

Revision 2

Deconvolution of the composition of fine-grained pyrite in sedimentary matrix by regression of time-resolved LA-ICP-MS data

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Abstract

Pyrite is a common mineral in sedimentary rocks and is the major host for many chalcophile trace elements utilized as important tracers of the evolution of the ancient hydrosphere. Measurement of trace element composition of pyrite in sedimentary rocks is challenging due to fine grain size and intergrowth with silicate matrix and other sulfide minerals. In this contribution, we describe a method for calculation of trace element composition of sedimentary pyrite from time-resolved LA-ICP-MS data. The method involves analysis of both pyrite and pyrite-free sediment matrix, segmentation of LA-ICP-MS spectra, normalization to total, regression analysis of dependencies between the elements and calculation of normalized composition of the mineral. Sulfur is chosen as an

23 explanatory variable, relative to which all regressions are calculated. The S content value
24 used for calculation of element concentrations from the regressions is calculated from the
25 total, eliminating the need for independent constraints. The algorithm allows efficient
26 measurement of concentrations of multiple chalcophile trace elements in pyrite in a wide
27 range of samples, including quantification of detection limits and uncertainties while
28 excluding operator bias. The data suggest that the main sources of uncertainties on pyrite
29 composition are sample heterogeneity and counting statistics for elements of low
30 abundance. The analysis of regression data of time-resolved LA-ICP-MS measurements
31 could provide new insights to the geochemistry of the sedimentary rocks and minerals. It
32 allows quantification of ratios of elements that do not have reference material available
33 (such as Hg) and provides estimates on content of non-sulfidic Fe in the silicate matrix.
34 Regression analysis of the mixed LA-ICP-MS signal could be a powerful technique for
35 deconvolution of phase compositions in complex multicomponent samples.

36

37 Highlights:

- 38• Sedimentary pyrite is an important host of many trace elements
- 39• LA-ICP-MS analysis of sedimentary pyrite results in variable, mixed data
- 40• Pyrite composition is calculated from the mixed data using an algorithm based on linear
41 regression
- 42• The algorithm involves segmentation of the time-resolved signal, normalization to total,
43 calculation of the regression equations and estimation of explanatory variable
44 normalizing to total

- 45• Regression analysis of time-resolved mixed LA-ICP-MS data is a powerful technique for
- 46 analysis of complex substances
- 47

48

49 **Introduction**

50 Pyrite is a common mineral in sedimentary rocks and can host a wide range of
51 trace elements (e. g., Abraitis et al. 2004; Gregory et al. 2015). Trace element
52 compositions of pyrite have been applied to various geochemical problems ranging from
53 tracing chemistry of paleo oceans, nutrient availability and evolution of the atmosphere
54 and biosphere (Huerta-Diaz and Morse 1990; Large et al. 2014, 2015b, 2015a; Gregory et
55 al. 2015; Zhou et al. 2017; Rickard et al. 2017) to release of economically-important
56 elements during metamorphism (Pitcairn et al. 2010; Large et al. 2011; Tomkins 2013;
57 Goldfarb and Groves 2015). The content of pyrite and its trace element composition in
58 sedimentary rocks can be measured by selective digestions by acids (Huerta-Diaz and
59 Morse 1990; Raiswell et al. 1994); however, such methods yield average compositions of
60 pyrite and cannot resolve different generations of the mineral. The challenges for *in situ*
61 analysis of pyrite in sedimentary rocks include fine grain size, low concentration levels of
62 many key elements, abundant inclusions of matrix minerals and intergrowth with other
63 sulfides (Fig. 1).

64 LA-ICP-MS is a powerful analytical technique commonly applied for analysis of
65 pyrite (e.g. Large et al. 2009). Typical analyses are performed by a round laser spot with
66 diameter 20-30 μm but in many sedimentary rocks the size of pyrite is smaller than
67 lateral spatial resolution of the laser (Fig. 1). Analysis could be performed with smaller
68 spot sizes, however, this is rarely practical due to decrease of signal and increase of
69 downhole elemental fractionation (Gilbert et al. 2014a).

70 Analysis of sedimentary pyrite by LA-ICP-MS results in the acquisition of
71 complex mixed data with variable contributions from phases other than pyrite (Fig. 2).
72 There are two principal methods of recalculation of the mixed compositional data:
73 subtraction and regression (Stepanov et al. 2012). When applied to pyrite, both methods
74 involve analysis of matrix of the rock with minimal pyrite content and mixtures of the
75 pyrite and matrix (Large et al. 2014). For the subtraction method, the integration interval
76 with high pyrite content is selected from the transient signal of the mixed analysis. The
77 weight fractions of the pyrite in the mixture is then calculated from pyrite stoichiometric
78 Fe content, followed by the subtraction of the matrix component from the mixed analysis.
79 This method has a number of limitations such as an assumption of a constant composition
80 of the matrix across the sample, subjective selection of the integration interval and the
81 complexity of propagation of the uncertainties into the final compositions.

82 Regression analysis is a statistical method for quantification of the dependencies
83 between variables, which previously has been applied to deconvolution of mixed LA-
84 ICP-MS analyzes of fine grains of accessory minerals in glasses for experimental work
85 on mineral/melt partitioning of trace elements (Rubatto and Hermann 2007; Stepanov et
86 al. 2012; Taylor et al. 2015). In this contribution we present new developments in the
87 application of regression analysis to measurement of pyrite trace element composition in
88 black shales by LA-ICP-MS, which overcome the major problems of the subtraction
89 method. We describe the algorithm which is capable of generating self-consistent
90 compositions of pyrite in complex matrixes and is able to estimate uncertainty of the data
91 and detection limits without assumptions on the composition of the sulfide or its grain

92 size. We also discuss the application of the regression approach to analysis of some
93 geochemical features of the sedimentary rocks.

94

95 **Analytical technique**

96 The samples of black shales were cut by diamond saw and mounted into 25 mm
97 round epoxy blocks and polished. Samples were analyzed for trace elements at CODES
98 Analytical Laboratories, University of Tasmania. A RESOLution 193 nm excimer laser
99 ablation system coupled with an Agilent 7700 ICP-MS was used for all analyses. The
100 instrument is equipped with a Laurin Technic S155 ablation cell capable of housing
101 sample holders with up to 20 epoxy blocks. Samples were ablated in an atmosphere of
102 pure He flowing at a rate of 0.35 L/min. Helium carrier gas was mixed with Ar make-up
103 gas flowing at 1.05 L/min outside the ablation chamber. The ICP-MS was optimized to
104 maximize sensitivity on mid to high-masses (in the range 80–240 amu) and minimize
105 production of molecular oxide species ($^{232}\text{Th}^{16}\text{O}^+ / ^{232}\text{Th}^+$) at < 0.2%. During each analysis
106 a background signal was recorded for 30 seconds, after which the laser was turned on and
107 the sample was ablated for 30-60 seconds. The laser repetition rate was 5 Hz and the laser
108 fluence was set to 2.7 J cm^{-2} for sulfides and SDGL2b2 reference material
109 (Danyushevsky et al., 2011) while fluence 3 J cm^{-2} was used for GSD-1g glass. Moisture
110 and air absorbed to the sample could cause significant deleterious effect on LA-ICP-MS
111 analyses by increasing drift and interferences (Thompson et al. 2018). All samples
112 analyzed in this study were degassed under vacuum for at least 12 hours prior to analysis.

113 During analytical sessions two varieties of spot analyses were collected on the
114 sample. One type was analysis of the sedimentary matrix which contained minimal

115 amounts of pyrite and is referred to hereafter as matrix analyses. Another type was
116 analysis of pyrite grains of different textural types referred to hereafter as pyrite analyses.
117 The laser beam size used for the ablation was between 10 and 50 μm in diameter;
118 however, most pyrite analyses were performed with 20 to 30 μm beam, depending on the
119 grain size. The matrix analyses were acquired with either the same beam size as the pyrite
120 or with a 50 μm beam, in order to improve the detection limits for elements at low
121 abundance, providing pyrite-free areas in the matrix were large enough.

122 The ICP-MS method used in this study comprised 42 masses (Table 1). The major
123 elements were measured to allow quantification of concentrations by normalization to a
124 total. Tantalum and Gd were measured to assess interferences on Au. The dwell times
125 ranged from 5 ms for major elements to 50 ms for low abundance chalcophile elements
126 such as Au, Se and Te (Table 1).

127 An in-house pyrite reference material (PPP-1, Gilbert et al. 2014b) was analyzed
128 with the same spot size as unknown pyrite for sulfur calibration in order to account for
129 downhole fractionation during ablation of pyrite (Gilbert et al. 2014a). Reference
130 material GSD-1g (Guillong et al. 2005) was used for calibration of lithophile elements
131 and an in-house reference material STDGL2b2 (Danyushevsky et al. 2011) was used for
132 calibration of chalcophile and siderophile elements (see Table 1 for details). Each batch
133 of analyses included two sets of calibration reference materials at the beginning and end
134 of the batch, and 15 to 30 analyses on the samples. Each set of reference materials
135 included 6 analyses (each reference material was analyzed twice). On each sample, 5-7
136 analyses were performed on the “matrix” and 10-15 analyses on pyrite.

137 **Algorithm**

138 The derivation of pyrite composition from the mixed signal involves segmentation
139 of the signal, primary standardization, recalculation to the total, calculation of the linear
140 regression, estimation of the internal standard value for regression, calculation of the final
141 values, estimation of uncertainties from regressions and filtering of values below the
142 detection limits. Each step is described in detail below, followed by a summary of the
143 data reduction procedure.

144 **Segmentation of the signal**

145 Ablation of a sulfide grain in the matrix results in highly variable time-resolved
146 spectra (Fig. 2). The time-resolved LA signal was divided into five segments of identical
147 duration (Fig. 2a). The number of segments has been chosen to obtain a sufficient number
148 of points for regression analysis, and also to ensure that the number of data points in each
149 segment is sufficient to yield a meaningful average signal for elements of low abundance.
150 The segmentation of the signal was performed using an automated script which sets the
151 same integration interval for all reference materials and unknowns. The effect of
152 downhole fractionation on analytical results was therefore minimized by quantifying
153 every segment using the identical time interval on reference materials.

154 **Primary quantification and normalization of the analyses**

155 Quantification of LA-ICP-MS data requires knowledge of the concentration of at
156 least one element – the internal standard (Longerich et al., 1996). However, by the very
157 nature of analysis of a heterogeneous aggregate of grains of pyrite in matrix,
158 concentrations of all elements vary both between analyses and during ablation. Therefore,
159 internal standardization was performed in two stages. At the first stage when preliminary

160 concentrations were calculated, stoichiometric content of Fe in pyrite was used as the
161 internal standard value. Preliminary concentrations were calculated following the
162 protocol established by Longerich et al. (1996):

$$163 \quad C_{El}^{Sam} = \frac{R_{El}^{Sam}}{S} \quad (1)$$

164 Where C_{El}^{Sam} is the concentration of element El in the sample, R_{El}^{Sam} is count rate of the
165 element and S in the normalized sensitivity calculated as:

$$166 \quad S = \frac{R_{El}^{Cal}}{C_{El}^{Cal}} \left(\frac{R_{Fe}^{Sam}}{R_{Fe}^{Cal}} * \frac{C_{Fe}^{Cal}}{C_{Fe}^{Sam}} \right) \quad (2)$$

167 Where C_{El}^{Cal} is the concentration of the element in the reference material used for the
168 calibration of the element; R_{El}^{Cal} is measured count rate of the element in the reference
169 material; R_{Fe}^{Sam} is count rate of Fe measured on sample and R_{Fe}^{Cal} measured on the
170 calibration standard; C_{Fe}^{Sam} is the concentration of Fe in the sample, which was set to the
171 stoichiometric Fe content in pyrite; and C_{Fe}^{Cal} is content of Fe in the reference material
172 used for calibration of the element El. The instrumental drift and mass bias were
173 constrained at this stage by using a linear fit to standards for each analytical batch.

174 The second stage quantification was performed by normalising the sum of all
175 elements to 1,000,000 ppm (Leach and Hieftje 2000; Liu et al. 2010). In the case of
176 mixed pyrite-sediment data, we assume that lithophile elements (Na, Mg, Al, Si, K, Ca,
177 Ti, V, Cr, Mn, Rb, Sr, Zr, Ba, Gd, Hf, Ta, Th and U) are present as oxides in silicate
178 matrix and inclusions and other elements (Co, Ni, Cu, Zn, As, Se, Mo, Ag, Cd, Sn, Sb, Te,
179 W, Pt, Au, Tl, Pb, Bi) occur as sulfides. Therefore, the second stage normalization was
180 performed using the coefficient defined as:

181

$$c = 1,000,000 / (\sum_{all} C_{El} + 16 * \sum_{lith} C_{El} N_{El}^O / M_{El}) \quad (3)$$

183 Where the first member $\sum_{all} C_{El}$ is sum of the concentrations of all measured elements and

184 the second member $16 * \sum_{lith} C_{El} * N_{El}^O / M_{El}$ is mass of oxygen in oxides of lithophile

185 elements, where C_{El} is the concentration of lithophile element, M_{El} is atomic mass of

186 the element and N_{El}^O is number of oxygen atoms per atom of metal. All Fe is assumed to

187 reside in sulfide at this stage of calculations.

188

189 **Regression analysis of the data**

190 After the normalization of the data, the segmented analyses commonly show

191 strong correlations between chalcophile elements residing in pyrite (Fig. 3). For

192 regression analysis, one needs to choose an explanatory (independent) variable. For

193 pyrite, both S and Fe could be used as explanatory variables; however, S was chosen as

194 the more versatile component present in all sulfides. Linear minimal squares regression

195 implemented in Excel was used for quantification of the regression dependencies (see

196 Supplementary table for an example of calculation).

197 The relation between an element and sulfur is described by the regression

198 equations:

$$C_{El} = C_S A_{El} + B_{El} \quad (4)$$

200 Where C_{El} is the concentration of an element and C_S is sulfur content in the mineral. A_{El}

201 and B_{El} are the slope and the intercept of the regression.

202 There are several options for choosing the data for regression. Our default option
203 is a regression through segmented pyrite analyses and segmented matrix analyses (Fig 3),
204 which is referred to below as “pyrite+matrix”. An alternative option is calculation of the
205 regression from five points of the segmented pyrite analysis alone, which is referred as
206 “pyrite only” hereafter. Since there are more analyses of the matrix in the dataset, the
207 low-S end of the “pyrite+matrix” regression is well constrained. Segments rich in pyrite
208 mainly define the slope of the regression line.

209 However, there are instances when a regression based on pyrite and matrix
210 analyzes could give erroneous results:

- 211 • The distribution of an element is highly variable in the matrix (Fig. 4a). In this case
212 the regression based on pyrite and matrix data would pass through the average of the
213 matrix data, whereas the local matrix measured in such a case might be better constrained
214 by the regression calculated from “pyrite only”.
- 215 • Pyrite is mixed with a matrix component which is different from the analyzed matrix
216 (Fig. 4b). This situation also could be resolved by calculating a “pyrite only” regression.
- 217 • Significant variation in the proportion of pyrite and matrix during the analysis is
218 necessary for constraining the regression equations. However, in some samples pyrite is
219 ubiquitous, and analysis of “pure” matrix without pyrite is not possible (Figs. 1, 4c). In
220 cases when each segment of the pyrite analysis has a similar composition (i.e., the
221 segments are characterized by a relatively constant proportion of the matrix and pyrite),
222 an unconstrained regression may be obtained, when the slope of an Fe-S regression
223 becomes negative, precluding a meaningful set of regression equations (Fig. 5). Non-
224 robust regressions may also occur when there is a difference in the extent of downhole

225 fractionation of Fe and S between the reference material and pyrite sample. The set of
226 non-robust regressions could predict meaningless negative concentrations of elements. In
227 such situations pyrite compositions also could be calculated by “pyrite only” regression
228 by forcing the “pyrite only” regression line through origin (Fig. 5).

229

230 **Calculation of concentrations from regression equations**

231 To calculate the concentration of an element from equation 4, a value for the
232 explanatory variable (sulfur in our case) is required. Such value is analogous to the
233 internal standard value, which is commonly estimated from electron probe micro analyzer
234 (EPMA) data or from mineral stoichiometry (Longerich et al., 1996). The explanatory
235 variable value for the set of regression equations could be obtained from the condition
236 that the sum of the mass concentrations of all elements in the analysis is expected to
237 equal 1,000,000 ppm:

$$238 \quad \sum C_{El} = 1,000,000 \quad (5)$$

239 Summation of regression equations (Eq. 4) for all elements leads to:

$$240 \quad \sum C_{El} = C_S \sum A + \sum B = 1,000,000 \quad (6)$$

241 Where $\sum A$ is sum of the regression slopes for all elements and $\sum B$ is sum of the
242 regression intercepts. This equation could be rearranged for calculation of the value of
243 explanatory variable C_S as:

244
$$C_s = \frac{1,000,000 - \sum B}{\sum A} \quad (7)$$

245 Notably this formula is independent of the data set used for estimation of the regression
246 equations and regression for different elements, based on either “pyrite+ matrix” or
247 “pyrite only”, could be used.

248 **Estimation of uncertainties**

249 The regression analysis allows estimation of uncertainties on the calculated
250 parameters, including the slope and intercept and the regression prediction. Uncertainty
251 of prediction could be calculated either as the standard error on the mean or the standard
252 error on the forecast (e. g., Montgomery et al. 2012). The standard error on the forecast
253 estimates uncertainty of the next measurement, whereas the standard error on the mean
254 estimates uncertainty on the average. Since the aim of our calculations is to estimate
255 average composition of the ablated pyrite grains, the standard error on the mean is a more
256 suitable measure of uncertainty.

257 The uncertainty from the regression was calculated as the standard error on the
258 mean from the well known statistical formulae
259 (<http://people.duke.edu/~rnau/mathreg.htm>, e.g., Desalvo, 1971):

260
$$\delta_{EL} = \frac{s}{\sqrt{N}} \sqrt{1 + (C_s - \text{Average}(C_s))^2 / \text{Stdev}(C_s)^2} \quad (8)$$

261 Where s is standard error of regression calculated as
262 $\sqrt{(N-1)/(N-2)} * (1-r^2) * \text{Stdev}(C_{El})$, N number of measurements, r is regression
263 coefficient, C_s is the concentration of sulfur from Eq. 7, $\text{Average}(C_s)$ is the average
264 sulfur of the analyses, C_{El} is the concentration of an element for which regression is

265 calculated and $\text{Stdev}(C_{\text{El}})$ is the standard deviation of the concentration of an element in
266 the data set.

267 The regression uncertainties have the lowest values at the average S content and
268 increase away from this value (Fig. 6). In the case of “pyrite+matrix” regression, the
269 average is close to the matrix values, which are more numerous than segmented pyrite
270 analyses. This results in an uncertainty estimate which is defined by the heterogeneity of
271 the matrix (Fig. 6). However, the segmented pyrite analyses should define the regression
272 uncertainty since our aim is calculating the pyrite composition. Therefore, the Eq. 8 was
273 modified to:

$$274 \quad \delta_{EL} = \frac{s}{\sqrt{N}} \sqrt{1 + (C_s - \text{Average}(C_s^{\text{Py}}))^2 / \text{Stdev}(C_s^{\text{Py}})^2} \quad (9)$$

275 Where C_s^{Py} is sulfur in “pyrite only” data.

276 **Assessing compositional variations of elements for which reference** 277 **materials are not available**

278 Quantification of LA-ICP-MS analyses is based on reference materials with the
279 known concentration of the analyte (Longerich et al., 1996). The requirements for
280 reference materials include homogeneity, sufficiently high concentration, limited amount
281 of isobaric interferences and low uncertainty on the accepted value. Unfortunately, such
282 materials are still not available for a number of elements, one of which is Hg.

283 Mercury is an extremely volatile element and it could diffuse from many
284 compounds especially at high temperature. So far, a glass reference material with
285 homogeneous and known Hg content is not available because Hg is mobile during glass
286 preparation. Hg-bearing minerals also are not suitable as reference materials because the

287 high volatility of Hg results in increased Hg background hampering analysis of Hg in the
288 unknowns. Additionally, high volatility and high density of Hg results in its accumulation,
289 migration and redeposition in the ablation cell.

290 Mercury is a chalcophile element and it can occur in significant concentrations in
291 pyrite (Wells and Mullens 1973; Griffin et al. 1991; Abratis et al. 2004). Some analyses
292 of sedimentary pyrite show strong correlations between count ratio $^{202}\text{Hg}/^{57}\text{Fe}$ and other
293 chalcophile elements (Fig. 7a, b) which is clear evidence for the concentration of Hg in
294 the sedimentary pyrite. Other samples show negative correlations, suggesting that Hg is
295 hosted by the matrix (Fig. 7c). Considering low variability of Fe content in pyrite, Hg/Fe
296 count-rate ratio can be used for comparison of Hg abundances between samples.

297 Isotope ^{202}Hg was measured along with other isotopes and after the subtraction of
298 the gas background the signal was corrected for instrumentation drift during each session.
299 Mass bias variation between sessions was corrected by calculating $^{206}\text{Pb}/^{57}\text{Fe}$
300 fractionation using background corrected signal ratio measured on STDGL2b2 reference
301 material. The calculated Hg/Fe ratios were corrected for yield by the coefficient used for
302 normalization to total (Eq. 3). Regression equation of the segmented signal relative to S
303 was calculated, and the Hg/Fe ratio was extrapolated to the S content calculated by Eq. 7.
304 The ratios have been multiplied by factor of 1000 for reporting (Fig. 7). The calculated
305 Hg/Fe*1000 ratios could be used for qualitative comparison of samples, indicating which
306 samples have higher and lower contents of Hg.

307 **Estimation of limits of detection**

308 Our calculation of the detection limits in compositions of sedimentary pyrite was
309 based on the approach of Longerich et al. (1996). The detection limits for individual

310 segmented analyzes were calculated from the noise on the gas blank (laser off) using the
311 following formula:

$$312 \quad DL_{EL} = i * Stdev(BG_{El}) / S_{El} * \sqrt{1 / N^{BG} + 1 / N^{SM}} \quad (10)$$

313 Where DL_{EL} is the limit of detection of the element, i is a coefficient defining the
314 threshold above which the signal is accepted as exceeding noise level (3 in this study),
315 S_{El} is normalized sensitivity as calculated by Eq. 2, $Stdev(BG_{El})$ is the standard
316 deviation of the gas blank, N^{BG} is number of measurement (sweeps) of the gas blank and
317 N^{SM} in number of measurements during the ablation of the sample. Masses with low gas
318 background could have all readings zero in some analyses. In order to alleviate this
319 situation and also to obtain more robust estimates of the noise, the average gas blank
320 standard deviation for all analyses measured on sample mounts was used in Eq. 10.

321

322 **Data reduction procedure**

323 The algorithm presented was implemented in an in-house Excel spreadsheet.
324 Time-resolved spectra were segmented and then the preliminary concentrations were
325 calculated. The validity of the calibration and background correction was checked by an
326 operator. At the next stage, normalization to a total of 1,000,000 ppm was performed,
327 followed by the generation of regression plots for individual analyses (Fig. 3-5). The
328 matrix analyses were inspected for the presence of pyrite, and analyses with S content in
329 the matrix significantly deviating from the rest of the analyses were excluded. Analyses
330 with insufficient pyrite to generate robust regressions were also excluded on the
331 operator's discretion. At this stage decisions on whether to use "pyrite+matrix" or "pyrite
332 only" regressions were made, and the majority of elements were processed with the

333 default “pyrite+matrix” option. For elements of low abundance, “pyrite+matrix” was
334 consistently the most robust option.

335 The last step involved calculating the final compositions of pyrite as well as other
336 parameters such as uncertainties, detection limits, pyrite-free mixed compositions, and
337 averages of the matrix which were exported to the database. An example of data
338 reduction for a sample with an average composition of pyrite-matrix mixtures and the
339 calculated compositions of pyrite grains are presented in Figure 8 and the average data
340 and calculated pyrite compositions are presented as supplementary material.

341

342 **Discussion**

343 **Existing algorithms for deconvolution of mineral-matrix mixtures by** 344 **regression analysis**

345 Regression analysis is routinely applied at different stages of reduction of LA-
346 ICP-MS data, including calibration curves and modelling downhole fractionation (e. g.,
347 Paton et al. 2010). Regression is also extensively used during post-processing of data
348 obtained by LA-ICP-MS, including estimations of compositions when proper
349 quantification of analyses is not possible. For example, Izmer et al. (2013) used
350 regression analysis of LA-ICP-MS data of an Archean impact spherule layer to constrain
351 Pt/Ir ratio of the impactite. Another study has derived the compositions of ash particles
352 by applying regression to LA-ICP-MS analyses of mixtures of ash and mineral phases
353 (Spears 2004).

354 In a pioneering study, Rubatto and Hermann (2007) used regression analysis for
355 calculation of compositions of small experimental zircon grains in a silicate glass matrix.

356 Mixes of zircons with glass and other minerals were analyzed by a laser beam 19-32 μm
357 in diameter. The preliminary compositions of mixed analyzes were calculated assuming
358 CaO content of the glass measured by EPMA. The regression equations were calculated
359 relative to Zr and extrapolated to Zr contents in zircon measured by EPMA. A test
360 experiment was conducted, where a mix of crushed zircon crystal of a known
361 composition was melted with granitic melt by fast heating, which precluded reaction of
362 melt and the mineral. The regression analyses of the mixtures reproduced the
363 composition of the zircon starting material. In some experiments, zircons were analyzed
364 both in the glass matrix and included in garnet; the compositions calculated by a
365 regression for both cases produced consistent results. LA-ICP-MS analyses averaged
366 multiple grains and zones within individual grains, producing consistent results from
367 multiple analyses. The averages obtained matched well the averages of multiple EPMA
368 analyses, and also had significantly smaller uncertainty compared to the EPMA data.
369 Therefore, Rubatto and Hermann (2007) concluded that the compositions of zircons
370 obtained from regression of LA-ICP-MS data were especially suitable for estimation of
371 partitioning of elements between zircon and melt.

372 The study by Stepanov et al. (2012) applied regression analysis for calculation of
373 the composition of monazite grains, which ranged in size from $<1 \mu\text{m}$ to $>1 \text{mm}$, in
374 silicate glass. The preliminary compositions of mixes of monazite and glass were
375 calculated by using Si content in glass measured by EPMA. The regressions were
376 calculated relative to Ce, which is the major element in monazite. The linear regression
377 equations were calculated using all monazite and glass analyses from an experimental
378 charge. Cerium concentration was used as the explanatory variable for calculating

379 monazite compositions. The precision of calculated compositions was estimated from
380 regression statistics and accuracy was assessed by EPMA of several REE in monazite and
381 also by a comparison of results of regression-based calculations with spot analyses of
382 larger monazite grains. The estimated errors of regressions increased with decrease of
383 grain size of monazite and abundance of elements; however, the EPMA and LA-ICP-MS
384 data showed good agreement.

385 The applications of regression analysis to deconvolution of compositions of small
386 grains of accessory minerals is not limited to LA-ICP-MS. Taylor et al. (2015) employed
387 regression analyses for calculation of compositions of small zircons crystals (<15 μm
388 long and 1-3 μm wide), which were analyzed together with glass by Secondary Ion Mass
389 Spectrometer (SIMS) with a 20-25 μm spot. Silicon content in glass measured by EPMA
390 was used as the internal standard for calculating preliminary compositions.
391 Concentrations of REEs were calculated by a regression relative to Zr contents. Another
392 approach involved calculating the amount of zircon in the ablated material using digital
393 processing of SEM images of the analyzed spots with an assumption that the amount of
394 zircon remained constant due to shallow depths of ablation pits. Comparison of zircon
395 composition obtained from regression and image analysis demonstrated a satisfactory
396 agreement though the latter showed slightly overestimated the REE concentrations.

397 **New developments and future directions in deconvolution of mixed** 398 **LA-ICP-MS data**

399 The previous studies demonstrated the utility of regression analysis for estimating
400 compositions of small grains in homogeneous matrixes by LA-ICP-MS and SIMS and
401 now this approach appears to be one of the main methods for measurement of

402 experimentally-derived partitioning coefficients between accessory minerals and melts.
403 The review of previous works summarized in Table 2 demonstrates that when regression
404 analysis is applied to the mixed LA-ICP-MS data, common issues arise such as the
405 necessity to choose which element will be used as the explanatory variable, internal
406 standards for primary quantification, and the concentration of an explanatory variable
407 used for calculating the mineral composition from the regressions.

408 This contribution further develops the regression method by applying it to
409 analysis of sulfide minerals in a complex sedimentary matrix. Sedimentary matrix in
410 black shales is different from experimental glasses by higher micro-scale heterogeneity,
411 and is thus unsuitable for analysis by EPMA. Without external EPMA data, an alternative
412 method of primary quantification based on normalization to a total was applied in this
413 work.

414 The principal differences of the new algorithm from the previous work (Rubatto
415 and Hermann 2007; Stepanov et al. 2012; Taylor et al. 2015) include: automatic
416 segmentation of the data to intervals of identical length; normalization to a total for
417 preliminary internal standardization; and calculation of the final values by choosing the
418 value of the independent variable such that the total composition of the mineral of interest
419 sums to 100%. The advantages of our algorithm include: less subjective data processing;
420 elimination of the need for EPMA data; and internal consistency of the results. An
421 additional benefit of normalization to a total is the distribution of analytical uncertainties
422 between all measured elements (Leach and Hieftje 2000). Measurement of phase
423 compositions by deconvolution of mixed data could find multiple applications to various
424 Earth science problems and data could be collected with wide diversity of

425 instrumentation. Recent models of mass spectrometers (e.g., triple quad and sector field
426 ICP) provide advantages in reduction of interferences and increase of sensitivity.
427 However, it seems that instruments with fast scan across wide range of masses and ability
428 to record quickly changing transient signal are most suitable for the acquisition of mixed
429 data. In this context time-of-flight (TOF) ICP-MS instruments with their ability of
430 simultaneous measurement of multiple masses and extremely fast acquisition could be
431 especially interesting for these applications.

432 This study deals with data collected in spot mode, however, LA-ICP-MS
433 imaging/mapping is another approach for analysis of heterogeneous materials (e. g.,
434 Large et al. 2009). Segmentation of the data could be achieved from content of an
435 element strongly concentrated in the mineral of interest or from more complicated
436 algorithms such as the Monocle add-on in the Iolite program (Petrus et al. 2017).
437 However, calculation of regressions, limits of detection and uncertainties from data with
438 different numbers of contributing pixels/sweeps is more complicated and requires further
439 development.

440

441 **Sources of analytical uncertainty**

442 The overall uncertainty of analyses arises from several sources, some of which are typical
443 for all LA-ICP-MS analyses, whereas others are specific to the data reduction algorithm
444 presented in this work. Regressions of different elements show characteristically different
445 regression uncertainties (Fig. 9). The best regressions are typically observed for Ni, Co,
446 As, Se and Sb with the estimated concentration uncertainties commonly < 10% (Fig. 9).
447 These are elements which are most efficiently concentrated in sedimentary pyrite

448 (Gregory et al. 2015). Low abundance elements such as Au, Ag, Bi and Te commonly
449 have larger uncertainties, which could be attributed to the limitations of counting
450 statistics. The elements which commonly demonstrate poor regression relative to Fe and
451 S include Pb, Zn, Cd and Cu and regression uncertainties could be >100 % (exceptions
452 are also common; e.g. Figure 3 shows a perfect correlation of Pb and S in pyrite). Poor
453 regressions might be related to the concentration of these elements in micro-inclusions of
454 other sulfides such as galena, sphalerite and chalcopyrite within pyrite (Gregory et al.
455 2015; Hu et al. 2018). Despite the occurrence of these minerals as inclusions, our
456 algorithm can estimate the composition of the compound sulfide, because we utilize S as
457 the explanatory variable. The large uncertainties on individual measurements can be
458 mitigated by increasing the number of pyrite analyses within a sample, which would
459 lower the uncertainties of weighted averages and better constrain statistical distributions
460 of elements in pyrite and matrix. Another approach could be pooling of multiple pyrite
461 analyses for a single regression; however, in this case the data on pyrite heterogeneity
462 will be lost.

463 Counting statistics is the principal source of uncertainty in the measurement of
464 elements of low abundance. Plots of relative regression uncertainty versus concentration
465 show negative correlations (Fig. 10). The minimum observed error increases linearly with
466 decrease of element abundance in pyrite. Many analyses of As and Cu show substantially
467 higher error than the minimum value defined by the counting statistics. These analyses
468 could be explained by internal heterogeneity of these elements in pyrite. Uncertainties on
469 Se show a good correlation with Se content suggesting homogeneous Se distribution in
470 pyrite in agreement with observations by Large et al. (2019).

471 A potential source of systematic uncertainty could be different matrix-controlled
472 fractionation in analyses rich in silicate minerals. These effects would affect the intercept
473 of regression line. However, considering that the slope is mostly defined by the pyrite
474 analyses, the effect of the matrix effects on pyrite composition is expected to be smaller
475 than the fractionation.

476 Sulfur is a difficult element for LA-ICP-MS due to high ionization potential,
477 interferences from dimers of oxygens and high volatility, which leads to fractionation
478 from other elements (Gilbert et al. 2014b). Because S is used as the explanatory variable
479 in the calculation of regressions and by the normalization to a total content of all
480 chalcophile elements, the uncertainty in sulfur contents affects all other calculated
481 concentrations. Individual pyrite compositions calculated by regression may show
482 significant variations in Fe and S contents (Fig. 11), however, the agreement of the
483 averages with the stoichiometric composition of pyrite validates the analytical approach.

484 In LA-ICP-MS analyses of minerals and glasses, heterogeneity of reference
485 materials and instrumental drift are significant sources of uncertainty (e. g.,
486 Danyushevsky et al. 2011). The comparison of these values with the regression
487 uncertainties (Table 3), however, suggests that heterogeneity of reference materials is
488 minor relative to other uncertainties during the analysis of sedimentary pyrite.

489

490 **Analysis of non-sulfidic Fe content by LA-ICP-MS in the black shales**

491 Sedimentary rocks contain Fe in a range of minerals which could be important for
492 understanding of geochemistry of these rocks. For example, total Fe_T content in the rock
493 could be divided into highly reactive Fe (Fe_{HR}), which then could be subdivided to Fe in

494 pyrite (Fe_{Py}), carbonate (Fe_{carb}) and oxide (Fe_{ox}) (Canfield et al. 1986). The non-reactive
495 Fe component could be hosted by the silicate minerals (Canfield et al. 1986; Huerta-Diaz
496 and Morse 1990). Iron speciation of sedimentary rocks is one of characteristic parameters
497 used for estimating the conditions of sedimentation and diagenesis and contents of
498 different Fe species could be determined by wet chemistry methods such as sequential
499 acid digestions (Canfield et al. 1986; Huerta-Diaz and Morse 1990; Liu et al. 2016).

500 Spot LA-ICP-MS analyses of pyrite-free matrix could also be used for estimating
501 *in situ* Fe contents in the sedimentary matrix. However, analyses of the matrix commonly
502 contain S and display correlations between Fe and S. These correlations suggest that
503 analyses of such matrixes also contain nano-grains of pyrite and the Fe content of these
504 analyses is higher than $\text{Fe}_{\text{non-Py}}$. Regression analysis could be used for correcting for the
505 presence of micro-grains of pyrite in the matrix analyses and calculating the true *in situ*
506 $\text{Fe}_{\text{non-Py}}$. The regression of normalized Fe versus S produces a line whose slope is
507 controlled by the ratio of Fe and S in pyrite and extrapolates to the composition of pyrite.
508 The value of the intercept of the regression line is the content of Fe at zero S content and
509 hence it represents $\text{Fe}_{\text{non-Py}}$ value. Therefore, Eq. 4 could be simplified to:

$$510 \quad C_{\text{Fe non-Py}} = B_{\text{Fe}} \quad (11)$$

511 If the regression line intercept is negative, then regression might suggest that some S is
512 present in a non-sulfidic state (for example as sulfate). However, rocks with such
513 characteristics seem to be rare among black shales as they were not encountered by our
514 study.

515 The advantages of estimating the non-sulfidic Fe by LA-ICP-MS include high
516 spatial resolution in comparison with whole rock methods, and sensitivity to low

517 concentrations of Fe_{non-Py}. The sensitivity for low Fe concentrations might be of particular
518 interest because the wet chemistry methods are reliable at Fe_T>0.5% wt. (Clarkson et al.
519 2014), whereas the detection limit for Fe in LA-ICP-MS analysis of the matrix is
520 typically in the range 20-100 ppm. Whole rock Fe speciation could be estimated by
521 combining the *in situ* regression data with independent measurements of the abundance
522 of pyrite in the samples.

523

524 **Advantages of analysis of pyrite in matrix by LA-ICP-MS**

525

526 Several analytical methods are used for trace element analysis of small grains
527 including EPMA, SIMS and synchrotron micro-XRF. However, LA-ICP-MS has some
528 distinctive advantages:

529 • Most of the elements occurring as impurities in pyrite can be measured by LA-ICP-
530 MS over a large range of concentrations. The ability of ICP-MS to measure wide ranges
531 of concentrations makes it possible to quantify compositions of both pyrite and matrix
532 mixtures.

533 • Requirements for sample preparation are relatively modest for LA-ICP-MS. The
534 method is suitable for analysis of porous samples with a wide range of grain sizes,
535 provided that the samples are properly degassed prior to analysis.

536 • One approach used for analysis of sedimentary pyrite involved calculation of the
537 concentrations using stoichiometric Fe content as the internal standard (Scott et al. 2009).
538 This treatment results in apparent presence in the analyses of various lithophile elements.
539 For example (Scott et al. 2009) observed correlations of Al and V contents in pyrite

540 analyzed by this method and concluded that these elements occur as silicate inclusions.
541 The regression method allows corrections for the presence of silicate inclusions in a
542 sensible and consistent way.

543 • LA-ICP-MS analyses sample larger volumes compared to EPMA and SIMS. The
544 larger sampling volume might be an advantage in the case of sulfides and accessory
545 minerals. Sulfides are often highly heterogeneous (Gregory et al. 2015) and analyses
546 utilizing small sampling volumes may require large numbers of measurements to achieve
547 representative results. Huerta-Diaz and Morse (1990) proposed that acid digestion
548 methods for estimating pyrite trace element compositions also have an advantage of
549 averaging significant volumes of the mineral. However, LA-ICP-MS is preferred to the
550 bulk methods because it allows targeted analysis of various morphological and genetic
551 types of pyrite in the sample. This ability is especially important for applications
552 involving analysis of sedimentary pyrite with an aim of reconstructing ocean chemistry,
553 because a rock could contain multiple generations of pyrite including recrystallized and
554 metamorphic pyrite (Large et al. 2007; Meffre et al. 2016; Gregory et al. 2019).

555 • When reference materials are not available such as in the case of Hg, regression of
556 LA-ICP-MS data still allows calculation of ratios, which could be used for sample
557 comparison.

558 • Throughput is another important parameter. The RESolution S-155 ablation cell could
559 host 20 mounts and with 2-3 positions reserved for reference materials 17-18 sample
560 mounts could be analyzed in a single analytical session including measurement of
561 multiple calibration standards during the analytical run.

562 **Summary**

563 We describe an algorithm for quantification of LA-ICP-MS analyses of
564 heterogeneous samples composed of small grains of pyrite in a heterogeneous matrix.
565 The algorithm is based on linear regressions and can accommodate the effects of matrix
566 heterogeneity without requiring external data for quantification. The method is suitable
567 for quantification of composition of pyrite and mixtures of pyrite with other sulfide
568 minerals in sedimentary rocks. The output of the method includes concentrations of most
569 chalcophile elements, uncertainties of individual measurements and detection limits.
570 Regression analysis utilizes the wealth of information recorded in the transient signal of
571 LA-ICP-MS analyses of complex samples and represents a new step towards the
572 comprehensive characterization of complex geological materials.

573

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581 **Supplementary materials**

582 Supplementary materials related to this manuscript can be found online.

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715

716 **Figure captions**

717

718 Figure 1: Examples of textures in pyrite-bearing black shales in reflected light: a) Pyrite
719 (Py) grains of variable size disseminated in the matrix and replacing fragments; b) Partly
720 recrystallized framboidal pyrite typical for sedimentary rocks; c) Centimeter sized pyrite

721 nodule (concretion) with significant content of inclusions of sedimentary matrix in pyrite
722 and abundant inclusions of chalcopyrite (Cpy).

723

724 Figure 2: Examples of time-resolved LA-ICP-MS profiles of fine-grained pyrite in
725 silicate sedimentary matrix. a) Analysis with high pyrite content at the beginning
726 decreasing over the course of the analysis. The results of segmentation of this analysis are
727 presented in the Figure 3. b) Analysis containing both pyrite and matrix. c) Analysis of
728 relatively clean pyrite with small contribution of matrix.

729

730 Figure 3: An example of correlations between chalcophile elements and sulfur in
731 segmented pyrite and matrix analyses calculated by normalization on total. Sulfur is
732 plotted on horizontal axis and the concentrations of elements are shown on vertical axis.
733 The matrix analyses are characterized by the low S contents and lower content of
734 majority of chalcophile elements, which results in regressions with positive slopes. For
735 Zn and Cd the calculated regressions point to values lower than the estimated detection
736 limits. Fe-S regression has a positive intercept (Fe at zero S content), which suggests that
737 some of Fe in the matrix is in non-sulfide state.

738

739 Figure 4: Examples of special regression cases. Top diagrams show Fe-S plots, which
740 demonstrate mixtures of pyrite and matrix in the analyses, bottom diagrams show
741 examples of trace elements where “pyrite+matrix” and “pyrite only” regressions produce
742 different results. a) Tl content in matrix is highly heterogeneous and ‘pyrite+matrix’
743 regression predicts Tl content below the limit of detection. The regression based on

744 'pyrite only' analyses is preferred. b) Correlation of As and S suggests that As content in
745 the matrix mixed with pyrite was higher than measured. Therefore, As content in pyrite
746 was calculated by 'pyrite only' regression. c) Analyses of pyrite in a sample without
747 clean matrix but with inclusions of silicates in pyrite. Regression of segmented analyses
748 could be used for calculation of pyrite composition even when both pyrite and silicates
749 are heterogeneous, however see Figure 5 for the case of non-robust regression.

750

751 Figure 5: Negative correlation between Fe and S could result in set of non-robust
752 regression equations which are unsuitable for prediction of mineral composition. Forcing
753 the regression through the origin could solve this problem.

754

755 Figure 6: The error envelope of uncertainty on the mean. The uncertainty increases away
756 from the average of the population. The uncertainty calculated from matrix and mixed
757 pyrite-matrix analyses could overestimate the uncertainty; however, the uncertainty based
758 on mixed pyrite-matrix analyses is considered to be realistic. Here and elsewhere the
759 uncertainty is calculated for 90% confidence interval.

760

761 Figure 7: Examples of Hg in the sedimentary rocks as revealed by LA-ICP-MS of pyrite
762 bearing black shales. The vertical axis shows background corrected Hg/Fe count ratios
763 multiplied by 1000 for convenience. a) drilling through pyrite reveals significant Hg
764 content in pyrite and lower Hg/Fe ratios in the matrix. b) analysis of almost clean pyrite
765 with higher Hg/Fe ratios than sedimentary matrix. c) drilling through pyrite (with very

766 low Hg/Fe ratios) into the matrix with higher Hg/Fe ratios; the regression suggests Hg
767 below the limit of detection in this case.

768

769 Figure 8. An example of a set of pyrite analyses calculated for a sample of black shale.
770 Empty circles show the compositions of pyrite-matrix mixtures and the filled circles
771 show compositions of pyrite calculated by the regression analysis. The error bars show
772 regression uncertainties for 90% confidence intervals.

773

774 Figure 9. Relative uncertainty of the set of pyrite analyses from a sample of black shale.
775 The uncertainties are relatively low for elements strongly concentrated in pyrite.
776 However, they could reach very high values (over 100%) for such elements as Mn, Zn,
777 Cd and Tl, which are present mainly in minerals other than pyrite (Gregory et al. 2015).

778

779 Figure 10. Relative uncertainty versus concentration of an element in the analyses of
780 pyrite (all analyses were obtained with 30 μm spot size) from the data base of pyrite
781 compositions from black shales (Large et al. 2014). Good correlation of uncertainty and
782 concentration suggest Se concentration is usually homogeneous in pyrite and the
783 uncertainty is controlled by the counting statistics. For As and Cu, the correlation is much
784 weaker, suggesting that heterogeneous distribution of these elements in pyrite commonly
785 contributes to uncertainty.

786

787 Figure 11. Histograms of Fe and S contents in pyrite calculated by the regression method
788 shows distribution is close to stoichiometric for both.

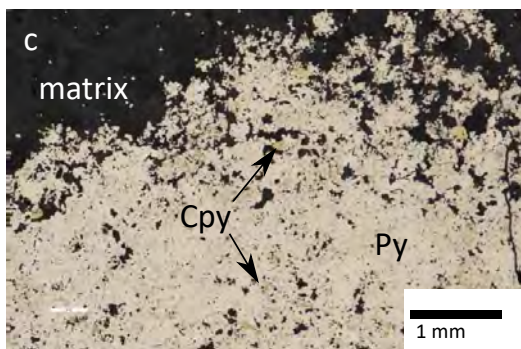
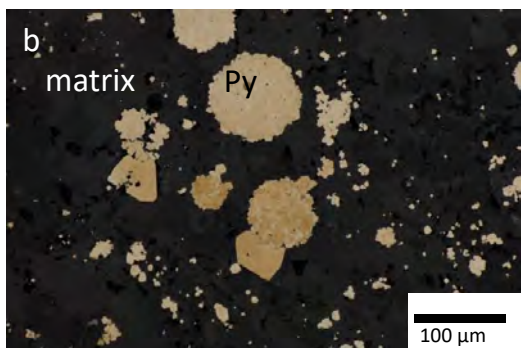
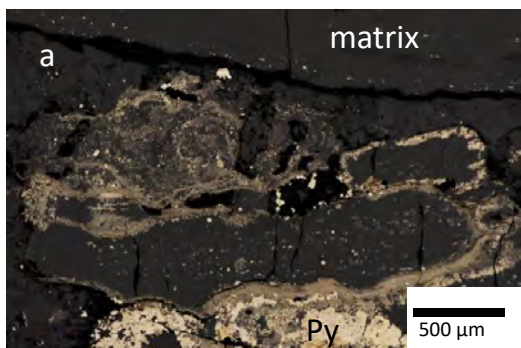


Figure 1

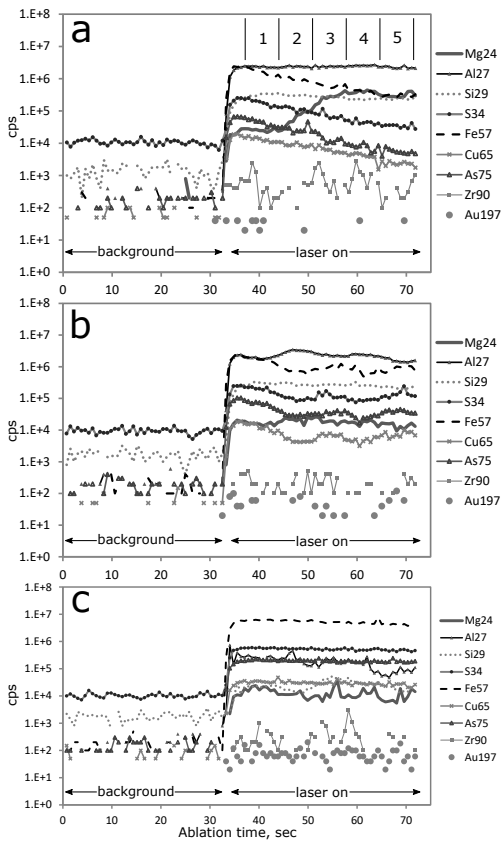


Figure 2

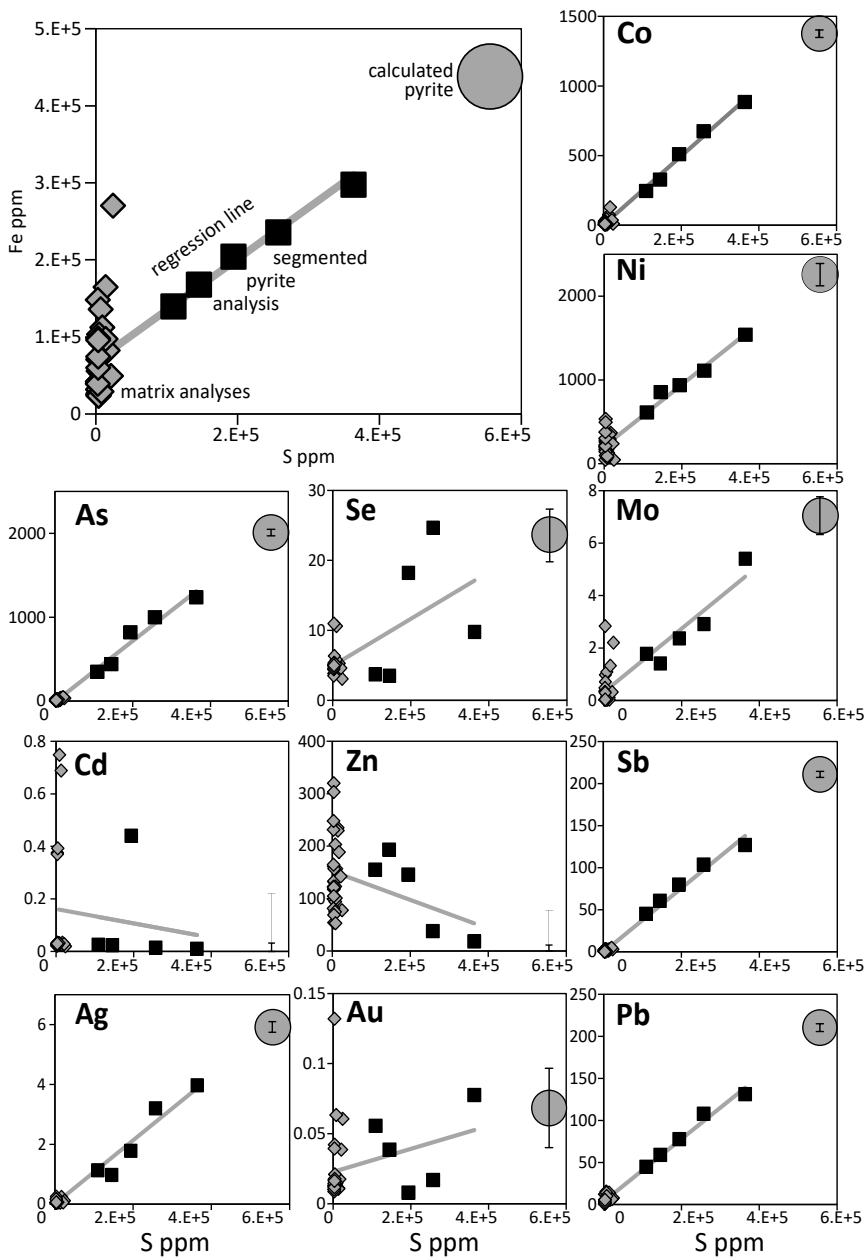


Figure 3

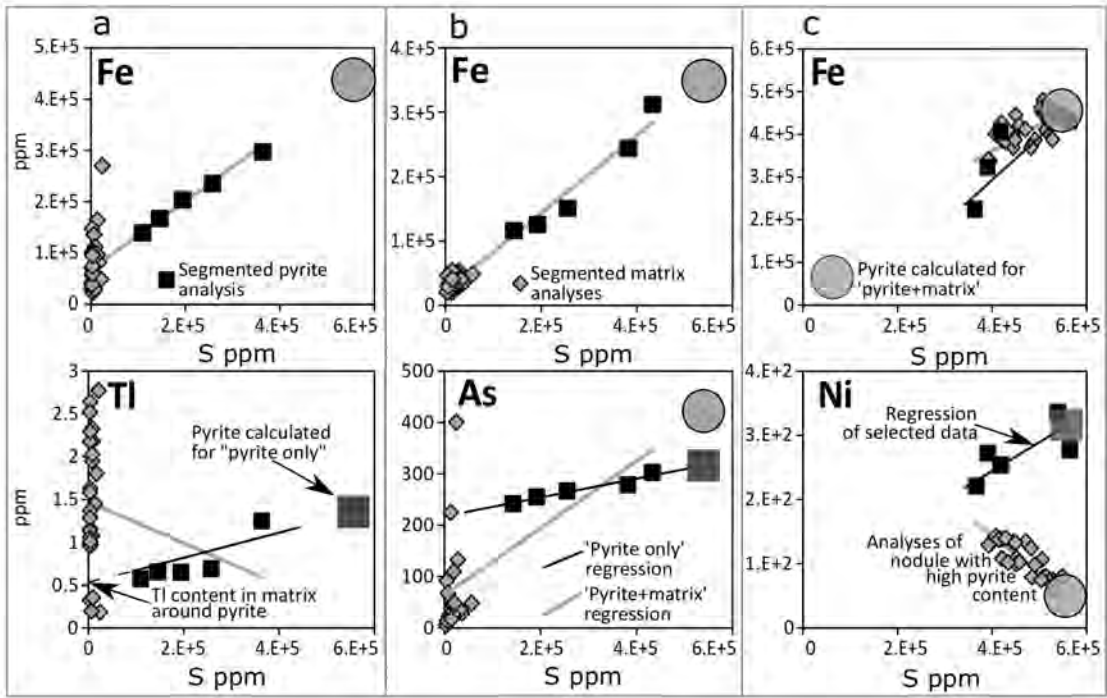


Figure 4

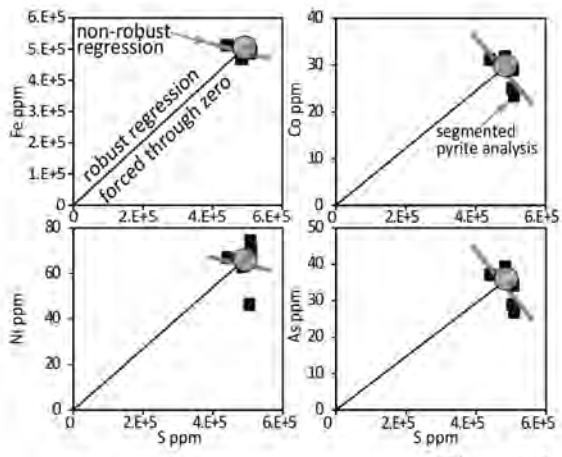


Figure 5

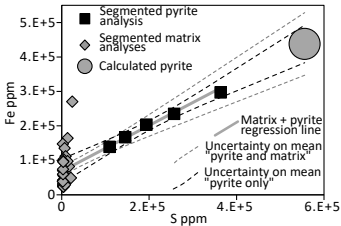


Figure 6

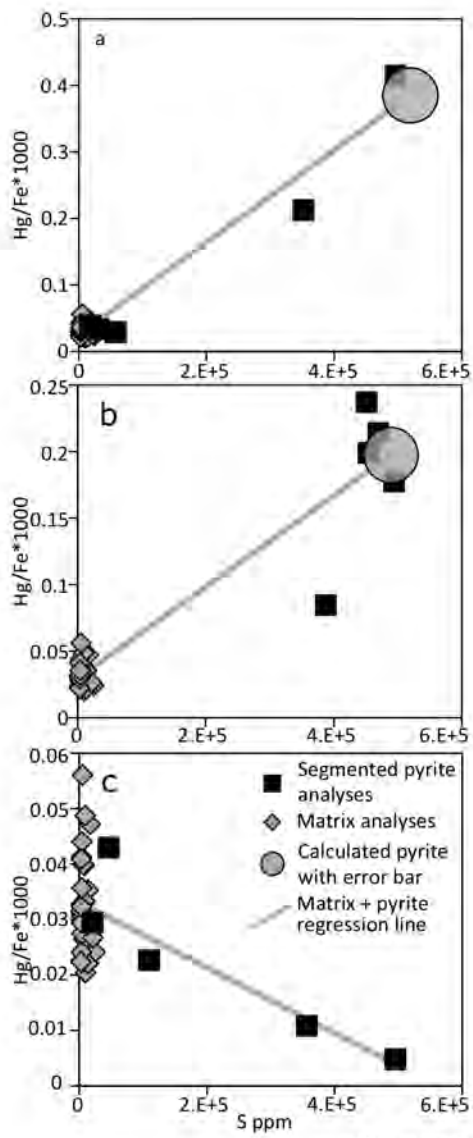
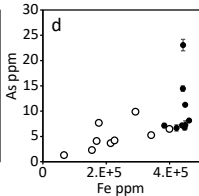
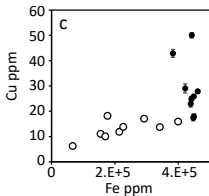
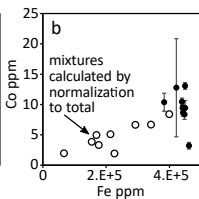
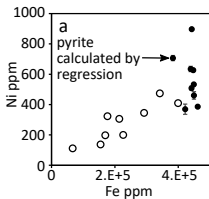


Figure 7



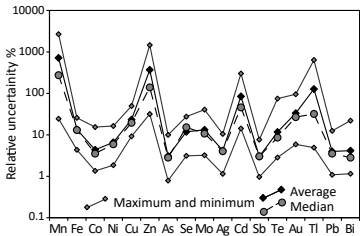


Figure 9

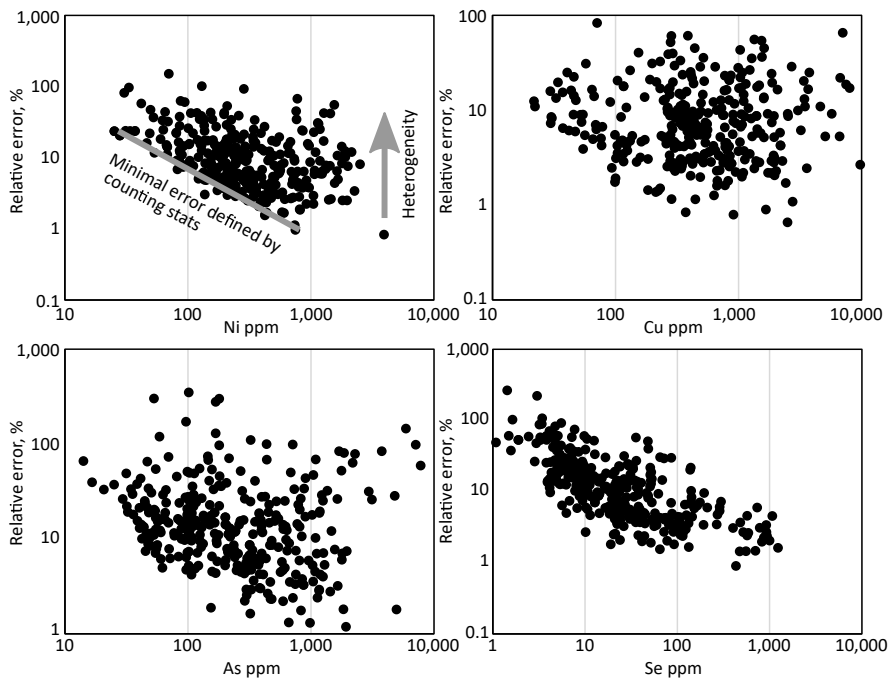


Figure 10

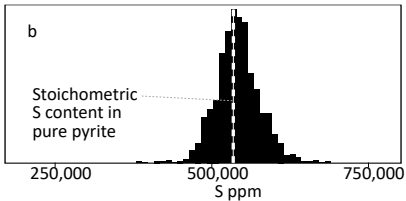
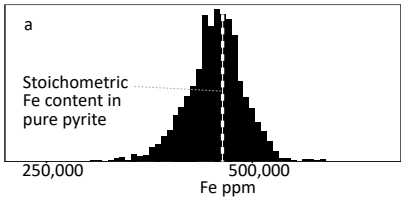


Figure 11

Tables

Table 1: Layout of analyzed elements, dwell times and standards used for quantification.

Elements	Dwell time	Isotope	Reference	Calculated
			materials used for calibration	in pyrite by regression
Na	0.005	23	GSD-1G	
Mg	0.005	24	GSD-1G	
Al	0.005	27	GSD-1G	
Si	0.005	29	GSD-1G	
S	0.005	34	PPP-1	+
K	0.005	39	GSD-1G	
Ca	0.005	43	GSD-1G	
Ti	0.005	49	GSD-1G	
V	0.005	51	GSD-1G	
Cr	0.005	53	GSD-1G	
Mn	0.005	55	GSD-1G	+
Fe	0.01	57	GSD-1G	+
Co	0.04	59	STDGL2b2	+
Ni	0.02	60	STDGL2b2	+
Cu	0.02	65	STDGL2b2	+
Zn	0.02	66	STDGL2b2	+
As	0.01	75	STDGL2b2	+
Se	0.04	77	STDGL2b2	+
Rb	0.02	85	GSD-1G	
Sr	0.02	88	GSD-1G	
Zr	0.01	90	GSD-1G	
Mo	0.02	95	STDGL2b2	+

Ag	0.02	107	STDGL2b2	+
Cd	0.02	111	STDGL2b2	+
Sn	0.02	118	STDGL2b2	
Sb	0.02	121	STDGL2b2	+
Te	0.05	125	STDGL2b2	+
Ba	0.01	137	GSD-1G	
Gd	0.005	157	GSD-1G	
Hf	0.01	178	GSD-1G	
Ta	0.01	181	GSD-1G	
W	0.02	182	STDGL2b2	
Pt	0.05	195	STDGL2b2	+
Au	0.05	197	STDGL2b2	+
Hg	0.02	202		
Tl	0.02	205	STDGL2b2	+
Pb	0.02	206	STDGL2b2	+
Pb	0.02	207	STDGL2b2	+
Pb	0.02	208	STDGL2b2	+
Bi	0.02	209	STDGL2b2	+
Th	0.01	232	GSD-1G	
U	0.01	238	GSD-1G	

Table 2: Summary of principal parameters used in different studies which applied regression analysis to the deconvolution of mixed LA-ICP-MS data.

	Mineral of interest	Matrix	Method	Internal standard or primary quantification	Explanatory (independent) variable	Explanatory variable value for regression	Data used to constrain regression
Rubatto and Hermann (2007)	Zircon	Glass and garnet	LA-ICP-MS	Ca	Zr	EPMA	Mineral plus matrix
Stepanov et al. (2012)	Monazite	Glass	LA-ICP-MS	Si	Ce	EPMA	Mineral plus matrix
Taylor et al. (2015)	Zircon	Glass	SIMS	Si	Zr	EPMA	Mineral plus matrix
Present study	Pyrite	Sediment	LA-ICP-MS	Fe for preliminary quantification; followed by normalisation to 100%	S	Calculated from set of regressions with 100% total requirement.	Two options: 'pyrite+matrix' and 'pyrite only'

Table 3: Sources of uncertainty of the data on the example of set of pyrite analyses.

Elements	Standard uncertainty %	Drift	Drift Error %	Average error %	Maximum error %	Minimum error %
				Regression errors		
Na	0.2	0.97	0.4			
Mg	0.04	0.98	0.6			
Al	0.3	0.97	0.7			
Si	0.8	0.94	1.2			
S			10.0			
K	0.1	0.98	0.6			
Ca	0.1	0.98	2.7			
Ti	0.06	1.01	1.1			
V	2	1.01	1.5			
Cr	3	1.06	3.8			
Mn	20	1.00	1.7	717	2672	24
Fe	0.1			13	26	4
Co	2	0.97	1.5	4	15	1
Ni	4	0.99	1.1	7	16	2
Cu	2	0.96	1.9	23	50	9
Zn	2	1.00	0.8	368	1465	32
As	8	1.05	1.4	3	10	1
Se	1	0.64	1.8	12	27	3
Rb	0.4	1.06	2.7			
Sr	0.7	1.08	0.5			
Zr	2	1.02	0.9			
Mo	3	1.10	4.4	13	40	3
Ag	3	1.02	4.2	4	11	1
Cd	4	1.05	1.9	84	301	14
Sn	6	1.07	1.9			
Sb	7	1.04	1.3	3	8	1

Te		0.89	0.3	12	75	3
Ba	1	1.06	2.3			
Gd	0.5	1.28	3.5			
Hf	2	1.14	1.4			
Ta	4	1.10	0.6			
W	4	1.17	2.1			
Pt		0.77	43.7			
Au		0.82	14.7	33	95	6
Hg						
Tl	0.1	0.70	11.7	127	636	5
Pb	2	1.11	1.3	4	12	1
Bi	4	1.06	1.5	4	22	1
Th	2	1.06	2.8			
U	2	1.09	2.4			
