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2	Radiation-Induced Changes in Vanadium Speciation in Basaltic Glasses: Implications for
3	Oxybarometry Measurements Using Vanadium K-edge X-ray Absorption Spectroscopy
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9	ABSTRACT
10	Magmatic oxygen fugacity (fO_2) exerts a primary control on the discrete vanadium (V)
11	valence states that will exist in quenched melts. Vanadium valence proxies for fO ₂ , measured
12	using X-ray absorption near-edge spectroscopy (XANES), can provide highly sensitive
13	determinations of the redox conditions in basaltic melts. However, X-ray beam-induced changes
14	in V speciation will introduce uncertainty in the calculated average V valence (V*) that must be
15	properly evaluated in order to make meaningful interpretations of the igneous evolution of the
16	system. The study presented here showed that beam-induced modifications in V speciation are
17	observed in silicate glasses that are dependent on the radiation dose rate used during analysis.
18	Changes in V speciation are observed to be most pronounced at the highest flux density tested,
19	9.25 $x10^{11}$ ph/s/µm ² (photons per second per square micrometer), with rapid changes occurring
20	in the first 200s of analysis. The high dose rate conditions result in changes in calculated V* ~ 0.3
21	valence unit for the most oxidized glass analyzed (V* = 4.94), which can correspond to $\sim 0.5 \log$
22	unit reduction in calculated fO_2 . However, at flux densities $\leq 1.13 \times 10^9$ ph/s/µm ² , measured
23	changes in V* were found to be < 0.03 for all standard glasses analyzed. The degree of reduction
24	observed during analysis is also found to be progressively smaller as the initial V* of the glass

decreases, such that magmatic glasses with V* values ≤ 3.7 show no statistically-significant change in calculated valence during analysis at any flux density tested. For most terrestrial magmatic glasses, where V* is found to be < 4, beam induced changes in V* can be effectively minimized (< 0.04), within analytical uncertainty of the XAFS analysis, by limiting flux densities to be $\leq 1 \times 10^9$ ph/s/µm².

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INTRODUCTION

Accurate estimates of the intrinsic oxygen fugacity (fO_2) of magmas are important for 31 constraining source composition and understanding the conditions under which melts are 32 33 generated, their sub-liquidus evolution and crystallization history, and the composition and evolution of volcanic gases released to the atmosphere (Sato 1978; Mathez 1984; Christie et al. 34 1986; Ballhaus et al. 1990; Carmichael 1991; Kelley and Cottrell 2009). Oxybarometers of 35 varying types provide geologists analytical approaches for indirectly constraining magmatic fO_2 36 through chemical analysis of varying primary igneous phases. Of the approaches currently 37 available for oxybarometry, those based on X-ray absorption fine structure (XAFS) spectroscopy 38 (Sutton et al. 2005, 2005, 2020; Cottrell et al. 2009; Bunker 2010; Jugo et al. 2010; Henderson et 39 al. 2014; Trail et al. 2015; Lanzirotti et al. 2018) are particularly attractive due to the high 40 sensitivity of XAFS in detecting subtle changes in the chemical speciation of redox-sensitive 41 elements in igneous minerals and/or glasses in response to changing magmatic fO₂. Various 42 XAFS oxybarometers have been developed for magmatic systems, including those based on 43 44 measuring changes in chemical speciation of multivalent elements such as Fe, S, Ti, Mn, Cr, Eu, Ce and V (reviewed in Sutton et al. 2020 and references therein). Vanadium XAFS 45 oxybarometry has broad applicability to terrestrial and extraterrestrial igneous materials since its 46 valence varies continuously from V^{3+} to V^{5+} over a wide range of fO_2 (Giuli et al. 2004; Sutton et 47

48 al. 2005; Karner et al. 2006; Lanzirotti et al. 2018; Nakada et al. 2020). In terrestrial basaltic melts, the $V^{3+} \leftrightarrow V^{4+}$ and $V^{4+} \leftrightarrow V^{5+}$ multivalent couples span an oxygen fugacity range relative 49 to the nickel-nickel oxide (NNO) buffer from about NNO-6 to NNO+3.5 (Papike et al. 2005). 50 An analytical complication in the application of XAFS oxybarometers with respect to 51 mineral and glass analysis is the potential for inducing changes in the valence and/or crystal-52 chemical environment of the element of interest due to exposure of samples to the synchrotron 53 X-ray beam. Such phenomena have been documented in several systems and have been found to 54 be particularly pronounced for hydrous magmatic glasses. For example, sulfur valence state 55 changes have been observed during XAFS analysis of volcanic glasses and melt inclusions 56 (Rowe et al. 2007; Wilke et al. 2008; Klimm et al. 2012); most commonly reported is the 57 generation of S^{4+} species in glasses, generally attributed to reduction of S^{6+} during irradiation 58 (Wilke et al. 2008). In another example, Cottrell et al. (2018) showed that Fe^{2+} in hydrous 59 silicate glasses can undergo rapid oxidation upon exposure to radiation. In the most hydrous 60 basaltic glasses they analyzed, they showed that radiation-induced increase in $Fe^{3+}/\Sigma Fe$ ratio of 61 up to 0.16 can be observed after several minutes of irradiation at X-ray flux densities of $\sim 2 \times 10^9$ 62 ph/s/µm². They also showed that for Fe, the rate of oxidation also correlates with the product of 63 water concentration and ferrous/ferric iron oxide ratio in the glass. Interestingly, in anhydrous 64 glasses, they observed no statistically significant change in Fe valence under any beam 65 conditions, which led them to infer that it is radiation-induced hydrolysis which may drive 66 oxidation of Fe²⁺ in these magmatic glasses. Other studies have noted that anhydrous silicate 67 glasses with high concentrations of Fe often do not show evidence of beam-induced changes in 68 Fe valence (Camara and Oel 1984). However, in soda-lime silicate glasses where Fe is doped at 69 trace element concentrations (150-5000 ppm), Fe^{3+} has been shown to undergo X-ray induced 70

reduction (Camara and Oel 1984; Gonçalves Ferreira et al. 2013). Such reduction was

72 hypothesized to result from the production of localized defects within the glass matrix during

radiation, where electron-hole pairs associated with the defect sites act to change the valences

of incorporated trace Fe. We hypothesize that other multi-valent trace elements in natural silicate

75 glasses, such as V, may display similar behavior on irradiation.

This study presents new observations regarding X-ray beam-induced changes in V valence in 76 silicate glasses. Although the potential for radiation-induced changes in V valence in the analysis 77 of natural silicate glasses has been alluded to in published studies (Lanzirotti et al. 2018), we are 78 79 unaware of any systematic study that has sought to describe and quantify the magnitude of any observed changes with radiation dose rate that are relevant to geochemical studies. For V XAFS 80 oxybarometry to provide accurate and precise estimates of changing oxygen fugacity in 81 magmatic systems, it is important to understand and quantify how irradiation during XAFS 82 analysis can modify the spectral features used to calculate the average V valence of igneous 83 84 systems.

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SAMPLES AND ANALYTICAL PROCEDURES

86 Four synthetic glasses and two natural volcanic glasses were analyzed by V K-edge XAFS at 87 varying incident flux densities. The synthetic samples are from a suite of anhydrous forsteriteanorthite-diopside (FAD) composition glasses doped to provide V concentrations of ~ 3000 ppm. 88 89 This suite of glasses were synthesized at 1 atm pressure, at equilibration temperatures of 1310 -90 1320°C and at log fO₂ ranging from -15.5 to -4.55. Their synthesis is fully described in Hanson and Jones (1998). For this study, four glasses in the suite were evaluated for radiation-induced 91 changes in V valence, chosen because they were equilibrated over a broad range of ~7 log units 92 fO₂, (expressed here relative to nickel-nickel oxide buffer, NNO, Huebner and Sato, 1970): 93

94	FAD_V2a (log fO ₂ -11.6, ΔNNO-5.3), FAD_V4 (log fO ₂ -9.6, ΔNNO-3.3), FAD_V1a (log fO ₂ -
95	7.1, Δ NNO-0.7) and FAD_V3 (log fO_2 -4.6, Δ NNO+1.7). The two natural glasses studied are
96	Smithsonian Institution NMNH National Rock Collection samples (Jarosewich et al. 1980)
97	designated A-99 (NMNH 113498-1), a sample of fresh basaltic glass collected from the
98	Makaopuhi Lava Lake, Hawaii, and VG-2 (NMNH 111240-52), a sample of fresh basaltic glass
99	dredged from the Juan de Fuca Ridge. Published estimates for NNO-relative fO_2 for A-99 and
100	VG-2, based on V XAFS oxybarometry, were Δ NNO-1.28 and Δ NNO-0.08 respectively
101	(Lanzirotti et al. 2018). Absolute vanadium abundances have not been reported for these two
102	glasses, but natural basaltic glasses generally have V concentrations between \sim 200–400 ppm,
103	which is consistent with the X-ray fluorescence intensities measured for V K emission lines. All
104	samples were prepared as polished (using 0.5 micron diamond-embedded polishing paper),
105	epoxy embedded chips where the glass thickness was large compared to the sampling depth of V
106	K-edge X-rays. Previous V K-edge XAFS analyses of these samples are described in Lanzirotti
107	et al. (2018), Sutton et al. (2005) and Karner et al. (2006).
108	Vanadium K-edge XAFS spectra were measured using the 13-ID-E undulator-based
109	microprobe at the GeoSoilEnviro-CARS sector, Advanced Photon Source (APS), Argonne
110	National Laboratory, USA. The optical and instrumental configuration of the beamline are
111	described in Lanzirotti et al. (2016, 2018) and (Sutton et al. 2017). Monochromatic radiation was
112	provided by a cryogenically-cooled, double-crystal monochromator using a Si(111)
113	monochromator crystal set. Beam focusing to the sample is provided by a pair of 240 mm long,
114	highly polished, dynamically-bent bare silicon mirrors in a Kirkpatrick-Baez (KB) geometry
115	capable of generating focused spot sizes of ~1 x 2 μm (FWHM) and of providing incident
116	monochromatic flux (I ₀) in excess of $5x10^{12}$ ph/s, which is measured in a helium-filled, 200 mm-

117 long ion chamber just upstream of the KB mirror optics. For the experiments described here, Xray flux density delivered to the sample was controlled by dynamic defocusing of the beam, 118 using the KB optics, and by filtering of the incident beam, using high-purity aluminum filters. 119 For consistency with previous published data from the beamline, the sample-impinging flux 120 densities that are reported here are calculated based on measured I₀, which is monitored upstream 121 of the mirror focusing optics. Flux is calculated using measured current in the 200 mm long ion 122 chamber, assuming 100% He gas at atmospheric pressure and effective ionization potential for 123 He from Knoll (2010). The KB mirrors have reflection efficiencies of ~80%, so flux delivered to 124 the samples is estimated to be 60% of the measured I_0 flux. The energy of the first derivative 125 peak of V metal foil provided a V K-edge energy of 5463.76 eV, consistent with value 126 determined by Kraft et al. (1996). Fluorescence spectra were collected using a Vortex ME4 and 127 Canberra SX7 silicon-drift diode detector arrays coupled to a high-speed digital spectrometer 128 system (Quantum Xpress3). 129

Full V K-edge XAFS spectra were collected at variable flux densities, 2.56×10^8 , 7.75×10^8 , 130 6.25×10^{11} , and 9.25×10^{11} ph/s/µm², with each analysis conducted on a fresh sample spot. This 131 allowed us to compare the final shape of the XAFS spectra with increasing radiation dose and 132 evaluate any potential trends in changing V speciation over time. The XAFS spectra collected at 133 flux densities of 2.56 x10⁸ and 9.25 x10¹¹ ph/s/ μ m² scanned the incident beam from 5365-5455 134 eV in 2.5 eV steps, from 5455-5490 eV in 0.15 eV steps, and from 5490-5645 eV in 2.0 eV steps 135 136 at 1 s per energy point. This allowed us to compare spectra with optimal signal/noise at the highest and lowest flux densities tested, with one spectrum being collected in ~6 minutes (Fig. 137 1). More rapid XAFS spectra were also collected at flux densities of 7.75 $\times 10^8$ and 6.25 $\times 10^{11}$ 138 ph/s/µm², scanning the incident beam from 5420-5455 eV in 3.5 eV steps, from 5455-5475 eV in 139

0.15 eV steps, from 5475-5500 eV in 2.5 eV steps and from 5500-5700 eV in 5.0 eV steps at 0.5 140 s per energy point. This allowed us to evaluate changes in spectral shape more quickly at two 141 flux densities that differ by a factor of ~ 800 , with slightly poorer signal/noise, but with one 142 spectrum being collected in only ~ 1.5 minutes (Figs. 2, 3 and 4). 143 V* values were calculated using the Sutton et al. (2005) calibration, derived from XAFS 144 spectra measured in synthetic glass standards produced under known fO_2 and temperature 145 conditions. Following the Sutton et al. protocol, each XAFS spectrum is first normalized to the 146 count rate measured at 5600 eV. The pre-edge peak intensity (I) is then measured as the 147 148 maximum intensity of the net pre-edge peak multiplet after normalized intensities are multiplied by 1000 and after subtracting the edge step contribution. The effective vanadium valence, V* is 149 then calculated as: 150

$$I = -153 + 199(V^*) - 106(V^*)^2 + 22.4(V^*)^3$$
(1)

Tests were also conducted to constrain rates of reduction in glasses showing evidence of 151 decreases in V* as a function of radiation dose rate using four flux densities covering more than 152 4 orders of magnitude: 1.69×10^6 , 1.38×10^8 , 1.13×10^9 , and 9.25×10^{10} ph/s/µm². A protocol 153 similar to that described by Cottrell et al. (2018) for evaluating beam-induced changes in Fe 154 valence was used for this purpose. For evaluating beam-induced changes in Fe valence in silicate 155 glasses Cottrell et al. measured the change in the peak amplitude of the second multiplet pre-156 edge peak of the Fe K-edge with time at differing flux densities. Here, we evaluated the time 157 158 dependence of the measured V spectral intensity at an energy of 5468.15 eV, which sits on the pre-edge peak, as a function of flux density (i.e., dose-rate). Each time series consists of 159 measurements at 1s intervals over 1500s of total irradiation and with each curve measured at a 160 161 fresh spot on the sample that had not been previously irradiated. The 13-ID-E monochromator

162 has been demonstrated (Sutton et al. 2017) to have no significant energy drift (<0.05 eV) as a function of time, so that it is unlikely that any intensity changes measured are affected by energy 163 drift. Although pre-edge peak energies also increase with increasing valence, the energy shifts 164 are small (~1 eV from V^{3+} to V^{5+} , Sutton et al. 2005). Thus, although we are solely monitoring 165 changing intensity of the pre-edge peak at an energy of 5468.15 eV in constructing the rate-of-166 change curves, this is the most convenient method for evaluating *potential* changes in valence at 167 very short accumulation times of 1 s. Thus, for the purposes stated here, the uncertainties 168 introduced by collecting intensities at a single energy are likely to be negligible. 169

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RESULTS

171 Observed Spectroscopic Changes Following Irradiation

In V K-edge XAFS spectroscopy, increases in V valence are generally indicated by increases 172 in the net, normalized intensity of the pre-edge peak and shifting of the main absorption edge to 173 higher energy, although other changes in spectral shape are also noted (Wong et al. 1984; Sutton 174 175 et al. 2005; Lanzirotti et al. 2018). The pre-edge peak results from the dipole forbidden $ls \rightarrow 3d$ 176 transition and is coupled to metal-ligand symmetry; V species in regular octahedral coordination with the O²⁻ ligand have the smallest observable pre-edge intensity while species in tetrahedral 177 178 coordination have the largest intensity. In glasses, these changes in coordination are coupled to changes in V valence. Figure 1 shows edge-step normalized V K-edge XAFS spectra collected 179 180 for FAD V3, the most oxidized glass analyzed, and FAD V2a, the most reduced glass analyzed, measured at flux densities of 2.56 $\times 10^8$ (in red) and 9.25 $\times 10^{11}$ (in blue) ph/s/ μ m². In oxidized 181 glass FAD V3, a decrease in the energy of the main absorption edge (E_0) of -1.1 eV and a 182 decrease in normalized, pre-edge peak intensity of -0.26 is observed (i.e., 26% decrease of the 183 above edge normalization level). These spectral changes and the observed changes in intensity 184

185	and energy of spectral resonances in the main absorption peak, the "white line," are consistent
186	with reduction in V valence (Lanzirotti et al. 2018) as the radiation dose rate increases. Reduced
187	glass FAD_V2a displays a roughly similar decrease in E_0 of - 0.8 eV, but a small increase in pre-
188	edge peak intensity is also observed with increasing dose rate if the spectroscopic background is
189	not subtracted from the pre-edge peak. The inset plots in Fig. 1 show the relative magnitude of
190	the measured pre-edge peaks for both glasses after subtracting the edge step contribution via
191	spline fitting of the rising absorption edge. Although the background subtracted spectra for
192	oxidized glass FAD_V3 indicate an ~30% reduction in pre-edge peak intensity after subtraction
193	of the spectral background, the background subtracted peak intensities for reduced glass
194	FAD_V2a displays a <1% difference at the two flux densities tested.
195	Both the reduced and oxidized glass display increases in the intensity of a spectroscopic low
196	energy shoulder at \sim 5475 eV with increasing dose-rate. This feature sits on the rising absorption
197	edge just above the pre-edge peak (Fig. 1, dashed line). The increasing intensity of this shoulder
198	at \sim 5475 eV has the potential for elevating the pre-edge peak baseline, an effect that is more
199	significant in highly reduced glasses where the pre-edge peak intensities are small. In V XAFS,
200	the spectral shoulder that is observed on the rising absorption edge has been attributed to a
201	$1s \rightarrow 4p$ shakedown transition (Wong et al. 1984) and these features are thought to be associated
202	with charge transfer from the ligand to the metal. Thus observed changes in the absolute intensity
203	of these shoulder peaks potentially indicate that changes in vanadium-oxygen complexing are
204	occurring in these glasses with increasing radiation dose-rate.
205	To better evaluate the rate at which any observed spectroscopic changes occur, a series of
206	more rapid XAFS analyses were conducted at both low (7.75 $x10^8$ ph/s/µm ²) and high (6.25
207	$x10^{11}$ ph/s/ μ m ²) flux density. Results of these analyses are shown in Fig. 2 for the reduced

208	synthetic glasses (FAD_V2a and FAD_V4), Fig. 3 for the oxidized synthetic glasses (FAD_V1a
209	ad FAD_V3) and in Fig. 4 for natural MORB glass VG-2. Shown are a series of 10 edge-step
210	normalized V K-edge XAFS spectra that were measured sequentially, with each scan acquired in
211	~1.5 minutes, collected at low (left) and high (right) flux density. Acquisition of 10 total spectra
212	requires the glass to undergo irradiation at each flux density for a total time of ~15 minutes. The
213	inset plots on each figure show the pre-edge peak portion of each scan and provide calculated V*
214	for each scan using the Sutton et al. (2005) calibration, which subtracts the pre-edge background.
215	We estimate that the uncertainty in calculated V* from fitting the pre-edge peak intensity for any
216	individual scan under the conditions used is $\sim 1.5\%$ (1 σ error, which is what is reported for
217	uncertainties below).
218	Observed trends in these spectra are consistent with those from Fig. 1. The most reduced
219	glass analyzed, FAD_V2a (log fO_2 -11.6), gives a calculated mean V* of V* 3.13 (± 0.01) at low
220	flux density and V* 3.10 (\pm 0.03) at high flux density (Fig. 2, top). Fig. 5 shows how the relative
221	calculated V* for this series of analyses changes with time. For FAD_V2a, the first two spectra
222	collected at the high flux condition yield slightly more reduced V*, but these are still within error
223	of the calculated mean for all 10 spectra in the series. Glass FAD_V4 (log fO_2 -9.6), which was
224	equilibrated under slightly more oxidizing conditions, gives a calculated mean V* of 3.62 (\pm
225	0.01) at low flux density and V* 3.47 (\pm 0.01) at high flux density (Fig. 2, bottom). After 200
226	seconds of irradiation, the FAD_V4 high flux sequence yields V* values that are clearly more
227	reduced than what is measured at low flux, outside of the fitting uncertainties (Fig. 5).
228	In more oxidized glasses, this difference is accentuated with increasing valence. FAD_V1a
229	(log fO ₂ -7.1) gives a calculated mean V* of 4.30 (\pm 0.01) at low flux density and V* 4.02 (\pm
230	0.03) at high flux density (Fig. 3, top). FAD_V3 (log fO_2 -4.6) gives a calculated mean V* of

4.94 (\pm 0.01) at low flux density and V* 4.65 (\pm 0.08) at high flux density (Fig. 3, bottom). For these glasses, all the high flux analyses are statistically distinct from what is measured at low flux at all measured time points (Fig. 5). Natural MORB glass VG-2 provides calculated mean V* of 4.33 (\pm 0.03) when measured at low flux density and 4.14 (\pm 0.05) at high flux density, very similar to results given by synthetic glass FAD_V1a, although uncertainties are slightly higher due to the lower V abundance in MORB glass.

Other studies have also utilized time series measurements of pre-edge intensities measured 237 rapidly at a single energy to help evaluate rates of speciation changes for multi-valent elements 238 239 in silicate glasses as a function of flux density (Cottrell et al. 2018; Moussallam et al. 2019; Gaborieau et al. 2020). By monitoring intensity changes at a single incident energy, rapid 240 changes that occur at the onset of X-ray exposure can be evaluated. Figure 6 shows such time 241 series analyses for the two most oxidized synthetic glasses analyzed (Fig. 6 top) and for the two 242 natural basaltic glasses (Fig. 6 bottom). Changing spectral intensities of the V pre-edge were 243 monitored at an incident energy of 5468.15 eV for four different flux densities (described above). 244 Accumulated counts were collected at 1s intervals over 1500s of total irradiation, and then 245 plotted as the relative fractional change in measured change intensity over time. It should be 246 noted, however, that although such time series can be used to monitor changes at very short time 247 intervals, such analyses can have large associated uncertainties due to low signal/noise 248 (particularly when absorber concentrations are low) and can be difficult to accurately correlate to 249 valence (V*) due to an inability to evaluate potential changes in pre-edge spectral background. 250 At the highest flux density tested for these time series, $9.25e^{10}$ ph/s/µm², the two oxidized 251 synthetic glasses again show clear decreases in V pre-edge peak intensities with increasing 252 accumulated dose. An intensity change of ~ -11% is noted for FAD V1a and -18% for FAD V3 253

254	after 1500s of irradiation (Fig. 6 top). These are consistent with the full XAFS results. Although
255	the full XAFS scans were collected using up to a factor of 10 higher flux density, the shorter
256	accumulation time used in collecting the time series means that roughly similar levels of total
257	photons were delivered to the analysis point. The time series measurements, however, more
258	clearly show that the decreases in measured intensity changes at the pre-edge energy are non-
259	linear with continued irradiation. The curves measured for both glasses flatten over time,
260	suggesting they may have reached an equilibrium condition after 400s of exposure at the highest
261	flux densities tested. Natural basaltic glasses VG-2 and A-99 both show small decreases in count
262	rates at the pre-edge at the highest tested flux density of ~ -5%. However, at the lower tested flux
263	densities, intensity changes are difficult to discern in these natural glasses within the analytical
264	uncertainty of measuring V K α fluorescence at the short 1s integration times used.
265	DISCUSSION
266	Evaluating Potential Changes in Vanadium Valence During XAFS Analysis
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267 268 269 270 271 272	We used the empirically-derived formula of Sutton et al. (2005) to evaluate the potential impact of the observed beam-induced changes in V speciation on the calculated average V valence (V*) of the glasses studies. The Sutton calibration calculates V* on the basis of measured, edge-step normalized, pre-edge peak intensities after subtraction of the rising edge background. For V XAFS oxybarometric studies, the pre-edge peak intensity is found to have high valence sensitivity, although it is important to recognize that changes in V coordination
267 268 269 270 271 272 272	We used the empirically-derived formula of Sutton et al. (2005) to evaluate the potential impact of the observed beam-induced changes in V speciation on the calculated average V valence (V*) of the glasses studies. The Sutton calibration calculates V* on the basis of measured, edge-step normalized, pre-edge peak intensities after subtraction of the rising edge background. For V XAFS oxybarometric studies, the pre-edge peak intensity is found to have high valence sensitivity, although it is important to recognize that changes in V coordination within the glass network will also alter pre-edge peak intensities. The study by Sutton et al.
267 268 269 270 271 272 273 273	We used the empirically-derived formula of Sutton et al. (2005) to evaluate the potential impact of the observed beam-induced changes in V speciation on the calculated average V valence (V*) of the glasses studies. The Sutton calibration calculates V* on the basis of measured, edge-step normalized, pre-edge peak intensities after subtraction of the rising edge background. For V XAFS oxybarometric studies, the pre-edge peak intensity is found to have high valence sensitivity, although it is important to recognize that changes in V coordination within the glass network will also alter pre-edge peak intensities. The study by Sutton et al. (2005) shows that in silicate glasses changes in the V pre-edge peak intensity changes are

277 to provide slightly more intense pre-edge peak intensities, so although the intensity changes are dominated by changes valence, the magnitudes of valence effects reported here will be 278 overestimates if some of the observed intensity changes are also due to coordination changes. 279 It is clear from these analyses that V in silicate glasses will undergo measurable beam-280 induced reduction at high flux densities when the average V valence in the glass, V*, is greater 281 than ~ 3.7 . The degree of reduction is also observed to increase with increasing V* of the glass. 282 The rapid XAFS measurements of the most oxidized glass analyzed, FAD V3 (Fig. 3), gives a 283 mean V* of 4.94 when measured at a low dose condition of 7.75 $\times 10^8$ ph/s/ μ m² and 4.65 at a 284 high dose condition of 6.25 $\times 10^{11}$ ph/s/ μ m². Therefore, an observed reduction in calculated V 285 valence (ΔV^*) of -0.3 when the flux density is increased by a factor of ~800. With decreasing V* 286 287 in the glass, the degree of observed reduction as a function of flux density at these conditions is also observed to be smaller. In FAD V1a the change in mean V* similar, from 4.30 to 4.02 288 (ΔV*=0.28), in FAD V4 from 3.62 to 3.47 (ΔV*=0.15), and in FAD V2a from 3.13 to 3.10 289 $(\Delta V^*=0.03)$. The results of both rapid XAFS scanning (Figs. 3 and 5) and from time series 290 analyses at the pre-edge peak (Fig. 6) also show that at flux densities at or below 7.75 $\times 10^8$ 291 $ph/s/\mu m^2$, changes in V* are not statistically discernable over 1500s of irradiation. Extrapolation 292 of the observed trends suggests that glasses with starting V* values near ~ 3.7 will display only 293 minimal change in pre-edge peak intensity with increasing dose. 294 Highly reduced glasses with calculated V* less than 3.7 also likely undergo changes in V 295 speciation with irradiation. This is reflected by measured changes in overall spectral shape, shifts 296 in the energy of the rising absorption edge and changes in the intensity of the spectral 297 background near the pre-edge peaks. However, it remains unclear in these samples to what 298

degree these changes in speciation also involve changes in valence state. For the most reduced

glass analyzed, FAD V2a, it was noted that calculated mean V* only changes from 3.13 to 3.10

300

 $(\Delta V^*=0.03)$, arguably within statistical uncertainty. We also noted from the rapid XAFS analysis 301 (Figs. 2 and 5) that the first two spectra in this series, collected at the high flux condition, yield 302 slightly more reduced V*, but still within error of the calculated mean for all 10 spectra in the 303 304 series. Thus some degree of beam-induced oxidation (as opposed to reduction) in these highly reduced glasses cannot be discounted. However, the impact on calculated average valence 305 306 appears small. 307 In the natural basaltic glasses studied the degree of reduction observed appears comparable to what is observed in the synthetic FAD glasses. In MORB glass VG-2 the mean V* decreases 308 from 4.33 to 4.14 ($\Delta V^*=0.19$), very similar to what is observed in FAD V1a, which yields a 309 generally similar mean V*. Observed trends are proportionally similar for both synthetic and 310 311 natural glasses even though there are large differences in Fe concentration in the glasses analyzed, which is significant in that it suggests that differences in Fe concentration have little 312 impact on the observed effect. Although Fe is present at only trace element concentrations in the 313 FAD glasses analyzed here (Lanzirotti et al. 2018), in natural basaltic glasses Fe concentrations 314 typically exceed 10 wt% and will be the most abundant multi-valent element present. It is worth 315 noting, however, that it is unknown if the degree or direction of beam-induced changes in V 316 valence in silicate glasses are impacted by the water content of the glass. All the glasses analyzed 317 here are relatively anhydrous, including the natural glasses. In studies of Fe valence in natural 318 glasses there is a clear correlation between water content and increasing degrees of Fe oxidation 319 with increasing flux density (Cottrell et al. 2018), 320

321 Discussion of Potential Mechanisms for Beam-Induce Changes in V Valence

322 Studies examining the structural and chemical changes that occur in glasses exposed to ionizing radiation have shown that irradiation can generate free electrons and holes which can 323 then be trapped in defect sites (Bishay 1970). These electron-hole pairs can act to change the 324 valences of incorporated trace elements spatially correlated with such defects (Camara and Oel 325 1984; El Batal and Naf 2005; Zhang et al. 2007; Abdelghany and ElBatal 2012; Goncalves 326 Ferreira et al. 2013). It has been hypothesized that the free electrons can reduce multivalent 327 transition metals while reduced ions in the glass can potentially capture the positive hole centers 328 to transform to more oxidized species. 329

The XAFS spectral changes observed here in glasses with starting V* greater than \sim 3.7 are 330 consistent with beam-induced reduction of V valence and is the dominant change observed in V 331 speciation due to irradiation. The pronounced decreases in pre-edge peak intensity, the shifting 332 of E₀ to lower energy and the changes observed in white line structure are all expected to occur 333 with V reduction (Lanzirotti et al. 2018). We hypothesize that V^{4+} and V^{5+} ions in these glasses 334 are able to capture available free electrons produced during irradiation to convert to V^{3+} and V^{4+} 335 ions, respectively. Such valence reductions are potentially accounted for by electron capture 336 reactions, consistent with what has been suggested from previous glass irradiation studies (El 337 Batal and Naf 2005; Abdelghany and ElBatal 2012). That the degree of beam-induced reduction 338 decreases with decreasing V*, so that glasses with initial V* value < 3.7 show only small 339 changes in V XAFS pre-edge peak intensity with increasing dose, imply that this is the point at 340 which the fraction of V^{3+} becomes significant and the effects of continued oxidation of V are 341 near the uncertainty level. 342

The possibility that some small degree of beam-induced oxidation of V^{3+} is occurring in the most reduced glasses analyzed cannot be conclusively discounted. Studies have shown that

irradiation of synthetic silicate, borate and soda lime glasses may result in the potential oxidation of reduced multivalent elements such as Fe^{2+} , Ce^{3+} , Cu^+ and Mn^{2+} (Bishay 1970; El Batal and Naf 2005), hypothetically attributed to the reduced ions trapping holes via photochemical reactions. However, any such oxidation in the glasses studied here is within analytical uncertainty in fitting the pre-edge peaks. Oxidation of V³⁺ also appears to be inconsistent with the measured decreases of the E_0 absorption edge in the V XAFS with increasing dose, by as much as -0.8 eV for the most reduced glass analyzed here.

Even so, these more reduced glasses do display continued spectroscopic changes with 352 353 irradiation, even though not clearly associated with changes in valence. There are observed increases in spectral intensity of the low energy shoulder at ~ 5475 eV, most likely reflecting 354 changes in vanadium-oxygen complexing and coordination within the glass network during 355 irradiation. Such effects can elevate the spectral backgrounds on the rising absorption edge and 356 may account for the small increases measured in instantaneous count rate near the pre-edge peak 357 with increasing dose. These changes in speciation seem to occur in all the glasses analyzed, 358 regardless of valence. 359

360

IMPLICATIONS

Although V K-edge XAFS is highly sensitive to detecting small variations in V valence due to changing magmatic redox conditions, beam-induced changes in V speciation are likely to introduce systematic errors in calculated V* estimates, particularly in oxidized systems, that must be properly evaluated and/or mitigated in order to make meaningful interpretations with regards to the igneous evolution of the system.

The study presented here demonstrates that there are clear changes in V speciation that can occur during XAFS analysis of silicate glasses and that the magnitude of the observed beam-

368 induced modifications is dependent on both the radiation dose over the period of analysis and the starting valence of the samples being analyzed. The highest observed changes in valence 369 occurred at flux densities > $6 \times 10^{11} \text{ ph/s/} \mu \text{m}^2$, resulting in reductions in calculated V* on the 370 order of 0.3 valence units for the most oxidized glasses analyzed ($V^* = 4.94$). A difference in 371 assumed V* of 0.3 valence units can translate to differences in calculated fO_2 of as much as ~ 0.5 372 log units for terrestrial magmatic glasses equilibrated at ~1200 °C (Sutton et al. 2005), 373 depending on redox conditions at equilibration, illustrating the importance of identifying the 374 potential for such alterations to occur at the time of analysis. However, for the majority of 375 terrestrial basaltic melts where V* is expected to be < 4, the impact on calculated oxygen 376 fugacity will be significantly smaller. 377 Our low flux density data show that the radiation alteration effects can be mitigated by doing 378 the measurements under these conditions. At the lower flux densities that we tested, $\leq 1 \times 10^9$ 379 $ph/s/\mu m^2$, measured changes in V* were found to be < 0.04 for repeated analyses in a series, 380 typically within the analytical uncertainty in calculating V* for any individual spectra using the 381 Sutton et al. (2005) calibration. It was also found that magmatic glasses with V* values near 3.7 382 show the smallest degree of change in speciation, and thus calculated V*, with dose. This is an 383 average V valence that is generally similar to that often measured for terrestrial basaltic glasses. 384 385 For example, the terrestrial basaltic glasses analyzed in Sutton et al. (2005) and Lanzirotti et al. (2018), equilibrated at fO₂ from NNO-1.6 and NNO-0.5, provide V* values between 3.7 and 4.1. 386 These results allow us to infer negligible impact of potential radiation-induced effects on 387 388 previously published results, studies which didn't account for possible beam-induced changes in

389

beamline configuration used for the majority of their analyses, with a roughly 4 x 4 μ m focused

V valence. For the glasses analyzed as part of the Sutton et al. (2005), we estimate that the

beam, would have been conducted using a flux density of $< 2x10^9$ ph/s/um². At this flux density 391 the most oxidized glasses analyzed may be expected to experience reductions in V* of < 0.05. 392 within the quoted uncertainty of the analyses used to build their calibration curve. A study by 393 Righter et al. (2006) examining the oxidation state of vanadium in coexisting spinel and basaltic 394 melt reports V K-edge XAFS data collected at the same beamline at roughly the same time. The 395 three standard spinels presented in the Righter et al. (2006) study, which are expected to be fairly 396 resistant to beam-induced changes in speciation, are shown to lie very close to the Sutton 2005 397 calibration curve. Consistent with the hypothesis that at the flux density conditions used in the 398 399 Sutton 2005 study, beam-induced reduction in the glass data was likely negligible. The study by Lanzirotti et al. (2018) developed V oxybarometry calibration models using multivariate analysis 400 to predict the fO_2 of equilibration in glasses of basaltic composition and then tested the models 401 on a suite of natural basaltic glasses. Recognizing the potential for beam-induced changes in V 402 valence, the XAFS data collected on reference glasses for this study used a defocused beam of 403 30 x 30 µm and purposely reduced incident flux, so that flux densities to the standards was 404 estimated to be $\sim 1 \times 10^8$ ph/s/µm², so that changes outside of analytical uncertainty are not 405 expected. Some of the analysis of natural glasses in this study did use a focused beam of $\sim 2 \times 2$ 406 μ m, which would increase flux density to $\sim 2 \times 10^{10}$ ph/s/ μ m². The natural glasses analyzed in that 407 study, however, yield V* close to 4.1, so that beam reduction effects would be expected to 408 reduce pre-edge peak intensities by <5%. A study by Head et al. (2018) included some V XAFS 409 analyses of magmatic melt inclusions from Nyamuragira volcano, Africa using stated flux 410 density of $\sim 2.5 \times 10^{10}$ ph/s/µm², with calculated V* values between ~ 3.1 and 3.8. Although some 411 reduction in V* would be expected at these flux densities, given calculated V* are < 4 it would 412 413 be expected to reduce pre-edge peak intensities by <5%.

414	The results presented here provide researchers a framework for conducting V K-edge XAFS
415	oxybarometry studies of natural magmatic glasses that will yield the most accurate estimate of
416	the fO_2 at which the magma quenched. For most terrestrial magmatic glasses, where V* is found
417	to be near 4, beam induced changes in valence can be effectively minimized to be within
418	analytical uncertainty of the XAFS analysis if incident flux densities are kept at $\leq 1 \times 10^9$
419	$ph/s/\mu m^2$. These results illustrate the importance of documenting not only the focused spot size
420	used in V XAFS analysis of silicate glasses but the incident flux as well, so that the flux density
421	can be estimated. This is a recommendation that has also been made for all XAFS analyses of
422	magmatic melt inclusions (Rose-Koga et al. 2021).
423	The results of this study also have potential implications for the irradiation behavior of other
424	multivalent trace elements in silicate glasses during XAFS analysis. Potential valence proxies for
425	fO_2 in igneous systems include several multivalent elements that can be measured using XAFS.
426	These are summarized in a review by Papike and coworkers (Papike et al. 2005, 2016) and
427	include elements such S, Ti, Mn, Cr, Eu, and Fe in addition to V. With the increasing availability
428	of brighter synchrotron sources and improved focusing optics capable of providing sub-
429	micrometer focused X-ray beams for spectroscopic analysis (and thus potentially higher flux
430	densities), it is important for researchers to accurately evaluate potential beam-induced changes
431	in valence state and speciation for the elements being measured in each geochemical system,
432	whether targets are magmatic silicate glasses or minerals.
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FIGURE CAPTIONS

559 Figure 1: Measured edge-step normalized V K-edge XAFS spectra collected for FAD V3, the most oxidized glass analyzed (top), and FAD V2a, the most reduced glass analyzed (bottom), 560 measured at flux densities of 2.56 $\times 10^8$ (in red) and 9.25 $\times 10^{11}$ (in blue) ph/s/ μ m². The inset plots 561 on each show the relative magnitude of the measured pre-edge peaks for both glasses after 562 subtracting the edge step contribution via spline fitting of the rising absorption edge. Although 563 the background subtracted spectra for oxidized glass FAD V3 indicate an ~30% reduction in 564 pre-edge peak intensity after subtraction of the spectral background, the background subtracted 565 peak intensities for reduced glass FAD V2a displays a <1% difference at the two flux densities 566 tested. Glass FAD V3 also displays a decrease in the energy of the main absorption edge (E_0) of 567 1.1 eV with increasing radiation dose. Glass FAD V2a displays a decrease of 0.8 eV in E₀ with 568 increasing dose. The dashed line shows a low energy shoulder at ~ 5475 eV on the rising 569

570	absorption edge that increases in intensity with increasing dose in both samples. This increasing
571	shoulder in FAD_V2a may be responsible for the increasing background under the pre-edge peak
572	and the decrease in E_0 .

573 Figure 2: Series of rapid V K-edge XAFS spectra collected on reduced synthetic glasses

- 574 FAD_V2a (top) and FAD_V4 (bottom) conducted at a low density of 7.75 $\times 10^8$ ph/s/ μ m² (left)
- and a high flux density of 6.25 $\times 10^{11}$ ph/s/ μ m² (right). Shown are a series of 10 edge-step

576 normalized V K-edge XAFS spectra that were measured sequentially, with each scan acquired in

577 ~1.5 minutes. Acquisition of 10 total spectra requires the glass to undergo irradiated at each flux

density for a total time of \sim 15 minutes. The inset plots on each figure show the pre-edge peak

portion of each scan and provide calculated V* for each scan using the Sutton et al. (2005)

580 calibration, which subtracts the pre-edge background.

581

- 582 Figure 3: Series of rapid V K-edge XAFS spectra collected on oxidized synthetic glasses
- 583 FAD_V1a (top) and FAD_V3 (bottom) conducted at a low and high flux density as described in

584 Fig. 2.

585

Figure 4: Series of rapid V K-edge XAFS spectra collected on natural MORB glass VG-2 at a
low and high flux density as described in Fig. 2.

588

- 589 Figure 5: Plots of the relative calculated V* for the rapid synthetic glass XAFS analyses
- 590 FAD_V2a (blue), FAD_V4 (red), FAD_V1a (gray) and FAD_V3 (green) shown in Fig. 2 and 3
- as a function of time. Calculated V* from the low flux density series are shown as open squares

- and from the high flux density series as filled circles. Error bars show estimated 1σ uncertainties in calculating V*.
- 594
- 595 Figure 6: V pre-edge intensity rate-of-change curves for oxidized FAD V1a, and FAD V3 and
- 596 for natural basaltic glasses VG-2 and A-99, collected at varying flux densities. Flux densities
- tested are labeled, covering more than 4 orders of magnitude. Intensities were measured at the V
- pre-edge peak at an energy of 5468.15 eV. Each time series consists of measurements at 1s
- intervals over 1500s of total irradiation. The vertical axis shows the relative change in the pre-
- edge peak intensity over time, so that for each curve the intensity of the first analysis point plots
- 601 at a value of 1.0.

Figure 1







Figure 3



Figure 4



Figure 5



Figure 6



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