

1 **Revision 2**

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4 **Magnetite-rutile symplectite in ilmenite records magma hydration in**
5 **layered intrusions**

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Abstract

20 The textures and geochemical characteristics of the rocks in layered intrusions
21 potentially provide insights into the physico-chemical processes that have taken place
22 in mafic magma chambers. Diverse exsolution textures of Fe-Ti oxides in layered
23 intrusions may record the variation of sub-solidus temperature and oxygen fugacity
24 (fO_2) of cooling magma chambers. Here we investigated ilmenite-hematite solid
25 solution (Ilm_{ss}) relationships evident in preserved intergrowths of magnetite-rutile and
26 ilmenite-hematite in the gabbro of the Xinjie layered intrusion. The crystallographic
27 orientation and 3-D morphology of the two intergrowth types constrain the
28 transformation mechanism of the exsolution textures from Ilm_{ss}. The results reveal
29 that the interface of the ilmenite-hematite intergrowth is more energetically favorable
30 than that of the magnetite-rutile symplectite when they are transformed from Ilm_{ss} on
31 cooling. The QUILF equilibria suggests that the magnetite-rutile symplectite can be
32 transformed from Ti-rich ilmenite with Ilm_{≥0.85} above 550°C when the sub-solidus
33 T- fO_2 trend is buffered by the biotite-ilmenite-feldspar-ulvöspinel (KUIIB) mineral
34 assemblages crystallized from hydrated mafic magmas. The magnetite-rutile
35 symplectite can be then taken as a unique texture indicator of magma hydration in the
36 evolution history of terrestrial, martian and lunar magmas.

37 **Key Words:** Magnetite-rutile symplectite; ilmenite-hematite solid solution (Ilm_{ss});
38 magma hydration; layered intrusion

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Introduction

41 Layered intrusions preserve the fully crystalline products of magmas that may
42 have experienced different physicochemical processes in mafic magma chambers (*e.g.*,
43 Holness et al. 2017 and references therein). One fundamental aspect on the

44 petrogenesis of layered intrusions that remains controversial is the link between rock
45 textures and magmatic processes (McBirney and Hunter 1995; Latypov et al. 2018;
46 Kruger and Latypov 2020). Despite intense studies in this area, studies concerning the
47 exsolution textures in minerals and the sub-solidus evolution of layered intrusions
48 have been rarely reported (Buddington and Lindsley 1964; McConnell 1975). The
49 diverse exsolution textures of Fe-Ti oxides in the rocks of layered intrusions have
50 been suggested to record the variation of temperature and oxygen fugacity (fO_2) of the
51 magma chamber during crystallization and sub-solidus cooling (Haggerty 1991; Frost
52 1991; Lattard et al. 2007; Brownlee et al. 2010). Understanding the transformation
53 mechanism of these exsolution textures is critical to constrain the sub-solidus cooling
54 processes of layered intrusions.

55 The ilmenite-hematite ($FeTiO_3$ - Fe_2O_3) solid solution (Ilm_{ss}) commonly occurs in
56 layered intrusions (Harrison et al. 2000). Ilm_{ss} tends to experience sub-solidus
57 re-equilibration and phase transformation during different T - fO_2 cooling paths,
58 forming hematite and/or magnetite exsolution and magnetite-rutile intergrowths in
59 ilmenite (Robinson et al. 2002; Tan et al. 2015, 2016; Guo et al. 2017). Experimental
60 results indicate that intergrowths of magnetite-rutile and ilmenite-hematite, which are
61 transformed from Ilm_{ss} , are thermodynamically equivalent over a large temperature
62 interval (Lindsley 1991). However, the magnetite-rutile intergrowth is rare in natural
63 rocks relative to the ilmenite-hematite intergrowth. The interfacial properties of
64 different phases are considered to be critical to the sub-solidus transformation
65 processes (Feinberg et al. 2004; Hammer et al. 2010; Wenk et al. 2011; De Yoreo et al.
66 2015; Xu et al. 2015, 2017), and may serve to solve this paradox. However, the
67 orientation relationships of Fe-Ti oxides have not yet been investigated so that the
68 effect of interfacial properties of different phases has not been fully understood. The

69 formation of the ilmenite-hematite intergrowth is usually ascribed to the
70 decomposition of Ilm_{ss} when temperature falls below that of the solvus (Harrison et al.
71 2000). In contrast, the magnetite-rutile intergrowth is likely related to fluids in layered
72 intrusions and metamorphic rocks (Southwick 1968; Tan et al. 2015; Guo et al. 2017).
73 However, there is no direct textural evidence for the oxidation of Ilm_{ss} reported so far.
74 Moreover, it remains enigmatic what controls the oxidizing T - $f\text{O}_2$ trends of the mafic
75 magmas from which the layered intrusions formed.

76 The $f\text{O}_2$ fluctuation and interfacial properties of the intergrowths have been
77 proposed to be potential factors affecting the sub-solidus transformation of Ilm_{ss}
78 (Lindsley 1991; Rohrer 2010). However, it remains unclear how different
79 intergrowths are developed during the transformation of Ilm_{ss} . In this study, we report
80 both magnetite-rutile symplectite and ilmenite-hematite intergrowth that are
81 transformed from the same Ilm_{ss} precursor in the Xinjie layered intrusion, SW China,
82 and examine the interfacial properties of the two intergrowths and the transformation
83 mechanisms involved. We use electron backscatter diffraction (EBSD) and
84 focused-ion beam–energy dispersive X-ray spectroscopy (FIB-EDS) tomography to
85 investigate the crystallographic orientation, 3-D morphology and texture of the two
86 intergrowths. We also use the compositions of the Fe-Ti oxides to constrain the
87 formation temperature (T) and $f\text{O}_2$ of different intergrowths in the QUILF equilibria
88 (Andersen et al. 1993). This study sheds light on the coherence between diverse
89 exsolution textures of Ilm_{ss} and sub-solidus T - $f\text{O}_2$ trends in a cooling mafic magmatic
90 system. As ilmenite is also ubiquitous in the lunar and martian magmatic rocks
91 (Raymond and Wenk 1971; Wang et al. 2004; Santos et al. 2015), the results in this
92 study can be helpful to the understanding of physicochemical conditions of magmatic
93 processes on the Moon and Mars.

94

Analytical methods

95 Electron backscatter diffraction

96 Thin sections from the analyzed samples were polished with 0.05 μm colloidal
97 silica for 3 hours to allow EBSD analysis. SEM imaging and EBSD analysis were
98 conducted on a Tescan MIRA3 Field Emission SEM, housed in the Microscopy &
99 Microanalysis Facility (John de Laeter Centre) at Curtin University, Perth, Western
100 Australia, and on a FEI Quanta 450 field emission gun SEM housed in the State Key
101 Laboratory and Geological Process and Mineral Resources (GPMR) of China
102 University of Geosciences (Wuhan). The EBSD measurement was performed with an
103 accelerating voltage of 20 kV and a working distance of ~ 20 mm. Electron
104 Backscatter Patterns (EBSPs) were automatically collected and indexed over a regular
105 grid with a 290 nm step size by using the Oxford Aztec 4.1 software. The CHANNEL
106 5+ software was used for plotting color-coded maps and the upper hemisphere
107 stereographic pole figures of the indexed mineral. Noise reduction was performed by
108 using a 'wildspike' correction and a five-neighbour zero solution extrapolation.

109 3D FIB-EDS tomography

110 The 3D tomography was performed using a Helios G4 Dual Beam Workstation at
111 the Thermo Fisher Scientific Inc., Shanghai. A selected volume was extracted from
112 the area of interest using the focused Ga-ion beam (acceleration voltage 30 kV) for
113 3D reconstruction. The chemical analyses of Fe, Ti, and O were carried out using
114 acceleration voltage 8 kV, beam current 13 nA. The energy-dispersive X-ray
115 spectroscopy (EDS) analysis was performed in mapping mode to investigate the two
116 dimensional distributions of Fe, Ti, and O. Serial cross-section slices were produced
117 by cutting the selected volume using focused Ga-ion beam, with a distance of 50 nm
118 between slices, and an EDS mapping was collected for every 3 milling steps. The

119 scripting routine is performed automatically with the “Auto slice and view 5.0”
120 software. After data collection, the 2D image sequences were aligned, cropped and
121 stacked into a 3D microstructure image. A total 3D volume of $31.6*11.8*13.25 \mu\text{m}^3$
122 with a voxel pixel of $11.53*11.53*50 \text{ nm}^3$ was reconstructed for further analysis.

123

124 **Results**

125 **Major petrographic features of the Xinjie layered intrusion**

126 The Xinjie intrusion is one of several layered intrusions in the Panzihua-Xichang
127 region in SW China (Fig. 1a). The intrusion is a NW-SE-striking, sill-like body
128 approximately 7.5 km long, 1-1.5 km wide and 1.2 km thick, and is divided, from the
129 base upwards, into a marginal zone and three lithological cycles (Unit I, II and III)
130 (after Wang et al. 2008). Unit I and II contain different modal proportions of olivine,
131 clinopyroxene, plagioclase and Fe-Ti oxides, forming interlayered wehrlite, olivine
132 gabbro, olivine clinopyroxenite, clinopyroxenite and melagabbro (Fig. 1b). Wehrlite
133 and olivine clinopyroxenite in Units I and II display similar texture, and contain <10
134 vol.% cumulus and intercumulus magnetite Fe-Ti oxides that are scattered in the
135 rocks (Fig. 2a, b). Hydrous silicates (*e.g.*, amphibole and biotite) are scarce in Units I
136 and II. Unit III is mainly composed of gabbro with < 30 vol.% Fe-Ti oxides (Fig. 2c,
137 d), but it hosts two thick (40-50 m thick) and one thin (~4 m thick) oxide gabbro
138 layers that contain 40-70 vol.% Fe-Ti oxides (Fig. 1b). The rocks of Unit III generally
139 contain 2-5 vol.% hydrous silicates, which are locally gathered and closely associated
140 with Fe-Ti oxides (Fig. 2d).

141 The exsolution textures in both cumulus and intercumulus ilmenite can be divided
142 into three types, which are distributed unevenly along the profile throughout the
143 intrusion (Fig. 1b). The ilmenite in Unit I and II is generally homogeneous in BSE

144 images and only displays local hematite lamellae (type-I, Fig. 3a, b). The ilmenite in
145 Unit III commonly contains symplectitic intergrowth of magnetite-rutile (type-II, Fig.
146 3c), and is closely associated with hydrous silicates (Fig. 3d, e). Both type-I and
147 type-II intergrowths are observed in the ilmenite of the melagabbro at the bottom of
148 Unit III, which was then selected to investigate in this study. In addition, the ilmenite
149 in the Fe-Ti oxide gabbro layer of Unit III contains magnetite exsolution (type-III, Fig.
150 3f).

151 **Appearance of magnetite-rutile symplectite and ilmenite-hematite intergrowth**

152 Magnetite-rutile symplectites in ilmenite comprise micro- to nano-scale anhedral
153 magnetite and rutile (Fig. 4a). The dendritic rutile tends to pinch outwards and is
154 truncated by magnetite (Fig. 4b). The symplectites have discrete and irregular
155 boundaries with the host ilmenite (Fig. 4c). Nano-scale hematite is evenly distributed
156 (Fig. 4d) and oriented parallel to the (0001) planes of the host ilmenite.

157 Most rutile grains are enveloped by continuous magnetite and show dendritic
158 shape in the 3D images (Fig. 5a and Supplementary movie1). The dendritic rutile
159 looks like isolated in the 2D backscattered electron (BSE) images, but is actually
160 interconnected in the 3D morphology (Fig. 5b and Supplementary movie2). Magnetite
161 appears as connected matrix in the symplectite (Fig. 5c and Supplementary Movie2).
162 Massive lens-like hematite lamellae have sharp contacts with the host ilmenite,
163 forming ilmenite-hematite intergrowth (Fig. 5d).

164 **Compositions of magnetite-rutile symplectite and ilmenite-hematite intergrowth**

165 The EMPA results indicate that rutile in the magnetite-rutile symplectite contains
166 95.97 to 98.63 wt.% TiO₂ and 1.38 to 3.59 wt.% FeO. Magnetite in the symplectite
167 contains 33.58 to 35.63 wt.% FeO, 59.50 to 63.70 wt.% Fe₂O₃ and 2.72 to 4.68 wt.%
168 TiO₂ (Table 1). The mineral mode of rutile in the symplectite is ~40 wt.% (~45 vol.%),

169 and magnetite is ~60 wt.% (~55 vol.%), so that the symplectite is estimated to contain
170 21.90 wt.% FeO, 36.82 wt.% Fe₂O₃, and ~41.19 wt.% TiO₂ in bulk composition
171 (Table 1).

172 The ilmenite-hematite intergrowth contains 41.39-43.05 wt.% FeO, 47.40-49.50
173 wt.% TiO₂ and 6.76-10.94 wt.% Fe₂O₃ in bulk composition (Table 2). Given that the
174 hematite lamellae mainly contain Fe and O based on the scanning transmission
175 electron mode with energy dispersive spectrometer (STEM-EDS) mapping (Fig. S1),
176 the variation of Fe₂O₃ is likely related to the uneven distribution of nano-scaled
177 hematite lamellae in the intergrowth.

178 **Crystallographic orientation of minerals**

179 The host ilmenite exhibits consistent crystallographic orientation (Fig. 6a, e). The
180 majority of magnetite in the symplectite shares a common $\{1\ 1\ 1\}_{\text{Mag}}$ plane, and a set
181 of corresponding $\langle 1\ 1\ 0 \rangle_{\text{Mag}}$ directions on the common $\{1\ 1\ 1\}_{\text{Mag}}$ plane. In detail,
182 there is an angular variation of ~4.5° across the magnetite (Fig. 6b). Similarly, the
183 majority of rutile shares a common $\{1\ 0\ 0\}_{\text{Rut}}$ plane and corresponding in-plane $\langle 0\ 0\ 1 \rangle_{\text{Rut}}$
184 $\langle 1\ 1\ 0 \rangle_{\text{Rut}}$ directions with an angular variation of ~16° (Fig. 6c). The lattice
185 orientations of a single rutile and magnetite grain record the progressive variation of
186 up to ~1.6° and ~3°, respectively (Fig. 6d). Note that the shared $\{1\ 0\ 0\}_{\text{Rut}}$ and $\{1\ 1\ 1\}_{\text{Mag}}$
187 planes fall into the same area as the $(0\ 0\ 0\ 1)_{\text{Ilm}}$ plane, and the shared $\langle 0\ 0\ 1 \rangle_{\text{Rut}}$
188 & $\langle 0\ 1\ 1 \rangle_{\text{Rut}}$ directions and $\langle 1\ 1\ 0 \rangle_{\text{Mag}}$ directions also fall into the same area as the $\langle 1\ 1\ 0 \rangle_{\text{Ilm}}$
189 directions (Fig. 6e-g). It is likely that the crystallographic orientations of the
190 majority of magnetite and rutile are controlled by the host ilmenite. Therefore, there is
191 an orientation relationship among the magnetite-rutile symplectites and the host
192 ilmenite, such that $\{1\ 0\ 0\}_{\text{Rut}} // \{1\ 1\ 1\}_{\text{Mag}} // (0\ 0\ 0\ 1)_{\text{Ilm}}$ and $(\langle 0\ 1\ 1 \rangle_{\text{Rut}} + \langle 0\ 0\ 1 \rangle_{\text{Rut}})$
193 $// \langle 1\ 1\ 0 \rangle_{\text{Mag}} // \langle 1\ 0\ -1\ 0 \rangle_{\text{Ilm}}$.

194 The crystallographic projections of both the host ilmenite and hematite lamellae
195 are parallel to each other along the (0 0 0 1) plane (Robinson et al., 2002). The lattice
196 fringes at the ilmenite-hematite interface run straightly across all the directions on the
197 high-resolution transmission electron microscopy (HRTEM) images (Fig. S2).

198

199

Discussion

200 Thermodynamic factors controlling the sub-solidus transformation of Ilm_{ss}

201 Ilmenite-hematite intergrowths are commonly interpreted as a sub-solidus
202 transformation product of Ilm_{ss} (Robinson et al. 2002). The HRTEM images reveal
203 that the ilmenite and hematite have the same crystallographic orientation and form
204 highly coherent interfaces in the intergrowth (Fig. S2), which can be attributed to their
205 crystallographic similarity (Robinson et al. 2002). The irregular morphologies of
206 magnetite and rutile indicate that the two minerals crystallized concurrently. In
207 addition, the orientation relationships between the magnetite, rutile and host ilmenite
208 indicate that their orientations are inherited from the Ilm_{ss} precursor (Fig. 6e-g). Thus,
209 the ilmenite-hematite intergrowth and the magnetite-rutile symplectite represent two
210 types of transformation products of an Ilm_{ss} precursor.

211 In general, the Fe-Ti oxides have distinctly different close-packed frameworks for
212 their oxygen atoms; hematite, ilmenite and rutile have "hexagonal close packing"
213 frameworks, whereas magnetite has a "cubic close packing" framework. Hematite and
214 ilmenite have oxygen atoms closely packed or nearly close-packed on the basal (0 0 0
215 1) plane and along the $\langle 1\ 0\ -1\ 0 \rangle$ direction (Fig. 7a, b). Magnetite has oxygen atoms
216 packed on the $\{1\ 1\ 1\}_{\text{Mag}}$ and along the $\langle 1\ 1\ 0 \rangle_{\text{Mag}}$ (Fig. 7c). Rutile has oxygen atoms
217 packed on the $\{1\ 0\ 0\}_{\text{Rut}}$ and along the $\langle 0\ 1\ 1 \rangle_{\text{Rut}} + \langle 0\ 0\ 1 \rangle_{\text{Rut}}$ (Fig. 7d). In this study,
218 the inherited orientations of the magnetite-rutile and ilmenite-hematite intergrowths

219 show that their oxygen atom frameworks are aligned consecutively along the
220 interfaces of the two intergrowths.

221 The sub-solidus transformation of Ilm_{ss} is thermodynamically determined by the
222 total Gibbs free energy change (ΔG) in a Fe-Ti oxide system, which can be expressed
223 as the equation:

$$224 \quad \Delta G = \Delta G_v + \Delta G_s + \Delta G_\xi$$

225 where ΔG_v refers to the Gibbs free energy change of phase transformation, ΔG_s
226 refers to the interfacial energy change due to new interface formation, and ΔG_ξ refers
227 to the interfacial strain energy change due to interface lattice misfit. Therefore, ΔG_v
228 is denoted as the driving force of the transformation, whereas ΔG_s and ΔG_ξ are
229 denoted as the energy barriers of the transformation (Rohrer 2010).

230 As the assemblage of magnetite + rutile are thermodynamically equivalent to that
231 of ilmenite + hematite (Lindsley 1991), the transformation of Ilm_{ss} into the
232 magnetite-rutile symplectite and ilmenite-hematite intergrowth would have the same
233 ΔG_v . Both ΔG_s and ΔG_ξ are determined by the interfacial properties of the Fe-Ti
234 oxides transformed from the Ilm_{ss} , and in turn, the interfacial properties of the Fe-Ti
235 oxides are mainly related to the symmetry and orientation of the oxygen atom
236 framework in each of the Fe-Ti oxides (Feinberg et al. 2004; Wenk et al. 2011). The
237 consecutive oxygen atom frameworks of the two intergrowths as shown in Figure 7
238 indicate that they share coherent or semi-coherent interfaces (Hammer et al. 2010; De
239 Yoreo et al. 2015). In this case, the ΔG_s can be treated as zero, the energy barrier ΔG_ξ
240 is then the key to determine the transformation path of Ilm_{ss} . The ΔG_ξ can be
241 estimated by the lattice misfit of the oxygen atom framework (δ) at their interfaces
242 (Feinberg et al. 2004; Wenk et al. 2011).

243 The lattice misfit of the oxygen atom framework along the interface of hematite

244 and ilmenite can be estimated using the oxygen atomic spacing of two minerals, *i.e.*,
245 $\delta_{\text{Ilm-Hem}} = (a_{\text{Ilm}} - a_{\text{Hem}}) / a_{\text{Ilm}}$, where a_{Ilm} and a_{Hem} refers to the oxygen atomic spacing of
246 ilmenite and hematite, respectively. The $\delta_{\text{Ilm-Hem}}$ is then estimated to be ~1% (Fig. 7a,
247 b). Likewise, the lattice misfit of the oxygen atom framework along the
248 magnetite-rutile interface ($\delta_{\text{Rut-Mag}}$) is estimated to be ~9.8% (Fig. 7c, d). The
249 relatively high $\delta_{\text{Rut-Mag}}$ value would increase the lattice misfit at the interface of
250 magnetite and rutile so that they need to adjust their orientations subtly during
251 coarsening. The intra-grain and inter-grain orientation variations of magnetite and
252 rutile (Fig. 6b-d) could produce high ΔG_{ξ} to hinder the transformation of Ilm_{ss} to the
253 magnetite-rutile intergrowth. In contrast, the low $\delta_{\text{Ilm-Hem}}$ value makes the formation
254 of ilmenite-hematite intergrowth energetically favorable when the temperature falls
255 below the solvus of Ilm_{ss} on sub-solidus cooling, in accordance with its high
256 frequency in natural occurrence.

257 **Transformation of magnetite-rutile symplectite from Ilm_{ss} precursor**

258 The textural relationship shown in 3D images indicates that the dendritic rutile in
259 the magnetite-rutile symplectite is likely the first phase exsolved from the Ilm_{ss} , and
260 predated the matrix magnetite (Figs. 5b, c). The bulk composition of the symplectite
261 is reconstructed to have ~37 wt.% Fe_2O_3 (Table 1), much higher than that for
262 coexisting ilmenite-hematite intergrowth (~8.5 wt.% Fe_2O_3 , Table 2), indicating that
263 the formation of the symplectite is related to the oxidation state rather than the
264 isochemical decomposition of the Ilm_{ss} precursor. The exsolution of rutile is ascribed
265 to the sub-solidus oxidation of Fe^{2+} to Fe^{3+} in Ilm_{ss} (Southwick 1968), which can be
266 expressed as $\text{Fe}_2\text{O}_3 \cdot 5\text{Fe}_2\text{TiO}_3_{\text{high Ti-Ilmss}} + \text{O}_2 = 3\text{Fe}_2\text{O}_3 \cdot \text{Fe}_2\text{TiO}_3_{\text{low Ti-Ilmss}} + 4\text{TiO}_2_{\text{rutile}}$.
267 The earlier exsolved rutile can act as a crystal seed, and significantly lower the
268 energy barrier needed for coarsening by absorbing Ti^{4+} in the Ilm_{ss} . The exsolved

269 rutile also creates an interface with the Ilm_{ss} . The lattice misfit ($\delta_{\text{Rut-Ilm}} \approx 6.9\%$) at the
270 rutile- Ilm_{ss} interface may cause segregation of Fe^{3+} from the Ilm_{ss} to the interface
271 (Zhang and Zhang 2020), resulting in Fe^{3+} enrichment at the interface. The Ti^{4+} loss
272 and Fe^{3+} enrichment along the rutile- Ilm_{ss} interface facilitate the growth of anhedral
273 magnetite along dendritic rutile (Fig. 5a). Therefore, the transformation of Ilm_{ss} to
274 the magnetite-rutile symplectites stems from the exsolution of rutile in Ilm_{ss} , which
275 is intrinsically attributed to $f\text{O}_2$ elevation during sub-solidus cooling.

276 **T- $f\text{O}_2$ trend for transformation of magnetite-rutile symplectite from Ilm_{ss}**

277 The transformation paths of Ilm_{ss} on the sub-solidus T- $f\text{O}_2$ trends depend on the
278 contents of Fe-Ti oxides, and components of coexisting silicates and fluids (*e.g.*,
279 H_2O and CO_2) in host rocks (Frost 1991). When the rocks are buffered by anhydrous
280 silicates (*e.g.*, clinopyroxene and olivine), Ilm_{ss} follows a slightly oxidizing T- $f\text{O}_2$
281 trend (QUILF, Fig. 8) and transforms to ilmenite-hematite intergrowths when the
282 temperature falls below the solvus (Harrison et al. 2000). When the rocks contain
283 hydrous silicates (*e.g.*, amphibole and biotite), Ilm_{ss} follows a steeply oxidizing T- $f\text{O}_2$
284 trend, *e.g.*, the cooling trend of KUIIB (Fig. 8), and transforms to the
285 magnetite-rutile symplectite.

286 In this study, the bulk composition of the Ilm_{ss} for the investigated sample of the
287 Xinjie intrusion is estimated to be $\text{Ilm}_{0.85}\text{Hem}_{0.15}$ (Table 2), and coexisting
288 titanomagnetite is $\text{Usp}_{0.45}\text{Mag}_{0.55}$ (Table S1). Our modeling results indicate that
289 $\text{Ilm}_{0.85}\text{Hem}_{0.15}$ and $\text{Usp}_{0.45}\text{Mag}_{0.55}$ can crystallize simultaneously at 952°C and
290 $\text{FMQ}+0.51$ (point “a” in Fig. 8). The $\text{Ilm}_{0.85}\text{Hem}_{0.15}$ may have experienced two-stage
291 transformation along the sub-solidus T- $f\text{O}_2$ trends. Increasing $f\text{O}_2$ of KUIIB would
292 induce “oxy-exsolution” of $\text{Usp}_{0.45}\text{Mag}_{0.55}$ at $\sim 825^\circ\text{C}$ (point “b” in Fig. 8), and the
293 oxidized $\text{Ilm}_{0.85}\text{Hem}_{0.15}$ at $\sim 550^\circ\text{C}$ would form the magnetite-rutile symplectite (point

294 “c” in Fig. 8) and Ti-rich solid solution of $\text{Ilm}_{0.92}\text{Hem}_{0.08}$. The $\text{Ilm}_{0.92}\text{Hem}_{0.08}$ is then
295 decomposed into the ilmenite-hematite intergrowth on subsequent cooling (Fig. 8).
296 This can well explain why the two intergrowths could occur in the same ilmenite
297 grains. We infer that the Ilm_{ss} with higher Ti content than $\text{Ilm}_{0.85}\text{Hem}_{0.15}$ tends to
298 transform to the magnetite-rutile symplectites at temperature above 550 °C when the
299 system is buffered by hydrous silicates (Fig. 8).

300

301

Implications

302 Magma hydration is key to increase the $f\text{O}_2$ of magma (Veksler and Hou, 2020)
303 and modify the crystallization sequence of evolved mafic magmas, triggering
304 crystallization of extensive chromite and Fe-Ti oxide in layered intrusions (Reynolds
305 1985; Pang et al. 2008; Boudreau 2016; Veksler and Hou, 2020). This study reveals
306 that the magnetite-rutile symplectite in Ilm_{ss} is essentially developed associated with
307 the mineral assemblages crystallized from hydrated magmas. Therefore, the
308 magnetite-rutile symplectite transformed from Ilm_{ss} can provide important clues of
309 magma hydration, which is critical to understanding the subsolidus cooling history
310 and related chromite/ Fe-Ti oxide mineralization of layered intrusions elsewhere.

311 Water is also of importance to the evolution and crystallization of lunar and
312 martian magmas (Gross et al. 2013; Hui et al. 2013; Filiberto et al. 2019). Although
313 primary hydrous minerals in the lunar and martian rocks are direct evidence for
314 hydrated magmas, they may be obscured by hydrothermal alteration, metamorphism,
315 weathering, solar wind implantation and meteorite impacts (Spandler et al., 2005;
316 Sharp et al. 2013; Hui et al. 2013; Jolliff et al. 2019). On the other hand, ilmenite is
317 ubiquitous in the martian and lunar rocks (Haggerty 1991; Wang et al. 2004; Santos et
318 al. 2015) and is less susceptible to subsequent overprints. Primary textures that

319 formed during crystallization and sub-solidus cooling can be well preserved in
320 ilmenite (*e.g.*, Fig. 3). Therefore, magnetite-rutile symplectites may be an indicator of
321 magma hydration in martian and lunar magmas even when other hydrous phases are
322 no longer present.

323

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452

453 **Figure captions:**

454

455 **Fig. 1. Geological background and lithological characters of the Xinjie intrusion.**

456 **(a)** A schematic geological map of the Xinjie intrusion in the Emeishan Large Igneous
457 Province (ELIP) in SW China; **(b)** A stratigraphic column that cuts through the Xinjie
458 intrusion showing the major rock types, distribution of Fe-Ti oxides, and exsolution
459 types in ilmenite in the intrusion. Note there are three types of exsolution textures in
460 ilmenite, including Type-I (ilmenite-hematite intergrowth), Type-II (magnetite-rutile
461 symplectite) and Type-III (magnetite exsolution).

462

463 **Fig. 2. The occurrences of silicates and Fe-Ti oxides from different lithological**
464 **units of the Xinjie intrusion. (a)** Anhedral and euhedral Fe-Ti oxides as interstitial

465 phases among clinopyroxene in **Unit I**, transmitted light; **(b)** Elongated and anhedral
466 Fe-Ti oxides as interstitial phases among silicates in **Unit II**, transmitted light; **(c)**
467 Cumulus Fe-Ti oxides in the Fe-Ti oxide gabbro at the bottom of **Unit III**, transmitted
468 light; **(d)** Anhedral and euhedral Fe-Ti oxides, amphibole and biotite as interstitial
469 phases among plagioclase in **Unit III**. Ol, olivine; Cpx, clinopyroxene; Pl, plagioclase;
470 Amp, amphibole; Bt, biotite.

471

472 **Fig. 3. BSE images of ilmenite grains hosting different types of exsolution**
473 **textures in the Xinjie intrusion. (a)** Euhedral ilmenite as inclusions in silicates

474 showing no exsolution (**Type-I**) in **Unit I**; **(b)** Elongated ilmenite showing
475 well-oriented hematite lamellae (**Type-I**) in **Unit II**; **(c)** Occurrence of
476 magnetite-rutile symplectite (**Type-II**) in ilmenite grain from the leucogabbro in **Unit**
477 **III**; **(d)** Biotite coexisting with plagioclase and ilmenite hosting magnetite-rutile
478 symplectite (**Type-II**) from the olivine gabbro at the bottom of **Unit III**; **(e)**
479 Amphibole coexisting with plagioclase, titanomagnetite and ilmenite hosting
480 magnetite-rutile symplectite (**Type-II**) from the olivine gabbro at the bottom of **Unit**
481 **III**, note the skeletal titanomagnetite (left) formed by sub-solidus reaction; **(f)**
482 Occurrences of magnetite (**Type-III**) in different ilmenite grains coexisted with
483 cumulus titanomagnetite in the Fe-Ti oxide gabbro at the bottom of **Unit III**. Ilm,
484 ilmenite; Hem, hematite; Tmt, titanomagnetite; Sym, symplectite; Mag, magnetite.

485

486 **Fig. 4. Textures in ilmenite from the melagabbro of Unit III in the Xinjie**
487 **intrusion, sample X-327. (a) magnetite-rutile symplectite in ilmenite (Ilm); (b)**
488 **vermicular rutile (Rut) occurs as a core and magnetite (Mag) presents as connected**
489 **matrix in the magnetite-rutile symplectite; (c) Miniscule magnetite and rutile present**
490 **near the boundary of the magnetite-rutile symplectite; (d) well-oriented nano-scale**
491 **hematite (Hem) lamellae parallel to the (0001) plane of the host ilmenite.**

492

493 **Fig. 5. Three-dimensional morphologies of magnetite-rutile symplectite in**
494 **ilmenite. (a) typical occurrence of rutile, magnetite and hematite in ilmenite; (b)**
495 **dendritic rutile and isolated rutile grains, note that the dendritic rutile is**
496 **interconnected; (c) magnetite surrounding large dendritic rutile grain; (d) nano-scale,**
497 **lens-like hematite lamellae homogeneously distributed in ilmenite.**

498

499 **Fig. 6. Microstructure and orientation for the major phases in the**
500 **magnetite-rutile symplectite and host ilmenite constructed from EBSD data. (a)**
501 **Phase-color map of magnetite (Mag, green), rutile (Rt, fuchsia) and host ilmenite**
502 **(Ilm); (b) magnetite lattice orientation variations to 4.5° from the red cross (TC_Mag,**
503 **texture component for magnetite); (c) rutile lattice orientation variations to 16° from**
504 **the red cross (TC_Rt, , texture component for rutile); (d) Grain reference orientation**
505 **deviation angle (GROD angle) showing the deviation angle from the average**
506 **orientation of a rutile grain and its surrounding magnetite; (e-g) lower hemisphere**
507 **equal area projection patterns of host ilmenite, magnetite matrix and vermicular rutile,**
508 **colored with their phase colors in (a). The circles and triangles indicate the parallel**
509 **planes and directions of different minerals, respectively. Note: the data on each model**
510 **indicate the periodic distance of every four oxygen atoms along $\langle 1\ 0\ -1\ 0 \rangle_{\text{Ilm}}$, $\langle 1\ 0$**
511 **$-1\ 0 \rangle_{\text{Hem}}$, $\langle 1\ 1\ 0 \rangle_{\text{Mag}}$, and $\langle 0\ 1\ 1 \rangle_{\text{Rut}} + \langle 0\ 0\ 1 \rangle_{\text{Rut}}$, respectively.**

512

513 **Fig. 7. Space filling models showing the symmetries of oxygen atom frameworks**
514 **of (a) ilmenite, (b) hematite, (c) magnetite and (d) rutile on their specific**
515 **orientations.**

516

517 **Fig. 8. The diagram of $\Delta \log fO_2$ (FMQ) versus Temperature showing the**
518 **isopleths of Fe-Ti oxide solid solution and the cooling trend of KUIIB buffer**
519 **(biotite-ilmenite-feldspar-ulvöspinel) (modified after Frost, 1991 and Harrison et**

520 **al., 2000**). Oxygen fugacity and temperature determined by QUILF-95 at $P = 5$ kbar
521 (Table S2); Usp₁₀ refers to solid solution of Ulvöspinel₁₀-Magnetite₉₀ (in molar
522 fraction) and Ilm₇₀ refers to Ilmenite₇₀-Hematite₃₀, and so on; $\Delta \log fO_2$ refers to the
523 FMQ buffer; “**a**” refers to the crystallization T - fO_2 condition for Ilm_{0.85}Hem_{0.15}
524 coexisting Usp_{0.45}Mag_{0.55}, “**b**” refers to the T - fO_2 condition for “oxy-exsolution” of
525 Usp_{0.45}Mag_{0.55}, and “**c**” refers to the T - fO_2 condition for transformation of
526 Ilm_{0.85}Hem_{0.15} to magnetite-rutile symplectite and ilmenite-hematite intergrowth.

TABLE 1. Major oxide compositions of rutile and magnetite in the magnetite-rutile symplectite (in wt.%)

Major oxides	Rutile						Magnetite							Bulk composition* (in average)
	1	2	3	4	5	6	1	2	3	4	5	6	7	
SiO ₂	0.02	0.01	0.02	0.00	0.01	0.02	0.02	0.01	0.00	0.00	0.27	0.01	0.00	0.03 (0)
MgO	0.01	0.02	0.01	0.04	0.03	0.03	0.05	0.02	0.00	0.00	0.00	0.00	0.01	0.02 (0)
Al ₂ O ₃	0.00	0.01	0.00	0.02	0.02	0.02	0.00	0.01	0.01	0.00	0.04	0.01	0.00	0.01 (0)
FeO*	2.43	2.91	1.38	2.63	3.59	2.87	34.91	33.58	34.01	35.06	34.85	35.63	35.12	22.14 (0.69)
Fe ₂ O ₃ *	-	-	-	-	-	-	59.50	63.70	62.15	61.68	61.41	60.42	60.68	37.28 (1.33)
MnO	0.02	0.01	0.00	0.00	0.00	0.00	0.04	0.01	0.00	0.00	0.00	0.01	0.02	0.01 (0)
NiO	0.00	0.00	0.00	0.00	0.00	0.00	0.08	0.09	0.09	0.01	0.07	0.05	0.05	0.04 (0)
Cr ₂ O ₃	0.00	0.00	0.00	0.03	0.03	0.00	0.02	0.03	0.02	0.03	0.02	0.01	0.00	0.01 (0)
TiO ₂	97.77	96.63	98.63	97.29	95.97	96.43	4.50	2.72	3.36	4.06	3.64	4.68	4.34	40.48 (2.02)
Total	100.24	99.59	100.03	100.01	99.64	99.36	99.11	100.16	99.64	100.84	100.30	100.81	100.21	100.02 (0.01)

*Notes: Redistribution of the measured ΣFeO between Fe₂O₃ and FeO is on the basis of charge balance and stoichiometry of magnetite; the average bulk composition of the rutile-magnetite symplectite is based on the modal proportion analysis of rutile/ magnetite ratios on BSE images, rutile takes up ~45 vol.% (~40 wt.%) and magnetite ~55 vol.% (~60 wt.%) in average; standard deviations have listed in the parentheses.

TABLE 2. Bulk major oxide compositions of ilmenite-hematite intergrowth (in wt.%)

Major oxides	1	2	3	4	5	6	7	8	9	10	Average	Bulk composition including two intergrowths
SiO ₂	0.00	0.00	0.00	0.02	0.01	0.01	0.01	0.03	0.02	0.00	0.01 (0.01)	0.02 (0)
MgO	0.33	0.34	0.35	0.35	0.35	0.33	0.35	0.34	0.33	0.32	0.34 (0.01)	0.16 (0.03)
Al ₂ O ₃	0.00	0.00	0.00	0.01	0.00	0.00	0.01	0.00	0.00	0.00	0 (0)	0.01 (0)
FeO*	42.06	43.03	42.77	42.90	42.31	43.24	42.32	41.39	41.75	43.05	42.54 (0.63)	39.55 (0.64)
Fe ₂ O ₃ *	9.42	7.54	8.14	7.86	8.93	6.76	8.56	10.94	10.66	7.59	8.45 (1.32)	14.78 (1.35)
MnO	0.67	0.65	0.74	0.66	0.71	0.67	0.65	0.66	0.61	0.65	0.67 (0.02)	0.35 (0.07)
NiO	0.03	0.01	0.00	0.02	0.00	0.05	0.00	0.01	0.03	0.00	0.02 (0.02)	0.02 (0.01)
Cr ₂ O ₃	0.03	0.00	0.00	0.04	0.03	0.00	0.01	0.01	0.00	0.00	0.01 (0.01)	0.02 (0.01)
TiO ₂	48.17	49.25	49.08	49.12	48.52	49.50	48.45	47.40	47.75	49.22	48.70 (0.70)	44.72 (0.87)
Total	100.70	100.83	101.08	100.97	100.86	100.55	100.35	100.78	101.14	100.83	100.73 (0.01)	99.65 (0.26)
											X_{Ilm} = 0.92	X_{Ilm} ≈ 0.85
											X_{Hem} = 0.08	X_{Hem} ≈ 0.15

*Notes: Redistribution of the measured Σ FeO between Fe₂O₃ and FeO is on the basis of charge balance and stoichiometry of ilmenite; the bulk composition including the ilmenite-hematite intergrowth and the rutile-magnetite symplectite; the modal proportions of symplectite in different ilmenite grains range from 34% to 58%, and the resulted standard deviations of the bulk composition are listed in the parentheses; X_{Ilm} and X_{Hem} refer to mole fractions of ilmenite and hematite, respectively; standard deviations have listed in the parentheses.

Fig.1

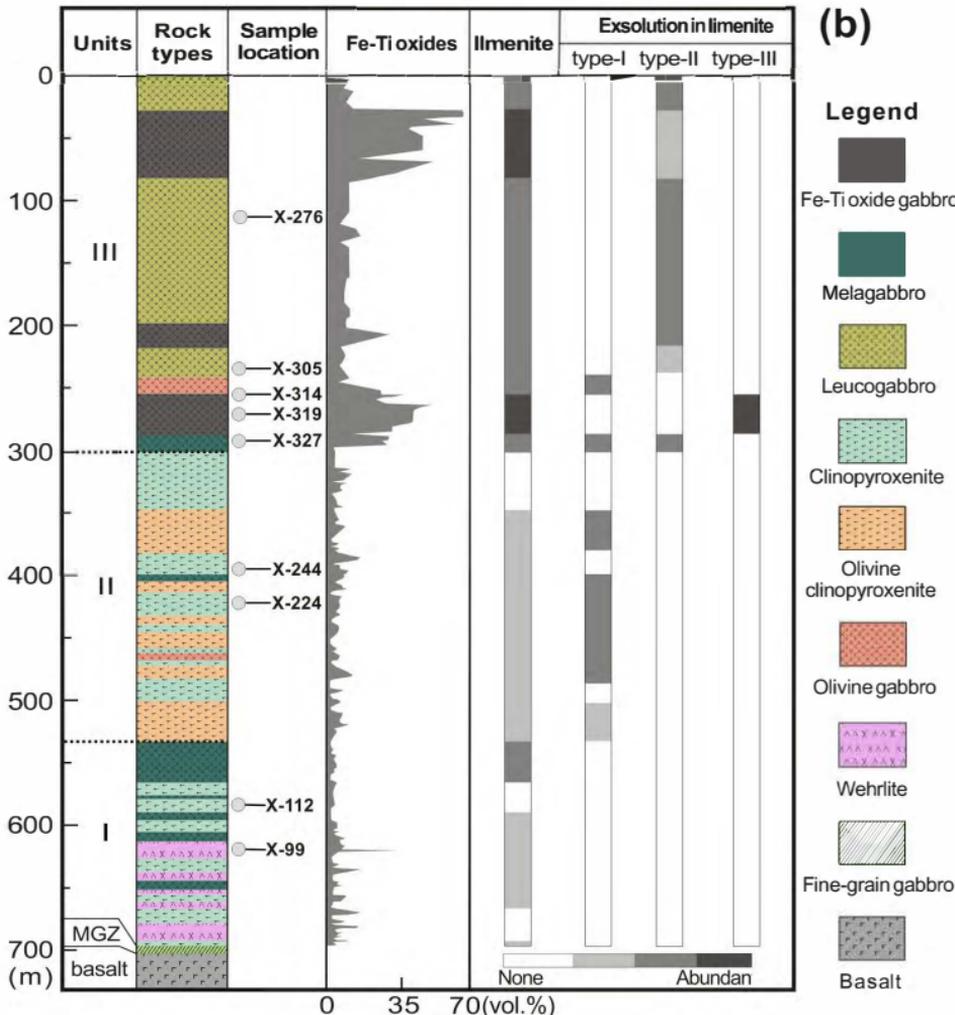
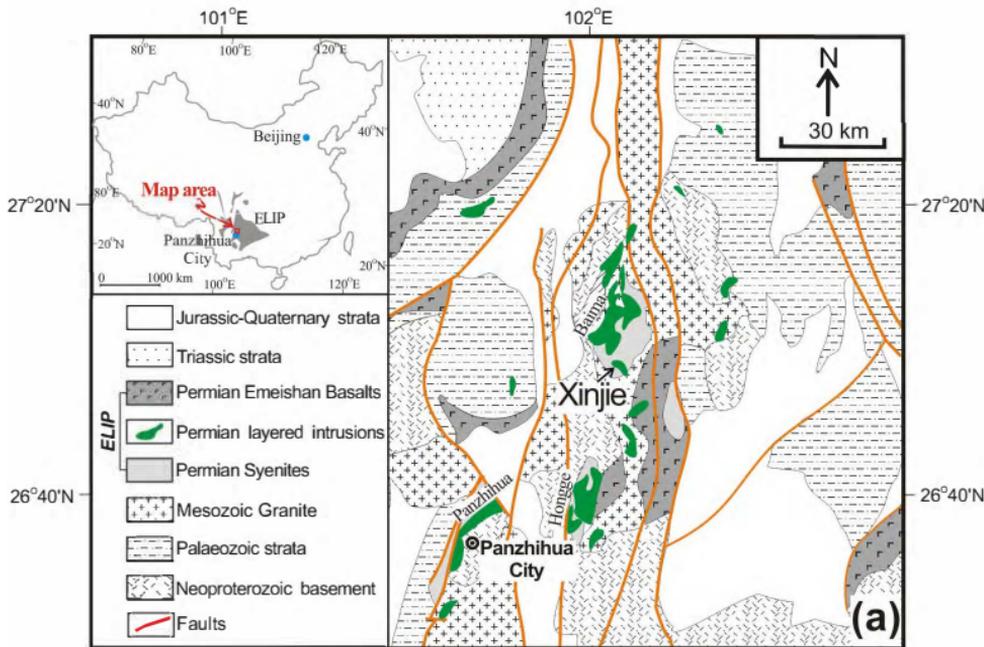


Fig.2

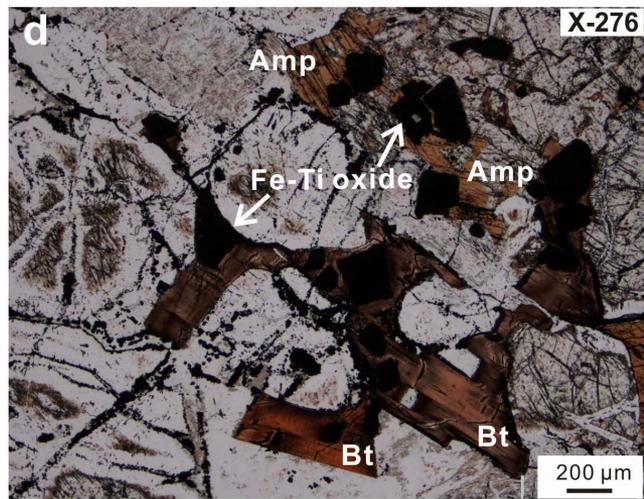
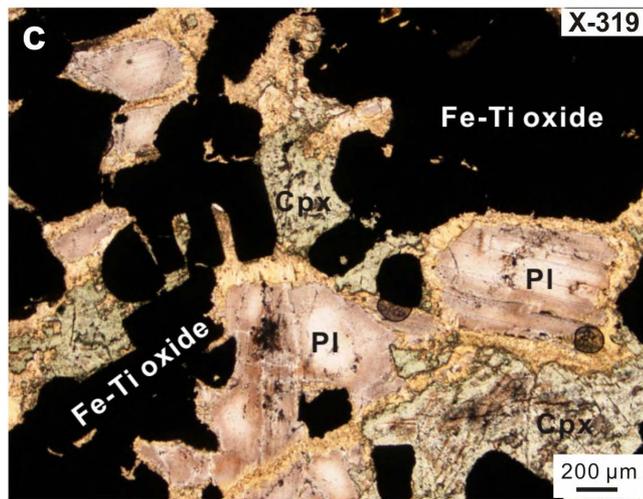
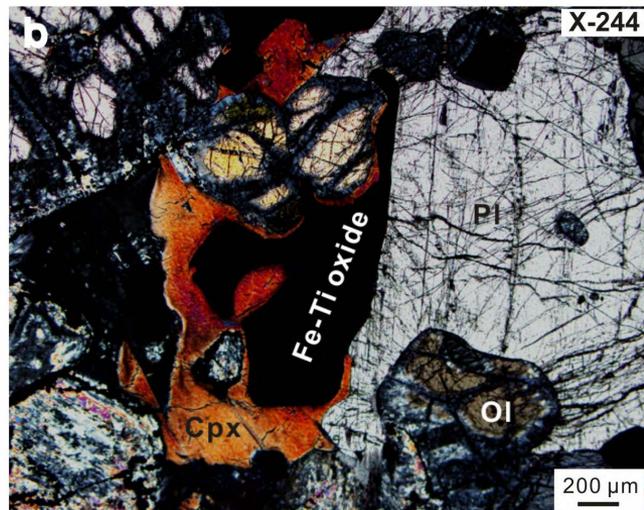
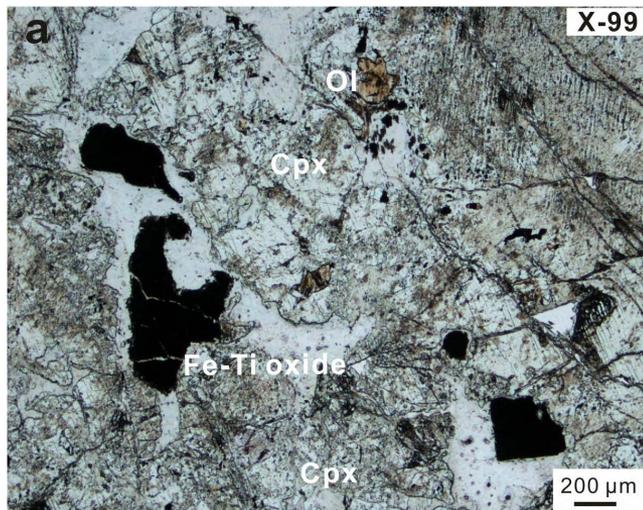


Fig.3

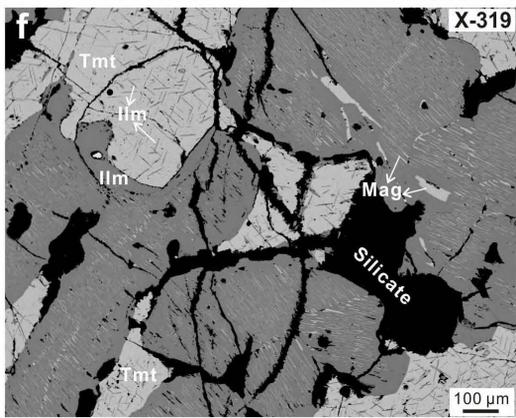
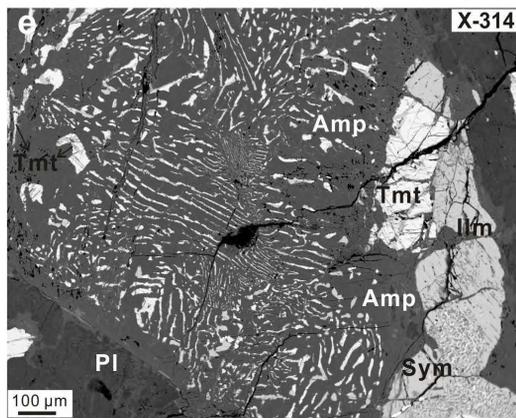
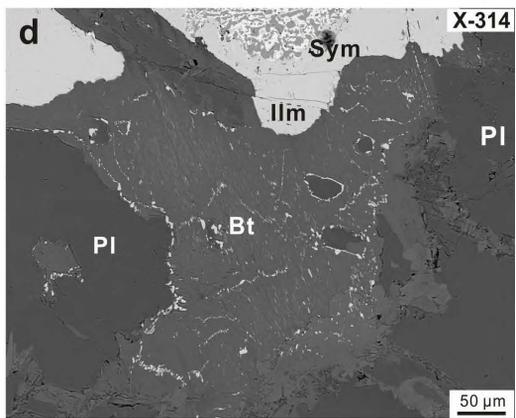
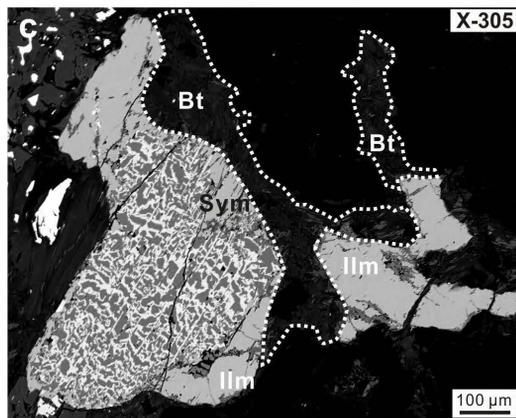
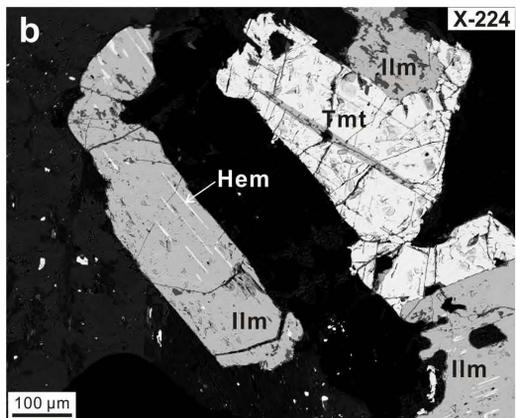
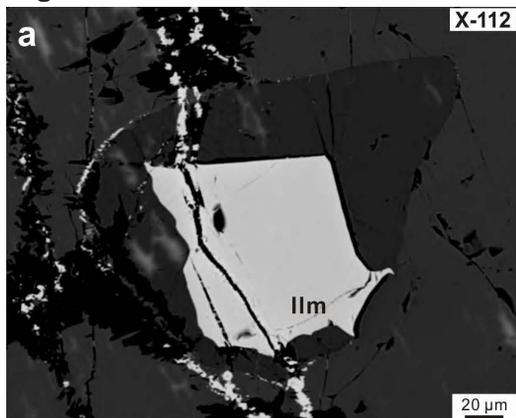


Fig.4

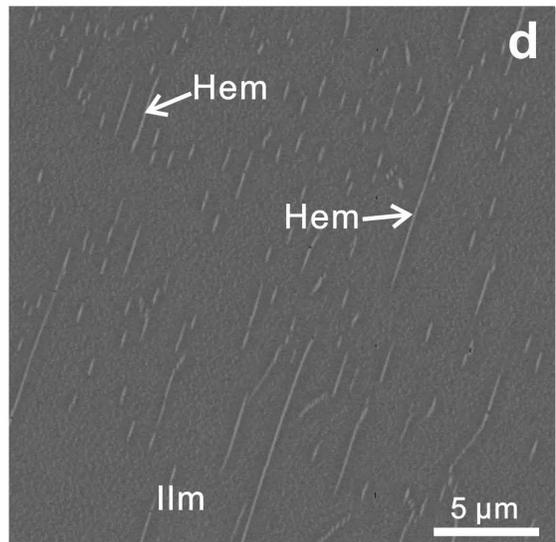
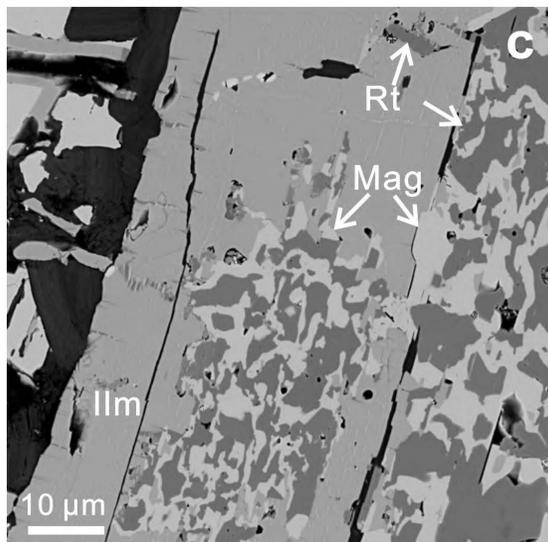
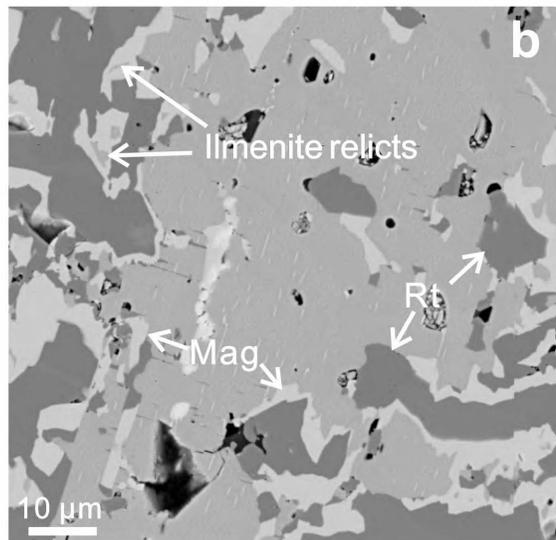
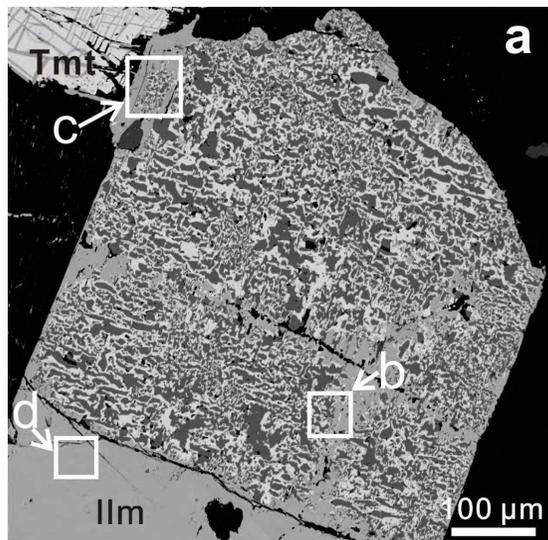


Fig.5

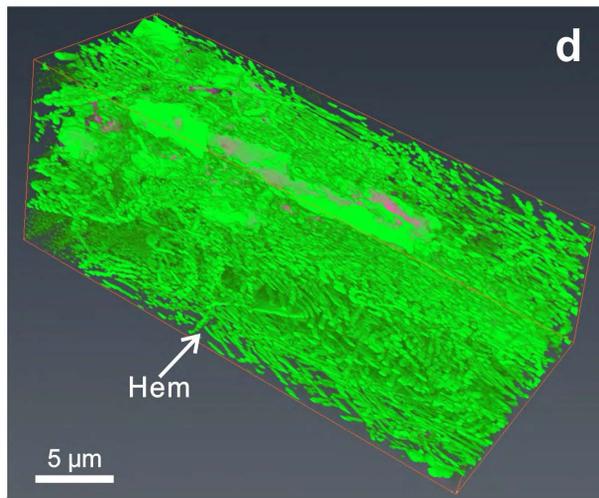
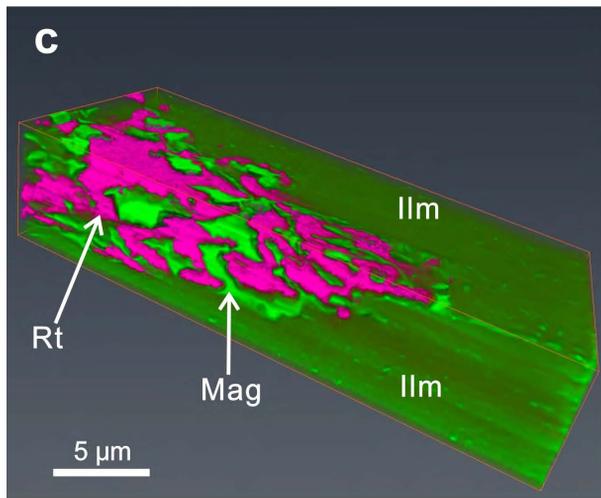
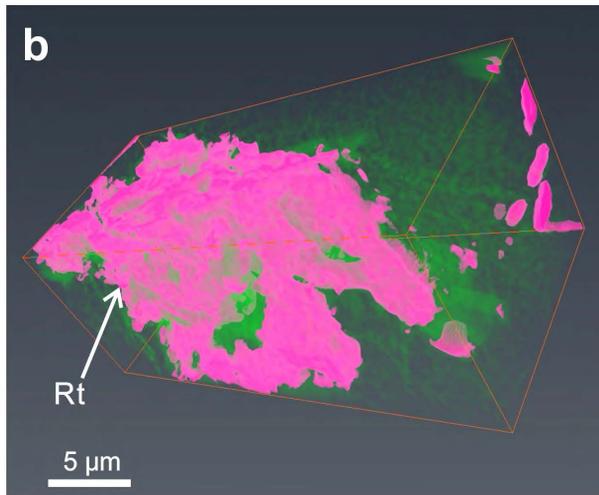
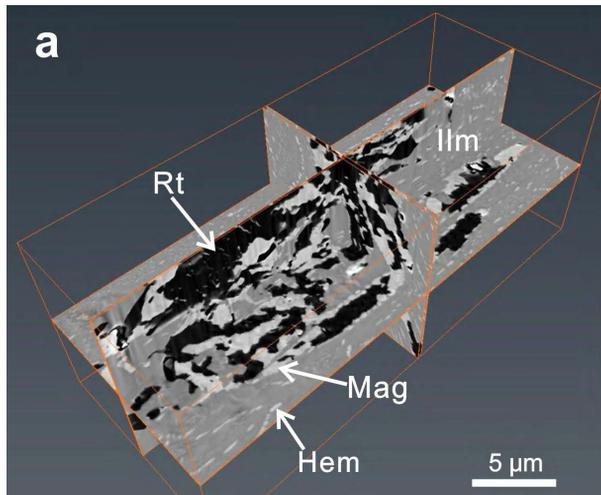


Fig. 6

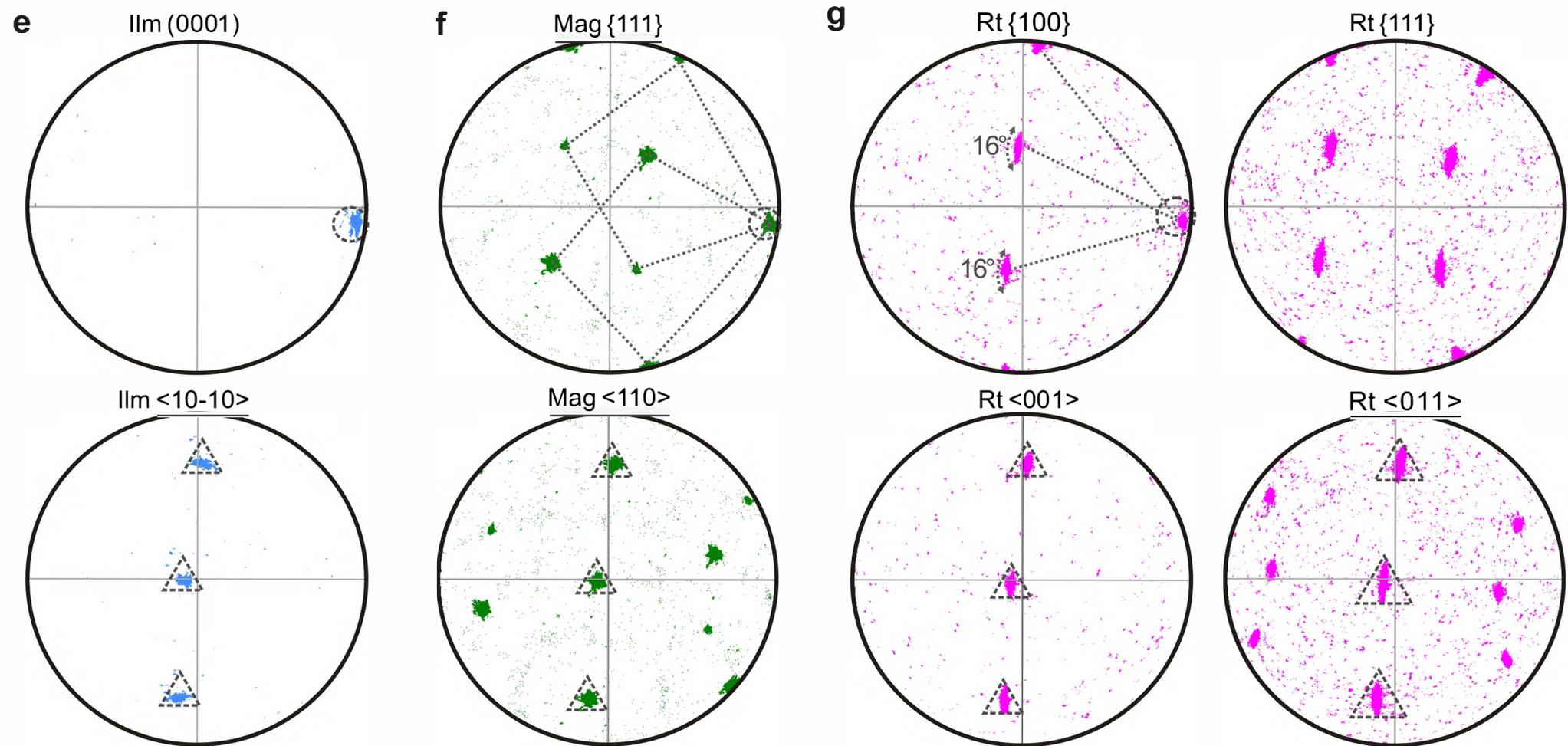
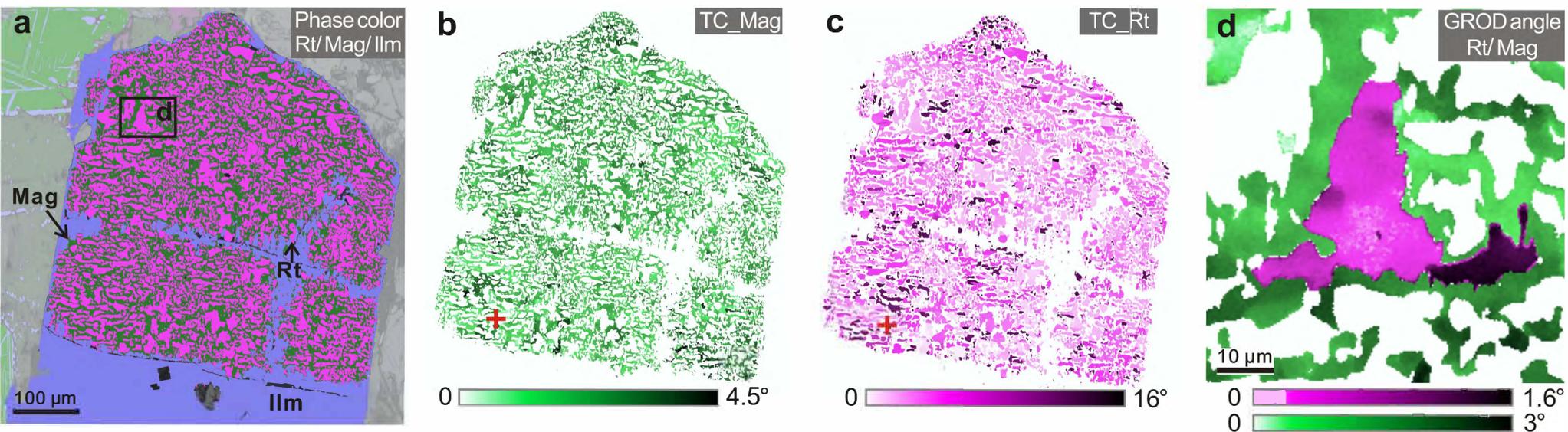


Fig.7

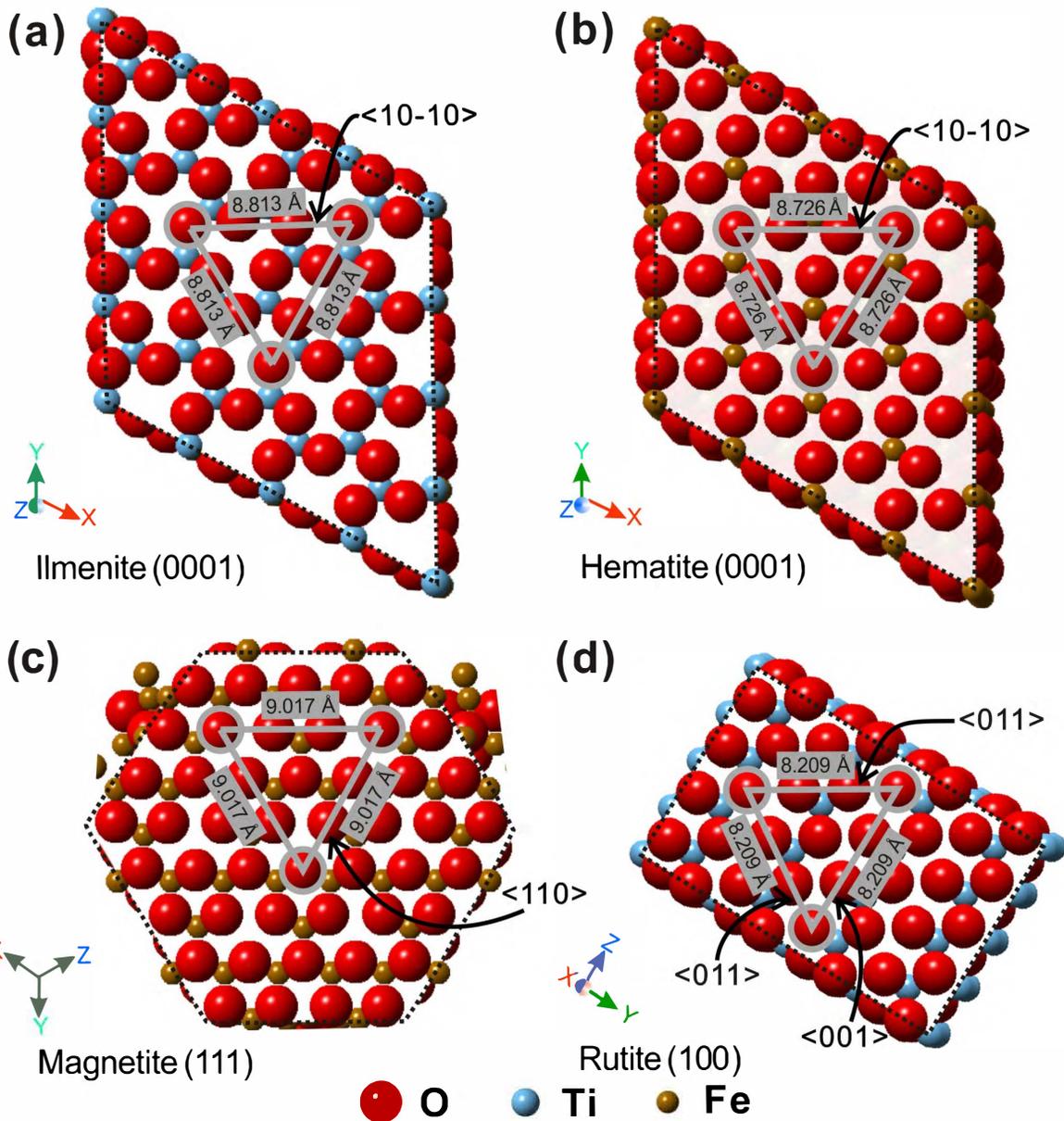


Fig.8