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Revision 2

Stability of fcc phase FeH to 137 GPa

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ABSTRACT

17 We examined the stable crystal structure of FeH_X ($X \sim 1$) (FeH hereafter) at high
18 pressure and temperature by X-ray diffraction up to 137 GPa. Results show that FeH
19 adopts a face-centered cubic (fcc) structure at pressures of 43 to 137 GPa and
20 temperatures of approximately 1000 to 2000 K. Our study revises a phase diagram of
21 stoichiometric FeH in which fcc has a wider-than-expected stability field at high
22 pressure and temperature. Based on our findings, the FeH endmember of the Fe-FeH
23 system is expected to be stable in the fcc structure at the P - T conditions of the Earth's
24 core, rather than in the double-hexagonal close packed (dhcp) structure as previously
25 reported. We compared experimentally determined FeH lattice volumes with those from
26 *ab initio* calculations. Additionally, we observed a change in compressibility at ~ 60 GPa,
27 which could be attributed to a magnetic transition - an interpretation supported by our
28 *ab initio* computations.

29 **Keywords:** FeH, fcc structure, Earth's core, X-ray diffraction, *ab initio* calculation

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INTRODUCTION

33 Hydrogen is a possible constituent light element within the Earth's predominantly
34 iron-nickel core (e.g., Poirier, 1994). Therefore, evaluation of the Fe–FeH system under
35 the high pressure (P) and high temperature (T) conditions of the Earth's deep interior is
36 important in the discussion of chemical and physical properties of the core. An
37 endmember of this system, stoichiometric FeH with double-hexagonal close packed
38 (dhcp) structure, has been observed in iron loaded in a hydrogen pressure medium at
39 pressures exceeding 3.5 GPa at room temperature (e.g., Badding et al. 1991). The
40 temperature-induced phase transition from dhcp to face centered cubic (fcc) was
41 observed up to 20 GPa in a multi-anvil apparatus (Ikuta et al. 2019; Saitoh et al. 2017;
42 Sakamaki et al. 2009). Locations of the melting curve and the dhcp/fcc phase boundary
43 predict the triple point of dhcp, fcc, and liquid phases FeH at about 60 GPa and 2000 K,
44 and the disappearance of the fcc stability field at higher pressures. Room-temperature
45 compression experiments observed no structural change in dhcp FeH up to 136 GPa, but
46 they did not involve thermal annealing during compression, and so lack concrete
47 evidence for the stability of dhcp at such a pressure range (Badding et al. 1991; Hirao et

48 al. 2004; Pépin et al. 2014; Shibazaki et al. 2012). On the other hand, there are
49 experimental and theoretical reports that suggest fcc FeH is stable above 80 GPa and up
50 to 1800 K (Isaev et al. 2007; Thompson et al. 2018). Free-energy calculations suggested
51 that, at room temperature, FeH undergoes a structural transition from dhcp to hexagonal
52 close packed (hcp) at 37 GPa, and a second transition from hcp to fcc occurs at 83 GPa
53 (Isaev et al. 2007). They also found that a ferromagnetic-paramagnetic transition takes
54 place at ~60 GPa in metastable fcc FeH, which leads to the change in compressibility
55 and stabilizes the fcc structure (Isaev et al. 2007).

56 In addition to endmember FeH, non-stoichiometric FeH_X (X<1) has been identified
57 in both the fcc structure (Thompson et al. 2018) and the hcp structure (Antonov et al.
58 1998; Yuan et al. 2018; Machida et al. 2019). Additionally, Fe polyhydrides with FeH₂,
59 FeH₃, and FeH₅ composition have been reported (Pépin et al. 2014; 2017), and FeH₄
60 and FeH₆ have also been suggested as potential Fe-H compounds at high pressure
61 (Bazhanova et al. 2012; Kvashnin et al. 2018).

62 In this study, we performed synchrotron X-ray diffraction experiments *in-situ* at
63 high *P-T* in a laser-heated diamond anvil cell (LHDAC) to pursue the stability field of

64 stoichiometric fcc FeH. These experiments confirmed that fcc FeH is stable up to 137
65 GPa and about 1000 K. We also collected lattice volume data of fcc FeH with changing
66 pressure and found anomalous compression behavior at around 60 GPa. Based on our
67 *ab initio* calculations, the magnetic transition in fcc FeH reasonably explains the
68 observed change in compressibility.

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METHODS

71 We employed the LHDAC technique to generate high *P-T* conditions. Diamond
72 anvils with 300 or 120 μm diameter culets were used, and gaskets were pre-indented
73 rhenium plate. We used two different methods to supply hydrogen to the iron foil
74 starting material (99.99% purity). In runs #1–3, iron hydride was formed by loading an
75 iron foil with a paraffin ($\text{C}_n\text{H}_{2n+2}$, $n>5$) pressure medium into the LHDAC with ruby or
76 KCl pressure markers (Narygina et al. 2011; Ohta et al. 2019; Thompson et al. 2018;
77 Hirose et al., 2019). In runs #4–8, liquid H_2 was introduced into the LHDAC and
78 compressed with an iron foil and a NaCl inner gasket at ~ 20 K in a helium refrigerator
79 (Chi et al. 2011; Ohta et al. 2015; Tagawa et al. 2016). The NaCl inner gasket prevents

80 hydrogen embrittlement of the Re gasket and the escape of hydrogen from a sample
81 chamber. The surface of the diamond anvils was coated with a thin Ti film to prevent
82 the hydrogen from escaping into the anvils. Supplemental Fig. S1 shows the sample
83 chamber after the liquid H₂ injection.

84 *In-situ* angle-dispersive X-ray diffraction (XRD) measurements were performed at
85 BL10XU, SPring-8. XRD patterns were collected on a flat panel detector. The
86 wavelength of the monochromatic X-ray was determined using a CeO₂ standard.
87 Two-dimensional XRD images were integrated over the Debye-Scherrer rings to
88 convert one-dimensional diffraction patterns as a function of 2-theta angle, and the
89 obtained peak profiles were fit to pseudo-Voigt peak shapes. These procedures were
90 performed on the programs of IPAnalyzer and PDindexer (Seto et al. 2010). The lattice
91 parameters were obtained by a least-squares fit of XRD peak positions. Sample pressure
92 was determined from ruby fluorescence spectrum (Dorogokupets and Oganov 2007) for
93 run #1 and from the lattice volumes of KCl (Dewaele et al. 2012) and NaCl (Ueda et al.
94 2008) for runs #2–3 and #4–8, respectively. Dewaele et al. (2012) used the Ruby
95 pressure scale by Dorogokupets and Oganov (2007) to obtain their equation of state of

96 KCl. To correct the systematic pressure difference between the equations of state of KCl
97 and NaCl, simultaneous lattice volume measurements of KCl and NaCl were also
98 performed (Table S1). The simultaneous lattice volume measurements of NaCl and KCl
99 demonstrated that the difference in pressure calculated from these two standards was 0.3
100 GPa at 52 GPa and 13.4 GPa at 133 GPa. We fitted such a pressure difference to the
101 following equation; $P_{\text{KCl}} - P_{\text{NaCl}} = a \times \exp(P_{\text{KCl}} / b)$, and obtained $a = 0.058(17)$ GPa and
102 $b = 26.9(15)$ GPa. Heating was performed from both sides of the sample by employing
103 a pair of 100 W single-mode Yb fiber lasers. Temperatures were measured by
104 spectroradiometric method (Ohishi et al. 2008).

105 The hydrogen content of FeH_X was estimated based on the volume expansion by
106 hydrogen incorporation at each pressure:

$$107 \quad X = [V(\text{fcc FeH}_X) - V(\text{fcc Fe})] / V_H. \quad (1)$$

108 Here we adopted the equation of state (EOS) of fcc Fe at 298 K (Boehler 1990). Note
109 that spin transition in fcc Fe has not been observed in previous experiments (Tsujino et
110 al. 2013), but is expected to occur within the studied pressure range. Volume expansion
111 per H atom, V_H , is calculated for each pressure by the Vinet equation of state using $K_0 =$

112 99.2 GPa and $K'_0 = 3.98$ from Fukai (1992). For V_{H0} , we employed the value of 2.22 \AA^3
113 based on neutron diffraction measurements of FeH_x up to 12 GPa and 1200 K (Ikuta et
114 al. 2019), assuming that the pressure and temperature terms are compensated and the
115 volume is close to that at 0 GPa and 300 K.

116 Our *ab initio* calculations produced lattice volumes of ferromagnetic (FM),
117 antiferromagnetic (AFM), and nonmagnetic (NM) fcc FeH at high pressures. We used
118 the Perdew-Burke-Ernzerhof (PBE) form (Perdew et al. 1996) of generalized-gradient
119 approximation (GGA) for the exchange-correlation functional. Vanderbilt's type
120 pseudopotentials were used (Vanderbilt 1990). The fcc unit cell contained 1 FeH. The
121 cutoff radii for Fe and H were 0.956 and 0.265 \AA , respectively. The plane-wave basis
122 set with a cutoff energy of 40 Ry was used. The \mathbf{k} -point mesh was $12 \times 12 \times 12$. A
123 smearing technique (Methfessel and Paxton 1989) was used for integration up to the
124 Fermi surface with a smearing parameter of 0.01 Ry. We used the quasi-harmonic
125 approximation to take phonon effects into account (Wallace 1972). Dynamical matrices
126 were computed at the $4 \times 4 \times 4$ \mathbf{q} -point mesh using the density-functional-perturbation
127 theory (Baroni et al. 2001; Giannozzi et al. 1991). Vibrational densities of states were

128 obtained by interpolation of phonon frequencies on the $24 \times 24 \times 24$ \mathbf{q} -point mesh. We
129 evaluated FM, AFM, and NM states. For AFM FeH, we considered two spin
130 configurations: one designed to mimic AFM FeO with iron atoms on the (111) plane in
131 the same spin and opposite spins on adjacent planes (Fig. S2a), and a second
132 configuration with iron atoms on the (100) plane in the same spin state and opposite
133 spins on adjacent planes (Fig. S2b). As the AFM spin configuration that mimics FeO
134 has lower enthalpy in the pressure interval in which it was evaluated (-5 to 20 GPa), it is
135 assumed to be more stable and is the only AFM structure discussed further in this text.
136 All calculations were performed using the Quantum-ESPRESSO package (Giannozzi et
137 al. 2009).

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RESULTS AND DISCUSSION

140 We performed eight separate XRD experiments to examine the stable phase of FeH
141 to 137 GPa (Fig. 1 and Table S2). At each P - T condition, we evaluated the stable phase
142 of FeH_X from appearance/disappearance and growth/weakening of XRD peaks by
143 increasing temperature with the fiber lasers. The hydrogen content (X) of each of the

144 synthesized fcc FeH_x samples after laser heating was almost unity (Fig. 2a).

145 In runs #1–3, the iron sample and paraffin reacted to form iron carbide soon after
146 the beginning of laser heating at around 1600 K and 60 GPa. After ~1-hour heating, iron
147 carbide disappeared, and fcc FeH_x and diamond formed as previously reported
148 (Narygina et al. 2011; Ohta et al. 2019; Thompson et al. 2018).

149 In the other five runs, we loaded H₂ cryogenically into the sample chamber with
150 iron foil and then observed the chemical reaction between iron and H₂ at high pressures
151 and temperatures. The occurrence of hydrogenation of the iron sample during the
152 cryogenic H₂ loading depends on the *P-T* pathway. After H₂ was cryogenically loaded,
153 but prior to heating, XRD from pure iron and the Raman peak of the H₂ vibration mode
154 could be confirmed in 2 of the 5 samples prepared using this method, while in the other
155 runs, we observed hcp or dhcp FeH_x after recovery from the helium refrigerator (Table
156 S2). To facilitate the reaction, we annealed the sample at around 1000 K by laser
157 heating. In four runs, FeH₂ or FeH₃ were observed occasionally, but weakened during
158 continuous heating (Table S2).

159 In runs #4 and #5, the samples were determined to be at 126 and 106 GPa,

160 respectively, when measured at room temperature subsequent to cryogenic H₂ loading.
161 We observed XRD peaks from pure hcp Fe and Raman peak from H₂ vibration mode
162 before heating. During laser heating, hcp Fe diffraction peaks weakened, and a diffused
163 XRD signal appeared, and further heating ended up with the sudden emergence of XRD
164 peaks from fcc FeH_x (Figs. 1a and S3). In run #5, after formation of fcc FeH_x, the
165 sample was further compressed and pressure-volume data was obtained. At 137 GPa,
166 the sample was heated again at around 1000 K. Growth of the peaks of fcc FeH_x was
167 observed during heating, and we therefore interpreted this as meaning that fcc FeH_x is
168 in a stable phase at this pressure and temperature.

169 In runs #6 and #7, hcp and dhcp FeH_x was already confirmed before heating at 85
170 and 70 GPa, respectively. The XRD peaks from dhcp FeH_x were replaced by those from
171 fcc FeH_x during heating, indicating the dhcp to fcc transition (Fig. 1b). In run #6, after
172 formation of fcc FeH_x, the sample was compressed and heated at 99 and 105 GPa, and
173 then decompressed and heated at 72, 66, 55 and 25 GPa. As with run #5, the growth of
174 peaks indicates that fcc FeH_x is stable at each pressure except 25 GPa. When the sample
175 was heated at 25 GPa, the fcc peaks were drastically weakened and new peaks, possibly

176 from dhcp FeH_X ($X < 1$), appeared. Therefore, fcc FeH_X should no longer be stable at 25
177 GPa. The mean value of X for fcc FeH_X in runs #1~7 was 1.01(6) (Fig. 2a).

178 In run #8, we first observed dhcp FeH_X just after H_2 loading. The consequent
179 reaction by laser annealing at 53 GPa was dhcp to hcp transition (Fig. S4). Hydrogen
180 content in the hcp FeH_X sample was about 0.3, which indicates hcp phase is stable when
181 hydrogen content is less than ~ 0.3 as reported by the literature (Antonov et al. 1998;
182 Gomi et al. 2018; Machida et al. 2019; Yuan et al. 2018).

183 Based on the present and previous results, we confirmed that fcc FeH has a wider
184 stability field than previously thought (Fig. 3). This suggests that prior studies observed
185 metastable dhcp phase to 136 GPa because of an absence of thermal annealing (Badding
186 et al. 1991; Hirao et al. 2004; Pépin et al. 2014; Shibazaki et al. 2012). A large energetic
187 barrier for dhcp-fcc transition in FeH would exist. One can consider that there is no
188 triple point of dhcp, fcc and liquid FeH, and that the dhcp phase has a closed stability
189 region similar to the hcp phase of cobalt (Yoo et al. 2000) (Fig. 3). While previous
190 theoretical studies have predicted the stability of hcp FeH at 300 K between about 40
191 and 80 GPa (Isaev et al. 2007), hcp FeH_X with $X \sim 1$ was not stable in this study.

192 The volumes of synthesized nearly stoichiometric fcc FeH_X were collected with
193 changing pressure, which is plotted in Figure 2b and summarized in Table S3. Our
194 experimental *P-V* data exhibits a discontinuous change in compressibility at ~60 GPa.
195 Note that the change in *X* at similar pressure can be attributed to this change in
196 compressibility and is not truly compositional. We fitted the third-order
197 Birch-Murnaghan EOS to the *P-V* data separately for the 15-57 GPa and 72-137 GPa
198 pressure intervals (Table 1). The bulk modulus K_0 and its pressure derivative K_0' of the
199 high-*P* EOS were larger and smaller than those of the low-*P* EOS, respectively.

200 The observed anomaly in compression behavior without structural change is
201 possibly due to magnetic transition in fcc FeH. Thus, we calculated the volume of fcc
202 FeH at high *P* considering its magnetism. In the static *ab initio* calculations for
203 stoichiometric FeH, we obtained the FM state up to ~50 GPa, which is always stable
204 with respect to the NM state. The AFM state is energetically never stable with respect to
205 both the FM and NM states at all pressures investigated. Under compression, the
206 magnetic moment (*M*) calculated for the FM state decreased gradually from 2.2 μ_B /FeH
207 at 13.5 \AA^3 /FeH (i.e., 0 GPa) to 1.8 μ_B /FeH at ~11 \AA^3 /FeH corresponding to ~50 GPa

208 (Fig. 4). By further compression, however, M in the FM state abruptly vanished and the
209 system became NM with an abrupt decrease in volume, a result qualitatively consistent
210 with Isaev et al. (2007) and Elsässer et al. (1998). For comparison, we also calculated M
211 in the dhcp and hcp phases under compression as shown in Figure 4. At low pressures
212 (i.e., $V > 11 \text{ \AA}^3/\text{FeH}$), M in all three phases are similar to each other; that in fcc is
213 marginally higher than those in the other two phases. In the dhcp phase, the
214 magnetization abruptly vanished at $\sim 10.3 \text{ \AA}^3/\text{FeH}$, slightly smaller than that in fcc. In
215 the hcp phase, on the other hand, magnetization behavior is rather different: the FM
216 state survives down to $\sim 8.5 \text{ \AA}^3/\text{FeH}$ ($\sim 210 \text{ GPa}$) with a continuous decrease in M . These
217 magnetization behaviors are generally consistent with previous calculations (Isaev et al.
218 2007; Tsumuraya et al. 2012).

219 Calculated parameters of static and 300 K third-order Birch-Murnaghan EOS of the
220 fcc phase are summarized in Table 1. As usual, the calculated 300 K volume is larger
221 than the static one, which is mainly due to the effect of zero-point motion. The static
222 FM and NM EOS are approximately consistent with earlier calculations (Bazhanova et
223 al. 2012; Isaev et al. 2007). The values of V_0 and K_0 in the FM state are larger and

224 smaller than those of NM, respectively. The 300 K P - V curve of FM and NM states are
225 shown in Figure 2b. Comparison of the present experimental and computational P - V
226 curves suggest that the change in compressibility can be attributed to a change in the
227 elastic properties due to the FM-NM transition predicted by *ab initio* calculations.
228 Another possibility is the change of site occupancy of H for the octahedral and
229 tetrahedral interstitial sites in FeH_x. However, such an argument is beyond the scope of
230 the present study because the site occupancy cannot be determined from XRD.

231 From the present experimental and computational results, the FM-NM transition
232 pressure in fcc FeH is 50-60 GPa. On the other hand, Narygina et al. (2011) argued that
233 fcc FeH was in the NM or AFM state at 26-47 GPa based on Mössbauer spectroscopy
234 measurements. We point out that their Mössbauer spectra may not have captured the
235 signal of fcc FeH alone, but was affected by the surrounding hcp Fe – which was not
236 heated and hydrogenated – because the spot size of their Mossbauer spectrometer was
237 $\sim 10^3$ times as large as the laser-heated spot, and $\sim 10^4$ times as large as the size of the
238 X-ray beam used in their XRD experiments. Recently, Thompson et al. (2018)
239 performed synchrotron Mössbauer spectroscopy on fcc FeH_x and found that spin

240 transition is unlikely to occur at 64-82 GPa, suggesting fcc FeH_x is not ferromagnetic at
241 these pressures.

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IMPLICATIONS

244 In this study, we demonstrated that near-stoichiometric fcc FeH (rather than dhcp
245 structure) with a nonmagnetic state was stable above ~70 GPa to 137 GPa (Fig. 3).
246 Experimental and theoretical studies of FeH_x suggested that it is able to satisfy the
247 density and sound velocity of the Earth's core, and is hence a strong candidate for a
248 light element in the core (Hirose et al. 2019; Sakamaki et al. 2016; Tagawa et al. 2016;
249 Umemoto and Hirose 2015). A precise *P-T-X* phase diagram of Fe-H at core pressures is
250 key to constraining the temperature of the core and density contrast at the inner
251 core-outer core boundary, but it has not been determined well because of the
252 experimental difficulties even at very low pressure (e.g., Shibazaki et al. 2014). Fukai
253 (1992) speculated a *P-T-X* phase diagram of the Fe-H alloy at about 100 GPa on the
254 basis of the well-known phase diagrams of metal-hydrogen systems and thermodynamic
255 functions. Fukai (1992) assumed that there was a continuous, subsolidus solid solution

256 between hcp Fe and dhcp FeH at ~100 GPa. Our results rule out dhcp FeH as an
257 endmember phase in the diagram because of the weakening of its XRD peaks in our
258 laser heating experiments (Fig. 1b). Our finding suggests that the subsolidus phases in
259 the Fe-FeH system at about 100 GPa change with hydrogen content. At least three
260 subsolidus phases would exist in the Fe-FeH binary region at the core conditions, since
261 hcp FeH_X (0.13<X<0.32) and dhcp FeH_X (0.41<X<0.90) were observed up to 113 GPa
262 and 2060 K as consequences of decomposition of hydrous mineral, and Fe and water
263 reaction (Nishi et al. 2017; Yuan et al., 2018) in addition to the present study. Melting
264 experiments on FeH_X by means of LHDAC demonstrated that FeH_{1.02} melted at 108
265 GPa and 2260 K, and a more hydrogen-rich Fe melt (FeH_{2.33}) was made at 127 GPa and
266 *T* less than 2120 K (Hirose et al. 2019) (Fig. 3). Some Fe polyhydrides could be stable
267 as subsolidus phases in a Fe-H system at Mbar pressure range (Bazhanova et al. 2012;
268 Kvashnin et al. 2018; Pépin et al. 2014, 2017). The melting *T* of hydrogen at 100 GPa
269 was determined to be 800~1000 K (Caillabet et al. 2011; Zha et al. 2017), while it was
270 assumed to be 2500 K in Fukai (1992). Therefore, the *P-T-X* phase diagram of the Fe-H
271 system at the core pressures range should be more complicated than we have ever

272 thought. Subsolidus phases of FeH_x with various hydrogen contents and their melting
273 temperatures at the core conditions are required to determine this in the future.

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- 418

419 **FIGURE CAPTIONS**

420

421 **FIGURE 1.** Representative XRD patterns taken in runs #5 and #7. **(a)** Lower pattern, at
422 106 GPa before laser heating; middle, at 103 GPa after heating; upper, at 99 GPa after
423 further heating. **(b)** Lower pattern, at 70 GPa before heating; upper, at 68 GPa after
424 heating. Unknown peaks are marked as ?.

425

426 **FIGURE 2.** **(a)** Hydrogen content of FeH_X as a function of pressure calculated using Eq.
427 (1). Closed and open symbols represent fcc and hcp FeH_X , respectively. Circles show
428 FeH_X synthesized from iron and paraffin (runs #1–3) and squares show FeH_X
429 synthesized from iron and H_2 (runs #4–8). **(b)** Volume per iron atom as a function of
430 pressure for fcc FeH_X ($X \sim 1.0$) and Fe. Symbols show experimental data for fcc FeH_X .
431 Open circles, synthesized from iron and paraffin (runs #1–3, this study); open squares,
432 from iron and H_2 (runs #4–7, this study); gray circles, from iron and paraffin (Narygina
433 et al. 2011). Black lines correspond to the fit of the present experimental data with a
434 third-order Birch-Murnaghan EOS for <60 GPa and >70 GPa. Light and dark gray lines

435 are the results of *ab initio* calculations for FM and NM fcc FeH. Dotted lines indicate
436 extrapolation. Black dashed line represents the compression curve of iron (Boehler
437 1990).

438

439 **FIGURE 3.** Proposed pressure–temperature phase diagram of stoichiometric FeH_X
440 (X~1). Closed and open symbols indicate the stability of fcc and dhcp structures,
441 respectively, from this study (circles) and earlier experimental results (squares,
442 Sakamaki et al. (2009); diamond, Narygina et al. (2011); triangles, Thompson et al.
443 (2018)). We recalculated the hydrogen content X for the data of Thompson et al. (2018)
444 by using the method we employed, and plot them with X~1.0. Asterisks indicate liquids
445 (Sakamaki et al. 2009). Melting curve of FeH_X (X~1.0) is from Hirose et al. (2019).

446

447 **FIGURE 4.** Magnetic moments calculated for fcc, dhcp, and hcp phases of
448 stoichiometric FeH as a function of lattice volume. Dashed lines are guides for the eyes.

449

450

TABLE and FIGURES

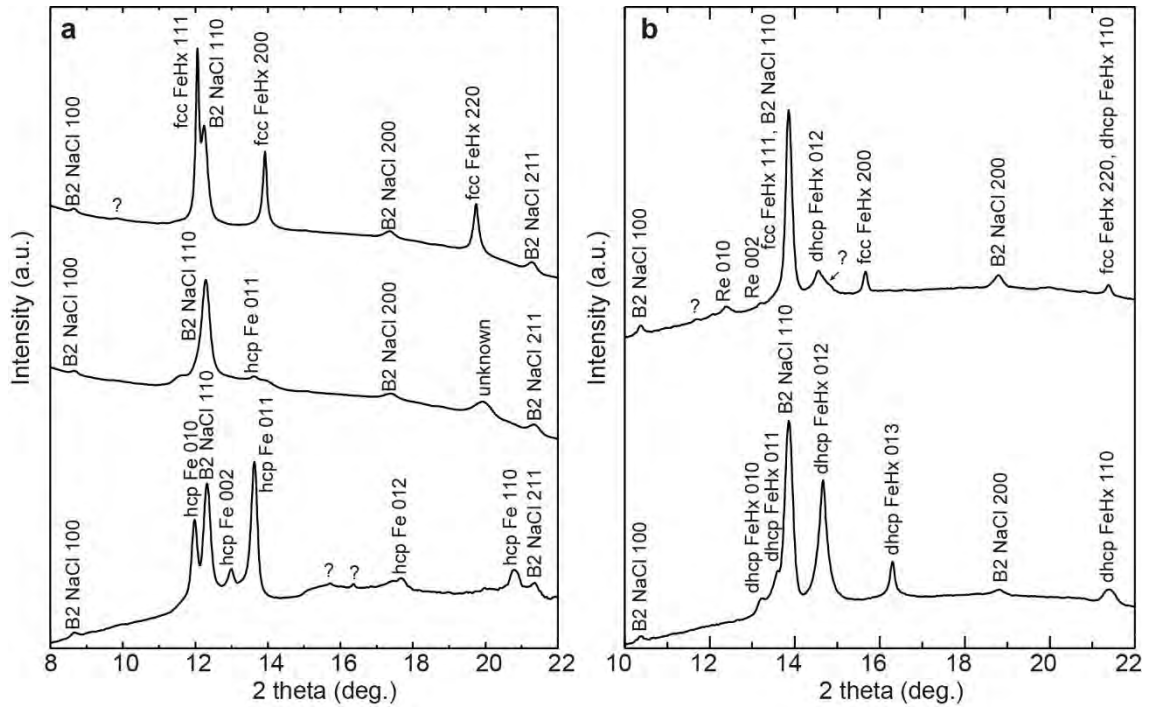
TABLE 1. Parameters for fcc FeH third-order Birch-Murnaghan equation of states from experimental and ab initio calculations compared with previous

	V_0 ($\text{\AA}^3/\text{Fe}$)	K_0 (GPa)	K_0'	Pressure range (GPa)	Temperature (K)	
experimental fcc FeH _{x-1}	13.8(7)	100(11)	9.3(16)	15-57	300	This study
	13.1(11)	227(8)	3.4(2)	72-137	300	This study
	13.5(1)	99(5)	11.7(5)	12-68	300	Narygina et al. (2011)
ab initio fcc FeH FM	13.5	168	4.5	<50 GPa	0	This study
	13.9	146	4.8	<50 GPa	300	This study
fcc FeH NM	13.9	155	3.7		0	Elsässer et al. (1998)
	12.3	262	4.2		0	This study
	12.6	242	4.3		300	This study
	12.2	271	4.3		0	Bazhanova et al. (2012)
	12.7	248	4.3		0	Elsässer et al. (1998)
Reference fcc Fe	11.3	165	5.5	0-16.8	300-1969	Boehler et al. (1990)

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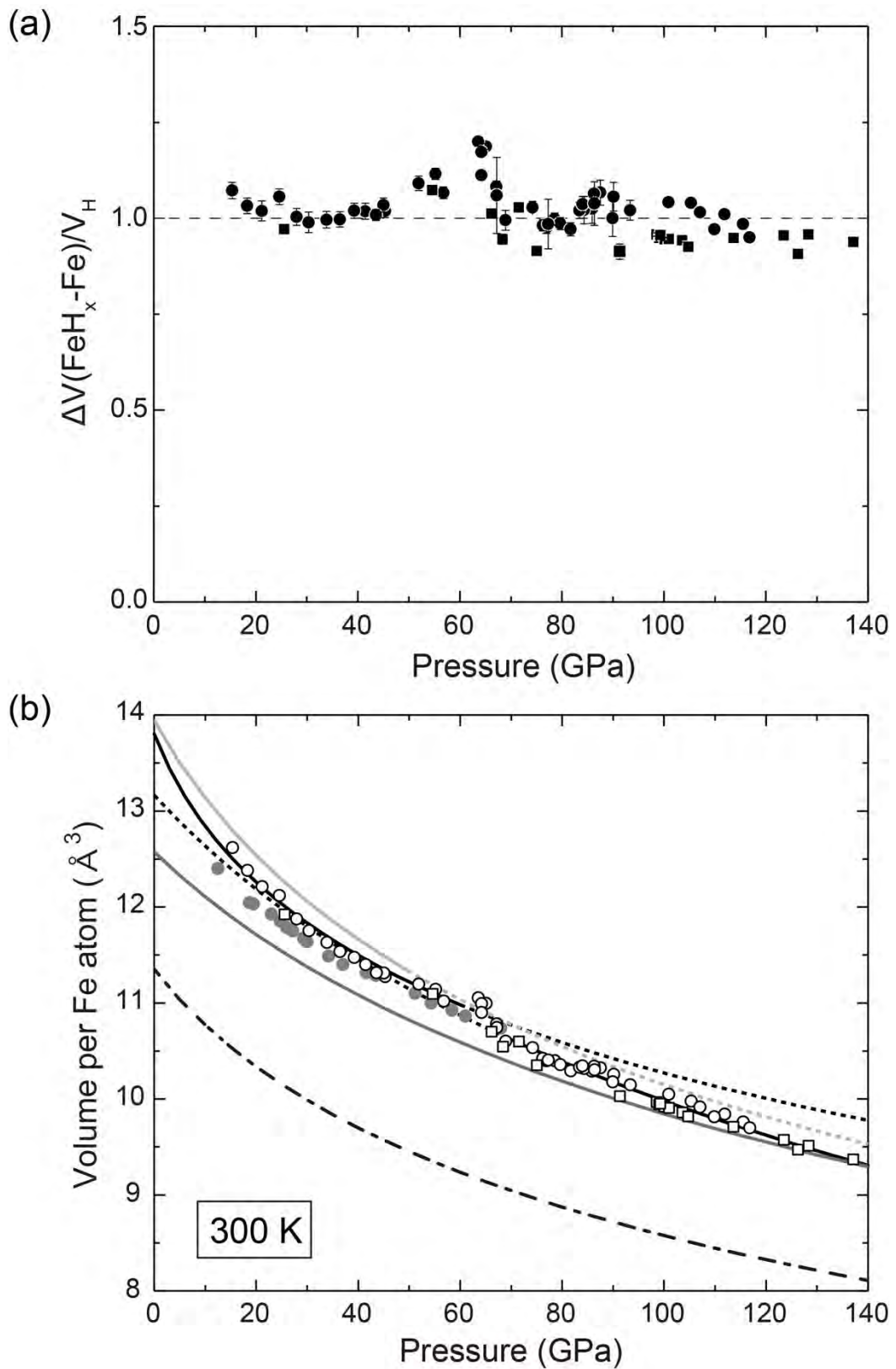


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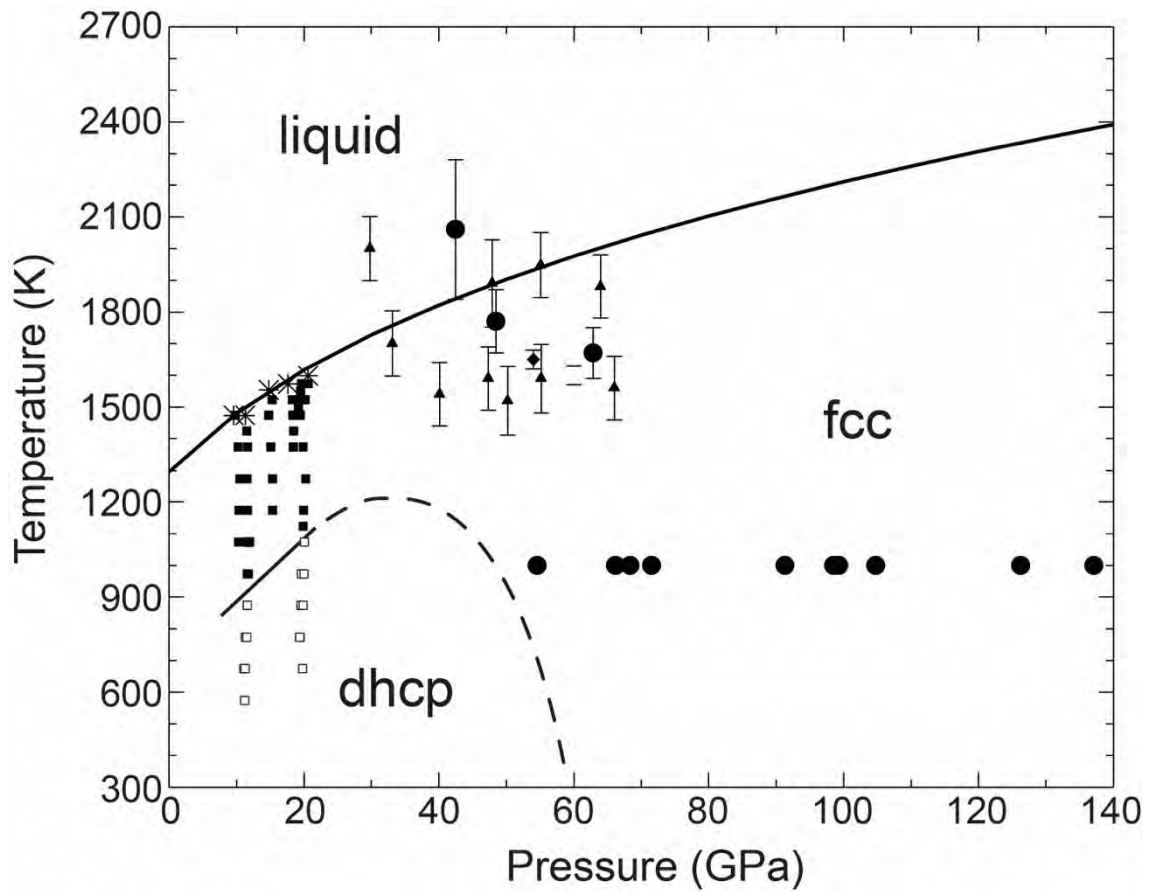
456 **FIGURE 1.**

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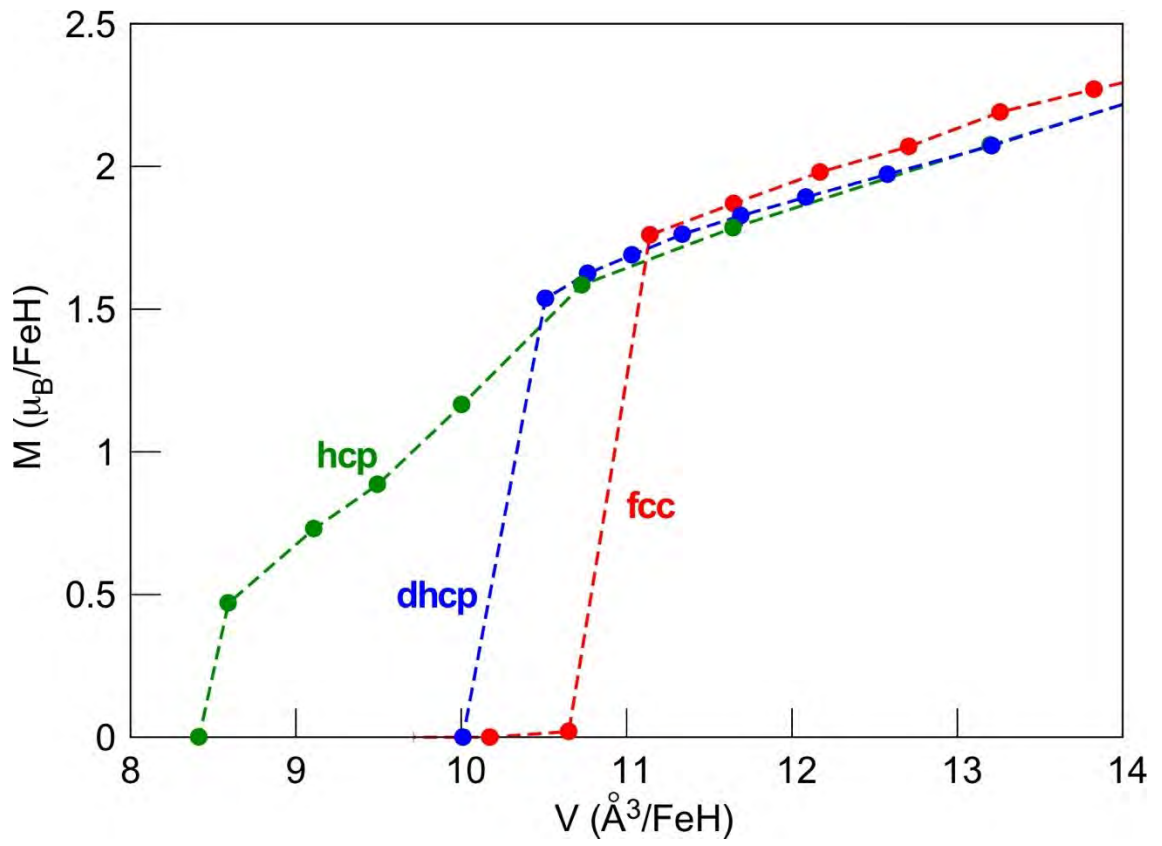
459 **FIGURE 2.**



460

461 **FIGURE 3.**

462



463

464 **FIGURE 4.**

TABLE 1. Parameters for fcc FeH third-order Birch-Murnaghan equation of states from exper

	V_0 (Å ³ /Fe)	K_0 (GPa)	K_0'	Pressure range (GPa)
experimental				
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	13.1(11)	227(8)	3.4(2)	72-137
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	12.2	271	4.3	
	12.7	248	4.3	
Reference				
fcc Fe	11.3	165	5.5	0-16.8

Experimental and ab initio calculations compared with previous studies.

Temperature (K)

300	This study
300	This study
300	Narygina et al. (2011)
0	This study
300	This study
0	Elsässer et al. (1998)
0	This study
300	This study
0	Bazhanova et al. (2012)
0	Elsässer et al. (1998)

300-1969 Boehler et al. (1990)