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3 **The effect of elemental diffusion on the application of olivine-composition-based**
4 **magmatic thermometry, oxybarometry, and hygrometry: A case study of olivine**
5 **phenocrysts from the Jiagedaqi basalts, northeast China**

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22 **Abstract**

23 Olivine compositions are widely used to constrain magmatic thermodynamic
24 conditions such as magmatic temperature, oxygen fugacity, and H₂O content.
25 However, elemental diffusion may change the initial compositions and lead to large
26 uncertainty on the estimation of these thermodynamic conditions. In this study, we
27 conducted LA-ICP-MS elemental mapping and EPMA analysis of olivine
28 phenocrysts and olivine-hosted spinel from the Jiagedaqi (JGD) alkaline basalts in
29 northeast China to evaluate the influence of elemental diffusion on olivine-
30 composition-based geothermometry, oxybarometry, and hygrometry. The JGD
31 olivines show normal Fo ($\text{Mg}/(\text{Mg} + \text{Fe}) \times 100$ in mole) zoning, with cores having Fo
32 of 77–87 and rims having Fo of 67–73. The constant P contents from core to rim
33 indicate that these compositional zonings were caused mainly by diffusion. Because
34 Al is a slow-diffusing element and its content is relatively constant from core to rim,
35 the temperature calculated by the Al-in-olivine thermometer is not influenced by
36 elemental diffusion and preserves the JGD olivine crystallization temperature up to
37 1150 °C. The temperatures calculated using the Sc/Y-in-olivine thermometer, the
38 oxygen fugacity calculated using the olivine–spinel oxybarometer, and the H₂O
39 content calculated on the basis of Ca partitioning between olivine and melt are
40 strongly influenced by the diffusion of Fo, Sc/Y, and Ca. However, the compositional
41 plateaus in olivine cores, which were not influenced by elemental diffusion, preserve
42 the magmatic temperature (1150 °C), oxygen fugacity (QFM + 1.4), and H₂O content
43 (4 wt.%) that applied during the formation of the JGD olivines. Together, these

44 findings suggest that the mantle source of the JGD basalts was metasomatized by
45 fluids released from the subducted slab. This study highlights that elemental diffusion
46 in olivine phenocrysts can strongly affect the application of olivine-composition-
47 based geothermometers, oxybarometers, and hygrometers. However, primitive olivine
48 cores that have not been influenced by diffusion preserve the initial magmatic
49 thermodynamic conditions.

50

51 **Keywords:** Olivine, diffusion, magma storage, compositional zoning, NE China,
52 Jiagedaqi basalts

53 **Introduction**

54 Basalt, originating from the Earth mantle, is considered as a probe to detect the
55 composition of the deep Earth and the material recycling between Earth surface and
56 deep spheres (Abraham et al., 2001; Fisk and Kelley, 2002; Herd, 2008; Herzberg,
57 2011a; Hofmann, 2003). The formation and evolution of basaltic magmas depends on
58 various factors, such as source lithology, water content, temperature, and redox state.
59 As olivine is an early crystallized silicate mineral from primitive basaltic magma,
60 several recent studies have suggested that olivine chemistry may be used to evaluate
61 these factors. For example, Ni and Mn contents and Mn/Zn ratios of primitive olivine
62 are suggested to be controlled by source lithology and can thus be used as
63 discriminators between pyroxenite and peridotite sources (Herzberg, 2011b; Howarth
64 and Harris, 2017; Liu et al., 2021; Sobolev et al., 2005). The partitioning of Ca
65 between olivines and their host magmas is affected by magmatic H₂O and can thus act
66 as a proxy for magma H₂O content (Gavrilenko et al., 2016; Hong et al., 2017). As the
67 partitioning of V, a multivalent element, is sensitive to oxygen fugacity (Canil, 1997;
68 Mallmann and O'Neill, 2009; Shearer et al., 2006; Shishkina et al., 2018; Wang et al.,
69 2019), it has been suggested that the partitioning of a ratio of V to a homovalent
70 element (e.g., V/Sc) between olivine and silicate melt is a proxy for oxygen fugacity
71 (Mallmann and O'Neill, 2013; Wang et al., 2019). Mallmann and O'Neill (2013)
72 proposed an empirical thermometer based on the partitioning of Sc/Y between olivine
73 and silicate melt. The partitioning behavior of Al between co-existing olivine and
74 spinel can be used to reconstruct magmatic temperature (Coogan et al., 2014; Wan et

75 al., 2008).

76 To use olivine compositions as thermodynamic proxies, an assumption is that
77 these element contents or ratios remain unchanged after the formation of olivine.
78 However, chemical zoning in olivine is a common phenomenon, and the presence of
79 this chemical zoning means that elemental diffusion is certain to have occurred
80 (Albert et al., 2019; Howarth and Gross, 2019; Kahl et al., 2011; Longpré et al., 2014;
81 Mutch et al., 2019; Ruprecht and Plank, 2013; Shea, et al., 2015a, 2015b; Thomson
82 and Maclennan, 2013). For example, olivine crystals in the lava of the April 2007
83 caldera-forming eruption of Piton de la Fournaise volcano have relatively constant
84 values of Fo of 85, Ni of 2170 ppm, and Ca of 1930 ppm in their cores, but show
85 normal zoning in the outermost (100 μm thick) rims, in which Fo and Ni decrease to
86 about 82 and 1650 ppm, respectively, and Ca increases to 2140 ppm (Albert et al.,
87 2019). The diffusion profiles depend on the initial element contents, diffusion
88 coefficients, and elapsed time. Therefore, element contents and ratios may change
89 owing to diffusion.

90 The Early Cretaceous basalts from the Jiagedaqi (JGD) area, northeast (NE)
91 China, are porphyritic and contain abundant subhedral to euhedral mm-sized olivine
92 crystals with compositional zonation. Thus, the JGD olivine phenocrysts are great
93 candidates to investigate the effect of elemental diffusion on the application of
94 olivine-composition-based magmatic thermometry, oxybarometry, and hygrometry. In
95 this study, major- and trace-elemental zoning in JGD olivine phenocrysts was
96 investigated by laser-ablation–inductively coupled plasma–mass spectrometry (LA–

97 ICP–MS) and electron microprobe element mapping techniques. In this paper, we
98 show that the elemental zoning was caused mainly by elemental diffusion and discuss
99 the influence of this compositional zoning on element- or element-ratio-based
100 thermometry, oxybarometry, and hygrometry.

101

102 **Geological background and rock characteristics**

103 Northeast China is located in the eastern Central Asian Orogenic Belt (CAOB),
104 with the Siberian Craton to the north and the North China Craton (NCC) to the south
105 (**Fig. 1a**). Northeast China consists of several Paleozoic micro-continental blocks
106 including, from northwest to southeast, the Erguna, Xing'an, Songnen–
107 Zhangguangcai Range, Jiamusi, and Khanka massifs, which collided and
108 amalgamated with each other during the Paleozoic (Wilde, 2015; Wilde and Zhou,
109 2015; Wu et al., 2011; Xu et al., 2013). From the Paleozoic to Mesozoic, NE China
110 was affected by the closures of the Paleo-Asian Ocean to its south and the Mongol–
111 Okhotsk to its north (Xiao et al., 2009; Xiao et al., 2003). From the Jurassic,
112 subduction of the Paleo-Pacific Plate beneath NE China played an important role in
113 the geological evolution of this area (Tang et al., 2018; Xu et al., 2013).

114 The Jiagedaqi (JGD) area is located in the central Great Xing'an Range, in which
115 abundant Early Cretaceous volcanic rocks are widely distributed (**Fig. 1b**). The
116 studied JGD volcanic rocks are exposed ~65 km to the south of Jiagedaqi county and
117 consist mainly of olivine-bearing basalts. Zircon U–Pb and Ar–Ar geochronology
118 have shown that the volcanic rocks in the study area are of Early Cretaceous age

119 (Wang et al., 2006; Zhang et al., 2008a). Enrichment in light rare earth elements
120 (LREEs) and large ion lithospheric elements (LILEs), depletion in heavy rare earth
121 elements (HREEs) and high field strength elements (HFSEs), and the enriched Sr–
122 Nd–Pb isotopes of the JGD basalts indicate that the mantle source of the JGD basalts
123 was metasomatized by fluids released from a subducted slab (Fan et al., 2003; Luo et
124 al., 2020; Zhang et al., 2008b).

125 The JGD basalts have a porphyritic texture, with 20–30 vol.% phenocrysts (**Fig.**
126 **2a and b**). Olivines are one of the main phenocryst phases, with grains having
127 subhedral to euhedral shapes and measuring 0.5–2.5 mm in length. Most olivine
128 crystals have cracks, and parts of them show weak serpentinization along rims and
129 cracks. Euhedral spinel and primary melt inclusions occur in many olivines.
130 Clinopyroxenes constitute the other main phenocryst phase and have short-columnar
131 shapes and measure 0.5–2.0 mm in length. The matrix of the JGD basalts comprises
132 small clinopyroxenes and plagioclases, with a few opaque minerals. A more detailed
133 petrographic description and geochemical characteristics (major and trace elements
134 and Sr–Nd–Pb isotopes) of the JGD basalts have been reported by Luo et al. (2020).
135 The olivine selection guidelines proposed by Shea et al. (2015a), avoiding small
136 grains, away from crystal corners, with clear concentration plateau, were followed to
137 select olivine crystals for elemental profile analyses.

138

139 **Analytical methods**

140 **Major- and trace-element mapping of olivine by LA–ICP–MS**

141 Major- and trace-element mapping of olivine was conducted using a 193 nm ArF
142 excimer LA system (RESOLUTION M50, Resonetics, USA) coupled to a sector-field
143 ICP–MS instrument (ELEMENT XR, Thermo Fisher, USA) housed at the State Key
144 Laboratory of Isotope Geochemistry, Guangzhou Institute of Geochemistry, Chinese
145 Academy of Sciences (SKLIG–GIG–CAS), Guangzhou, China. Detailed descriptions
146 of the two instruments have been reported in previous studies (Zhang et al., 2021;
147 Zhang et al., 2019). The RESOLUTION M50 LA system has a two-volume LA cell
148 (Laurin Technic), which enables the system to wash out 99% of the signal in less than
149 1.5 s (Müller et al., 2009). Element mapping was conducted on standard 30- μm -thick
150 polished rock thin-sections, mounted in an in-house rotatable holder. By adjusting the
151 longitudinal direction of the measured olivine crystal parallel to the X or Y axis, this
152 rotatable holder ensures that a minimum rectangle measurement area envelops the
153 whole crystal and thus improves the measurement efficiency. Before analysis, the
154 surfaces of thin sections were cleaned with ethanol and deionized water and then
155 dried with a nitrogen gun. A series of parallel rasters across the olivine crystal surface
156 were ablated to form a rectangular grid (**Fig. 2c and d**). The measured raster lines
157 were arranged in successive order, with no gaps or overlaps between each other. The
158 laser was operated with a 45 μm beam size, 15 Hz repetition rate, 8 J cm^{-2} energy
159 density, and 10 $\mu\text{m s}^{-1}$ moving speed for edge-to-edge sampling. Pre-ablation of each
160 raster was used to remove any contamination on sample surfaces, and each raster
161 started with 20 s gas blank measurement with the laser off.

162 The ELEMENT XR sector-field ICP–MS was set to low mass resolution mode.

163 Signals of the following masses were detected in olivine: ^7Li , ^{23}Na , ^{27}Al , ^{25}Mg , ^{29}Si ,
164 ^{31}P , ^{43}Ca , ^{45}Sc , ^{47}Ti , ^{51}V , ^{53}Cr , ^{55}Mn , ^{57}Fe , ^{59}Co , ^{60}Ni , ^{63}Cu , ^{66}Zn , ^{71}Ga , ^{89}Y , and ^{90}Zr .
165 Dwell time and settling time for each mass and other instrumental parameters are
166 given in **Table S1**. The total sweep time from ^7Li to ^{90}Zr was 0.55 s. Under these
167 instrumental settings, the total analysis time was approximately 2 h for an area of 1.5
168 mm \times 1.5 mm on the sample surface. The oxide molecular yield ($^{232}\text{Th}^{16}\text{O}/^{232}\text{Th}$) was
169 less than 0.3%. United States Geological Survey (USGS) reference glass GSD-1G
170 was used as the external calibration material, and BHVO-2G and an in-house olivine
171 standard (JHOL) were measured as unknown samples to evaluate the analytical
172 precision and accuracy. Element mapping of every five unknown samples was
173 preceded and followed by one analysis of GSD-1G. For olivine crystals, the sum of all
174 measured metal oxides was normalized to 100% to calculate the content of each
175 element (Liu et al., 2008; Wu et al., 2020). The limits of detection (LODs) for all
176 measured elements were estimated from the counting statistics of the gas blank
177 intensities and are listed in **Table S2**. An in-house written Python 3 software program,
178 *Fastline*, was used to process element content calibration and create elemental pseudo
179 color plots.

180

181 **Olivine and spinel composition analyses using an electron microprobe**

182 Compared with LA-ICP-MS, electron probe microanalysis (EPMA) has a higher
183 spatial resolution (beam size of several micrometers for EPMA vs. tens of
184 micrometers for LA-ICP-MS) and is more accurate and precise for major elements.

185 Core to rim concentration profiles for the elements Si, Mg, Fe, Ni and Ca of 20
186 olivine crystals were measured using a Cameca SXFive FE EPMA at SKLIG-GIG-
187 CAS. The operating conditions were 20 kV accelerating voltage, 100 nA beam current,
188 and 1 μm beam size. The step interval between two adjacent points was about 20 μm .
189 Si, Mg, Fe, Ni, and Ca were detected using LTAP, LTAP, LIF, LLIF, and LPET
190 crystals, respectively. Major-element compositions (Si, Mg, Al, Mn, Fe, Cr, and Ti) of
191 spinel inclusions in olivine phenocrysts and of their olivine hosts (Si, Mg, Fe, and Al)
192 were also measured with EPMA. All spinel grains selected for analysis are close to
193 the cores of their olivine hosts (**Fig. S1**). The operating conditions for spinel were a
194 beam current of 20 nA, a beam diameter of 1 μm , and an accelerating voltage of 15
195 kV. For olivine, the operating conditions were a beam current of 100 nA, a beam
196 diameter of 2 μm , and an accelerating voltage of 20 kV. MongOl olivine (Batanova et
197 al., 2019) and spinel standard BAR8601-10 (Cao et al., 2019) were used to monitor
198 the accuracy and precision of the measurements. Repeated analyses of both standards
199 show that the relative standard deviation (2RSD) for most elements is 0.1%–1.2%, for
200 CaO is 5.8%, and for Al_2O_3 (in olivine) is 6.6% (Table S5).

201

202 **Electron backscattered diffraction mapping**

203 Electron backscattered diffraction (EBSD) mapping was performed to obtain the
204 crystallographic orientation of olivine at the State Key Lab of High Performance
205 Ceramics and Superfine Microstructure, Shanghai Institute of Ceramics, CAS, using
206 an FEI Magellan 400 field-emission scanning electron microscope (SEM) equipped

207 with a Channel 5 EBSD system (Oxford Instruments) and a Nordly-S EBSD detector.
208 Detailed analytical procedures have been reported by Wang et al. (2017). Instrumental
209 conditions were an accelerating voltage of 15 kV, a working distance of ~5 mm, and a
210 sample tilt of 70°.

211

212 **ANALYTICAL RESULTS**

213 **Olivine compositions and zoning profiles**

214 Twenty olivine crystals were selected for major and minor element
215 measurements from rim to core by EPMA (**Table S3**). To better constrain the
216 populations of the compositions of olivine cores and rims, ten more olivine crystals
217 were measured their cores and rims (**Table S4**). The rim to core traverses show that
218 many crystals are characterized by compositional plateaus in their cores (**Fig. 3**) and
219 that the plateau range increases with increasing crystal size. The Fo content of olivine
220 cores varies from 77 to 87, with a main population around Fo = 83.7 (**Fig. 4**).
221 Compared with the wide variation of olivine cores, olivine rims have limited variation
222 in Fo content (67–73, **Fig. 4**). One olivine crystal was selected for major- and trace-
223 element mapping by LA–ICP–MS (**Fig. 2d**). **Figure 5** displays two-dimensional Fo,
224 Li, Mn, Ni, Sc, Zn, Y, and V distributions in the selected olivine crystal, which
225 indicate normal Fo zoning (Fo decreasing from core to rim). From core to rim, Ni
226 (from ~710 ppm to 1350 ppm) decreases and Li (from ~2 ppm to 12 ppm), Mn (from
227 ~730 ppm to 3350 ppm), V (~4 ppm to 15 ppm), Sc (~10 ppm to 14 ppm), Zn (~80
228 ppm to 190 ppm), and Y (~0.15 ppm to 0.75 ppm) increase. In contrast, Al₂O₃ (~0.03

229 wt%) and P₂O₅ (~0.2 wt%) are essentially uniform from core to rim, showing little
230 zoning (**Fig. 6a** and **b**). Sc/Y is ~50 in the core and changes to ~20 in the rim (**Fig.**
231 **6b**).

232

233 **Spinel compositions**

234 A total of 52 JGD spinel inclusions hosted in olivines and their host olivines
235 were analyzed for major elements (**Table S5**). Cr# values ($\text{Cr}/[\text{Cr} + \text{Al}] \times 100$ in
236 mol.%) vary from 32 to 41 and the Fo values of their host olivines vary from 75 to 87
237 (**Fig. 7**). TiO₂ ranges from 0.5 to 3.1 wt.% and displays little correlation with Cr#.
238 Al₂O₃ content varies from 25 to 36 wt.%.

239

240 **DISCUSSION**

241 **Genesis of the primitive olivine cores**

242 Before using the primitive olivine cores to constrain the thermodynamic conditions
243 (e.g., temperature and oxygen fugacity) of the JGD magma, the genesis of these
244 olivine cores first needs to be established. Most JGD olivines have euhedral shapes,
245 different from mantle xenocrysts, which normally have anhedral shapes (Kamenetsky
246 et al., 2006; Hong et al., 2012). The high Ca contents (> 720 ppm, **Fig. 8a**) of the JGD
247 olivines also indicates that they are not mantle-derived xenocrysts but crystallized
248 from magma (e.g., Hong et al., 2013; Simkin and Smith 1970). In addition, the
249 abundant primary melt inclusions hosted in JGD olivines support a magmatic origin
250 of these olivines (**Fig. S2**; Schiano, 2003). The fractional crystallization trend of Cr#

251 in spinels vs Fo in their host olivines (**Fig. 7**) and the continuous compositional
252 variations of all the measured primitive olivine cores (**Fig. 8**) suggest that these cores
253 share the same trends of evolution and crystallized from the same magma system.

254

255 **Origin of the olivine zoning**

256 There are several possible types of zoning (Costa et al., 2008; Oeser et al., 2015, Shea
257 et al., 2015b): inherited xenocryst cores overgrown by magmatic rims, crystal growth
258 zoning and diffusion zoning. The first situation can be identified by the sharp contrast
259 compositions between core and rim. The other two situations also have different
260 geochemical features, as follows. First, if elemental zoning is caused by crystal
261 growth, then both fast-diffusing and slow-diffusing elements should have
262 corresponding zonings. In contrast, if zoning is caused by diffusion, the elements with
263 different diffusion coefficients will have different diffusion profiles, and the slowest-
264 diffusing trace element (e.g., P in olivine, CaAl-NaSi in plagioclase) may have no
265 diffusion-induced chemical zoning (Costa et al., 2008; Neave et al., 2014; Yang et al.,
266 2019). Second, diffusions for many elements are anisotropic in different
267 crystallographic directions. If zoning is caused by diffusion, then the diffusion
268 distances will vary according to different crystallographic direction (e.g., Qian et al.,
269 2008). Third, if zoning is caused by diffusion, and the diffusion distance is much
270 smaller than the crystal size, then the diffusion profile can be reproduced by the one-
271 dimensional diffusion model for a semi-infinite medium with a source-maintained
272 constant concentration (Cherniak and Watson, 1994). For the JGD olivine phenocrysts

273 measured in this study, contents of the fast-diffusing elements (e.g., Mg, Fe, Li, and
274 Ni) show zoning (**Fig. 5**), whereas contents of the slow-diffusing elements (e.g., Al
275 and P) show limited variations across olivine crystals (**Fig. 6, Table S3**). The limited
276 variations of Al and P do not change continuously from core to rim like the fast-
277 diffusing elements (e.g., Mg, Ca, Mn) and are mainly caused by cracks in olivine. The
278 compositional profiles (Fo, Ni) of the JGD olivines are well reproduced by the one
279 dimensional diffusion model for a semi-infinite medium (see discussion below). The
280 above two lines of evidence indicate that elemental zonings in the JGD olivine
281 phenocrysts were caused by diffusion.

282 Numerous studies have discussed the formation of chemical zoning in mineral
283 phenocrysts (Albert et al., 2019; Boro et al., 2021; Costa et al., 2008; Costa and
284 Dungan, 2005; Costa et al., 2020; Girona and Costa, 2013; Gordeychik et al., 2020;
285 Kahl et al., 2011; Lynn et al., 2017, 2018; Shea et al., 2015b). The JGD olivine
286 phenocrysts have normal Fo zoning, whereby the cores have primitive compositions
287 (Fo = 77–87) and the rims have evolved compositions (Fo = 67–73). Two possible
288 magmatic scenarios can explain the chemical zoning of the JGD olivine phenocrysts.
289 Scenario 1: The JGD olivine cores record a pre-existing magma plumbing system and
290 the large variations of their Fo represent different magma evolving stages. The early
291 crystallizing olivine has higher Fo and the later crystallizing olivine has lower Fo.
292 Later, another evolved magma recharges into the plumbing system and captures the
293 pre-existing olivines crystallized at different stages. As rims with an evolved
294 composition formed, element diffusion started in these olivine crystals. Subsequently,

295 the mixed magma was erupted, and the olivine crystals preserved the compositional
296 zoning. Scenario 2: The JGD olivine cores, with higher Fo, crystallized from a deep
297 magma, which carried the JGD primitive olivine and mixed with a shallower and
298 evolved magma. Then the newly formed rims of the JGD olivine have lower Fo. Since
299 no reverse-zoned olivine was observed in the JGD basalts, the evolved magma was
300 possibly crystal free. Otherwise, some of the JGD olivines should have reverse zoning
301 (Fo increase from core to rim).

302

303 **Timescale for eruption of the Jiagedaqi magma**

304 Element diffusion profiles are a useful tool for establishing the timescale of the
305 evolution of magmatic systems (Brenna et al., 2018; Costa et al., 2008, 2020; Kahl et
306 al., 2011; Lynn et al., 2017; Oeser et al., 2015; Shea et al., 2015a). Here, to evaluate
307 the timescale, we used the one-dimensional diffusion model for a semi-infinite
308 medium with a source-maintained constant concentration at a fixed crystal–melt
309 boundary and a constant diffusion coefficient.

310

311 **Finding the initial and boundary conditions**

312 To conduct the modelling, initial and boundary conditions, which refer to the
313 shape of the zoning pattern before modification by diffusion, should be identified first.
314 The plateau compositions in cores of olivine phenocrysts show no gradient and were
315 treated as the initial compositions. We used two ways of fixing the initial composition
316 of the rim: 1) If plateaus existed near the rims, then the plateau compositions were

317 treated as the initial compositions; and 2) if rims had no compositional plateaus, then
318 we used the $erfc^{-1} \left[\frac{C_x - C_0}{C_i - C_0} \right]$ vs x plot to fit the initial compositions, where C_x is the
319 concentration at distance x , C_0 is the initial concentration, C_i is the concentration at
320 the interface ($x = 0$). The fitted line of $erfc^{-1} \left[\frac{C_x - C_0}{C_i - C_0} \right]$ vs x should cross the
321 coordinate origin and to determine the initial composition (C_i), C_i is changed until the
322 intercept of the fitted line is zero (Cherniak and Watson, 1994).

323

324 **Diffusion coefficient**

325 The Fe–Mg diffusion coefficient in olivine is a function of temperature, pressure,
326 composition, crystallographic direction, and oxygen fugacity, as established from
327 experimental measurements (Costa and Dungan, 2005; Dohmen et al., 2007; Petry et
328 al., 2004). In this study, temperature and oxygen fugacity were determined using the
329 Al-in-olivine thermometer (Coogan et al., 2014; Wan et al., 2008) and the olivine–
330 spinel oxygen barometer (Ballhaus et al., 1991), respectively. We used temperature
331 estimated from olivine rim compositions rather than that estimated from olivine core
332 compositions. This is because although high-Fo olivine cores crystallized from high
333 temperature, these high-Fo olivine were captured by an evolved magma, which had
334 lower temperature. Because there is no spinel-olivine pair in the matrix of the JGD
335 basalts for us to calculate the temperature of the evolved magma, we regressed a
336 relationship of crystallization temperature vs Fo value of olivine based on Al-in-
337 olivine thermometer (**Fig. S3**). Using this relationship, the extrapolated temperature

338 for the averaged olivine rim composition ($Fo = 69.46$) is 1092 °C, which was used in
339 the calculation of Fe–Mg diffusion coefficient. The averaged oxygen fugacity based
340 on the olivine–spinel oxygen barometer is QFM + 1.38 and was chosen for Fe–Mg
341 diffusion coefficient calculation. Crystallographic directions were measured using
342 EBSD (**Fig. S4**). Then, the diffusion coefficients were calculated with the DIPRA
343 program (Girona and Costa, 2013), which was based on the formulas of Dohmen et al.
344 (2007) and Dohmen and Chakraborty (2007). The DIPRA is a user-friendly program
345 to model multi-element diffusion in olivine (Girona and Costa, 2013). To use DIPRA,
346 the user needs to input an element concentration profile as a two-dimensional array,
347 temperature, pressure, fugacity and crystal orientation.

348 Twenty traversing concentration profiles of the JGD olivine phenocrysts were
349 measured to constrain the timescale (**Fig. 3 and 9** and **Table S3**) with the DIPRA
350 program. The timescales calculated from Fo profiles range from 66 to 491 days
351 (**Table S6**). The large variation of the calculated timescales may be due to some of the
352 measured olivine sections do not cross through their crystal centers. This is because
353 that if the section does not cross through the center of the measured crystal, the
354 measured diffusion profile will be shorter than the actual profile, resulting in under
355 estimation of the diffusion timescale. We also modeled the timescales by 17 NiO
356 profiles (Figure 8, Table S3) of the same olivines used for timescale modelling by Fo
357 profiles, which have a range from 76 to 588 days, generally larger than that modelled
358 by Fo profiles (**Fig. S5, Table S6**). This may be due to the mismatch between Fo and
359 Ni diffusion coefficients under relatively low temperature and the less well

360 characterized diffusion coefficient of Ni compared to diffusion coefficient of Fo
361 (Lynn et al., 2017). The larger analytical error of Ni than Fe-Mg may also lead to
362 larger uncertainty of the timescales. Based on the above reasons, although the
363 modelling results by both Fo and NiO profiles are in a same order of magnitude
364 (dozens of days to less than two years), the timescales derived from Fo profiles are
365 preferred in this study.

366

367 **The effect of elemental diffusion on the application of olivine-composition-based**
368 **magmatic thermometry, oxybarometry, and hygrometry**

369 **Olivine crystallization temperature**

370 Temperature is a fundamental property of magmas and can be used to identify
371 their tectonic setting (Coogan et al., 2014; Herzberg and Asimow, 2008; Herzberg,
372 2011a; Putirka, 2005). In this study, two geological thermometers, the Al-in-olivine
373 thermometer (Coogan et al., 2014; Wan et al., 2008) and the Sc/Y partitioning
374 thermometer (Mallmann and O'Neill, 2013), were used to calculate the olivine
375 crystallization temperature and evaluate the influence of elemental diffusion on both
376 thermometers. The Al-in-olivine geothermometer (Coogan et al., 2014; Wan et al.,
377 2008), which is based on the partitioning of Al between coexisting olivine and spinel,
378 was first established with experiments conducted under reducing conditions (~QFM–
379 1.5) by Wan et al. (2008). Recently, Coogan et al. (2014) expanded this
380 geothermometer to more oxidized conditions using the following equation:

381
$$T(K) = \frac{10000}{0.577 + 0.884Cr\# - 0.897\ln(Kd)} \quad (1)$$

382 where $Kd = \frac{Al_2O_3^{olivine}}{Al_2O_3^{spinel}}$. This geothermometer is independent of the melt composition,
383 crystallization pressure, and oxygen fugacity, and the uncertainty of the estimated
384 temperature is $<30^\circ C$ (Coogan et al., 2014). Given that Al^{3+} diffuses in the olivine
385 lattice far more slowly than Mg^{2+} and Fe^{2+} (Spandler and O'Neill, 2010), the Al
386 content can still record the initial olivine crystallization temperature.

387 The Sc/Y partitioning thermometer was calibrated with a series of experiments
388 under pressure of 1 bar to 2.0 GPa and temperature of 1200 – 1530 °C and is
389 expressed as (Mallmann and O'Neill, 2013):

390
$$\log_{10} Kd_{Sc/Y}^{ol/melt} = b_0 + b_1/T + b_2P/T + b_3Mg\#^{ol}/T + b_4X_{CaO}^{melt}/T$$

$$+ b_5(X_{NaO_{0.5}}^{melt} + X_{KO_{0.5}}^{melt})/T + b_6X_{SiO_2}^{melt}/T \quad (2)$$

391 where $Kd_{Sc/Y}^{ol/melt}$ is Sc/Y exchange coefficient between olivine and melt; b_0 to b_6 are
392 equation factors; T and P represent temperature and pressure, respectively; Xs
393 represent oxide molar content of melt.

394 Therefore, it is clear that the formula for the Al-in-olivine thermometer is related
395 only to spinel Cr# and the Al partition coefficient between olivine and spinel (Coogan
396 et al., 2014; Wan et al., 2008), whereas the formula of the Sc/Y partitioning
397 thermometer is related to the olivine Fo value and the Sc/Y partition coefficient
398 between olivine and its parental magma (Mallmann and O'Neill, 2013). Experimental
399 diffusion studies have shown that Al and Cr in spinel (Suzuki et al., 2008) and Al in
400 olivine (Spandler and O'Neill, 2010) are slow-diffusing elements, whereas Mg, Fe, Sc,
401 and Y in olivine diffuse much more quickly (Spandler and O'Neill, 2010), as

402 confirmed by the relatively constant Al content from core to rim and the zoning
403 profiles of Fo, Sc, and Y in the JGD olivine phenocrysts measured in this study (**Fig.**
404 **5 and 6**). To use the Sc/Y partitioning thermometer, we selected several discrete
405 points from the olivine profile in **Fig. 6**. The temperatures calculated using the Al-in-
406 olivine thermometer display a normal positive correlation with Fo content (**Fig. 10**),
407 whereas the temperatures calculated with the Sc/Y partitioning olivine thermometer
408 show a negative correlation with Fo content (**Fig. 10**). However, the temperatures
409 calculated with both thermometers for the core composition are consistent with each
410 other (around 1150 °C), which suggests that Fo, Sc, and Y in the core plateaus are less
411 affected by diffusion. Thus, the Al-in-olivine thermometer is more resistant to
412 diffusion than the Sc/Y partitioning thermometer.

413

414 **Oxygen fugacity**

415 Oxygen fugacity (fO_2) strongly affects the behavior of multivalent elements,
416 which in turn affect many key processes, such as magma evolution, ore formation,
417 and atmospheric compositional change (Cao et al., 2020; Frost, 1991; Mavrogenes
418 and O'Neill, 1999; Putirka, 2016; Scholtysik and Canil, 2021; Snyder et al., 1993;
419 Sun et al., 2015b; Wang et al., 2019; Wang et al., 2020). Therefore, many geochemical
420 oxybarometers have been proposed to calculate oxygen fugacity (Ballhaus et al., 1991;
421 Burnham and Berry, 2012; Mallmann and O'Neill, 2009; Mallmann and O'Neill,
422 2013; O'Neill and Wall, 1987; Rzehak et al., 2020; Smythe and Brenan, 2016). It is
423 suggested that degassing plays an important role in controlling oxygen fugacity of

424 basaltic magmas (Burgisser and Scaillet, 2007; Kelley and Cottrell, 2012; Mathez,
425 1984), though the study of de Moor et al. (2013) shows no change in $\text{Fe}^{3+}/\Sigma\text{Fe}$ or
426 $\text{S}^{6+}/\Sigma\text{S}$ with extent of S degassing at Erta Ale volcano. Because olivine and spinel are
427 the two early-crystallizing minerals in basaltic magma, oxybarometers based on
428 olivine and spinel compositions record the $f\text{O}_2$ of the relatively primitive magma, as
429 they are close to the liquidus. In this study, we used the widely applied olivine–spinel
430 oxybarometer of Ballhaus et al. (1991) to constrain the oxygen fugacity of the JGD
431 basaltic magma. The formula of the olivine–spinel oxybarometer of Ballhaus et al.
432 (1991) is related to olivine Fo content and spinel composition and expressed as:

$$\begin{aligned} \Delta \log(f\text{O}_2)^{\text{QFM}} = & 0.27 + \frac{2505}{T} - \frac{400P}{T} - 6 \log(X_{\text{Fe}}^{\text{Olv}}) - \frac{3200(1 - X_{\text{Fe}}^{\text{Olv}})^2}{T} \\ & + 2 \log(X_{\text{Fe}^{2+}}^{\text{Sp}}) + 4 \log(X_{\text{Fe}^{3+}}^{\text{Sp}}) + \frac{2630(X_{\text{Al}}^{\text{Sp}})^2}{T} \end{aligned} \quad 3$$

433 where $\Delta \log(f\text{O}_2)^{\text{QFM}}$ is the calculated oxygen fugacity relative to QFM buffer; T
434 and P represent temperature and pressure, respectively; $X_{\text{Fe}}^{\text{Olv}}$ and $X_{\text{Fe}^{2+}}^{\text{Sp}}$ are molar
435 ratios of $\text{Fe}^{2+}/(\text{Fe}^{2+} + \text{Mg})$ in olivine and spinel; $X_{\text{Fe}^{3+}}^{\text{Sp}}$ and $X_{\text{Al}}^{\text{Sp}}$ are molar ratios of
436 $\text{Fe}^{3+}/(\text{Fe} + \text{Al} + \text{Cr})^{3+}$ and $\text{Al}/(\text{Fe} + \text{Al} + \text{Cr})^{3+}$, respectively. As the equation shows, the
437 olivine–spinel oxybarometer needs Fe^{3+} content of spinel. Therefore, this
438 oxybarometer may not be applicable to reduction systems because of too low Fe^{3+} for
439 accurate determination in spinel.
440

441 Because the JGD spinel crystals measured in this study were too small (mostly
442 10–20 μm) to perform profile quantitative analysis to show whether they have Mg and
443 Fe zoning (Vogt et al., 2015) and the BSE images of JGD olivine-hosted spinels

444 displayed little zonation (**Fig. S6**), we considered only the influence of Fo diffusion in
445 olivine on the oxygen fugacity calculation. Discrete points on the Fo profiles from
446 core to rim were selected to calculate oxygen fugacity. The results calculated with the
447 Fo compositions of the core plateaus display oxidized oxygen fugacity, with fO_2
448 between QFM + 0.6 (QFM refers to the quartz–fayalite–magnetite buffer) and QFM +
449 2.0, and an average value of QFM + 1.4 (**Fig. 11**). As the used Fo contents are taken
450 progressively from core to rim, the calculated fO_2 becomes accordingly reduced.
451 Using the olivine rim compositions (assuming Fo = 70), the calculated values of fO_2
452 are between QFM – 0.9 and QFM + 0.8, with an average value of QFM + 0.1.
453 Because the Fo contents of the olivine cores have relatively constant plateaus, we
454 presume that the values of fO_2 calculated with the core compositions represent the
455 true oxygen fugacity, whereas the lower fO_2 values calculated with the rim
456 compositions were affected by Fo diffusion in the olivine phenocrysts.

457

458 **H₂O content in JGD magma**

459 Owing to degassing processes during magma upwelling and eruption, it is
460 difficult to directly measure the primary H₂O content in volcanic rocks. Some
461 hygrometers based on mineral geochemistry have been proposed by previous studies.
462 For example, many studies have measured H₂O content in clinopyroxene and used the
463 partitioning coefficient of H₂O between clinopyroxene and melt to calculate the H₂O
464 content in magma (Liu et al., 2017; Nazzareni et al., 2011; Wade et al., 2008; Xia et
465 al., 2013). Recently, Gavrilenko et al. (2016) calibrated a Ca-in-olivine

466 geohygrometer up to ~ 8 wt% H₂O content based on the partitioning of Ca between
467 olivine and silicate melt. We used this geohygrometer to calculate the H₂O content in
468 JGD magma. Although this geohygrometer does not consider the effect of other
469 volatiles (e.g., CO₂) and pressure on Ca partitioning coefficient and may not be
470 applicable to some magmatic systems (e.g., komatiite; De Hoog et al., 2010;
471 Gavrilenko et al., 2016), it does not influence the discussion of the elemental diffusion
472 effect on this geohygrometer. The results show that as Ca content increases from core
473 to rim (**Fig. 12**), the calculated H₂O content decreases from ~4 wt.% to <2 wt.% (**Fig.**
474 **12**). Because Ca has a plateau in the core, the calculated H₂O content also displays a
475 plateau (~4 wt.%), which is treated as the primary H₂O content in the JGD magma.

476

477 **IMPLICATIONS**

478 Compared with MORB, the low olivine crystallization temperature (1150 °C),
479 high oxygen fugacity (QFM + 1.4), and high H₂O contents (~4 wt%) of the JGD
480 basalts support a metasomatized mantle source, which may have been related to the
481 closure of the Mongolia–Okhotsk Ocean or the subduction of the Paleo-Pacific Plate
482 beneath the Eurasian continent (Fan et al., 2003; Sun et al., 2015a; Tang et al., 2018;
483 Xu et al., 2013). The released fluids from subducted slabs metasomatized the
484 overlying mantle wedge and lowered the solidus of the wedge. Then, partial melting
485 of this metasomatized mantle wedge resulted in H₂O-enriched and high-oxygen-
486 fugacity magma, like that of the JGD basalts. As discussed above, the compositional
487 zoning of the olivine phenocrysts indicates a recharge magma system, which led to

488 the eruption of the JGD basalts or a deep magma carrying high-Fo olivines mixed
489 with a shallower and evolved magma. In the recharge model, the last input of evolved
490 magma induced an increase in the pressure of the JGD magma chamber. After a
491 timescale of dozens of days to less than two years, the JGD magma was erupted, and
492 the olivine phenocrysts preserved the disequilibrium compositional zoning.

493 Olivine phenocrysts from the JGD basalts show compositional zoning caused by
494 elemental diffusion. Modellings of the one-dimensional Mg–Fe and Ni diffusion
495 profiles indicate that the diffusion timescale is dozens of days to less than two years.
496 The diffusions of Mg, Ca, Fe, Sc, and Y in the JGD olivines have a strong influence
497 on magmatic temperature calculated using a Sc/Y-in-olivine thermometer, magmatic
498 oxygen fugacity calculated using an olivine–spinel oxybarometer, and magmatic H₂O
499 content calculated using a hygrometer based on Ca partitioning between olivine and
500 melt. However, the compositional plateaus in olivine cores, which were not
501 influenced by elemental diffusion, preserve the magmatic temperature (1150 °C),
502 oxygen fugacity (QFM + 1.4), and H₂O content (4 wt.%) that prevailed during the
503 formation of the JGD olivines, supporting the inference that the mantle source of the
504 JGD basalts was metasomatized by fluids released from a subducted slab. Our study
505 clearly shows that the use of olivine geochemistry to investigate magmatic
506 thermodynamic conditions must consider the effects of elemental diffusion. For
507 volcanic rocks, although they cool down soon after eruption, elemental profile
508 analyses of phenocrysts are needed to evaluate the effect of diffusion. If phenocrysts
509 have compositional plateaus in their cores, then the thermodynamic conditions of

510 primitive magma can be derived from the core compositions. For large intrusions, the
511 cooling rates normally vary from 200 °C/Ma to 1-10 °C/Ma (Ortega-Gutiérrez et al.,
512 2014), much smaller than that for volcanic rocks. Except for the very slow-diffusing
513 elements (e.g., P in olivine, CaAl-NaSi in plagioclase), other elements possibly reach
514 equilibrium at low temperature and thus lose the thermodynamic information of their
515 primitive magmas.

516

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525

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867 **Figure captions**

868 **Figure 1** (a) Simplified geological map of East Asia. (b) Distribution of the
869 Cretaceous volcanic rocks in NE China. CAOB - Central Asian Orogenic Belt.

870

871 **Figure 2** (a-b) Photomicrographs of representative samples under crossed-polarizers.
872 The large olivine phenocryst in b was selected for LA-ICP-MS elemental mapping. (c)
873 Image taken under reflected light of the same olivine phenocryst in b. (d) BSE image
874 of the measured olivine phenocryst after LA-ICP-MS analysis. The crossed-polarizer
875 image and BSE image show the zonation. Laser ablation was conducted under raster
876 mode. Laser spot size, 45 μm ; repetition rate, 15 Hz; energy density, 8 mJ cm^{-2} ;
877 moving speed 10 $\mu\text{m s}^{-1}$. Ol, olivine; MI, melt inclusion; Sp, spinel.

878

879 **Figure 3** Olivine forsterite (Fo) profiles measured by EPMA. All measured olivine
880 phenocrysts display normal Fo zoning pattern (decrease from core to rim).

881

882 **Figure 4** Histogram of Fo contents for olivine rims and cores. The dashed lines
883 indicate the mean values of Fo contents for rims and cores.

884

885 **Figure 5** 2D elemental distribution plots for selected elements in the olivine
886 phenocryst in Figure 2b. Except Fo in percentage, other elements are in ppm.

887

888 **Figure 6** Concentration variations for Fo and P₂O₅ (a), Al₂O₃ and Sc/Y (b) across the
889 profile indicated by the red line in the BSE image. The peak or trench represents crack
890 in the measured olivine phenocryst. Dashed lines in (a) are modeled Fo content
891 profiles with DIPRA program. The blue dashed line indicates the inferred initial
892 profile shape before diffusion occurred. The red dashed line shows a diffusion
893 timescale of about 400 days. Modelling parameters: temperature = 1092 °C; fO₂ =
894 Δ QFM + 1.4; crystal orientation (α , β , γ) = (84, 85, 7).

895

896 **Figure 7** Plot of Cr# in spinels vs Fo in their host olivines. The olivine-spinel mantle
897 array is from Arai (1994).

898

899 **Figure 8** Plots of concentrations (in ppm) of Ca (a) and Ni (b) vs Fo for the JGD

900 olivine cores.

901

902 **Figure 9** DIPRA modelling examples of rim-to-core Fo (a, b) and Ni (c, d) profiles
903 for the JGD olivines. BSE images illustrate the locations of the analytical traverses.
904 Red lines show the modelling results from one dimensional diffusion model for a
905 semi-infinite medium with calculated timescales. C_0 is the initial concentration and C_i
906 is the boundary concentration. The error bars of all measured NiO are set to 6%.

907

908 **Figure 10** Olivine crystallization temperatures calculated with the Al-in-olivine
909 thermometer by Coogan et al. (2014) (triangle) and by Sc/Y-in-olivine thermometer
910 by Mallmann and O'Neill (2013) (circle). The temperatures calculated by Al-in-
911 olivine thermometer show a normal positive correlation with Fo contents. In contrast,
912 the temperatures calculated by Sc/Y-in-olivine thermometer show a negative
913 correlation with Fo contents. The Sc/Y-in-olivine thermometer used the bulk rock
914 Sc/Y (Luo et al., 2020) to calculate the Sc/Y partition coefficient. Data source: MORB,
915 Matthews et al. (2016), Yang et al. (2019) and Danyushevsky et al. (2003); Arc,
916 Elburg et al. (2007), Søger et al. (2015) and Nazarova et al. (2017).

917

918 **Figure 11** Calculated oxygen fugacity relative to the QFM buffer vs Cr# in spinel.
919 The oxygen fugacity calculated with olivine core composition (circles) records the
920 true magmatic redox condition. Other symbols represent the results calculated with
921 the olivine compositions (Fo = 80, 75 and 70) influenced by Mg-Fe diffusion.

922

923 **Figure 12** Concentration variation of CaO for the A-A' profile in Figure 5a and the
924 calculated H₂O content with the Ca-in-olivine geohygrometer (Gavrilenko et al.,
925 2016). The MgO (7.06 wt%) and CaO (9.50 wt%) of the JGD whole rock (Luo et al.,
926 2020) were used in the calculation. The peak or trench represents a crack in the
927 measured olivine phenocryst.

928

Figure 1

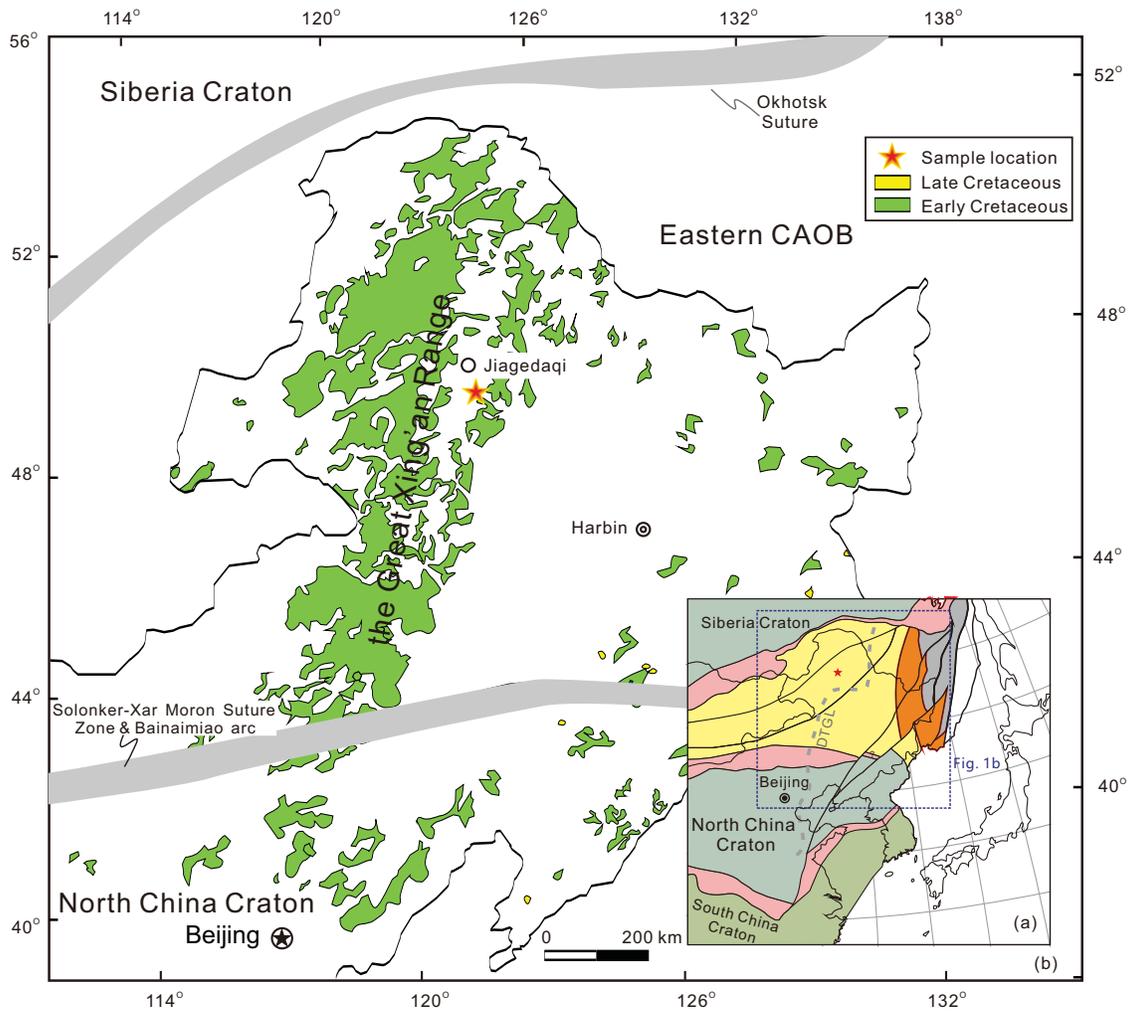


Figure 2

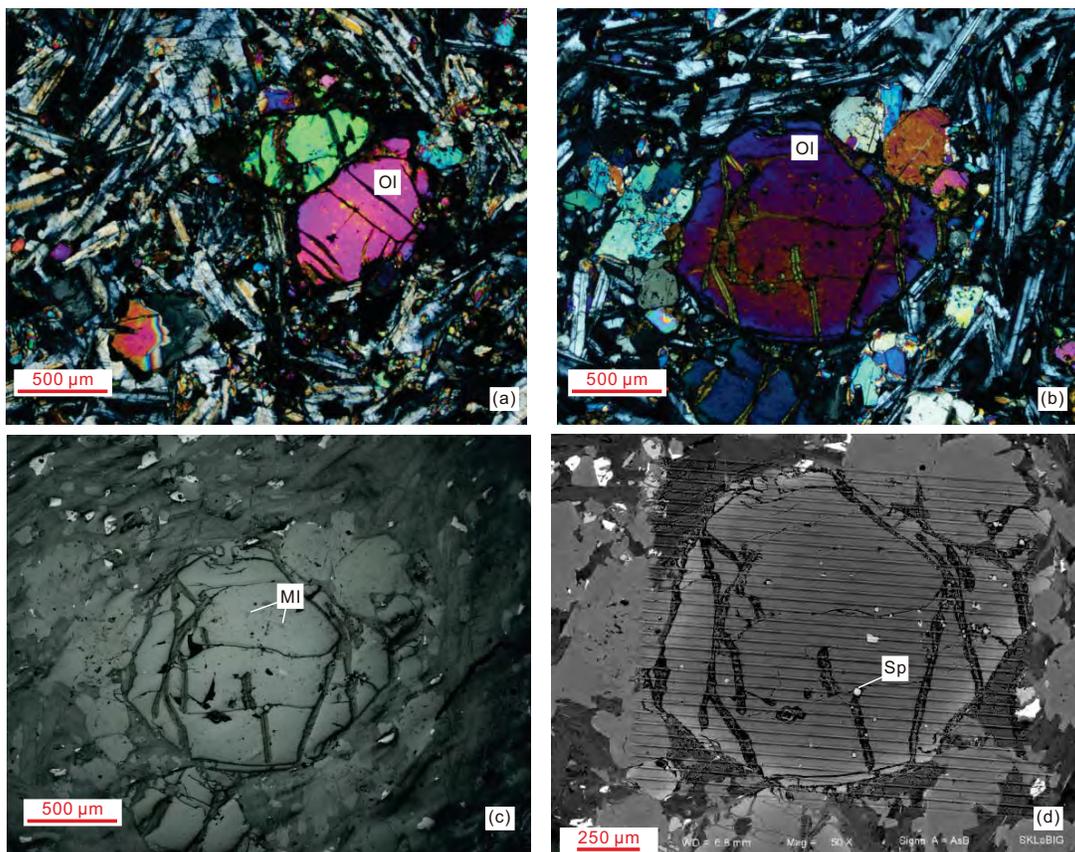


Figure 3

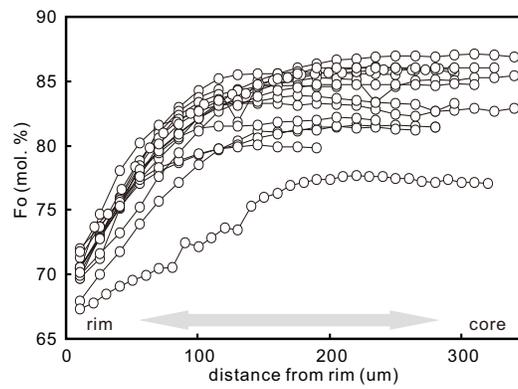


Figure 4

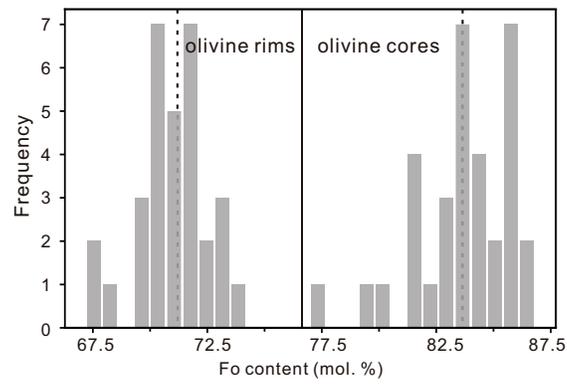


Figure 5

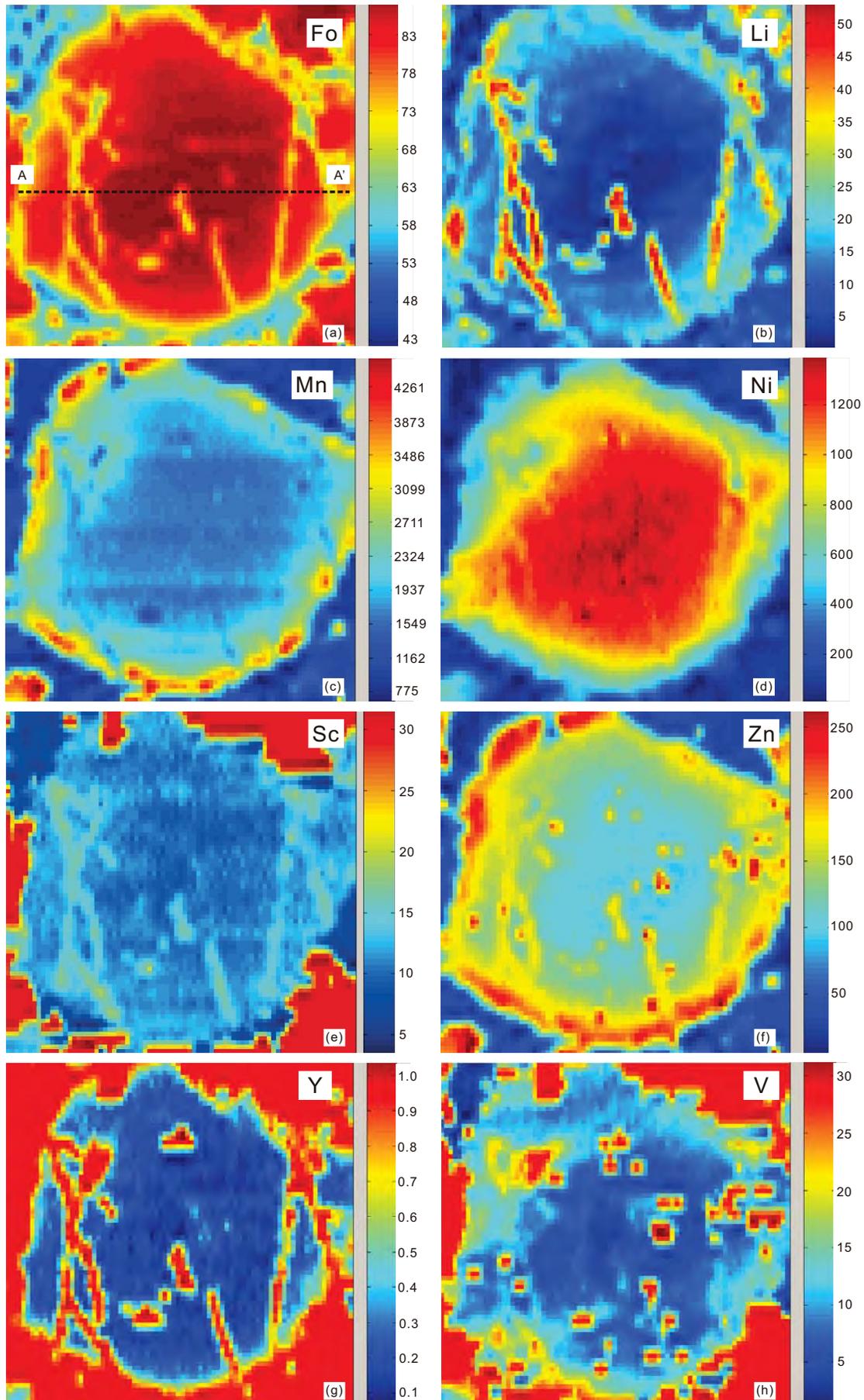


Figure 6

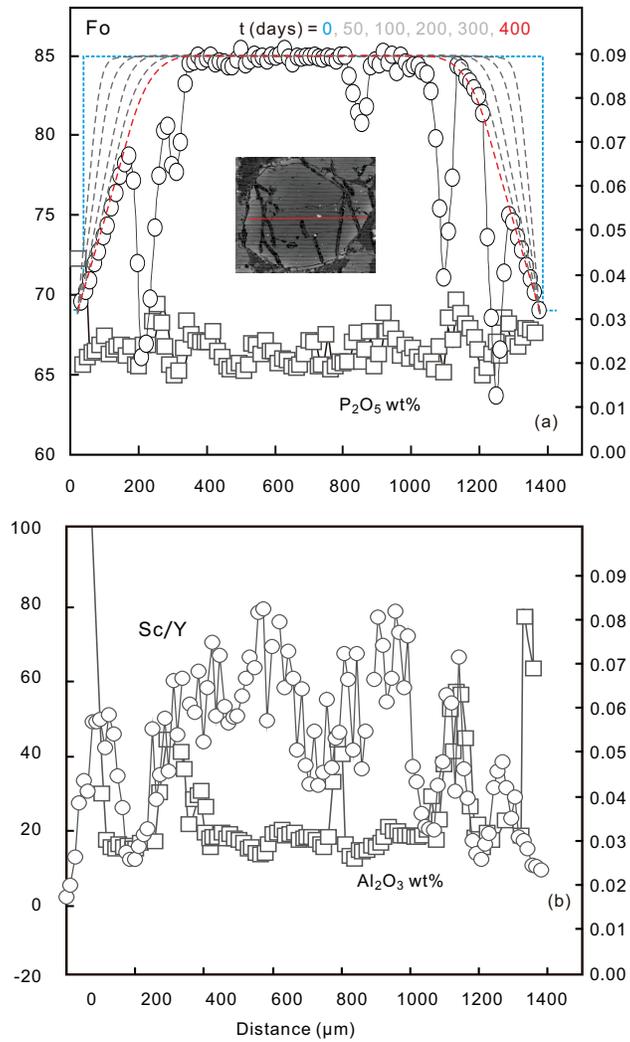


Figure 7

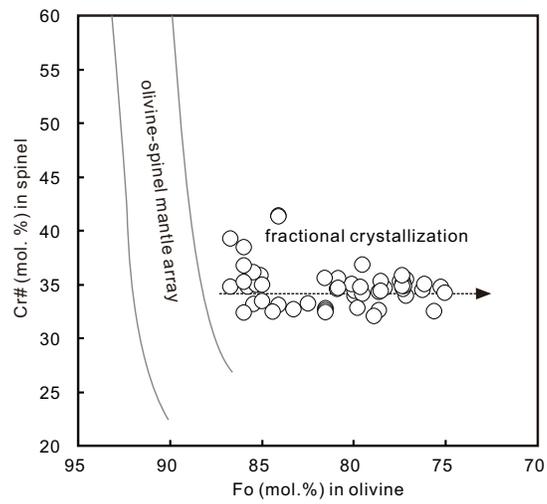


Figure 8

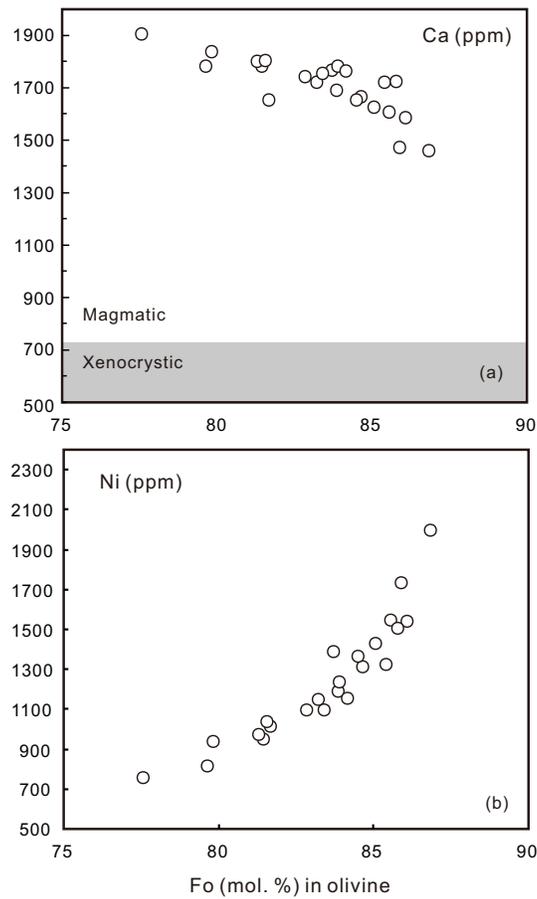


Figure 9

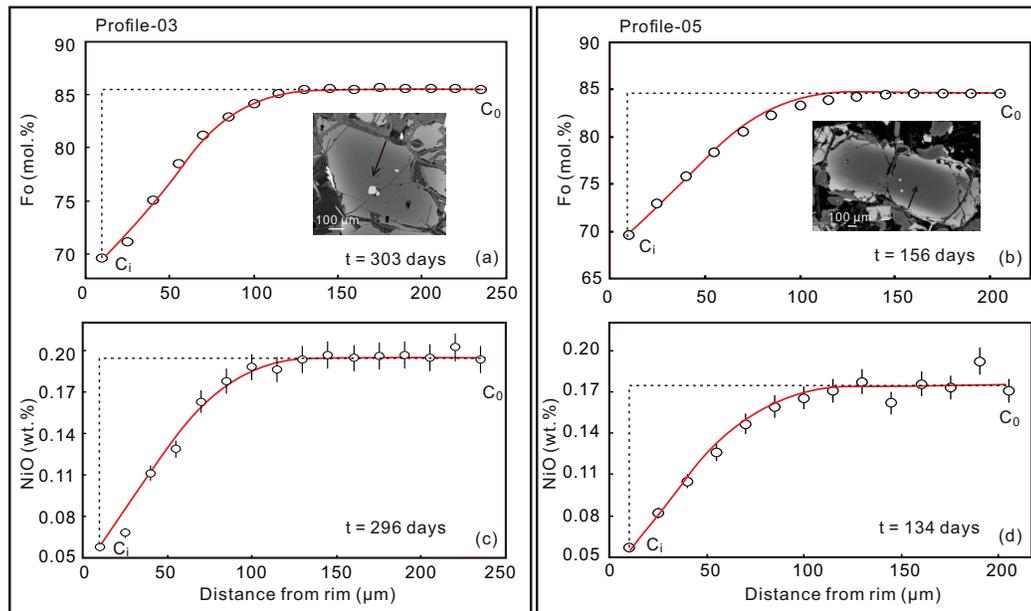


Figure 10

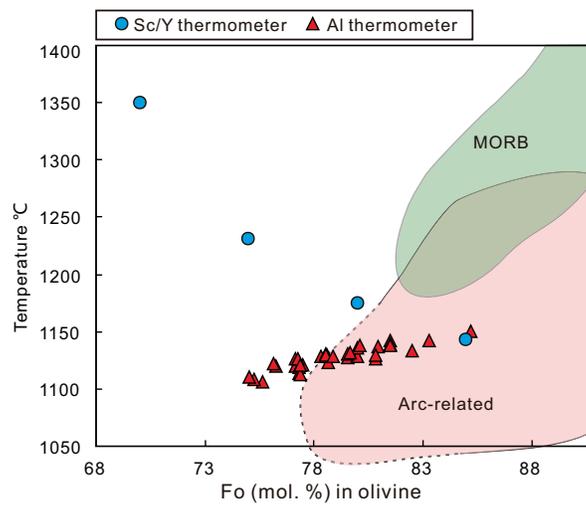


Figure 11

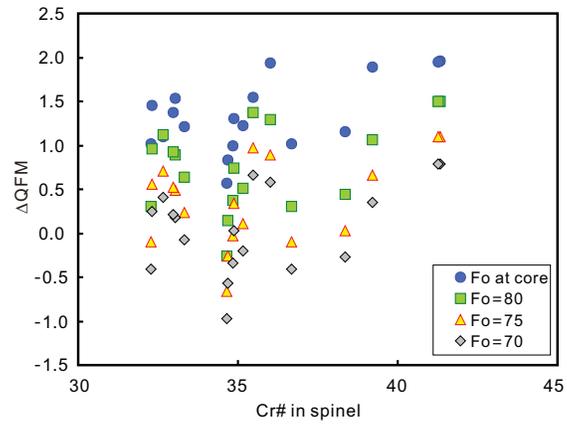


Figure 12

