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3 **Raman analysis of octocoral carbonate ion structural disorder along a natural depth**

4 **gradient, Kona coast, Hawai'i**

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14 **Abstract**

15 Both environmental and physiological factors cause carbonate ion structural disorder in

16 biogenic Mg-calcites. A major component of this disorder is driven by the incorporation of Mg

17 through environmental forcing and growth rate kinetics although non-Mg factors (e.g., other

18 cation/anion impurities, organic molecules) also contribute. Understanding the drivers of Mg

19 content in biogenic calcite and its effects on disorder has implications for octocoral Mg paleo-

20 proxies and the stability and diagenetic alteration of their calcitic skeletons. However, prior

21 studies of biogenic Mg-calcites have often been complicated by sampling inconsistencies over

22 space and time and potential intra-sample Mg variability. This study aims to analyze the relative

23 contributing factors of octocoral Mg-calcite structural disorder along gradients of both depth and

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24   growth rate. Calcitic octocorals (Corallidae and Isididae, N = 28) were collected from 221–823

25   m depths across a natural gradient in biogeochemical parameters (pH: 7.4–7.9, T: 5–16°C) off

26   the Kona coast of Hawaiʻi Island and analyzed using Raman spectroscopy. Samples were

27   collected during the same month, controlling for potential seasonal variability. Raman spectral

28   parameters from the ν_1 peak quantified total carbonate ion structural disorder (full width at half

29   maximum height [FWHM] of ν_1) and Mg content (ν_1 position, Raman shift). The total structural

30   disorder was then partitioned into Mg-driven and non-Mg driven components (residual ν_1

31   FWHM). The total structural disorder and Mg content decreased significantly with increasing

32   depth, correlating with temperature and carbonate system parameters. The Mg-temperature

33   relationships from this study were also consistent with prior studies. Non-Mg structural disorder

34   did not correlate to any environmental parameters. When measured across an intra-sample

35   gradient of ontogenetic growth rate, total structural disorder, Mg content, and non-Mg structural

36   disorder increased with growth rate for all but one taxon, demonstrating the kinetic effect of

37   growth rate as well as potential taxon-specific physiological effects. These results provide insight

38   into how environmental and growth rate kinetic effects independently affect different

39   components of carbonate ion structural disorder (Mg content and non-Mg factors). These

40   findings also suggest that Raman spectroscopy may be helpful in quantifying solubility within

41   biogenic calcites.

42   **Keywords:** Octocorals, magnesian calcite, carbonate ion disorder, Raman spectroscopy,

43   depth gradient, growth rate kinetics

44

45   **Introduction**

46   Environmental and physiological (i.e., vital effects) factors can influence carbonate ion

47   structural disorder (referred to hereinafter as structural disorder) within biogenic calcium

48   carbonates ($CaCO_3$) through the incorporation of elemental impurities as well as growth rate

49   kinetics and organic molecules involved in biogenic calcification. $CaCO_3$ structural disorder

50   involves the orientation of carbonate ions relative to the *c* (vertical) and *a* (horizontal) unit cell

51   axes (Bischoff et al. 1985). Ideally, carbonate ions are symmetrically aligned with their

52   respective basal planes along the *a* axis, however, deviations from this symmetry occur at

53   varying extents within synthetic and biogenic $CaCO_3$. The incorporation of non-constituent ion

54   impurities (divalent cations with radii differing from Ca) such as Mg, Sr, and Ba results in

55   altered cation-oxygen bond lengths that can rotate carbonate ions out of the basal plane towards

56   the *c* axis (Bischoff et al. 1985; Urmos et al. 1991; Perrin et al. 2016; Farfan et al. 2018, 2021).

57   Large anions like sulfate (relative to carbonate ions) can also be substituted into the $CaCO_3$

58   crystal structure and increase structural disorder (reviewed in Vielzeuf et al. 2018). Kinetic

59   effects influenced by growth rate as well as ambient temperature and pressure also promote

60   increased disorder because rapid crystallization reduces the symmetric alignment of carbonate

61   ions and increases the incorporation of impurities (Watson 2004; Morse et al. 2007; Mavromatis

62   et al. 2013; Farfan et al. 2018, 2021). $CaCO_3$ precipitation rates can be elevated through

63   favorable physicochemical conditions obtained either through the exterior environment or within

64   an organism's extracellular calcifying fluid (Al-Horani et al. 2003; Morse et al. 2006; McCulloch

65   et al. 2012). Organic molecules like skeletal organic matrices are crucial in regulating biogenic

66   crystal nucleation and can also introduce disorder by interacting with adjacent carbonate ions as

67   well as other crystallographic features (Tambutté et al. 2011; Mass et al. 2013; DeCarlo et al.

68   2018; Coronado et al. 2019). Biogenic $CaCO_3$ are known to contain larger unit cell volumes and

69 c/a axis ratios relative to synthetic $CaCO_3$ even with comparable concentrations of elemental

70 impurities such as Mg (Bischoff et al. 1985). However, it should be noted that Farfan et al.

71 (2021) observed similar unit cell volumes between biogenic (scleractinian coral) and synthetic

72 aragonites.

73   A major component of structural disorder within biogenic calcite (a polymorph of

74 $CaCO_3$) involves the incorporation of Mg cations, which is largely driven by temperature,

75 carbonate system parameters, and skeletal growth rate. Although the exact mechanisms

76 influencing biogenic Mg incorporation are not well understood (Long et al. 2014), the

77 connection between temperature and Mg content in organisms such as octocorals has been well

78 established (Weinbauer et al. 2000; Thresher et al. 2004, 2010; Sherwood et al. 2005). Carbonate

79 system parameters such as saturation state (Ω) have also been correlated with Mg content

80 (Thresher et al. 2011) due to their thermodynamic and kinetic influence on calcite solubility

81 (Mackenzie et al. 1983; Mucci 1987; Morse et al. 2007). Other studies have also highlighted the

82 complications of skeletal growth rate effects on Mg content for calcite (Vielzeuf et al. 2013,

83 2018; Robinson et al. 2014; Chaabane et al. 2019; Flöter et al. 2019) as well as aragonite

84 (Gagnon et al. 2007; Rollion-Bard and Blamart 2015; Bell et al. 2017). The main factor

85 controlling Mg incorporation is still debated, although a combination of different environmental

86 factors (e.g., temperature) simultaneously influencing biological response and growth rate

87 kinetics is generally thought to be the case (Vielzeuf et al. 2018).

88   Understanding the drivers of octocoral Mg content and resulting structural disorder has

89 implications for mineral solubility and Mg paleo-proxies, yet prior studies have been

90 complicated by inconsistencies in sampling location and timeframes as well as potential Mg

91 variability within individual organisms. Higher calcite Mg content generates structural disorder

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92   that decreases mineral stability and increases solubility across natural gradients in depth and

93   latitude (Andersson et al. 2008; Lebrato et al. 2016). Mg-temperature relationships are integral to

94   paleothermometry applications, yet the impact of vital effects on Mg incorporation and resulting

95   structural disorder confounds temperature reconstructions (Chaabane et al. 2019). Prior

96   geochemical studies on octocorals relied on limited or opportunistic sample sets lacking in

97   corresponding environmental data with specimens collected from a wide variety of locations and

98   timeframes (Robinson et al. 2014). As a result, attempts to distinguish Mg contributions from

99   different environmental and vital effects are fundamentally confounded. Specimens from

100   different oceanographic regimes contain a different combination of biogeochemical conditions

101   and food sources, which could impact the elemental geochemistry of octocoral skeletons

102   (Hasegawa et al. 2012). Not accounting for time-based variables such as seasonal growth rates

103   could also lead to inaccuracies if samples are collected at different times of the year. Spatial

104   variability in trace elements within individual octocoral skeletons has been analyzed through

105   radial cross section measurements for numerous species, revealing areas of relatively faster

106   growth (e.g., medullar zone for *Corallium rubrum*, Vielzeuf et al. 2008). However, intra-sample

107   variability has not been quantified on surface skeletons despite providing an opportunity to

108   observe the effects of variable growth rates (branches vs. base; Vielzeuf et al. 2008) for skeletons

109   precipitated concurrently under the same environmental conditions. Only a few studies have

110   analyzed octocoral geochemistry with respect to environmental gradients involving

111   corresponding oceanographic parameters (Thresher et al. 2011; Bostock et al. 2015), which have

112   better enabled the analysis of major environmental effects on measured Mg.

113   Raman spectroscopy can provide a fast, nondestructive method of quantifying octocoral

114   structural disorder from Mg and non-Mg sources as well as Mg content. The Raman spectra of

115   calcium carbonates contain six main peaks: two representing the intermolecular lattice vibration

116   modes (librational L peak and translational T peak) and four representing the internal vibration

117   modes (ν_1, $2\nu_2$, ν_3, ν_4) (Bischoff et al. 1985, [1]Fig. S1). The incorporation of smaller sized

118   (relative to Ca) Mg cations into calcite shortens cation-oxygen bonds, which then elevate the

119   vibrational frequencies and positional disorder of neighboring carbonate ions (Krishnamurti

120   1957; White 1974; Bischoff et al. 1985). Raman spectral parameters like peak position (Raman

121   shift, cm^{-1}) and width (full width at half magnitude, FWHM, cm^{-1}) increase predictably as a

122   function of Mg content (mol%, Bischoff et al. 1985; Perrin et al. 2016). Incorporated divalent

123   cations larger than Ca such as Ba and Sr generate longer, weaker bonds with oxygen that should,

124   in theory, oppose and suppress the Raman signal influenced by Mg. Analogous work on

125   aragonite minerals reported that ν_1 Raman shift values decrease 0.0544 cm^{-1} per mmol/mol Sr/Ca

126   and 0.0004 cm^{-1} per μmol/mol B/Ca while increasing 0.0262 cm^{-1} per mmol/mol Mg/Ca (Farfan

127   et al. 2021). However, findings from Kaabar et al. (2011) demonstrated that considerable Ba and

128   Sr content (up to 33% of total cations) did not cause significant changes to calcite ν_1 peak Raman

129   shift when the Ca fraction was 20% or higher. Previous Raman calibration of calcite Sr content

130   also observed no significant change in calcite ν_1 peak characteristics even across a 0–13% range

131   in Sr content (Shibano et al. 2017). Sulfate anions thought to impact structural disorder did not

132   significantly change ν_1 Raman shift in calcites with up to 20% $CaSO_4$ (Kontoyannis et al. 1997).

133   While large fractions of Ba, Sr, and S can interact with Ca to significantly alter calcite FWHM

134   values (Kontoyannis et al. 1997; Kaabar et al. 2011), Ba, Sr, and S content within octocoral

135   calcite is negligible based on previous studies (Thresher et al. 2007; LaVigne et al. 2011; Sinclair

136   et al. 2011; Vielzeuf et al. 2018, see [1]Supplementary Info., Figures, and Tables). Observed

137   Raman spectral changes from octocoral calcite can therefore be predominantly attributed to Mg.

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138   The FWHM of biogenic calcite is consistently larger than that of synthetic calcites even at very

139   similar Mg content (Bischoff et al. 1985), suggesting that physiological factors aside from Mg

140   are contributing to overall disorder. As such, the overall ν_1 FWHM signal can be partitioned into

141   Mg and non-Mg driven components (e.g., residual ν_1 FWHM, Comeau et al. 2018; DeCarlo et al.

142   2019). The residual ν_1 FWHM parameter quantifies structural disorder after the effects of Mg

143   incorporation on overall FWHM have been removed and has been employed in studies of Mg-

144   calcite crustose coralline algae as a potential qualitative proxy of calcifying fluid Ω (Comeau et

145   al. 2018; Cornwall et al. 2018, 2020).

146   The goals of this study are to separate and analyze the relative contributions of structural

147   disorder with respect to environmental gradients and variable growth effects within octocoral

148   colonies to better understand the drivers of disorder within octocoral calcite. Structural disorder

149   (Mg and non-Mg) as well as Mg content were measured within the calcitic surface skeletons of

150   three Corallidae species and one Isididae genus collected along a natural gradient in depth and

151   environmental parameters (temperature, pH, Ω, salinity) off the Kona coast of Hawai'i Island

152   within the same month. The study area took advantage of the North Pacific's naturally

153   compressed vertical oceanographic gradients and shallow calcium carbonate saturation horizons

154   (Greenwood 2009). For the depth gradient Raman measurements, structural disorder and Mg

155   content were measured from the basal portion of skeleton using Raman spectroscopy and

156   compared to corresponding *in situ* oceanographic data to determine the effect of environmental

157   factors on structural disorder while controlling for seasonal and ontogenetic growth effects.

158   Intra-sample Raman measurements were conducted on specific octocoral specimens at varying

159   branch diameters to analyze disorder with respect to variable ontogenetic growth rates under

160   fixed environmental conditions.

161

Materials and Methods

Octocoral collection and oceanographic observations

Deep-sea octocorals of the families Corallidae and Isididae were collected live along a

natural depth gradient (221–823 m, nitrate maximum at ~800 m with pH = 7.35 and DIC = 2340

μmol/kg) off the leeward coast of Hawaiʻi Island (Fig. 1, Table 1). Sample collections were

made using *Pisces IV/V* submersibles as part of the Hawaiʻi Undersea Research Laboratory

(HURL) cruises in 2011, 2016, and 2017. All Kona coast specimens were collected within the

same four-week period (August to September), thereby minimizing the impacts of seasonal

growth variations on Mg content and structural disorder (Mavromatis et al. 2013; Vielzeuf et al.

2013, 2018; Flöter et al. 2019). Moreover, the collection of specimens from the same

oceanographic regime minimizes differences in food sources (via surface primary productivity

and export) and biogeochemical conditions that can inherently confound results. By controlling

for spatial (e.g., sampling location and corresponding oceanography) and temporal (e.g., seasonal

growth) variables, environmental factors and species-specific effects can be better isolated. An

additional sample (M3) collected off Makapuʻu, Oʻahu at 417 m depth was used for the

calibration of Raman-based Mg content. All relevant octocoral specimens (N = 28) were

taxonomically identified by specialists at the University of Colombia, Bogotá (Luisa Dueñas;

Ardila et al. 2012, Dueñas et al. 2014) and the University of Zürich (Bertalan Lendvay; Lendvay

et al. 2020). Octocoral species in this sample set include *Pleurocorallium* cf. *secundum*,

Corallium tortuosum, *Hemicorallium imperiale/laauense* (a single species complex; Dueñas and

Lendvay, pers. comm.), and *Acanella* spp. (likely multiple species). Prior to geochemical

analysis, organic matter was removed from the skeletons by soaking and rinsing them in a

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184   sodium hypochlorite (bleach) and water mixture and rinsing with fresh water. For a subset of

185   octocoral skeletons to be measured using Raman and LA-ICPMS (see "Calibration of Raman Mg

186   content" and Table 1), cross sections of the basal skeleton were also cut horizontally to the

187   growth axis using a diamond saw (Isomet 1000, Buehler) and then polished using imperial

188   polishing paper (30 μm, 9 μm, 3 μm, and 1 μm; 3M Inc., Maplewood, MN, USA). No additional

189   polishing of the surface skeleton was carried out (see [1]Supplementary Info., Figures, and Tables).

190   At each sampling site, oceanographic parameters were measured from the surface down

191   to 1000 m to fully characterize water column chemistry. In the sites visited during the 2011

192   cruise (Hoʻokena, Kealakekua, Waiʻahukini), CTD hydrocasts equipped with a Sea-Bird SBE-9,

193   SBE-18 pH probe, and a rosette of 24 Niskin bottles were conducted at 4–5 stations per site

194   along a 2 km transect perpendicular to the shore. Hydrocasts provided continuous measurements

195   of temperature, conductivity (salinity), and pH as well as discrete water samples (N = 89) that

196   were stored in borosilicate glass bottles in a cool dark container until chemical analysis.

197   Additional continuous temperature, salinity, and pH data were measured using SBE-25 CTD and

198   SBE-18 pH probes attached to the submersible during sample collection. At the Kailua-Kona

199   site, temperature loggers (Onset TidbiT v2) were deployed on the seafloor at nine stations from

200   200–900 m and collected data every 30 minutes for 12 months (Sept 2016 to Aug 2017, [1]Fig.

201   S2). This additional data enabled the quantification of vertical oscillations in isopycnal depths

202   from internal tides, which aided with accurately aligning environmental parameters with

203   octocoral samples (discussed in [1]Supplementary Info., Figures, and Tables).

204   Discrete seawater samples were analyzed for carbonate system parameters (pH, DIC, TA)

205   in accordance with standard operation procedures described in Dickson et al. (2007) and

206   Riebesell et al. (2010) with appropriate modifications for differences in our analytical equipment.

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207   Dissolved inorganic carbon (DIC) was measured using an APOLLO SciTech DIC analyzer

208   (model AS-C3). Total alkalinity (TA) was measured using an automated open-cell

209   potentiometric titration method described in Dickson et al. (2007). Certified reference materials

210   (CRM) were measured routinely to confirm measurement accuracy. Spectrophotometric

211   measurements of pH were made using an m-cresol purple dye indicator on a temperature-

212   controlled spectrophotometer (Thermo Scientific^{TM} Orion^{TM} AquaMate 7000 Vis

213   Spectrophotometer). Tris standards (Dickson Lab, SIO) were used to verify dye quality and

214   measurement accuracy. Necessary temperature and salinity measurements were conducted using

215   a YSI multiparameter meter (YSI 5563). CO2SYS (Excel; Lewis and Wallace 1998) was used to

216   calculate calcite saturation state (Ω_{Cal}), [CO_3^{2-}], and [HCO_3^-] using K_1 and K_2 dissociation

217   constants from Mehrbach et al. (1973) refit by Dickson and Millero (1987), K_{HSO4} from Dickson

218   (1990), total boron from Uppström (1974), and seawater-scale pH. Potential density (σ_θ) was

219   calculated in MATLAB using the Gibbs-Seawater toolbox (McDougall and Barker 2011).

220

221   **Raman measurement and analysis of octocoral surface skeleton**

222   **Micro-Raman system.** The micro-Raman system located in the Raman and Infrared

223   Spectroscopy Laboratory at the Hawai'i Institute of Geophysics and Planetology (HIGP) was

224   used to conduct Raman measurements in accordance with procedures used in Acosta-Maeda et

225   al. (2013). The instrument is comprised of a Kaiser Micro-Raman Spectrometer with a RAMAN

226   RXN1^{TM} Microprobe (Kaiser Optical Systems Inc.) and a Leica DMLB microscope with a 100x

227   maximum magnification objective, motorized x-y stage, Kaiser HoloSpec spectrometer, and

228   Andor CCD camera. The near-infrared excitation laser uses an external-cavity Invictus diode

229   laser with a maximum power of 66 mW. The laser wavelength was set at 785 nm to reduce

230   interference from octocoral skeletal organic matter via residual fluorescence (DeCarlo et al.

231   2017, 2018; Farfan et al. 2021). The system has a spatial resolution of ~1 μm (using the 100x

232   objective) with a Raman spectral lines measurement accuracy of ±0.5 cm^{-1} and a spectral range

233   of 150–3300 cm^{-1}. Cyclohexane Raman spectra (801.3 cm^{-1} peak) were frequently measured to

234   check for instrument drift over time. The micro-Raman objective was properly focused for every

235   measurement taken. GRAMS/AI software (Thermo Fisher Scientific Inc.) was used to curve fit

236   the Raman ν_1 peak with a mixed Gaussian-Lorentzian curve that allowed the extraction of

237   Raman shift and linewidth (peak width) parameters. Calculating the final FWHM required

238   deconvolving the Raman instrument linewidth from the raw measured linewidth. Deconvolutions

239   were done via Raman measurements of Neon lamp light and the equation:

240
$$\omega_{obs}{}^2 = \omega_{actual}{}^2 + \omega_{inst}{}^2 \qquad\qquad (1)$$

241   where ω_{obs} is the observed linewidth of the Raman peak, ω_{inst} is the instrument-based linewidth,

242   and ω_{actual} is the true linewidth (FWHM) from the analyzed sample. Using relevant ν_1 peak

243   equations from Perrin et al. (2016) (see [1]Supplementary Info., Figures, and Tables), octocoral

244   Mg content was predicted from Raman shift measurements. The resulting Mg value was then

245   used to predict the FWHM signal contributed by Mg (Mg-driven structural disorder, Perrin et al.

246   2016), which was then subtracted from the overall FWHM measurement to obtain the residual

247   FWHM value (non-Mg structural disorder).

248   **Calibration of Raman Mg content.** The accuracy of the calculated Mg values based on

249   the ν_1 Raman shift equation from Perrin et al. (2016) was first verified through the pairing of

250   measurements from Raman spectroscopy and Laser Ablation Inductively Coupled Plasma Mass

251   Spectrometry (LA-ICPMS) on polished octocoral basal cross sections. Radial Mg measurements

252   were conducted on a subset of depth gradient octocorals as well as sample M3 from Makapuʻu

253   (N = 8, Table 1) using an Agilent 7700cs connected with an excimer laser (NWR-193, New

254   Wave Research) at the Atmosphere and Ocean Research Institute, The University of Tokyo.

255   Setup conditions included a 100-μm diameter laser spot, a pulse rate of 10 Hz and pulse energy

256   of 3 mJ, using He as the carrier gas. The signal intensity of ions (^{26}Mg and ^{43}Ca) was calculated

257   by subtracting the background level of trace elements, which had been obtained empirically with

258   the laser output set to 0%. Subsequently, the ratio of trace elements to ^{43}Ca was obtained and

259   compared to the signal intensities of standard reference materials (discussed in [1]Supplementary

260   Info., Figures, and Tables). Observable round craters in the cross-section surface from the 100-

261   μm laser spot were used to align the subsequent Raman measurements. Raman measurements (N

262   = 10 per marker) were taken adjacent to each crater along the transects. Raman v_1 Mg estimates

263   were compared to LA-ICPMS Mg to generate a trendline between the two data sources to

264   validate the equations from Perrin et al. (2016) and adjust Raman v_1 Mg content for the octocoral

265   samples.

266   **Octocoral surface skeleton measurements.** Raman measurements were taken on the

267   outermost, up current side of the basal portion (>3 mm diameter) of the octocoral skeleton

268   surface to capture the geochemical conditions at the time of sample collection. Octocorals tend to

269   display skewed growth patterns in their basal cross sections due to consistent variations in radial

270   growth rate in different directions (Luan et al. 2013; Vielzeuf et al. 2018). Growth rates are

271   fastest on the up current side of the corals (opposite of the feeding polyps) and produce relatively

272   thicker growth rings (Kahng, pers. obs.). The central portion of the cross sections contains the

273   fast-growing medullar region (Corallidae) or the central axis (Isididae), which is known to have

274   naturally higher Mg content than the surrounding growth rings (Vielzuef et al. 2008; Perrin et al.

275   2015; Flöter et al. 2019, Fig. 2). All depth gradient Raman measurements (N = 15 per sample: 3

276   five-point transects with ~100 μm spacing) were taken from the up current side of the thicker

277   basal portion of skeleton (as opposed to younger, thinner branches that are likely comprised of

278   mostly medullar skeleton) to reduce potential growth rate bias in structural disorder and Mg

279   measurements across the depth gradient (Fig. 2). Preliminary sample size analysis demonstrated

280   that at least N = 9 Raman measurements per octocoral sample yield a negligible change in

281   standard error values of ν_1 peak parameters ([1]Fig. S3).

282   Intra-sample Raman measurements were also taken across the surface skeleton of several

283   entire octocoral samples (*P.* cf. *secundum* from 273 m; *C. tortuosum* from 280 m; *H.*

284   *imperiale/laauense* from 444 m; *H. imperiale/laauense* from 472 m; *Acanella* spp. from 823m)

285   to investigate intra-sample variability in structural disorder and Mg content with respect to

286   ontogenetic skeletal growth rates. Because the medullar region/central axis grows relatively

287   faster than the outer annular region, it is assumed that surface skeleton grows relatively faster at

288   thinner branches where the medullar region/central axis is exposed at the surface than thicker

289   basal branches containing annular skeleton at the surface (Vielzeuf et al. 2008; Flöter et al.

290   2019). Five-point transects were taken on 20 different locations along the octocoral surface

291   skeleton (N = 100) to provide a wide coverage of thicker (slower-growing) basal regions and

292   thinner (faster-growing) branches. Branch diameter, the distance between the up current and

293   down current sides, was measured using digital calipers (0.01 mm resolution; Neiko Tools, USA)

294   and then compared to corresponding Raman measurements (ν_1 FWHM, ν_1 Raman shift Mg

295   content, residual ν_1 FWHM) to ascertain variability in structural disorder relative to ontogenetic

296   growth rates.

297   Potential Mg and non-Mg disorder patterns along the depth gradient were compared with

298   corresponding oceanographic measurements to illuminate patterns between skeletal

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299   geochemistry and environmental conditions. Univariate models were emphasized as temperature

300   and carbonate system parameters were highly collineated. All relevant statistical analyses were

301   conducted using R Studio (version 4.0.5, R Core Team 2021). For this study, significance was

302   defined as $P < 0.05$.

303

304   **Results**

305   **Environmental context**

306   Oceanographic parameters (temperature, salinity, pH, Ω_{Cal}) decrease sharply at around

307   100 m depth, signifying the transition to denser water masses ([1]Fig. S4). Environmental

308   parameters begin to stabilize past 500 m. The calcite saturation horizon ($\Omega_{Cal} = 1$) is almost

309   reached at ~600 m. Collinearity was also observed in the oceanographic data, especially between

310   pH and temperature ([1]Fig. S4). The sampled octocorals displayed a clear depth zonation pattern

311   with *P.* cf. *secundum* and *C. tortuosum* inhabiting the shallowest areas from ~220 m to ~300 m

312   and *H. imperiale/laauense* observed in deeper regions from ~400 m to ~580 m. *Acanella* spp.

313   had a wider depth distribution from ~400 m to over 800 m. Depths from 221 m to 823 m

314   experienced temperature and pH values ranging from 16.37 to 4.99 °C and 7.94 to 7.41,

315   respectively. High-resolution temperature and potential density data are discussed further in the

316   [1]Supplementary Info., Figures, and Tables section.

317

318   **LA-ICPMS corrections to v_1 Raman shift Mg content**

319   Mg content displayed considerable scatter (~2.5 mol% range) around the linear trendline:

320   LA-ICPMS Mg = (1.1116 × v_1 Raman shift Mg) + 3.0703 ($R^2 = 0.77$, N = 90)   (2)

14

321   shown in Figure 3a. There was a noticeable grouping of data points corresponding with the two

322   octocoral species sampled (*H. imperiale/laauense* and *P.* cf. *secundum*). Although the LA-

323   ICPMS Mg values closely overlapped between the two species, Raman-derived Mg values

324   showed greater differences close to 2 mol%. The trendline slope did not differ greatly from the

325   1:1 line between ν_1 Raman shift Mg and LA-ICPMS Mg, although there was an offset in Mg

326   indicated by the y-intercept where ν_1 underpredicted Mg by 3.86 ± 0.74 mol% on average.

327   Henceforth, corrections based on the linear relationship between LA-ICPMS and Raman are

328   applied to the octocoral ν_1 Raman shift Mg data (depth gradient and intra-sample measurements).

329   Raman-based Mg content appeared to increase significantly towards the medullar region of the

330   cross-section as exemplified with the slow growth axis of *P.* cf. *secundum* from 238 m (Dev.

331   Expl. $= 0.94$, $R^2 = 0.92$), which had a strong relationship with LA-ICPMS Mg data ($R^2 = 0.79$, P

332   < 0.001, Fig. 3b, c).

333

334   **Octocoral basal surface structural disorder and Mg along depth gradient**

335   **FWHM, Mg, and residual FWHM.** Octocoral ν_1 FWHM ranged from a high of $10.06 \pm$

336   0.03 cm^{-1} detected from *P.* cf. *secundum* at 221 m to a low of 9.11 ± 0.04 cm^{-1} from *H.*

337   *imperiale/laauense* at 574 m. ν_1 FWHM across all octocoral species significantly decreases with

338   increasing potential density (Fig. 4). *H. imperiale*/*laauense* ν_1 FWHM displayed a significant

339   correlation with potential density, while *Acanella* spp. did not appear to display any notable

340   FWHM patterns ([1]Table S1). *P.* cf. *secundum* ν_1 FWHM had an especially strong correlation

341   with potential density although only four samples were collected. Only two samples of *C.*

342   *tortuosum* were acquired so no species-specific trendline could be generated for any of the

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343   Raman parameters. When only Corallidae octocorals are considered, the ν_1 FWHM depth

344   gradient correlation is further strengthened.

345   Octocoral Mg content (from ν_1 Raman shift) along the depth gradient ranged from a high

346   of 11.54 ± 0.04 mol% $MgCO_3$ detected from *C. tortuosum* at 280 m to a low of 8.97 ± 0.09

347   mol% $MgCO_3$ from *H. imperiale/laauense* at 582 m. Mg content also significantly decreases

348   with increasing potential density (Fig. 4). *H. imperiale/laauense* Mg displayed a significant yet

349   weaker correlation with potential density compared to ν_1 FWHM, while *Acanella* spp. Mg

350   displayed no significant pattern with potential density. *P.* cf. *secundum* Mg content was more

351   moderately correlated with potential density compared to ν_1 FWHM ([1]Table S2). The Mg depth

352   gradient correlation is again strengthened when only Corallidae octocorals are considered.

353   Residual ν_1 FWHM ranged from 1.73 ± 0.08 cm^{-1} for *Acanella* spp. at 451 m to $1.21 \pm$

354   0.04 cm^{-1} for *H. imperiale/laauense* at 399 m and did not change significantly with potential

355   density (Fig. 4, [1]Table S3). Moreover, two-way ANOVA tests yielded no significant species-

356   specific correlations between octocoral residual ν_1 FWHM and those oceanographic parameters

357   (best interaction was between Species and Ω_{Cal}; $F_{3,20} = 1.54$, $P = 0.23$).

358

359   **Correlations with environmental data.** ν_1 FWHM was most strongly correlated with

360    $[CO_3^{2-}]$ and Ω_{Cal} (both $R = 0.96$), although temperature and pH also displayed strong correlations

361   (Fig. 5, [1]Table S1). ν_1 FWHM from *Acanella* spp. displayed no significant correlations with any

362   environmental parameter. *H. imperiale/laauense* ν_1 FWHM was strongly correlated with all

363   environmental parameters except salinity, with the strongest covariance being with TA ($R = -$

364   0.86). *P.* cf. *secundum* ν_1 FWHM was well correlated with all environmental parameters.

16

365   Mg content had similar correlation patterns compared to ν_1 FWHM although R-values

366   were lower overall. $[CO_3^{2-}]$ and Ω_{Cal} (both R = 0.90) maintained the strongest positive

367   correlations with octocoral Mg content (Fig. 6, [1]Table S2). In *Acanella* spp., the Mg content

368   displayed no significant correlations with any environmental parameter. *H. imperiale/laauense*

369   was moderately correlated with all environmental parameters except for salinity. *P*. cf. *secundum*

370   Mg was well correlated with all environmental parameters except for pH (only moderately

371   correlated), but all linear models had no significant p-values due to low sample size.

372   Residual ν_1 FWHM was weakly correlated with all environmental parameters with the

373   highest correlation occurring with pH (R = 0.34, [1]Table S3).

374

375   **Intra-sample octocoral structural disorder and Mg**

376   ν_1 FWHM of *P*. cf. *secundum* collected at 273 m consistently increases at smaller branch

377   diameters from 9.70 ± 0.05 cm^{-1} at 5.18 mm to 9.95 ± 0.04 cm^{-1} at 0.95 mm (Fig. 7a, [1]S7a). Mg

378   content is mostly scattered between 11.22 ± 0.04 mol% and 11.63 ± 0.04 mol% with a notable

379   peak of 11.87 ± 0.08 mol% at 1.77 mm (Fig. 7a, [1]S7a). Residual ν_1 FWHM also increases

380   somewhat both at the smallest branch diameters (from 1.27 ± 0.03 cm^{-1} at 5.64 mm to $1.48 \pm$

381   0.04 cm^{-1} at 0.95 mm) and at high branch diameters (e.g., 1.43 ± 0.06 cm^{-1} at 10.13 mm). In *C*.

382   *tortuosum* from 280 m, both ν_1 FWHM (9.76 ± 0.03 cm^{-1} to 9.76 ± 0.03 cm^{-1}) and residual ν_1

383   FWHM (1.36 ± 0.02 cm^{-1} to 1.59 ± 0.05 cm^{-1}) increase nonlinearly with decreasing branch

384   diameter, while Mg content increases more linearly from 11.37 ± 0.09 mol% at 7.31 mm to

385   11.75 ± 0.13 mol% at 1.45 mm ([1]Fig. S7b).

386   Within the *H. imperiale/laauense* specimen collected at 444 m, ν_1 FWHM sharply

387   increases to 9.87 ± 0.03 cm^{-1} at less than 0.5 mm compared to values of around 9.2–9.4 cm^{-1}

388   from larger diameters (Fig. 7b, [1]S7c). Mg content experiences a similar pattern with values

389   increasing to greater than 10.23 ± 0.07 mol% at diameters less than 0.58 mm. Residual ν_1

390   FWHM increases with decreasing branch diameter at a consistent rate from 1.41 ± 0.01 cm^{-1} at

391   3.67 mm to 1.80 ± 0.02 cm^{-1} at 0.39 mm. In *H. imperiale/laauense* from 472 m, ν_1 FWHM (from

392   9.14 ± 0.02 cm^{-1} at 3.98 mm to 9.47 ± 0.04 cm^{-1} at 0.73 mm) and Mg content (9.02 ± 0.07 mol%

393   to 9.56 ± 0.02 mol% at those same diameters) also increase with decreasing branch diameter at a

394   consistent rate ([1]Fig. S7d). Its residual ν_1 FWHM values follow a similar yet more scattered

395   gradual increase from 1.48 ± 0.02 cm^{-1} at 6.82 mm to 1.70 ± 0.04 cm^{-1} at 0.73 mm. The sharp

396   step-like ν_1 FWHM and Mg spikes observed within the *H. imperiale/laauense* specimens contrast

397   with the more linear and scattered changes within *P.* cf. *secundum* and *C. tortuosum*.

398   In *Acanella* spp. 823 m, a similar pattern is observed between branch diameter and Mg

399   content, which rises from 9.25 ± 0.07 mol% at 8.61 mm to 9.88 ± 0.09 mol% at 1.05 mm (Fig.

400   7c, [1]S7e). However, ν_1 FWHM (9.23 ± 0.03 cm^{-1} at 6.76 mm to 9.03 ± 0.05 cm^{-1} at 1.05 mm) and

401   residual ν_1 FWHM (1.55 ± 0.04 cm^{-1} at 6.76 mm to 1.15 ± 0.04 cm^{-1} at 1.05 mm) contain

402   strikingly different patterns with branch diameter compared to ν_1 FWHM and residual ν_1 FWHM

403   from the other octocoral specimens.

404   Species-specific patterns between branch diameter and overall ν_1 FWHM, Mg-based ν_1

405   FWHM, and residual ν_1 FWHM are apparent (Fig. 8, [1]Table S4). Within *P.* cf. *secundum* from

406   273 m, Mg-based ν_1 FWHM is moderately correlated with overall ν_1 FWHM (R = 0.54), while

407   residual ν_1 FWHM and overall ν_1 FWHM are more strongly correlated (R = 0.77). FWHM values

408   from the *H. imperiale/laauense* specimen collected at 444 m were well correlated with both Mg-

409   based (R = 0.82) and residual ν_1 FWHM (R = 0.82). Strong correlations (R = 0.72 and R = 0.79,

410   respectively) are also observed for *H. imperiale/laauense* from 472 m. ν_1 FWHM has a very

18

411   strong correlation with residual ν_1 FWHM (R = 0.95) and a strong correlation with Mg-based ν_1

412   FWHM (R = 0.82) within *C. tortuosum* from 280 m. ν_1 FWHM values measured from *Acanella*

413   spp. at 823 m are weakly correlated with Mg-based ν_1 FWHM (R = 0.33) but strongly correlated

414   with residual ν_1 FWHM (R = 0.86).

415

416   **Discussion**

417   **Quantifying Mg content from Raman measurements**

418   Raman spectral measurements of octocoral calcite should be largely representative of

419   changes in Mg content as opposed to other incorporated cations and anions (discussed in

420   Introduction). It should be noted that the range in measured Mg content within the octocorals is

421   narrow, equating to a ν_1 Raman shift range of only 1.09 cm^{-1} for the LA-ICPMS corrections, 0.59

422   cm^{-1} for the depth gradient, and 0.65 cm^{-1} for the intra-sample measurements. However,

423   assuming similar elemental ratios between Mg and other ions like Sr, B, and Ba from Vielzeuf et

424   al. (2018), one would expect to see a range of around 1.92 mmol/mol Sr/Ca, 89.64 μmol/mol

425   B/Ca, and 0.012 μmol/mol Ba/Ca for a Mg/Ca range of 97 mmol/mol. Changes in Mg should

426   again dominate the ν_1 Raman signal (e.g., based on Farfan et al. 2021, +2.54 cm^{-1} from Mg with

427   only -0.10 cm^{-1} from Sr), yet the presence of non-Mg cations could lead to underestimated ν_1

428   Raman shift Mg values. For this reason, the LA-ICPMS Mg correction is needed to obtain more

429   accurate Mg values from ν_1 Raman shift. While the results of Farfan et al. (2021) provide a

430   potential detailed look into the effects of certain divalent cations on Raman spectral analysis of

431     $CaCO_3$, a detailed analogous analysis on calcite minerals is needed.

432   The trendline (Eqn. 2) comparing LA-ICPMS and ν_1 Raman shift Mg displays a

433   consistent offset of around 3 mol% from the 1:1 line with Raman shift underestimating LA-

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434   ICPMS Mg (Fig. 3a). The Raman-Mg calibration lines from Perrin et al. (2016) are based on

435   calcites synthesized under high pressure and temperature without the presence non-Mg

436   cations/anions, organic molecules, or other physiological factors that could impact the calcite

437   Raman signal. It is possible that the presence of non-Mg cations/anions in the octocoral skeleton

438   are contributing to the underestimated v_1 Raman shift Mg values in this case. Raman shift values

439   within biogenic Mg-calcites are also known to be slightly lower than their synthetic counterparts

440   at similar Mg content (Bischoff et al. 1985). The consistent offset of the LA-ICPMS and Raman

441   trendline suggests that these biogenic factors are largely similar across the octocoral species in

442   this study. There is a notable spread of data points around the trendline that could be due to the

443   differing spatial resolutions of LA-ICPMS (100 μm) and Raman (~2 μm). Octocoral skeletons

444   are known to contain fine-scale spatial variability in Mg content (Vielzeuf et al. 2008). While

445   smaller differences in Mg content are being masked by the spatial mismatch between the two

446   methods, larger Mg differences are better resolved. For instance, Figs. 3b-c highlight the sharp

447   increase in Mg content from the skeletal surface to the medullar region within a single cross

448   section sample of *P.* cf. *secundum*, resulting in a strong LA-ICPMS and v_1 Raman shift Mg

449   trendline with relatively less scatter. Nevertheless, a strong correlation exists between the two

450   methods with a consistent offset, thus allowing v_1 Raman shift Mg content to be corrected.

451

452   **Environmental drivers of structural disorder and Mg**

453   Overall v_1 FWHM and Mg content both decrease across the environmental gradient,

454   indicating an overall decrease in structural disorder correlating strongly with decreasing

455   temperature and carbonate system parameters. The relationship between potential density (depth)

456   and v_1 FWHM and Mg content agrees with prior findings along environmental gradients

457   (Andersson et al. 2008; Thresher et al. 2011). Lower temperatures at greater depth result in a

458   thermodynamically more soluble Mg-calcite structure relative to warmer, shallower waters

459   (Mucci 1987; Morse et al. 2007). Conditions less conducive to calcification at depth (reduced

460   pH, $[CO_3^{2-}]$, and Ω_{Cal}) would also lead to thermodynamically less stable calcite, which

461   discourages Mg incorporation (Mackenzie et al. 1983). Although the relationship between calcite

462   Mg content and temperature has received relatively more attention, carbonate system parameters

463   maintain the highest correlations with ν_1 FWHM and Mg in this study with strong temperature

464   correlations as well.

465   Non-Mg driven structural disorder measured using residual ν_1 FWHM displayed no clear

466   pattern with depth or environmental parameters. Residual ν_1 FWHM decreases slowly with depth

467   only among the shallowest *C. tortuosum* and *P.* cf. *secundum* specimens, yet the variable

468   clustering from deeper *H. imperiale/laauense* and Isididae disrupts any overall trends. The lack

469   of significant change in non-Mg disorder with depth suggests that the octocoral calcite stability

470   influenced by vital effects or growth rate kinetics not involving Mg are not sensitive to

471   environmental changes. In contrast, ν_1 FWHM and Mg display strong patterns with depth, which

472   implies that skeletal structural disorder is due largely to Mg incorporation. If the majority of non-

473   Mg structural disorder is driven by variations in growth rate, then Mg should dominate the ν_1

474   FWHM signal given that seasonal growth rates are controlled through a consistent sampling

475   protocol over time and location within the sample (basal portion of surface skeleton). Another

476   possibility is that basal skeleton growth rate itself does not change significantly with depth,

477   which has been observed by Thresher et al. (2016) for Isididae.

478   ν_1 FWHM could also be positively correlated with extracellular calcifying fluid Ω (Ω_{ECF})

479   due to the well-recognized positive relationship between calcite growth rates and Ω (Morse et al.

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480   2007). For biogenic calcifiers, higher Ω_{ECF} should drive faster skeletal accretion rates, which

481   would increase structural disorder and ν_1 FWHM values. While this relationship has been well

482   demonstrated for aragonite and used to quantify shallow water coral Ω_{ECF} (DeCarlo et al. 2019),

483   it has not yet been verified for Mg-calcite. Prior studies have used residual ν_1 FWHM as a

484   qualitative proxy for Ω_{ECF} within Mg-calcite calcifiers (e.g., crustose coralline algae, Comeau et

485   al. 2018; Cornwall et al. 2018, 2020) assuming that the Mg-calcite FWHM-Ω_{ECF} relationship

486   operates similarly to that of synthetic and coral aragonite (DeCarlo et al. 2017; Farfan et al.

487   2021). If true, then the results here would indicate that Ω_{ECF} is not sensitive to the environmental

488   gradient. While this result would have interesting implications for octocoral responses to

489   contemporary carbonate chemistry gradients as well as ongoing ocean acidification, a systematic

490   calibration between calcite Ω_{ECF} and Raman spectral parameters (e.g., FWHM) is needed to

491   confirm these results.

492   Mg-temperature relationships in this study align with relationships from prior studies of

493   synthetic and biogenic calcite (Fig. 9, [1]Table S6). The linear expression for skeletal Mg/Ca as a

494   function of temperature based on all available octocoral samples from this study ($R^2 = 0.70$, N =

495   28) as well as the one derived exclusively from Corallidae ($R^2 = 0.73$, N = 23) either overlapped

496   or were similar to those established for *Corallium* spp., *C. rubrum*, and *Corallium* spp. compiled

497   in previous studies (Chave 1954; Weinbauer et al. 2000; Yoshimura et al. 2011; Vielzeuf et al.

498   2013; Thresher et al. 2016; Chaabane et al. 2019). Only the model from Weinbauer et al. (2000)

499   ($R^2 = 0.99$, N = 3) had a very different slope and intercept. Isididae from this study ($R^2 = 0.20$, N

500   = 5) did not display a strong relationship albeit with a low sample size. The Isididae model

501   generated by Thresher et al. (2016) ($R^2 = 0.87$, N = 73) was similar to prior Corallidae models.

502   Inorganic high-Mg calcite experiments under varying temperatures (5–40°C) also yielded a slope

22

503   overlapping with the above octocoral studies although the intercept was much lower (Mucci

504   1987). The comparable Mg-temperature slopes between organic and inorganic calcite suggest a

505   similar kinetic Mg incorporation mechanism (Thresher et al. 2016). Meanwhile, differences in

506   intercept values among different octocoral Mg models could result from species-specific effects.

507   The lack of a consistent intra-annual Mg-temperature relationship in prior studies is likely driven

508   by interference from vital effects amplifying Mg content (Vielzeuf et al. 2013). Nevertheless,

509   vital effects can be a product of temperature as well as other environmental parameters. While

510   seasonal fluctuations in skeletal growth rate appear to drive intra-annual Mg variability, longer-

511   term mean temperature has been correlated with average octocoral Mg content (Chaabane et al.

512   2019), a result that could be observed here along the depth gradient.

513

514   **Growth rate effects on structural disorder and Mg**

515   Within individual octocoral specimens, the change in residual ν_1 FWHM with respect to

516   branch diameter differs from that of Mg content despite experiencing the same environmental

517   history. Intra-sample Raman measurements can examine structural disorder as a function of

518   changes in skeletal growth rate driven entirely by organismal physiology and not by

519   environmental conditions. Octocoral branches grow lengthwise and radially (branch

520   width/diameter) over time. As branches increase in length, the medullar or central axis skeleton

521   is precipitated first. As the branch thickens via radial growth, the medullar skeleton eventually

522   transitions to an annular skeleton containing observable growth rings (Fig. 2). Because growth

523   rates are relatively higher within the interior medullar zone and central axis of Corallidae and

524   Isididae skeleton, respectively (Vielzeuf et al. 2008; Flöter et al. 2019), radial growth rates of

525   surface skeleton should be faster for smaller branch diameters and relatively slower at larger

23

526   ones. This pattern is further implied by the increase in residual ν_1 FWHM (non-Mg disorder)

527   with decreasing branch diameter (Fig. 7). Linear changes in residual ν_1 FWHM with respect to

528   branch diameter, as opposed to stepwise changes observed for Mg discussed later, could be a

529   result of ontogenetic growth effects where radial growth rates naturally and gradually decrease

530   with age and size (Sinclair et al. 2011; Farmer et al. 2015; Flöter et al. 2019). Thicker, older

531   sections of the branch with relatively slower surface skeletal growth rates would have less non-

532   Mg structural disorder compared to thinner, younger branches. The differing residual ν_1 FWHM

533   pattern observed within *Acanella* spp. (Fig. 7c, [1]S7e) is puzzling since greater non-Mg disorder

534   was also expected within the faster growing central axis of Isididae. This inverted pattern is also

535   well manifested in the overall ν_1 FWHM signal despite Mg content increasing with decreasing

536   branch diameter. Further intra-sample analysis of *Acanella* spp. surface skeleton is needed to

537   confirm whether this outcome is simply an anomaly or the result of a consistent taxon-specific

538   mineralogical difference. Ω_{ECF} could also at least partially contribute to these intra-sample

539   patterns although such interpretations must be approached with caution. It should also be noted

540   that ontogenetic skeletal growth rates are not necessarily the same as crystal growth rates which

541   influence structural disorder and Mg incorporation. For instance, crystal growth rates could occur

542   at the same gradual rate as the bulk skeleton or in spontaneous pulses (Vielzeuf et al. 2018). As

543   branch diameter (i.e., skeletal growth rate) varies, correlations between residual ν_1 FWHM and ν_1

544   FWHM are comparable to that of Mg content and ν_1 FWHM (Fig. 8, [1]Table S4). Non-Mg growth

545   rate kinetics appears to contribute more towards overall intra-sample structural disorder relative

546   to the environmental gradient, where only Mg and ν_1 FWHM were significantly correlated (Fig.

547   4).

24

548   Stepwise increases in Mg content with decreasing octocoral branch diameter are

549   indicative of faster-growing medullar or central axis skeleton, which typify ontogenetic growth

550   patterns in Corallidae and Isididae (Vielzeuf et al. 2008; Sinclair et al. 2011). Significant

551   increases in Mg content were observed with decreasing branch diameter, especially at diameters

552   smaller than 2 mm (Fig. 7). This value agrees with spatially mapped Electron Microprobe

553   Analysis (EMPA) data of Corallidae (*C. rubrum*) cross sections where the diameter of the

554   medullar region containing higher Mg was 1.0–2.0 mm with an irregular shape (Vielzeuf et al.

555   2008; Fig. 2, 3b). For Isididae, the central axis is known to be much rounder with a maximum

556   diameter of 1.0 mm (Flöter et al. 2019). The outer edge of the central axis skeleton in *Acanella*

557   spp. was likely detected through the sharp Mg peak at 1.05 mm diameter. As observed with

558   residual v_1 FWHM, the early growth medullar zone would intuitively be exposed at the branch

559   tip surface, resulting in Mg spikes at the smallest branch diameters. Elevated Mg content in these

560   regions is likely a product of kinetic effects driven by skeletal growth rates (reviewed in Vielzeuf

561   et al. 2018). In scleractinian skeletons, Gagnon et al. (2007) noted that high Mg regions (also

562   known as Centers of Calcification) were influenced by non-Rayleigh processes in contrast to the

563   rest of the skeleton. Mechanisms such as the growth entrapment model (Watson 2004) as well as

564   a more modified impurity incorporation/repulsion model (Vielzeuf et al. 2018) have been

565   proposed to explain the connection behind precipitation rate and Mg incorporation. The presence

566   of differing organic molecules within rapid growth zones is also thought to selectively enhance

567   Mg incorporation (Gagnon et al. 2007). Regardless of the exact mechanism, these differing

568   growth regions demonstrate the need to focus comparative octocoral surface Mg measurements

569   at the thicker basal portions of skeleton to avoid accidental sampling of the medullar zone.

570

Species-specific patterns

571

572   Analysis of intra-sample Mg and non-Mg structural disorder reveals potentially different

573 biomineralization patterns among closely related Corallidae species (*H. imperiale/laauense, C.*

574 *tortuosum, P.* cf. *secundum*; Ardila et al. 2012) as well as Isididae. Throughout most of the intra-

575 sample specimens, Mg-based v_1 FWHM and residual v_1 FWHM are both well correlated with

576 overall v_1 FWHM although the strength of these correlations varies by species. Some species

577 patterns (e.g., *P.* cf. *secundum*) appear to be more strongly influenced by non-Mg sources

578 compared to others (e.g., *H. imperiale/laauense*) (Fig. 8, [1]Table S4). Mg and residual v_1 FWHM

579 are positively correlated within *C. tortuosum* and *H. imperiale*/*laauense* while being negatively

580 correlated in *P.* cf. *secundum*. The high variability between the two parameters, however,

581 resulted in no significant trends except for *C. tortuosum*. *P.* cf. *secundum* and *C. tortuosum* also

582 displayed relatively lower residual v_1 FWHM values despite having the highest Mg content. Mg

583 content and residual v_1 FWHM were expected to be positively correlated since both are

584 positively impacted by skeletal growth rates (DeCarlo et al. 2017; Vielzeuf et al. 2018). This

585 outcome is only seen within *C. tortuosum* where both Mg and non-Mg structural disorder

586 strongly contribute to overall structural disorder pattern. The lower residual v_1 FWHM values

587 within the *P.* cf. *secundum* and *C. tortuosum* specimens could be evidence of a species-specific

588 biomineralization effect given the concurrent similarity in v_1 FWHM results between the two *H.*

589 *imperiale/laauense* specimens. Despite this similarity in v_1 FWHM, *C. tortuosum* is

590 taxonomically more closely related to *H. imperiale/laauense* than *P.* cf. *secundum* (Ardila et al.

591 2012). Another instance of taxon-specific biomineralization is observed with *Acanella* spp.,

592 which displayed unforeseen v_1 FWHM and residual v_1 FWHM patterns with branch diameter.

593 Although these results were surprising considering the known radial growth patterns for Isididae

594 (Flöter et al. 2019), *Acanella* spp. resides in a different taxonomic family compared to the

595 remaining specimens.

596 LA-ICPMS and ν_1 Raman shift Mg data also display indications of species-specific

597 patterns in octocoral Mg and non-Mg disorder. *H. imperiale/laauense* and *P.* cf. *secundum* cross

598 sections from multiple depths were used in the analysis of LA-ICPMS and ν_1 Raman shift Mg

599 data. Despite an overlap of LA-ICPMS Mg content between the two species (~10.5–11.2 mol%),

600 Mg content based on ν_1 Raman shift forms two distinct species-specific groups (Fig. 3a).

601 Grouping is also observed with corresponding ν_1 FWHM data, although residual ν_1 FWHM does

602 not appear to show any significant patterns ([1]Fig. S8). Like the intra-sample analysis, residual ν_1

603 FWHM values within *P.* cf. *secundum* are similar to those of *H. imperiale/laauense* despite the

604 former having greater ν_1 FWHM and Mg content than the latter ([1]Fig. S9 and [1]Table S5). Overall

605 structural disorder and vibrational frequencies are greater in *P.* cf. *secundum* compared to *H.*

606 *imperiale/laauense* even at areas with the same LA-ICPMS Mg content. Increasing calcite unit

607 cell volumes are thought to lead to lower vibrational frequencies (lower Raman shift) and

608 structural disorder (Bischoff et al. 1985). In this case, *H. imperiale/laauense* specimens could

609 have characteristically larger unit cell volumes compared to *P.* cf. *secundum* specimens. As

610 previously mentioned, the general discrepancies between LA-ICPMS and ν_1 Raman shift Mg

611 content could be attributed to physiological mechanisms absent from the synthetic calcites used

612 in Perrin et al. (2016). Both ν_1 Raman shift and ν_1 FWHM-based trendlines are mostly offset

613 from the 1:1 line through the y-intercept, suggesting that physiological factors contributing to the

614 Mg discrepancy could be largely similar across different octocoral individuals and species.

615 However, the slope of the ν_1 FWHM trendline (0.79 ± 0.04) is offset more compared to that of ν_1

616 Raman shift (1.11 ± 0.06), indicating that different parts of the LA-ICPMS/Raman FWHM

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617   signal (e.g., from different species) are being impacted differentially. The Mg-driven disorder

618   appears to be relatively higher for *P.* cf. *secundum* compared to *H. imperiale/laauense*.

619   Skeletal morphologies could also explain observed differences in Mg and non-Mg driven

620   disorder with respect to changes in branch diameter and growth rate. Stepwise increases in Mg

621   content are apparent within *H. imperiale/laauense* and *Acanella* spp. but occur to a lesser extent

622   within *P.* cf. *secundum* and *C. tortuosum* (Fig. 7, [1]S7).For instance, *C. tortuosum* exhibits linear

623   (or close to linear) trendlines for Mg and non-Mg structural disorder with respect to branch

624   diameter, which could be caused by a more gradual change in disorder and Mg content between

625   the medullar and annular zones (as opposed to sharper transitions in *H. imperiale/laauense*).

626   Faster-growing medullar zones may not be well defined for some individual octocorals, thereby

627   resulting in a less obvious change in structural disorder and Mg. While the intra-sample *P.* cf.

628   *secundum* specimen displayed inconsistent Mg patterns, one specimen of *P.* cf. *secundum*

629   analyzed by LA-ICPMS and Raman displayed a sharp radial increase in Mg content (Fig. 3b, c).

630   A possible explanation of this phenomenon could be connected to skeletal morphology rather

631   than inherent species-specific differences in medullar skeleton characteristics. For instance, the

632   robust skeletons of *P.* cf. *secundum* (as well as *C. tortuosum*) quickly transition from thick basal

633   branches to warped irregular branch tips compared to the smoother transitions in *H.*

634   *imperiale/laauense* (Fig. 7, [1]S7). Less consistent Mg patterns with branch diameter could be a

635   product of inconsistencies in the assumed relationship between ontogenetic growth rate and

636   branch diameter caused by the irregular skeletal morphology itself. Based on the patterns

637   observed with *P.* cf. *secundum* and *C. tortuosum*, branch diameter may not be as reliable of a

638   growth rate metric for the irregular morphologies of those species compared to *H.*

639   *imperiale/laauense*. Whether the formation of this differing skeletal morphology significantly

28

640   alters any of the mechanisms (e.g., growth kinetics, non-Rayleigh mechanisms, biomolecules)

641   thought to influence structural disorder and Mg content within the medullar zone is unknown

642   (Gagnon et al. 2007; Tambutté et al., 2011).

643

644   **Implications**

645   **ν_1 FWHM, Mg, and calcite mineral stability**

646   Mg measurements alone do not appear to entirely capture the factors impacting biogenic

647   Mg-calcite disorder and therefore stability. Mg has been demonstrated to be a major driver of

648   structural disorder within synthetic and biogenic calcites, although other sources of disorder can

649   originate from non-Mg sources such as organic molecules and other physiological effects from

650   calcification (Bischoff et al. 1985; Perrin et al. 2016). Data from this study has shown instances

651   where octocoral specimens of very similar Mg content differed in their overall ν_1 FWHM, a

652   result likely to occur from additional structural disorder from non-Mg sources (residual ν_1

653   FWHM). For example, while the intra-sample Mg content from *P*. cf. *secundum* and *C*.

654   *tortuosum* overlapped heavily, ν_1 FWHM values from the latter often exceeded that of the former

655   (Fig. 7, [1]S7). ν_1 Raman shift Mg and ν_1 FWHM data from the depth gradient samples are well

656   correlated but express a nonlinear pattern indicating that there are factors influencing ν_1 FWHM

657   differently from Mg. The EMPA-based Mg and ν_1 FWHM calibration lines from Perrin et al.

658   (2016) are also nonlinear as opposed to their strongly linear ν_1 Raman shift Mg calibration lines.

659   Intra-sample ν_1 FWHM and Mg correlations also varied strongly among different species, which

660   is likely driven by variable species-specific skeletal growth rates. Clearly, non-Mg structural

661   disorder can impact the overall ν_1 FWHM signal independently from Mg.

662   Raman spectroscopic measurements of calcite ν_1 FWHM could theoretically provide a

663   quantitative proxy for solubility although empirical validations must first be performed. Mg has

664   served as a prominent proxy for synthetic and biogenic calcite solubility through its correlations

665   with carbonate chemistry along depth and latitudinal gradients (Andersson et al. 2008; Lebrato et

666   al. 2016). While Mg-based metrics of solubility are valid for synthetic calcites, biogenic calcites

667   are subject to significant variability in solubility estimates even for specimens of similar Mg

668   content (Morse et al. 2006). Because ν_1 FWHM can quantify overall structural disorder within

669   carbonate minerals (Bischoff et al. 1985), the spectral parameter could provide a robust method

670   to incorporate non-Mg factors influencing calcite stability and therefore solubility. Increased

671   accuracy in determining biogenic calcite solubility could help determine which calcifying

672   organisms are most at risk from stressors like ocean acidification. Despite the logical connection

673   between structural disorder and mineral solubility, no experiments to date have quantified the

674   relationship between calcite solubility and ν_1 FWHM.

675

676   **Species-specific biomineralization**

677   Biomineralization patterns differ by species, especially between those residing in

678   shallower and deeper water, respectively. Certain specimens (*H. imperiale/laauense*) displayed

679   structural disorder and Mg values anticipated from a sharp radial change in skeletal growth zones

680   (i.e., from annular to medullar). However, changes in structural disorder were less striking in

681   other specimens (*P.* cf. *secundum*, *C. tortuosum*) with some patterns (*Acanella* spp.) being

682   entirely different from expectations. Such differences could be linked to vital effects involving

683   organic molecules or larger calcification mechanisms that influence the specific manner of

684   calcite precipitation and subsequent Mg incorporation. Further clarification of these species-

685   specific biomineralization patterns could provide insight as to how certain species precipitate

686   skeleton differently than others as well as how such biominerals will react to future

687   environmental change. These inter-species differences in biomineralization may also impact how

688   useful different species are for paleo-proxy applications (reviewed in Thompson 2022).

689

690   **Mg paleo-proxies**

691   A better understanding of calcite structural disorder will clarify applications of Mg paleo-

692   proxies. Vital effects interfere with the correlation between skeletal Mg and temperature since

693   faster growth also enhances Mg content (Vielzeuf et al. 2018). This interference is particularly

694   apparent in temperate waters where growth rate and temperature covary seasonally. The results

695   from this study demonstrate the effect that growth rates on different segments of octocoral

696   skeleton can have on Mg content despite experiencing the same environmental conditions.

697   Correcting Mg content for vital effects will likely require the quantification and removal of

698   growth rates with respect to changes in Mg to isolate the purely abiotic environmental effects.

699

700

31

708   and L. Dueñas at the Universidad de Los Andes in Colombia and B. Lendvay at the University of

709   Zürich for morphological and genetic identification of the octocoral specimens. Sample M3 was

710   provided by F. Parrish from the Pacific Islands Fisheries Science Center.

711

712

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948

949   **Figure Captions**

950

951 **Figure 1.** Octocoral and oceanographic sampling sites including the Makapuʻu site for sample

952 M3.

953

954 **Figure 2.** Spatially mapped relative Mg content (color-coded, count/count) of a Corallidae

955 octocoral cross section taken using Electron Microprobe Analysis (EMPA, see Supplementary[1]

956 Info., Figures, and Tables) and a corresponding photograph of the sample. The central medullar

957 region of skeleton contains elevated Mg content and is surrounded by annular skeleton

958   (annotated in the EMPA image). Up current and down current sides of axial skeleton are

959   annotated in the photograph.

960

961   **Figure 3. (a)** Relationship between Mg content calculated from ν_1 Raman shift (Perrin et al.

962   2016) and Mg measured using LA-ICPMS (N = 90). The 1:1 line is shown in black. **(b)** Radial

963   measurements of octocoral Mg content from *P*. cf. *secundum* 238 m (slow-axis) taken using the

964   Raman ν_1 peak (black, N = 26) and LA-ICPMS (blue, N = 26) **(c)** as well as the relationship

965   between the two. All error bars represent ±1 SD with the shaded regions representing the 95%

966   CI. Note that the data from subplot **(b)** was taken along a radial transect from the cross section

967   surface to the inner medullar region (see Fig. 2 for a visual example).

968

969   **Figure 4.** Mean ν_1 FWHM, Mg content (from ν_1 Raman shift), and residual ν_1 FWHM of

970   octocoral surface skeleton (N = 28) with respect to potential density (an analog of depth). Error

971   bars represent ±1 SD and the shaded regions represent the 95% CI.

972

973   **Figure 5.** Mean ν_1 FWHM of octocoral surface skeleton with respect to major oceanographic

974   variables covered in this study (N = 28). Error bars represent ±1 SD and the shaded regions

975   represent the 95% CI.

976

977   **Figure 6.** Mean Mg content (from ν_1 Raman shift) of octocoral surface skeleton with respect to

978   major oceanographic variables covered in this study (N = 28). Error bars represent ±1 SD and the

979   shaded regions represent the 95% CI.

980

981   **Figure 7.** Mean intra-sample octocoral v_1 FWHM, Mg content (from v_1 Raman shift, abbreviated

982   as "RS"), and residual v_1 FWHM with respect to octocoral branch diameter for three of the five

983   specific samples (each with N = 100) **(a)** *P.* cf. *secundum* from 273 m; **(b)** *H. imperiale/laauense*

984   from 444 m; **(c)** *Acanella* spp. from 823 m. All specimens are shown in [1]Fig. S7. Each

985   measurement location (a five-point transect) is shown by the red markers. Branch diameter

986   serves as a proxy for ontogenetic growth rates (e.g., smaller branches = faster skeletal growth). A

987   general additive model was used for the trendlines where error bars are ±1 SD, shaded regions

988   are the 95% CI, and DevExpl represents the deviance explained by the model.

989

990   **Figure 8. (a)** Mean v_1 FWHM compared to mean Mg-based v_1 FWHM, **(b)** mean v_1 FWHM

991   compared to mean residual v_1 FWHM, and **(c)** mean Mg content (from v_1 Raman shift) compared

992   to mean residual v_1 FWHM for the five octocoral specimens used during the intra-sample

993   variability measurements (N = 100 for each specimen). Error bars are ±1 SD while shaded

994   regions are the 95% CI.

995

996   **Figure 9:** Mg-temperature relationships from this study and other relevant octocoral or inorganic

997   calcite studies (authors annotated). Data from this study are shown with triangle markers. Shaded

998   regions represent the 95% CI around the trendlines. The red dashed line represents the

999   relationship from Mucci (1987). N-values and trendline statistics are displayed in [1]Table S6.

1000   A: Chave 1954; Weinbauer et al. 2000; Yoshimura et al. 2011; Thresher et al. 2016

1001   B: Yoshimura et al. 2011; Vielzeuf et al. 2013; Chaabane et al. 2019

1002

1003

1004 **Deposit Items**

1005 **[1]Deposit Item, Supplementary Info.**

1006 **Oceanographic data adjustments using seawater potential density**

1007   Oceanographic data were adjusted as the CTD casts and discrete bottle samples only

1008 provide a depth-based snapshot of mid-water environmental conditions. While sessile benthic

1009 organisms like octocorals are fixed at depth, oceanographic parameters are tightly fixed to

1010 density-specific water masses, which can undergo constant vertical oscillations from physical

1011 forcing, thereby exposing octocorals to a range of environmental conditions over short

1012 timescales. Inertial oscillations in seawater masses are apparent through high-resolution

1013 temperature data collected from Kailua-Kona, which can be driven by factors such as lunar tides

1014 and larger mesoscale eddies. Such oscillations also visibly attenuate with increasing depth and

1015 can influence oceanographic measurements conducted at a fixed depth depending on the time of

1016 sampling. The temperature time-series data, as well as correlations between temperature and

1017 potential density (σ_θ, kg/m^3) from the CTD casts, provided insight on the magnitude of water

1018 mass oscillations over different depths and times and allowed for CTD measurement

1019 adjustments. A temperature and σ_θ correlation converted high-resolution temperature data over

1020 multiple depths into high-resolution σ_θ data which was then averaged by depth over a 24-hour

1021 composite period (e.g., changes in σ_θ relative to mean σ_θ ($\Delta\sigma_\theta$) as a function of time of day). The

1022 depth and timing of specific oceanographic measurements were inserted into the 24-hour

1023 averaged curves to calculate $\Delta\sigma_\theta$. A more accurate σ_θ ($\sigma_{\theta,\text{True}}$) was then calculated through

1024 $$\sigma_{\theta,\text{True}} = \sigma_{\theta,\text{Measured}} + \Delta\sigma_\theta \qquad \text{(S1)}$$

1025 which approximated a more accurate corresponding oceanographic parameter depending on the

1026 relationship between said parameter and $\sigma_{\theta,\text{True}}$.

45

1027   Temperature data from multiple depths ranging from 220 to 900 m depict short-term (M_2

1028   semidiurnal tidal cycles; Merrifield and Holloway 2002) and long-term (mesoscale eddies; Calil

1029   et al. 2008) variability resulting from the oscillations of water masses with different densities

1030   over a yearlong interval ([1]Fig. S2). Deviations from the average potential density were relatively

1031   small (~0.01 kg/m^3 at most) and resulted in few adjustments in depth-based oceanographic data

1032   ([1]Fig. S5). Power spectral density analysis revealed consistent tidal patterns dominated by the M_2

1033   semidiurnal tide, which decreased rapidly with increasing depth ([1]Fig. S6). Average temperatures

1034   decreased rapidly after the 220 m logger (average temperature of 16.29 ± 1.32 °C compared to

1035 11.02 ± 0.79 °C from 302 m) but then stabilized at 506 m and greater (6.80 ± 0.29 °C).

1036

1037   **LA-ICPMS Reference Materials**

1038   For LA-ICPMS measurements, the reference material NIST SRM 612 was used as an

1039   external standard, with the reference materials coral powder JCp-1 and giant clam powder JCt-1

1040   (National Institute of Advanced Industrial Science and Technology, Tsukuba, Japan) measured

1041   to confirm accuracy. The reference materials were analyzed three times before and after every

1042   50–100 sample spots to monitor instrument drift. The Mg/Ca ratio calculations were based on the

1043   analyses of NIST SRM 612, standard glass, and reproducibility (percent relative standard

1044   deviation [%RSD]), measured at <5%. The Mg/Ca ratio of JCp-1 (certified value of Mg/Ca = 4.2

1045   mmol/mol) was determined to be Mg/Ca = 4.5 ± 0.2 (1σ, N = 26) mmol/mol. Detailed analytical

1046   conditions are reported in the literature (e.g., Murakami-Sugihara et al. 2019).

1047

1048   **Raman measurements of octocoral Mg and FWHM using the v_1 peak**

46

1049   The v_1 peak is the strongest in the Mg-calcite spectrum and has been measured in

1050   numerous studies of biogenic aragonite (Kamenos et al. 2013, 2016; Hennige et al. 2015; Pauly

1051   et al. 2015; DeCarlo et al. 2017, 2018, 2019; Comeau et al. 2018; Cornwall et al. 2018, 2020;

1052   Farfan et al. 2018, 2021) and Mg-calcite (Bischoff et al. 1985; Urmos et al. 1991; Borromeo et

1053   al. 2017; Comeau et al. 2018; Cornwall et al. 2018, 2020). v_1 has also been measured for

1054   amorphous calcium carbonate (Wang et al. 2012) and was noted by Borromeo et al. (2017) to be

1055   the most reliable peak (along with v_4) for distinguishing Mg content in biogenic calcites. While

1056   the v_1 peak is a non-degenerate (singlet) peak corresponding to the symmetric stretching

1057   vibrational mode of oxygen with respect to the carbon atom, high Mg content (~30 mol%) is

1058   known to cause asymmetry in the v_1 peak leading to the fitting of two peaks (Perrin et al. 2016

1059   supplementary materials). However, the octocoral Mg content measured in this study does not

1060   exceed 14.5 mol%, meaning that the v_1 peak doubling phenomenon and resulting inaccuracies in

1061   fitting Raman shift and FWHM parameters should be minimal.

1062   The intensities of the Raman spectral lines are sensitive to crystallite orientation within

1063   the analyzed sample. However, the positions (Raman shift) and line widths (FWHM) are

1064   independent of crystallite orientation. Because of this reason, Raman spectroscopy is considered

1065   a suitable technique for analyzing octocoral samples without polishing the surface skeleton.

1066   Polishing the surface skeleton would require removing sample material which would disrupt the

1067   connection between skeletal geochemistry and environmental to be measured at the time of

1068   octocoral sample collection.

1069   The Mg content of the octocoral surface skeleton was calculated using the v_1 peak

1070   calibration lines from Perrin et al. (2016). The calibration line for v_1 Raman shift was

1071   Raman shift $(cm^{-1}) = 0.256 \times [MgCO_3 \text{ (mol\%)}] + 1085.71$ ($R^2 = 0.988$, N = 20)   (S2)

1072   which covers a range of 0 to 50 mol% $MgCO_3$. Overall ν_1 FWHM can be partitioned into a Mg-

1073   driven component (Mg-driven ν_1 FWHM) and a non-Mg component (residual ν_1 FWHM). Mg-

1074   driven ν_1 FWHM is calculated by applying the Mg content from the ν_1 Raman shift calibration

1075   line to the ν_1 FWHM calibration line:

1076   FWHM (cm^{-1}) = -0.00787 × [$MgCO_3$ (mol%)]^2 + 0.51 × [$MgCO_3$ (mol%)] + 3.61

1077   (R^2 = 0.973, N = 15)   (S3)

1078   Residual ν_1 FWHM is the difference between the overall ν_1 FWHM signal and Mg-driven ν_1

1079   FWHM.

1080   Since concentrations and variability of Mg in octocoral skeleton are notably greater than

1081   that of other seawater ions (e.g., Sr, Ba, B), observed changes in ν_1 Raman shift should be

1082   predominantly driven by incorporated Mg. For instance, octocoral Sr/Ca and B/Ca had ranges of

1083   only around 1.26 mmol/mol and 59 μmol/mol, respectively (Farmer et al. 2015; Vielzeuf et al.

1084   2018). Using the ratios from Farfan et al. (2021), this translates to Raman shift decreases of

1085   0.069 cm^{-1} and 0.023 cm^{-1}, which are small compared to the expected 1.67 cm^{-1} increase from

1086   Mg based on Mg/Ca ranges (~64 mmol/mol) from Vielzeuf et al. (2018). A similar outcome is

1087   reached with respect to FWHM for those ranges in Mg/Ca (+1.91 cm^{-1}), Sr/Ca (-0.047 cm^{-1}), and

1088   B/Ca (-0.018 cm^{-1}). Farfan et al. 2021 did not measure Ba/Ca, but its concentration in octocorals

1089   (range of 8.1 μmol/mol) is less than that of B/Ca (Vielzeuf et al. 2018).

1090

1091   **EMPA measurement of octocoral cross section**

1092   High-resolution spatial patterns in Mg were measured within single cross section sample

1093   of Corallidae (species unknown, Taiwan). The octocoral cross section was first polished and

1094   coated with Pt-Pd before analysis. The two-dimensional distribution of Mg and Ca of the cross

48

1095   section was observed using an electron microprobe analyzer (JXA-8900, JOEL) at the

1096   Atmosphere and Ocean Research Institute, The University of Tokyo. The analysis was carried

1097   out with the following settings: 15 kV accelerating voltage; 200 nA probe current; 150 sec dwell

1098   time; and 10 μm probe diameter. The distribution of Mg/Ca count ratios was visualized using

1099   ImageJ software (NIH, USA).

1100

1101   **Tables**

1102   **Table 1:** Metadata for all depth gradient octocoral samples and M3 from Makapuʻu.

Sample ID#	Date/Time Sampled	Depth (m)	Location	Species	LA-ICPMS?
108	9/5/2016, 3:34pm	221	Kailua-Kona	*Pleurocorallium* cf. *secundum*	Yes
107	9/5/2016, 2:29pm	238	Kailua-Kona	*Pleurocorallium* cf. *secundum*	Yes
204	9/5/2016, 3:12pm	269	Kailua-Kona	*Pleurocorallium* cf. *secundum*	No
106	9/5/2016, 2:06pm	273	Kailua-Kona	*Pleurocorallium* cf. *secundum*	No
1105	9/28/2011, 12:33pm	280	Kealakekua	*Corallium tortuosum*	No
105	9/5/2016, 1:40pm	293	Kailua-Kona	*Corallium tortuosum*	No
1124	9/30/2011, 2:00pm	395	Waiʻahukini	*Hemicorallium imperiale/laauense*	Yes
1126	9/30/2011, 2:44pm	396	Waiʻahukini	*Acanella* spp.	No
1101	9/28/2011, 1:45pm	399	Kealakekua	*Hemicorallium imperiale/laauense*	Yes
1112	9/29/2011, 11:28am	401	Hoʻokena	*Hemicorallium imperiale/laauense*	No
1114	9/29/2011, 12:22pm	402	Hoʻokena	*Hemicorallium imperiale/laauense*	No
104	9/5/2016, 1:16pm	405	Kailua-Kona	*Acanella* spp.	No
1121	9/30/2011, 9:44am	444	Waiʻahukini	*Hemicorallium imperiale/laauense*	No
1120	9/30/2011, 8:56am	445	Waiʻahukini	*Hemicorallium imperiale/laauense*	No
1104	9/28/2011, 10:00am	446	Kealakekua	*Hemicorallium imperiale/laauense*	No
314	8/30/2017	447	1919 Lava Flow*	*Hemicorallium imperiale/laauense*	No
1109	9/29/2011, 9:18am	449	Hoʻokena	*Hemicorallium imperiale/laauense*	No
1110	9/29/2011, 10:01am	449	Hoʻokena	*Hemicorallium imperiale/laauense*	No
1117	9/29/2011, 3:29pm	449	Hoʻokena	*Hemicorallium imperiale/laauense*	No
1103	9/28/2011, 9:47am	451	Kealakekua	*Acanella* spp.	No
103	9/5/2016, 12:30pm	472	Kailua-Kona	*Hemicorallium imperiale/laauense*	No
202	9/5/2016, 1:49pm	506	Kailua-Kona	*Hemicorallium imperiale/laauense*	No
304	8/29/2017, 12:57pm	544	Kailua-Kona	*Hemicorallium imperiale/laauense*	No
313	8/30/2017	560	1919 Lava Flow*	*Hemicorallium imperiale/laauense*	Yes
312	8/30/2017	574	1919 Lava Flow*	*Hemicorallium imperiale/laauense*	Yes
102	9/5/2016, 11:45am	582	Kailua-Kona	*Hemicorallium imperiale/laauense*	Yes
303	8/29/2017, 11:27am	753	Kailua-Kona	*Acanella* spp.	No
101	9/5/2016, 10:56am	823	Kailua-Kona	*Acanella* spp.	No
M3	2/5/2000	417	Makapuʻu	*Hemicorallium imperiale/laauense*	Yes

1103   Note: The 1919 Lava Flow site was just south of the Hoʻokena site.
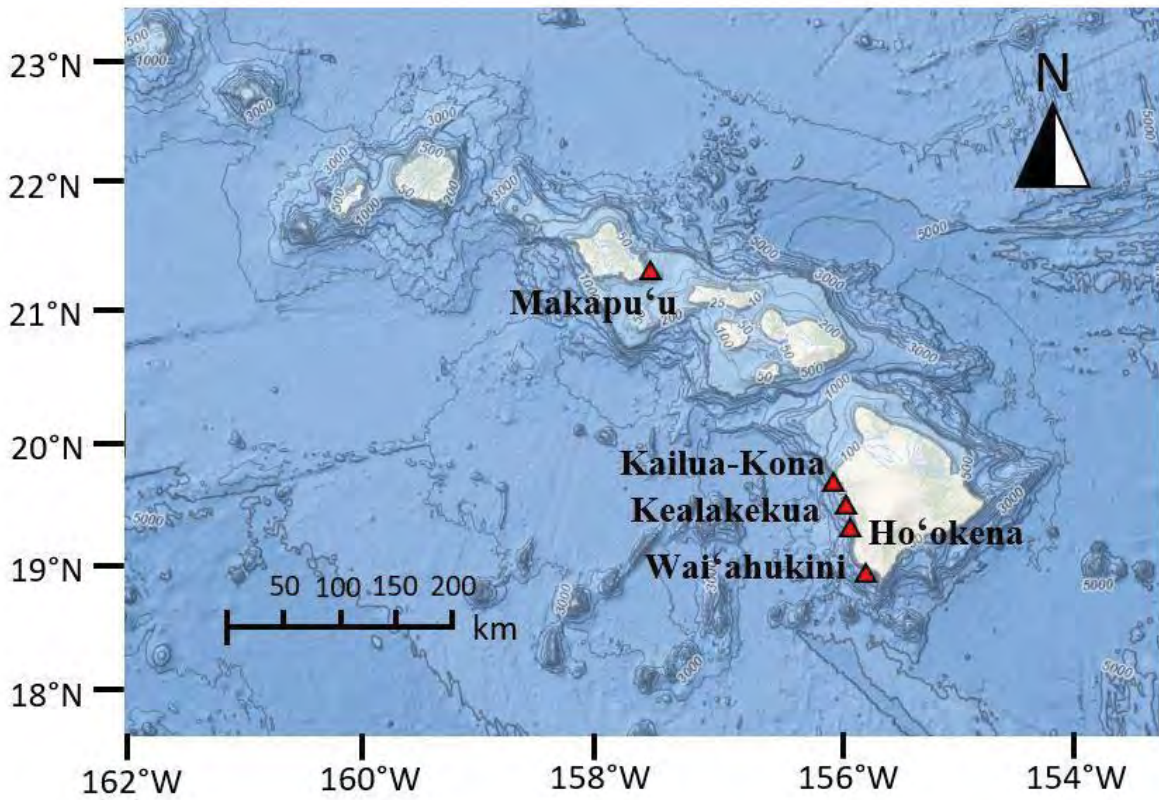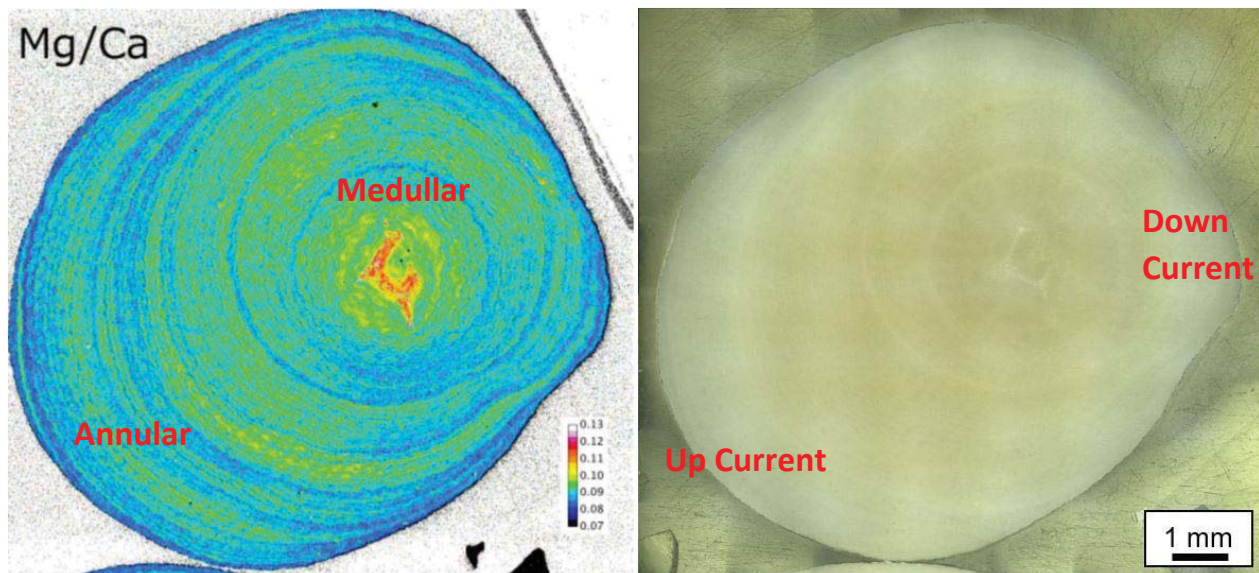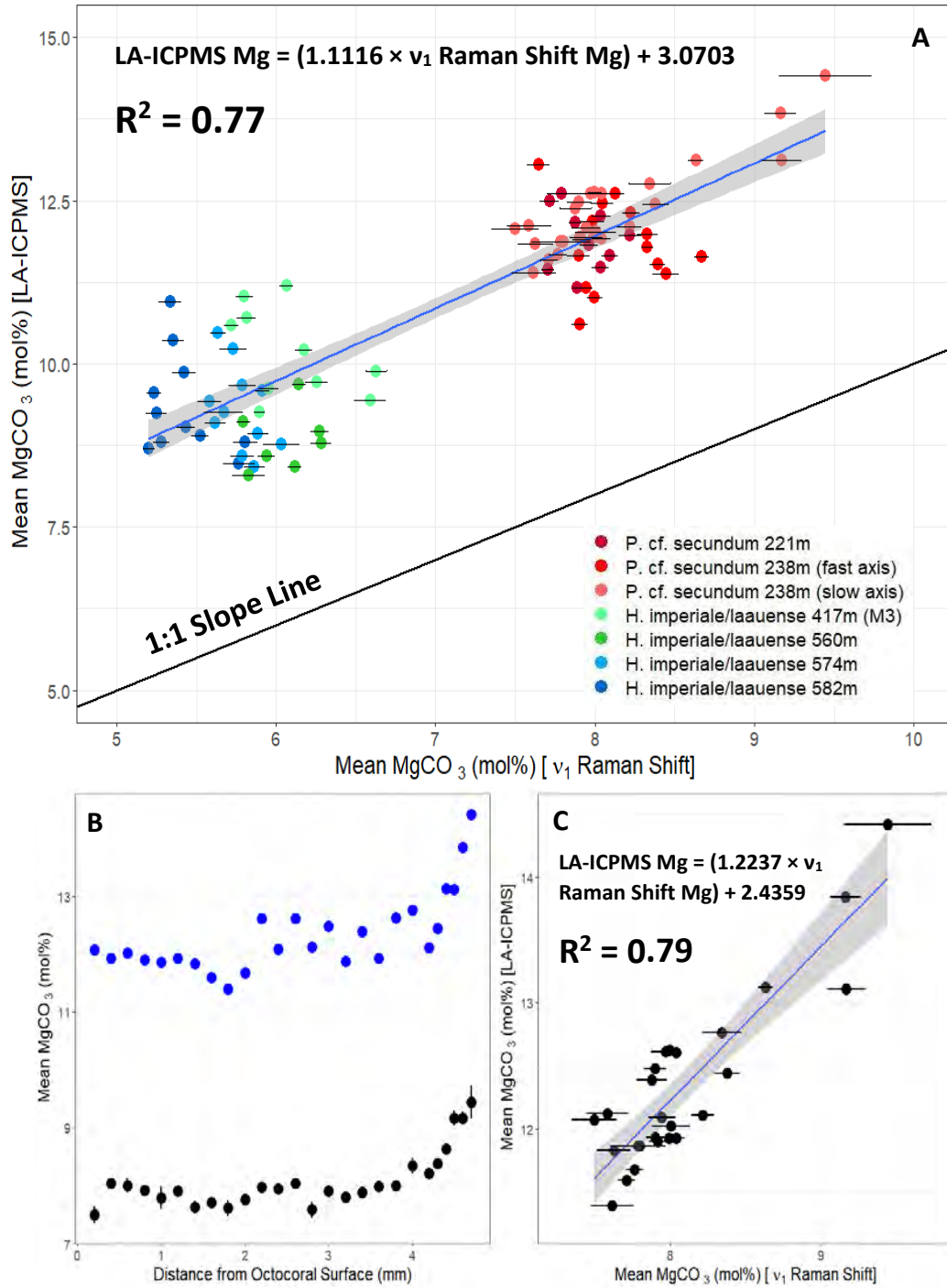1104

1105

49

Figure 1

Figure 2

Figure 3
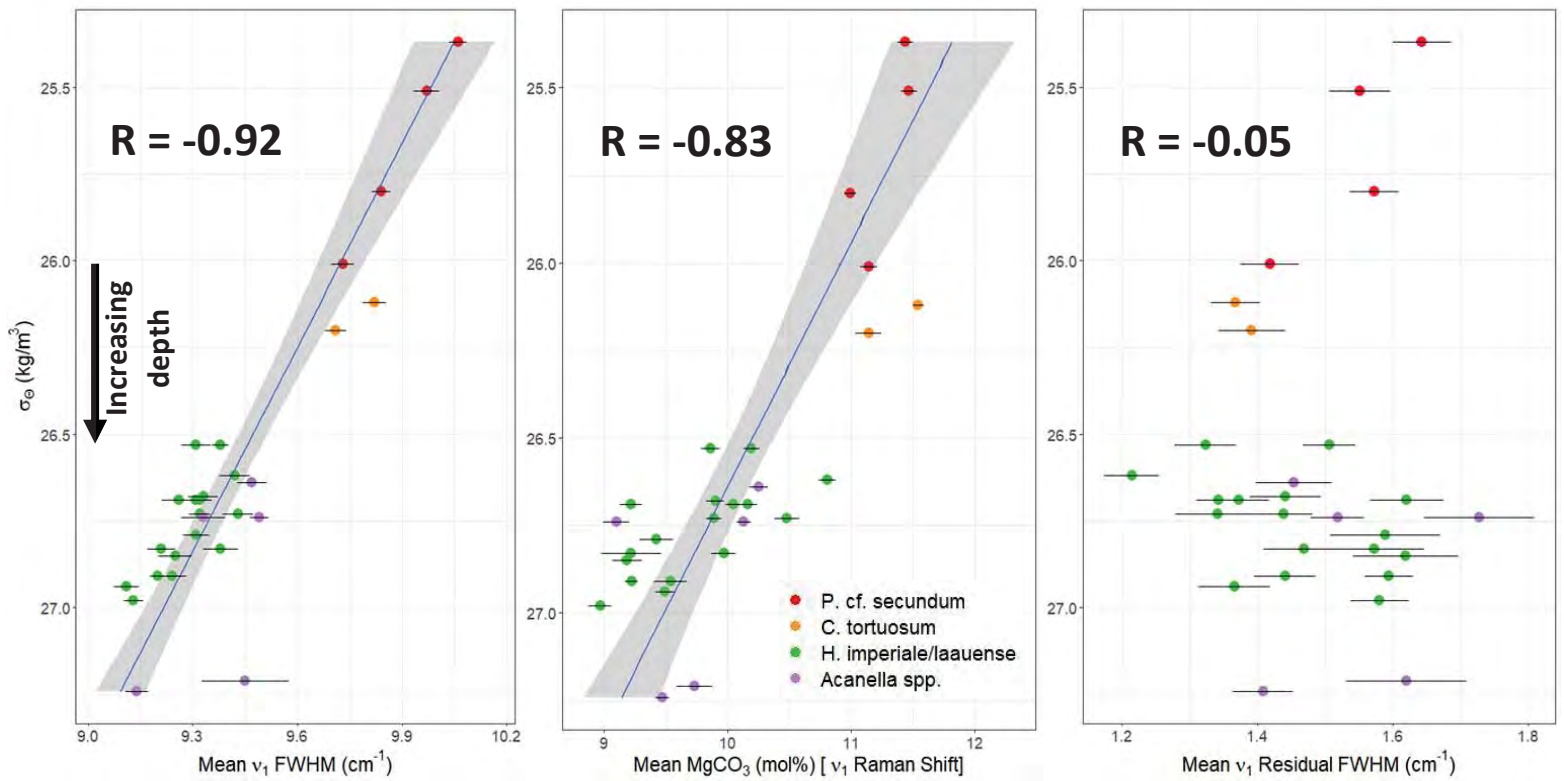
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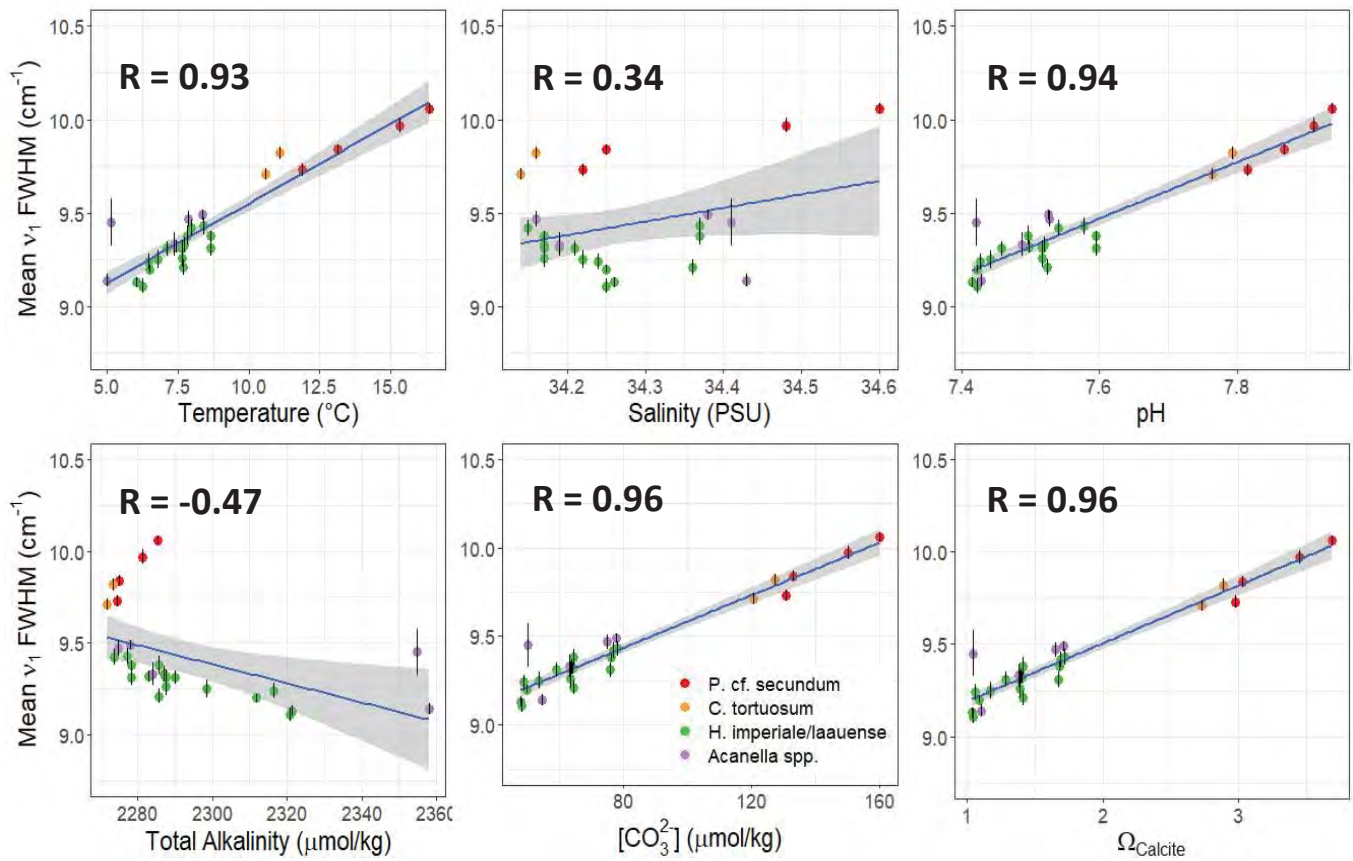
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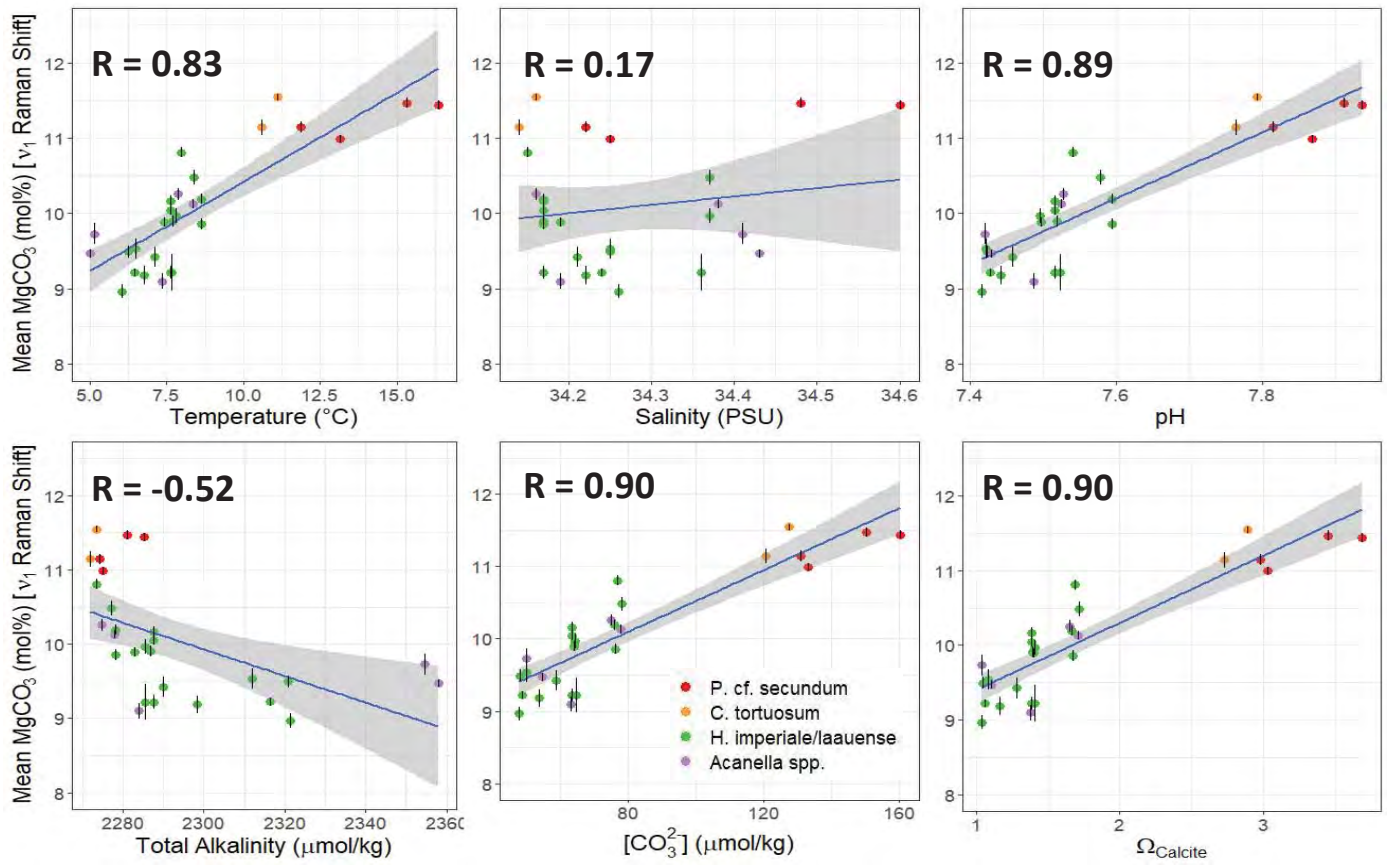
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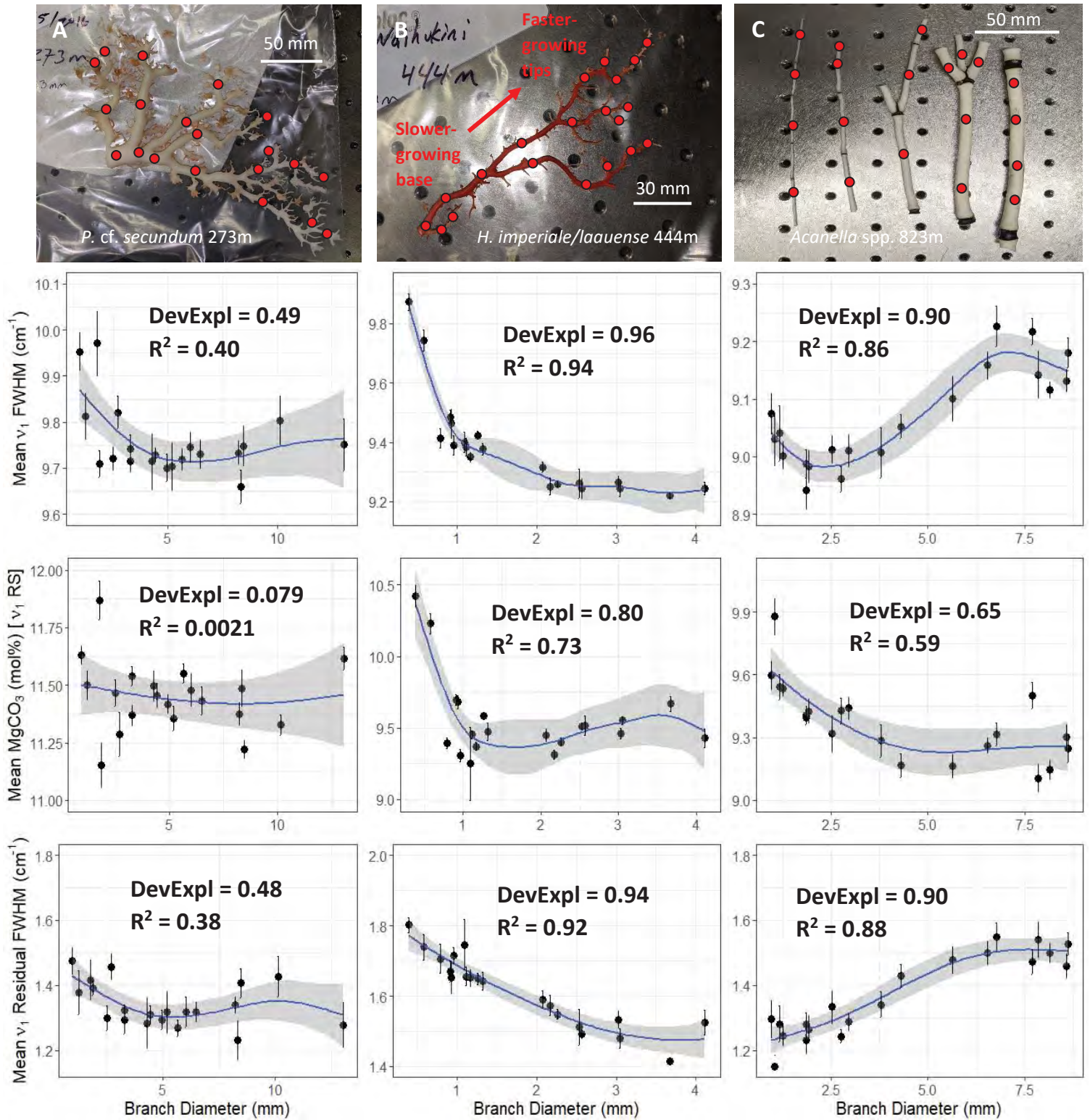
Figure 7

Figure 8

Figure 9