe^{2+} -
69 bearing garnet (almandine) and an Fe^{3+} -bearing garnet (andradite). The *x*-axis of Figure 1
70 represents the analyser crystal position of a JEOL EPMA spectrometer. The $\text{Fe}^{3+}/\Sigma\text{Fe}$ of a
71 mineral such as garnet is quantified by the count-rates obtained at two fixed analyser crystal
72 positions: one at the flank of the $\text{FeL}\alpha$ line, and one at the flank of the $\text{FeL}\beta$ line (vertical
73 dashed lines in Fig. 1). As the spectrum of a garnet changes with changing $\text{Fe}^{3+}/\Sigma\text{Fe}$ and ΣFe ,
74 so do the count-rates at these fixed analyser crystal positions. Hence, the count-rates at these
75 flanks of garnets correlate with their $\text{Fe}^{3+}/\Sigma\text{Fe}$ and ΣFe . The sensitivity of this correlation is at

76 its maximum, where the change of the count-rate is maximal. This is where the difference
77 between the count-rate of an Fe³⁺-poor and Fe³⁺-rich garnet is maximal (dashed lines in Fig. 1).
78 Out of convention, we are not using the difference in count-rates between the FeL α and FeL β
79 flank measurements, but their ratio FeL β /FeL α , which is subsequently abbreviated as L β /L α .

80 The measured FeL β /FeL α ratios and Fe concentrations (= Σ Fe) of several flank method
81 garnet standards with their known Fe³⁺/ Σ Fe are used to calculate a 2D linear regression (from
82 now on called ‘regression’). This regression provides a set of parameters to (i) calculate the
83 Fe³⁺/ Σ Fe in minerals with unknown Fe³⁺/ Σ Fe, as well as (ii) lines of equal Fe³⁺/ Σ Fe in a plot of
84 Σ Fe vs. L β /L α (Fig. 2) to visualise the measured data.

85 The flank method is particularly suitable when mineral grains are zoned or too small for
86 other methods such as Mössbauer spectroscopy, or XANES, both with with insufficient spatial
87 resolution, or when these methods are unavailable. An advantage of the flank method is the
88 simultaneous Fe³⁺/ Σ Fe and quantitative elemental analysis at each measured spot.

89 Applying the flank method requires two workflows:

90 (i) *Determining analyser crystal positions*: Höfer and Brey (2007) use the FeK α 9th order
91 line to determine the analyser crystal positions to measure the Fe L-line flanks, as this
92 line sits in between the FeL α and FeL β lines. This required specialised software to
93 determine the peak of the FeK α 9th order line, as well as a dedicated Excel spreadsheet.

94 (ii) *Data reduction to determine Fe³⁺/ Σ Fe*: Höfer and Brey (2007) describe two options for
95 data reduction, one quicker, but less precise, and a second – the regression mentioned
96 above – which is a more complex and precise procedure.

97 Both options require a complex combination of numerous Excel spreadsheets, and, for
98 the regression, the conversion of selected spreadsheets to text files which are then fed
99 to and processed with the Matlab clone Octave. This workflow required a trained expert
100 and can quickly take up to several days. Re-running it – *e.g.*, to test variations of certain

101 parameters or standards – is almost equally time-consuming. Finally, the procedure
102 requires proprietary software and is not openly accessible.

103 Both workflows remained largely unchanged over the years, even though additions have been
104 made to the spreadsheets. These patches made the entire process increasingly complex. Hence,
105 the data reduction process could not be performed without extensive hands-on training,
106 although the measurement itself could be reproduced fairly straightforwardly.

107 Here we present an entirely new data reduction tool we call *Flank Reduction* that uses
108 the precise regression method (eq. 2 in Höfer and Brey 2007). The entire data reduction is
109 processed within a simple-to-use web-application that has a graphical user interface (GUI) as
110 well as many new and additional functionalities. The entire data reduction workflow now takes
111 only minutes and can be re-run with changed parameters, as often as necessary within seconds.
112 We further reimagined a simple procedure for the positioning of the analyser crystals, for which
113 the web-application provides a dedicated helper tool. Finally, we embedded the web-
114 application in a data reduction platform.

115

116 **2. Methods**

117

118 **Coding**

119

120 The code for the new data reduction procedure and analyser crystal positioning tool is
121 written in Python. The GUI of the web application, including all interactive elements, styling,
122 *etc.* is implemented using the Streamlit (streamlit.io) programming framework and published
123 through the Streamlit community cloud. All diagrams are realised with the Bokeh and Plotly
124 libraries *i.e.*, all plots can be interactively controlled, downloaded as image files, and
125 informative features such as tooltips (context information when hovering over a data point)
126 have been added. The entire code, requirements, data files, *etc.* are hosted and published under

127 the MIT License, *i.e.*, openly accessible GU Frankfurt, Institut für Geowissenschaften GitHub
128 repository (GeoSci-FFM/flank-method). Finally, the Quarto web-publishing framework is used
129 to provide in-depth flank method background information, code and workflow documentation,
130 training material, video tutorials (hosted on YouTube), test datasets, a flank method section for
131 publications, literature, and more information surrounding the flank method and the entire data
132 reduction web-application. Everything can be accessed through the free and open
133 geoplatform.de website. We follow and support with this approach the call for FAIR (Findable,
134 Accessible, Interoperable, Reproducible) open-source publication (*e.g.*, Lehnert et al. 2021;
135 Klöcking et al. 2023; Wilkinson et al. 2016 and references therein), and more specifically for
136 FAIR4RS (for Research Software; Barker et al. 2022) publication.

137

138 Measurement program, standards & parametrisation

139

140 In the quantitative analysis measurement program, the Fe L-lines are measured on the
141 flanks of the FeL α and FeL β peaks as described above, and the Fe-concentration is measured
142 by EMPA, using conventional methods.. At least all major elements need to be measured for
143 an accurate matrix correction, and optional elements of interest – such as trace elements – may
144 be added. Each sample is measured with a grid of *e.g.*, 5x5 (*i.e.*, total of 25) spot analyses.
145 Details and examples on how to set-up the measurement program, naming conventions for the
146 samples and flank method standards (*i.e.*, standards with known Fe-concentration and Fe³⁺/ Σ Fe
147 ratio), structure, and requirements of the sample file, *etc.* can be found on the flank method
148 website.

149 It is required to measure at least four flank method standards at the beginning and end
150 of a measurement campaign. For a better precision, we recommend using more than the
151 required four flank method standards, and ideally with a substantial spread in Fe³⁺ and Σ Fe that
152 covers the expected range in Fe³⁺ and Σ Fe of the samples. The used flank method standard data

153 need to be part of the Flank Reduction standard file that is stored alongside the program on
154 GitHub. Alternatively, a self-produced file with additional standard compositions can be
155 uploaded. The standard file, its entries and structure are detailed on the Flank Reduction
156 website.

157 Mathematically, the flank method rests on the assumption that the Fe^{3+} concentration
158 can be modelled as

$$159 \quad Fe^{3+} = -A - B \frac{L\beta}{L\alpha} + (1 - C)\Sigma Fe + D \frac{L\beta}{L\alpha} \Sigma Fe \quad (Eq. 1),$$

160 which is identical to Eq. 2 in Höfer and Brey (2007), but rearranged to have Fe^{3+} on the left-
161 hand side. In the interest of a more condensed notation we rewrite Eq. 1 as

$$162 \quad z = a + bx + cy + dxy \quad (Eq. 2),$$

163 where we define $z = Fe^{3+}$, $a = -A$, $b = -B$, $c = 1 - C$, $d = D$, $x = \frac{L\beta}{L\alpha}$ and $y = \Sigma Fe$. To estimate
164 the four coefficients a , b , c , and d (or, equivalently A , B , C , and D), we rely on $N \geq 4$
165 measurements of x , y , and z from flank method standards, which we denote by x_i , y_i , z_i ,
166 respectively. The subscript $i = 1, \dots, N$ is the running index of the standard samples. Due to
167 measurement errors and unmodelled effects in Eq. 1, it is neither possible nor desirable to find
168 coefficients a , b , c , and d such that the observations z_i are explained exactly. Instead, we try to
169 minimise the least-squares misfit

$$170 \quad \chi = \frac{1}{2} \sum_{i=1}^N \frac{1}{\sigma_i^2} (a + bx_i + cy_i + dx_i y_i - z_i)^2 \quad (Eq. 3),$$

171 where the σ_i denote the standard deviations of the measurement errors in Fe^{3+} of the individual
172 standards (*i.e.*, the standard deviation of z_i). Forcing the partial derivatives of Eq. 3 with respect
173 to the coefficients a , b , c , and d to 0 leads to a system of four linear equations,

$$174 \quad \sum_{i=1}^N \frac{z_i}{\sigma_i^2} = a \sum_{i=1}^N \frac{1}{\sigma_i^2} + b \sum_{i=1}^N \frac{x_i}{\sigma_i^2} + c \sum_{i=1}^N \frac{y_i}{\sigma_i^2} + d \sum_{i=1}^N \frac{x_i y_i}{\sigma_i^2} \quad (Eq. 4.1),$$

$$175 \quad \sum_{i=1}^N \frac{x_i z_i}{\sigma_i^2} = a \sum_{i=1}^N \frac{x_i}{\sigma_i^2} + b \sum_{i=1}^N \frac{x_i^2}{\sigma_i^2} + c \sum_{i=1}^N \frac{x_i y_i}{\sigma_i^2} + d \sum_{i=1}^N \frac{x_i^2 y_i}{\sigma_i^2} \quad (Eq. 4.2),$$

$$176 \quad \sum_{i=1}^N \frac{y_i z_i}{\sigma_i^2} = a \sum_{i=1}^N \frac{y_i}{\sigma_i^2} + b \sum_{i=1}^N \frac{x_i y_i}{\sigma_i^2} + c \sum_{i=1}^N \frac{y_i^2}{\sigma_i^2} + d \sum_{i=1}^N \frac{x_i y_i^2}{\sigma_i^2} \quad (Eq. 4.3),$$

177
$$\sum_{i=1}^N \frac{x_i y_i z_i}{\sigma_i^2} = a \sum_{i=1}^N \frac{x_i y_i}{\sigma_i^2} + b \sum_{i=1}^N \frac{x_i^2 y_i}{\sigma_i^2} + c \sum_{i=1}^N \frac{x_i y_i^2}{\sigma_i^2} + d \sum_{i=1}^N \frac{x_i^2 y_i^2}{\sigma_i^2} \text{ (Eq. 4.4).}$$

178 The sums in Eqs. 4.1 - 4.4 are the knowns of the linear system and can be computed directly
179 from the flank method standard measurements and their errors. Provided that the standard
180 measurements are independent, the linear system of equations can be solved for a , b , c , and d .
181 This is implemented using standard functions for linear system solutions with widely used
182 Python packages.

183 A solution for a , b , c , and d is acceptable, when the differences between modelled and
184 measured Fe^{3+} concentrations ($a + bx_i + cy_i + dx_i y_i - z_i$) are on average similar to the
185 measurement standard deviations (σ_i), *i.e.*, when according to eq. 3 we have $\chi \approx \frac{N}{2}$. For $\chi \gg \frac{N}{2}$,
186 the error standard deviations are either under-estimated or additional effects need to be
187 considered. Conversely, $\chi \ll \frac{N}{2}$ typically signals that measurement errors have been over-
188 estimated.

189 The $\text{Fe}^{3+}/\Sigma\text{Fe}$ ratio of the used flank method standards have been determined using
190 Mössbauer spectroscopy and have similar standard deviations of about ± 0.01 (1σ). We
191 therefore simplify the four eq. 4.1 to 4.4. by assuming that the standard deviations of all standard
192 samples are identical. The standard deviations – *i.e.*, all σ_i^2 – can then be removed from eq. 4.1
193 to 4.4. For the same reason, we are not calculating χ to test whether the determined Fe^{3+} values
194 are plausible. Instead, we determine the $\text{Fe}^{3+}/\Sigma\text{Fe}$ ratios of at least four flank method standards
195 and compare these to the reported $\text{Fe}^{3+}/\Sigma\text{Fe}$ ratios determined by Mössbauer spectroscopy. In
196 Höfer et al. (2017), it has been shown that the uncertainties in $\text{Fe}^{3+}/\Sigma\text{Fe}$ determination are
197 similar for both methods.

198
199
200

3. Results

201 Figure 3 shows the schematic workflow of the flank method. Steps 2 and 3 remain
202 largely unchanged to how flank method data have been reduced before, except for a few
203 adjustments, of which the most important are explained below. Step 1 and in particular the
204 entire step 4 of the data reduction process are new and described in the following two sections.

205

206 Determining the analyser crystal positions on the flanks of FeL α and FeL β

207

208 Analyser crystal positions need to be determined at the beginning of a measurement
209 campaign. Figure 1 shows FeL α and FeL β spectra from an almandine (Fe²⁺-rich) and an
210 andradite (Fe³⁺-rich) together with their difference spectrum. The collection of such spectra are
211 time-consuming and unnecessary, and it is sufficient to measure small intervals across the
212 minima and maxima positions of the difference spectrum (Fig. 4), which then usually takes less
213 than 10 min. Such 'interval spectra' can be uploaded to a helper tool in Flank Reduction to
214 identify the analyser crystal positions. The requirements for the file structure of the interval
215 spectra are documented on the Flank Reduction website.

216 The analyser crystals need to be set at the minima and maxima of the difference
217 spectrum (vertical, dashed lines in Fig. 1) to achieve a maximum dispersion of the lines of equal
218 Fe³⁺/ Σ Fe on the Fe³⁺ Results Plot (Fig. 2). A large dispersion directly translates to a better
219 resolution and accuracy of the determined Fe³⁺/ Σ Fe. Therefore, the L β /L α of a standard with
220 high Fe²⁺ and Σ Fe (*e.g.*, almandine) is useful as a quality check for the chosen positions as well
221 as a long-term monitor for L β /L α .

222

223 The flank data reduction procedure

224

225 The following describes the general workflow for the Flank Reduction program to
226 illustrate how it works and highlights its core aspects such as ease of use, versatility, data

227 inspection capabilities, expandability, and speed. Further documentations are detailed on the
228 Flank Reduction website.

229 Flank Reduction requires comma-separated value (csv) files in the UTF-8 format,
230 which is an export format in almost all spreadsheet or similar programs. The spreadsheet with
231 the measurements requires columns with the sample name, measured FeO-concentrations, and
232 the flank FeL α and FeL β intensities, as well as a copy of the sample name column, which is
233 renamed to 'Inspected'. The benefit of this additional column is explained towards the end of
234 section 3. Any additional measured element concentration column will be part of the results
235 output file, while any other additional column is ignored. This spreadsheet is uploaded *via* the
236 'upload' button on the 'Data Upload & Reduction' page of Flank Reduction (Fig. 5) and can be
237 displayed by expanding the box below the upload button. All tables of Flank Reduction are
238 interactive, *e.g.*, sortable by clicking on a header, or searchable by using the cmd+F or ctrl+F
239 keyboard shortcuts. Information on how to use each sub-page are provided in info-boxes on the
240 sidebar of each sub-page.

241 If not selected otherwise, the first four standards from the uploaded spreadsheet are
242 automatically selected and used to calculate the regression parameters according to eq. 1. As
243 stated above, 25 spots are typically measured on a standard or sample. The averages of these
244 25 analyses of each sample are automatically calculated, and the Fe³⁺/ Σ Fe ratios are calculated
245 using the regression parameters. Similarly, the averages of the 25 analyses for all other elements
246 in each sample are calculated. The results are immediately ready as a downloadable csv output
247 file on the 'Output' page on the app webpage. This data reduction is done within seconds,
248 compared to hours with the previous Excel sheets.

249

250 Visualisation and inspection of the results

251

252 The individual analyses of the uploaded data should be checked for outliers or other
253 anomalies before calculating the results as described above. The ‘Data Inspect’ section is
254 dedicated to inspecting the measurement results – drift during the measurements, the result
255 from the regression – the results from the $\text{Fe}^{3+}/\Sigma\text{Fe}$ calculation, as well as to perform several
256 quality checks. The Fe^{3+} Results Plot as shown in Figure 2 serves as the first quality check for
257 the regression and Fe^{3+} determination. It immediately shows whether the standards chosen for
258 the regression are suitable for the unknown samples. The plot should principally look as
259 displayed, and with an ideally large dispersion of the lines of equal $\text{Fe}^{3+}/\Sigma\text{Fe}$. The flank method
260 standards used for the regression can be selected and the selection instantly varied to identify
261 the optimal regression. This means, the regression and dispersion of the lines of equal $\text{Fe}^{3+}/\Sigma\text{Fe}$
262 in the Fe^{3+} Results Plot depend on the selected set of flank method standards. Hence, this
263 procedure is used to select the set of flank method standards that best represent the $\text{Fe}^{3+}/\Sigma\text{Fe}$ of
264 the chosen flank method standards. To illustrate how an additional flank method standard
265 changes the regression, Figure 2b shows the extreme and unreasonable case – given the set of
266 Fe^{3+} -poor samples in this example – with an additional andradite standard. This changes the
267 positions of the lines of equal $\text{Fe}^{3+}/\Sigma\text{Fe}$ in the region of the samples by about 0.02. The
268 comparison of Figure 2a and b illustrates the importance of using standards with similar
269 $\text{Fe}^{3+}/\Sigma\text{Fe}$ and ΣFe as the samples to produce accurate results.

270 The calculated $\text{Fe}^{3+}/\Sigma\text{Fe}$ can be displayed against various parameters such as element
271 concentration or beam current to check whether these changes depend on the selected
272 parameter, or the selected standards. This provides profound insights into the measurements,
273 calculated results, or the suitability of the used flank method standards.

274 As mentioned, each sample is often analysed with a grid of a total of 25 points. Flank
275 Reduction automatically calculates the average of this number of points, independent from how
276 many analyses per sample have been acquired. To reduce the measuring time, the optimum

277 number of analyses can be found by testing how many analyses per sample are sufficient to still
278 obtain an acceptable uncertainty of the results. To do this, only few samples – maybe 10 or 20
279 – are measured with 25 or even more analyses per sample. It can then be selected (cf. Fig. 5)
280 how many randomly picked analyses out of these 25 will be used to calculate the $\text{Fe}^{3+}/\Sigma\text{Fe}$ of
281 each sample. Figure 6 displays the result of such a calculation using an existing measurement
282 series of 63 samples that has been measured with 25 analyses per sample. The shaded areas
283 represent the 0.02 uncertainty of the method. As can be seen, 9-16 analyses per sample are
284 already sufficient to obtain the same result as with 25 analyses per sample. Such an insight can
285 significantly reduce the measurement time, *i.e.*, significantly increase the number of analyses
286 per time. This tool therefore helps identify an optimum relationship between measurement time
287 and number of analyses per sample. Flank Reduction allows to produce a plot such as Figure 6
288 within minutes from the dataset of a single measurement campaign.

289 In cases, some analyses obtained from a sample are poor, *e.g.*, have largely deviating
290 element concentrations or totals that are either too low or too high.. This might happen when a
291 sample is small, poorly polished or carbon-coated, heterogeneous or zoned on a small scale, or
292 if the grain size is too small for a grid of points and requires placing the individual analyses
293 points manually on different grains. The inspection tool allows identifying poor analyses by
294 displaying all analyses of each sample in individual plots either for one element (Fig. 7) or plots
295 of all elements from an individual sample. A colour coded box for each sample with the average
296 value and its standard deviation provides quick information whether a sample might contain
297 poor analyses. Each identified poor analysis can be indicated as such in the 'Inspected' column
298 of a measurement spreadsheet. Simply adding an *e.g.*, 'x' before the sample name will
299 essentially make this a new sample for the program. If the sample name is alternatively replaced
300 by 'ignore', the entire analysis will be ignored. After all poor analyses have been identified and
301 marked in the 'Inspected' column of the measurement spreadsheet, this needs to be uploaded
302 again. It is then possible to select the 'Inspected' column as the sample name column. If names

303 in the 'Inspected' column were marked with *e.g.*, an 'x', and say three analyses were marked
304 like this, then these three will be treated as a separate sample, *i.e.*, an $\text{Fe}^{3+}/\Sigma\text{Fe}$ will be calculated
305 for this, and it will be displayed in its own plot. This has the advantage that the $\text{Fe}^{3+}/\Sigma\text{Fe}$ of the
306 poor analyses can directly be compared to the analyses of the remaining good analyses. This
307 might be informative to better understand the sample, *e.g.*, if individual spots were placed
308 across a zonation, crack, or else. If an analysis name was replaced by 'ignore', it will not show
309 up in the results or in a plot.

310 Each sample is bracketed by drift monitor measurements (*e.g.*, 5) to detect a possible
311 change in $L\beta/L\alpha$, *e.g.*, due to temperature variations in the lab or a spectrometer drift during
312 long measurement campaigns. An ideal drift monitor standard is high in Fe^{2+} (*e.g.*, almandine),
313 which corresponds to a high $L\beta/L\alpha$. The stability of the $L\beta/L\alpha$ over the measurement
314 campaign, as well as any other element or measured value can be inspected in 'Results
315 Inspection' and 'Drift Monitor'. Unstable values of $L\beta/L\alpha$ directly translate into a variation in
316 $\text{Fe}^{3+}/\Sigma\text{Fe}$. The drift monitor analyses might be used to correct the sample measurements
317 accordingly.

318

319

4. Discussion

320

321 Flank Reduction allows for a fast, simple and GUI driven data reduction to determine
322 $\text{Fe}^{3+}/\Sigma\text{Fe}$ in minerals such as garnet using microprobe data, and in addition provides valuable
323 and relevant new insights. The Fe^{3+} Results Plot allows for a quick test which flank method
324 standards are the most suitable for the studied samples. For example, if five standards were
325 measured, four can be used to calculate the regression parameters, while the fifth is treated as
326 unknown. After each of the five standards was treated as an unknown, the deviation of their
327 determined $\text{Fe}^{3+}/\Sigma\text{Fe}$ from their real values provide clues towards which standard might skew

328 the results towards incorrect values. Further, systematic shifts on the dispersion of the lines of
329 equal $\text{Fe}^{3+}/\Sigma\text{Fe}$ (cf. Fig. 2) can be identified when standards and/or samples with largely
330 different $\text{Fe}^{3+}/\Sigma\text{Fe}$ are studied. The Fe^{3+} Results Plot then helps to determine how to possibly
331 split such measurements into sub-campaigns, each of which measuring only standard-sample
332 combinations that have similar ΣFe and $\text{Fe}^{3+}/\Sigma\text{Fe}$. Finally, the Fe^{3+} Results Plot illustrates the
333 increasing precision of $\text{Fe}^{3+}/\Sigma\text{Fe}$ with increasing ΣFe , as the dispersion of the lines of equal
334 $\text{Fe}^{3+}/\Sigma\text{Fe}$ increase with increasing ΣFe .

335 The precisions of the determined $\text{Fe}^{3+}/\Sigma\text{Fe}$ depend on the number of analyses per
336 sample. Flank Reduction allows to find the minimum required number of analyses per sample
337 without compromising on analytical precision (cf. Fig. 6). Choosing the right number of
338 analyses per sample can reduce the measurement time significantly, thereby allowing to either
339 save time and cost, or measure additional or repeated points, which will increase the reliability
340 of the results.

341 The wide range of optional sample as well as result visualisations provide insights to
342 analysis variations or trends by correlating elements, $L\beta/L\alpha$, probe current, $\text{Fe}^{3+}/\Sigma\text{Fe}$, and more.
343 These tools are informative to identify and exclude outliers, instantly test the influence of such
344 exclusions in re-runs, or understand the homogeneity of the studied samples. This straight-
345 forward workflow of selecting the best analyses should allow for more robust results.

346 An exciting option of Flank Reduction is its expandability to determine the $\text{Fe}^{3+}/\Sigma\text{Fe}$ in
347 additional minerals *e.g.*, pyroxene or amphibole. This is straight-forward, but requires a set of
348 well-defined flank method standards with known values of $\text{Fe}^{3+}/\Sigma\text{Fe}$. These are then used in the
349 same manner as *e.g.*, garnet standards. All tools of Flank Reduction can be used to test the
350 validity of the standards in the same way as described above. Hence, Flank Reduction is not a
351 dedicated app to determine $\text{Fe}^{3+}/\Sigma\text{Fe}$ in garnets, but universally applicable to other minerals
352 after the flank method itself has been proven to be applicable to these minerals.

353 Finally, other visualisations, new functionalities, or simply suggestions for an enhanced
354 workflow can be added to Flank Reduction. In fact, the program is constantly being developed
355 and tweaked.

356

357

5. Implications

358

359 Studying the $\text{Fe}^{3+}/\Sigma\text{Fe}$ in minerals such as garnets provide important information about the
360 formation of these minerals and their host rocks. Determining their $\text{Fe}^{3+}/\Sigma\text{Fe}$, however, was so
361 far often hampered by either the availability of a suitable technique (Mössbauer, XANES) or
362 required a complex data reduction process when using the EPMA flank method. The
363 implementation of Flank Reduction as an open online tool written in a widespread open coding
364 language makes the so far challenging to use data reduction process that was based on a
365 proprietary program, intuitive, quick, simple, understandable, and easily expandable. This
366 emphasises the potential and importance of research software, as well as the need for open and
367 FAIR4RS publication. Clearly defining file formats and thereby following international
368 conventions allows for synergies across software, for example, the measurement template as
369 well as required Flank Reduction spreadsheet structure will both soon become part of
370 ProbeSoftware¹.

371

372 The comprehensive documentation is embedded in a platform for modular or even
373 distributed software. Additional research software is easily added, without requirements for
374 specific formats. The documentation of such research software could be readily added to the
platform as markdown files.

¹<https://probesoftware.com>

375

Acknowledgements

376

377

We are grateful to two anonymous reviewers for their support in making this a better

378

publication. We similarly thank the AE to improve this paper with his thoroughly review.

References

- 379
380
381 Barker M., Hong N. P. C., Katz D. S., Lamprecht A.-L., Martinez-Ortiz C., Psomopoulos F.,
382 Harrow J., Castro L. J., Gruenpeter M., Martinez P. A. and Honeyman T. (2022)
383 Introducing the FAIR Principles for research software. *Scientific Data* 9, 622.
- 384 Enders, M., Speer, D., Maresch, W.V., and McCammon, C.A. (2000) Ferric/ferrous iron ratios
385 in sodic amphiboles: Mössbauer analysis, stoichiometry-based model calculations and the
386 high-resolution microanalytical flank method. *Contributions to Mineralogy and*
387 *Petrology*, 140, 135-147.
- 388 Gudelius D, Aulbach S, Braga R, Höfer H, Woodland AB, Gerdes A (2019): Element transfer
389 and redox conditions in continental subduction zones: New insights from peridotites of
390 the Ulten Zone (N Italy). *Journal of Petrology* 60: 231-268
- 391 Höfer H. E. (2002) Quantification of Fe²⁺/Fe³⁺ by Electron Microprobe Analysis – New
392 Developments. *Hyperfine Interactions* 144–145, 239–248.
- 393 Höfer H. E. and Brey G. P. (2007) The iron oxidation state of garnet by electron microprobe:
394 Its determination with the flank method combined with major-element analysis.
395 *American Mineralogist* 92, 873–885.
- 396 Höfer, H. E., Brey, G. P., Schulz-Dobrick, B., and Oberhänsli, R. (1994) The determination of
397 the oxidation state of iron by the electron microprobe. *European Journal of Mineralogy*,
398 6, 407-418.
- 399 Höfer, H. E., Brey, G. P., and Oberhänsli, R. (1996) The determination of the oxidation state
400 of iron in synthetic garnets by X-ray spectroscopy with the electron microprobe. *Physics*
401 *and Chemistry of Minerals*, 23, 241.
- 402 Höfer H. E., Weinbruch S., McCammon C. A. and Brey G. P. (2000) Comparison of two
403 electron probe microanalysis techniques to determine ferric iron in synthetic wüstite
404 samples. *European Journal of Mineralogy* 12, 63–71.

- 405 Höfer H. E., Brey, G. P., and Hibberson, W. O. (2004) Iron oxidation state determination in
406 synthetic pyroxenes by electron microprobe. *Lithos*, 73, 551.
- 407 Höfer, H.E., Brey, G.P., Shu, Q., Heckel, C., & Vasilyev, P. (2017). Redox state of Archean
408 kyanite/corundum eclogites and garnet pyroxenites from Bellsbank, South Africa.
409 International Kimberlite Conference: Extended Abstracts, 11.
- 410 Klöcking M, Wyborn L, Lehnert KA, Ware B, Prent AM, Profeta L, Kohlmann F, Noble W,
411 Bruno I, Lambart S, Ananuer H, Barber ND, Becker H, Brodbeck M, Deng H, Deng K,
412 Elger K, Franco GdS, Gao Y, Ghasera KM, Hezel DC, Huang J, Kerswell B, Koch H,
413 Lanati AW, Maat Gt, Martínez-Villegas N, Yobo LN, Redaa A, Schäfer W, Swing MR,
414 Taylor RJM, Traun MK, Whelan J, Zhou T (2023) Community recommendations for
415 geochemical data, services and analytical capabilities in the 21st century. *Geochimica et*
416 *Cosmochimica Acta* (in press) [doi: 10.1016/j.gca.2023.04.024]
- 417 Lehnert, K., Wyborn, L., Bennett, V.C., Hezel, D.C., McInnes, B.I.A., Plank, T., Rubin, K.,
418 2021. Onegeochemistry: Towards an interoperable global network of fair geochemical
419 data URL: <https://zenodo.org/record/5767950>
- 420 Longo, M., McCammon, C.A., Jacobsen, S.D. (2011): Microanalysis of the iron oxidation state
421 in (Mg,Fe)O and application to the study of microscale processes. *Contributions to*
422 *Mineralogy and Petrology*, 162, 1249-1257.
- 423 Malaspina, N., Poli, S., Fumagalli, P. (2009): The Oxidation State of Metasomatized Mantle
424 Wedge: Insights from C-O-H-bearing Garnet Peridotite. *Journal of Petrology* 50, 1533-
425 1552.
- 426 Malaspina, N., Scambelluri, M., Poli, S., Van Roermund, H.L.M., Langenhorst, F. (2010): The
427 oxidation state of mantle wedge majoritic garnet websterites metasomatized by C-bearing
428 subduction fluids. *Earth and Planetary Science Letters* 298, 417-426.

- 429 McCammon, C.A., Chaskar, V., and Richards, G.G. (1991) A technique for spatially resolved
430 Mössbauer spectroscopy applied to quenched metallurgical slags. *Measurement Science*
431 *and Technology*, 2, 657-662.
- 432 Tang M, Lee C-TA, Costin G, Höfer HE (2019): Recycling reduced iron at the base of
433 magmatic orogens. *Earth and Planetary Science Letters* 528, 115827.
- 434 Tao R., Fei Y., Bullock E.S., Xu C., Zhang L., (2018): Experimental investigation of Fe³⁺-rich
435 majoritic garnet and its effect on majorite geobarometer. *Geochimica et Cosmochimica*
436 *Acta* 225, 1-16
- 437 Tao R., Zhang L., Tian M., Zhu J., Liu X., Liu J., Höfer H.E., Stagno V., Fei Y. (2018):
438 Formation of abiotic hydrocarbon from reduction of carbonate in subduction zones:
439 Constraints from petrological observation and experimental simulation. *Geochimica et*
440 *Cosmochimica Acta* 239, 390-408
- 441 Wang C, Tao R, Walters JB, Höfer HE, Zhang L (2022): Favorable P–T–fO₂ conditions for
442 abiotic CH₄ production in subducted oceanic crusts: A comparison between CH₄-bearing
443 ultrahigh- and CO₂-bearing high-pressure eclogite. *Geochimica et Cosmochimica Acta*
444 336, 269-290.
- 445 Wilkinson, M.D., Dumontier, M., Aalbersberg, I.J., Appleton, G., Axton, M., Baak, A.,
446 Blomberg, N., Boiten, J.W., da Silva Santos, L.B., Bourne, P.E., Bouwman, J., Brookes,
447 A.J., Clark, T., Crosas, M., Dillo, I., Dumon, O., Edmunds, S., Evelo, C.T., Finkers, R.,
448 Gonzalez-Beltran, A., Gray, A.J., Groth, P., Goble, C., Grethe, J.S., Heringa, J., 't Hoen,
449 P.A., Hooft, R., Kuhn, T., Kok, R., Kok, J., Lusher, S.J., Martone, M.E., Mons, A.,
450 Packer, A.L., Persson, B., Rocca-Serra, P., Roos, M., van Schaik, R., Sansone, S.A.,
451 Schultes, E., Sengstag, T., Slater, T., Strawn, G., Swertz, M.A., Thompson, M., van der
452 Lei, J., van Mulligen, E., Velterop, J., Waagmeester, A., Wittenburg, P., Wolstencroft,
453 K., Zhao, J., Mons, B., 2016. The FAIR guiding principles for scientific data management
454 and stewardship. *Scientific Data* 3.

455 Zhang C., Almeev R.R., Hughes E.C., Borisov A. A., Wolff E.P., Höfer H.E., Botcharnikov R.
456 and Koepke J (2018): Electron microprobe technique for the determination of iron
457 oxidation state in silicate glasses. American Mineralogist 103, 1445-1454.
458

507

Figure captions

508

509 Fig. 1: EPMA FeL α and FeL β -spectra from an almandine (Fe³⁺-poor) and an andradite
510 (Fe³⁺-rich). For the orange line –the ‘difference spectrum’ –, the almandine spectrum was
511 subtracted from the andradite spectrum. The vertical, dashed lines indicate optimal analyser
512 crystal positions at the minima/maxima of the difference spectrum.

513 Fig. 2: a) Fe³⁺ Results Plot calculated from 4 flank method standards (large symbols)
514 and with an example set of 63 samples. Lines represent equal Fe³⁺/ Σ Fe. L β /L α ratios were
515 obtained redundantly from two spectrometers, hence, the plots contain a regression for each
516 spectrometer (blue and orange, respectively). b) Same as a), but the regressions were calculated
517 from 5 flank method standards, including one andradite.

518 Fig. 3: Schematic flow chart for the flank method workflow. Steps 1 & 4 are completely
519 redesigned and the core of this paper. In particular, the data reduction step 4 is now an open,
520 online tool and allows the fast, GUI driven calculation of Fe³⁺/ Σ Fe.

521 Fig. 4: The same as Figure 1, but demonstrating that small interval spectra are fully
522 sufficient to determine the optimum analyser crystal positions.

523 Fig. 5: The GUI to upload data and control how the Fe³⁺/ Σ Fe calculation is performed.

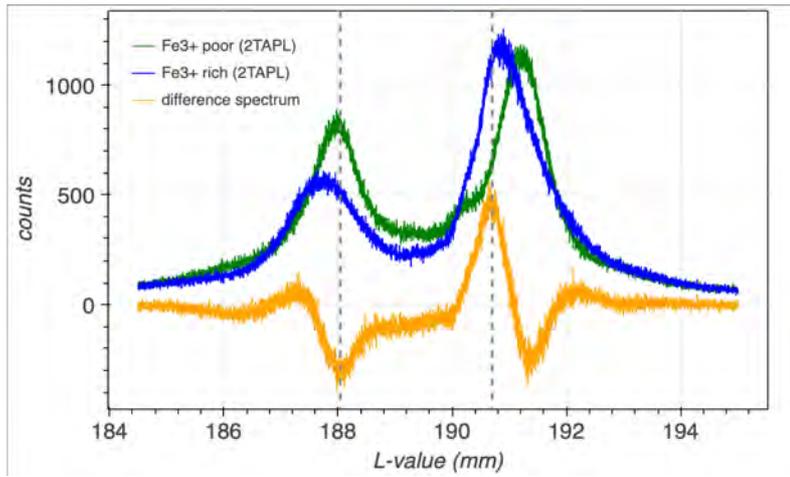
524 Fig. 6: The Fe³⁺/ Σ Fe of each sample is calculated from the average of 25 analyses. From
525 these 25 analyses, smaller numbers have been randomly chosen – these are indicated in the
526 legend –, and from these the Fe³⁺/ Σ Fe of each sample was recalculated. The shadowed area
527 represents a constant 0.02 error, which is a typical error of this method. a) and b) are the same
528 plots, with b) showing lesser calculated results.

529 Fig. 7: The quality of all analyses on each sample can be inspected on individual plots
530 with colour coded indications for the magnitude of the standard deviation. The x-axis are the
531 point numbers.

459

Figures

460



461

Fig. 1

462

463

464

465

466

467

468

469

470

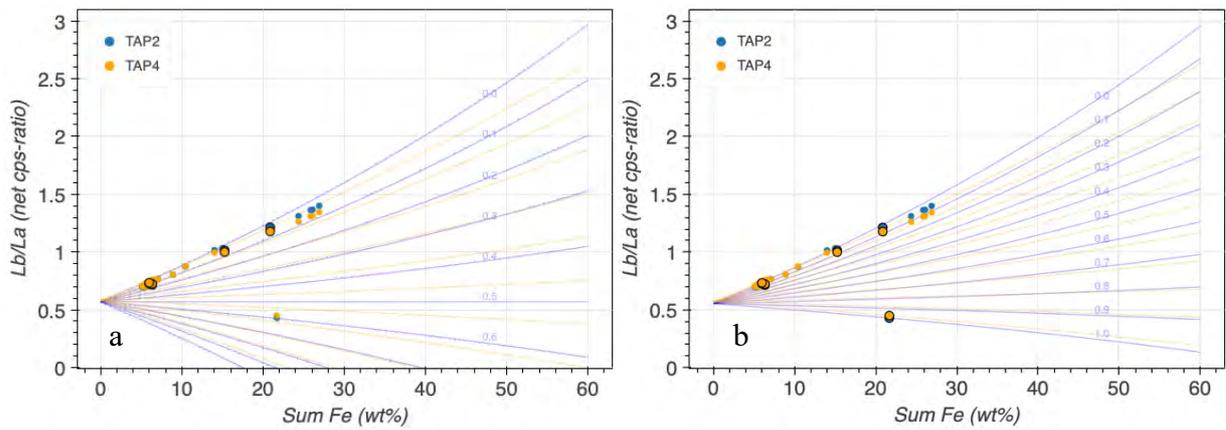
471

472

473

474

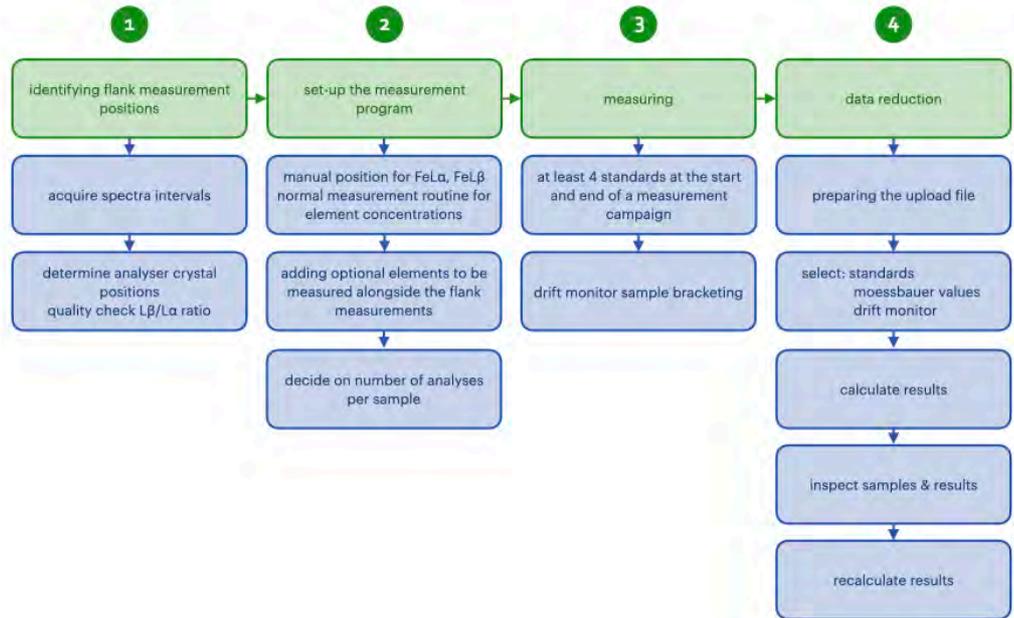
475



476

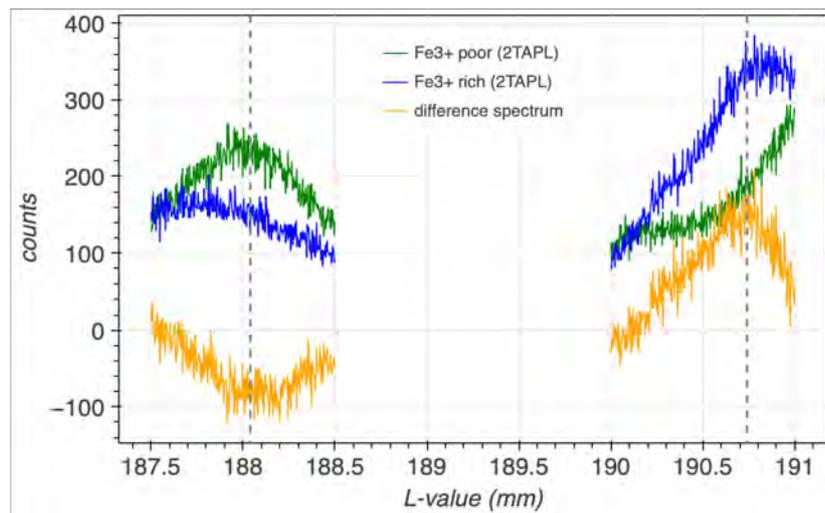
477

Fig. 2



478
479 Fig. 3

480
481
482



483
484 Fig. 4

485

Drag and drop file here
Limit 200MB per file

Browse files

test data with and.csv 2.1MB

Select whether to see the uploaded file with or without 'ignore' in rows
 see without 'ignore' see with 'ignore'

You uploaded the following data for flank reduction

Choose whether all data or only inspected data will be used
 All Inspected

Select standards used to calculate the Fit Parameters
AlmO_3R80_p3 x UA5_STDH1_p3 x UA10_STDH1_p3 x Damknolle_STD... x

Select the standard used as drift monitor.
AlmO_OX3

Select the Moessbauer values to be used.
Fe3+/SumFe

Number of random samples used for calculations (0 = all)
0

Calculate Results

Flank data successfully reduced!

486

487

488 Fig. 5

489

490

491

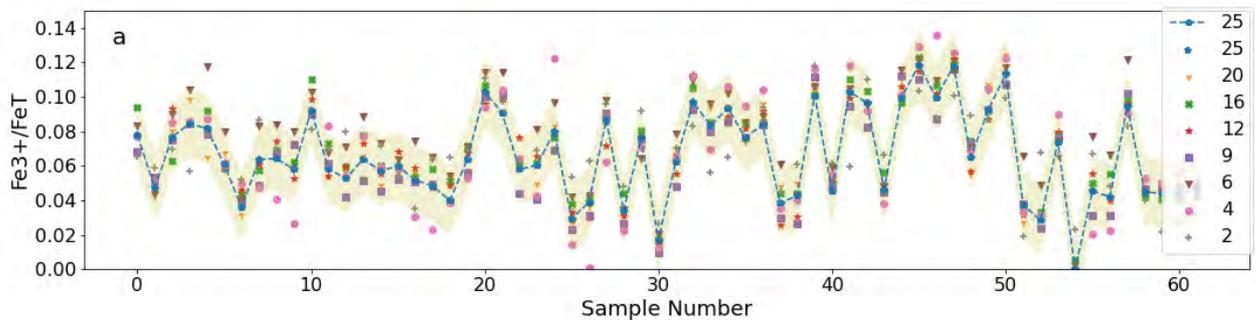
492

493

494

495

496



497

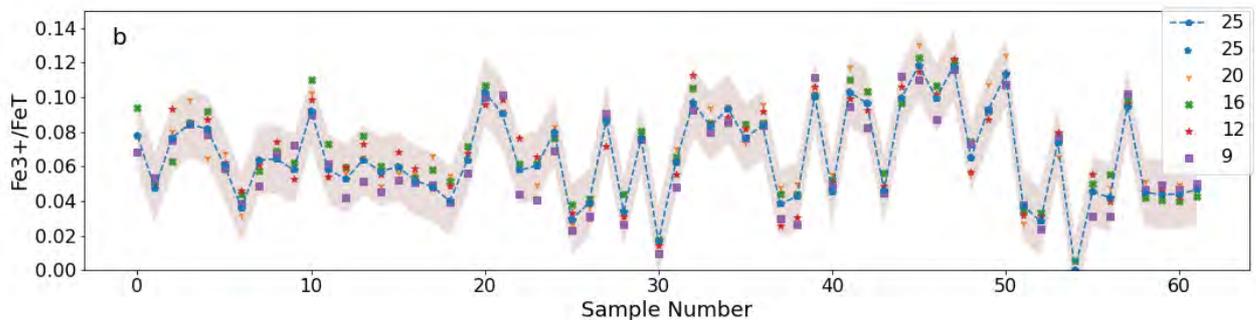
498

499

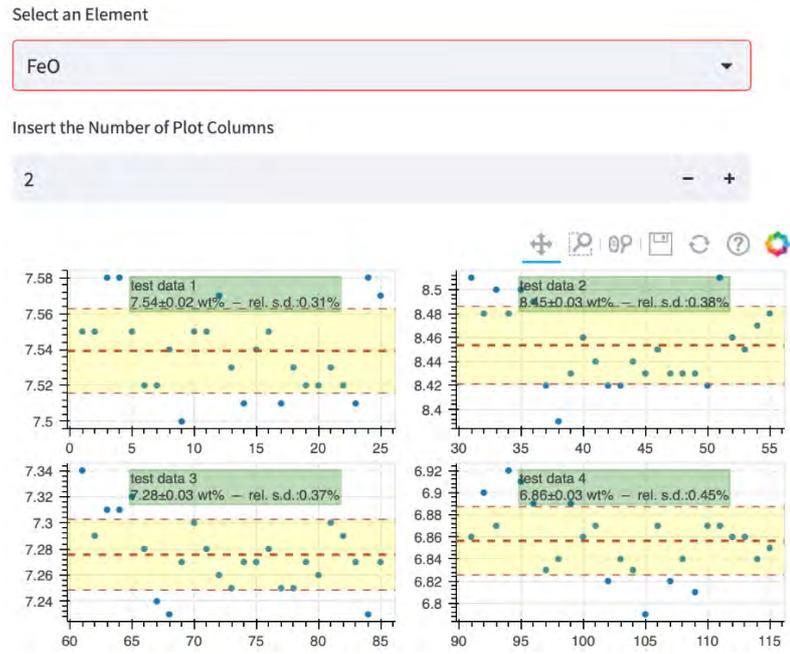
500

501

502



503 Fig. 6



504

505

506 Fig. 7