1	Revision 3
2	Chenmingite, FeCr <sub>2</sub> O <sub>4</sub> in the CaFe <sub>2</sub> O <sub>4</sub> -type structure, a shock-induced,
3	high-pressure mineral in the Tissint Martian meteorite
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14	ABSTRACT
15	Chenmingite (FeCr <sub>2</sub> O <sub>4</sub> ; IMA 2017-036), is a high-pressure mineral,
16	occurring as micrometer to submicrometer-sized lamellae within precursor
17	chromite grains along with xieite and Fe,Cr-rich ulvöspinel next to shock-induced
18	melt pockets, from the Tissint Martian meteorite. The composition of type
19	chenmingite by electron probe analysis shows an empirical formula of
20	$(Fe^{2+}_{0.75}Mg_{0.23}Mn_{0.02})(Cr_{1.60}Al_{0.29}Fe^{3+}_{0.06}Fe^{2+}_{0.04}Ti_{0.02})_{\sum 2.01}O_4$ . The general and
21	end-member formulas are (Fe,Mg)(Cr,Al) <sub>2</sub> O <sub>4</sub> and FeCr <sub>2</sub> O <sub>4</sub> . Synchrotron X-ray
22	diffraction reveals that chenmingite has an orthorhombic Pnma CaFe <sub>2</sub> O <sub>4</sub> -type
23	(CF) structure with unit cell dimensions: $a = 9.715(6)$ Å, $b = 2.87(1)$ Å, $c = 9.49$
24	(7) Å, $V = 264.6$ (4) Å, and $Z = 4$ . Both chenningite and xieite formed by solid
25	state transformation of precursor chromite under high pressure and high
26	temperature during the Tissint impact event on Mars. The xieite regions are
27	always in contact with melt pockets, whereas chenmingite lamellae only occur
28	within chromite, a few micrometers away from the melt pockets. This
29	arrangement suggests that chenmingite formed under similar pressures as xieite
30	but at lower temperatures, in agreement with experimental studies.
31	

Keywords: Chenmingite, FeCr<sub>2</sub>O<sub>4</sub>, high-pressure mineral, shock-induced phase, Tissint
 Martian meteorite, shergottite.

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# INTRODUCTION

37 The Tissint Martian meteorite is a fresh, highly-shocked, olivine-phyric shergottite 38 (e.g., Baziotis et al. 2013; Ma et al. 2015, 2016). Around shock-generated melt pockets and 39 veins in this meteorite, olivine is often transformed to ringwoodite or ahrensite and, in the 40 immediate vicinity of the melt veins or pockets, to bridgmanite plus wüstite (Ma et al. 2016). Plagioclase is transformed to maskelynite, locally melted and partially crystallized to 41 42 tissintite (Ma et al. 2015) and zagamiite-stishovite assemblages. These responses to shock 43 conditions on Mars provide clues to the nature of the ejection events for Martian meteorites 44 and to local shock conditions. Novel high-pressure minerals can narrow constraints on shock 45 metamorphism in meteorites and the scale of impact events on their parent bodies. As natural 46 phases, they contain minor chemical components which can guide experimental research 47 about deep mantle mineralogy and provide constraints in assessing chemical or textural 48 signatures that survive retrograde transformation in ultrahigh pressure terrains.

Here, we describe a new high-pressure mineral chenmingite,  $FeCr_2O_4$  with a *Pnma* orthorhombic CaFe<sub>2</sub>O<sub>4</sub>-type (CF) structure. It occurs in Tissint within chromite grains in contact with shock melt pockets (Fig. 1). Chen et al. (2003a) first reported FeCr<sub>2</sub>O<sub>4</sub>-CF in the Suizhou meteorite along with xieite but did not name this phase. It has also been synthesized at high pressures and moderate temperatures (e.g., Chen et al. 2003a; Ishii et al. 2014).

54 Preliminary results of this work were given by Ma et al. (2018).

55 The mineral chenningite (FeCr<sub>2</sub>O<sub>4</sub>; IMA 2017-036) has been approved by the 56 Commission on New Minerals, Nomenclature and Classification of the International 57 Mineralogical Association (Ma and Tschauner 2017). It is one of thirteen newly-approved 58 high-pressure minerals discovered in shocked meteorites since 2013 (Ma 2018). The name is 59 in honor of Ming Chen, a cosmochemist and mineralogist at the Guangzhou Institute of 60 Geochemistry, Chinese Academy of Sciences, for his outstanding contributions to research 61 on high-pressure mineralogy of meteorites, shock metamorphism and terrestrial impact 62 craters. Among his contributions is the discovery of natural FeCr<sub>2</sub>O<sub>4</sub> with a CF structure and  $x_{ieite}$  (the natural CaTi<sub>2</sub>O<sub>4</sub>-type polymorph of chromite) in the Suizhou meteorite, a highly shocked L6 chondrite (Chen et al. 2003a, b; 2008).

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# SAMPLE AND ANALYTICAL METHODS

67 The Tissint meteorite, which fell at Tata, Morocco on 18 July 2011, is a Martian 68 meteorite (olivine-phyric shergottite). The type material is in Tissint section UT2 deposited in 69 the Meteorite Collection of the Frank H. McClung Museum at the University of Tennessee, 70 Knoxville, Tennessee 37996, USA. Section UT2 also hosts type ahrensite (IMA 2013-028; 71 Ma et al. 2016) and type tissintite (IMA 2013-027; Ma et al. 2015). Tissint consists mostly of 72 olivine (microphenocrysts and magnesian macrocrysts with thin ferroan rims), pyroxene 73 (mostly pigeonite with some subcalcic augites), maskelynite (An58-69), with minor chromite 74 and ilmenite, and accessory pyrrhotite, merrillite and magnetite. The meteorite contains 75 scattered shock melt pockets and rare melt veins, and associated high pressure phases. 76 High-resolution scanning electron microscope (SEM), electron back-scatter 77 diffraction (EBSD), electron probe microanalysis (EPMA), and synchrotron X-ray diffraction 78 (SXRD) were used to characterize the composition, structure and petrography of chenmingite 79 and associated phases. Back-scatter electron (BSE) imaging was performed using a ZEISS 80 1550VP field emission SEM. EBSD analyses were attempted using an HKL EBSD system on 81 a ZEISS 1550VP SEM, operated at 20 kV and 6 nA in focused-beam mode with a 70° tilted 82 stage and variable pressure mode (25 Pa). However, no meaningful EBSD pattern was 83 obtained from chenmingite. Chemical analyses of chenmingite and associated phases were 84 carried out using a JEOL 8200 electron microprobe interfaced with the Probe for EPMA 85 program from Probe Software, Inc. and operated in focused beam mode at 15 kV and 15 nA with a probe diameter of ~150 nm. Standards were chromite (CrK $\alpha$ , FeK $\alpha$ ), spinel (AlK $\alpha$ , 86 87 MgK $\alpha$ ), TiO<sub>2</sub> (TiK $\alpha$ ), and Mn<sub>2</sub>SiO<sub>4</sub> (MnK $\alpha$ ). Quantitative elemental microanalyses were 88 processed with the CITZAF correction procedure (Armstrong 1995) and analytical results are 89 given in Table 1. 90 Synchrotron diffraction data were collected at the undulator beamline 13-IDD 91 (GSECARS, APS, Argonne National Laboratory) using a primary beam of wavelength

92 0.495936 Å, monochromatized by a double-crystal Si monochromator. The X-ray beam was

- 93 focused to  $2 \times 3 \,\mu\text{m}^2$  by vertical and horizontal Kirkpatrick-Baez mirrors of 200 mm focal
- 94 length. A MAR165 CCD area detector was used for collecting diffraction data, which were

95 obtained in forward scattering geometry. Calibration was conducted with GSE-ADA (Dera et 96 al. 2013) and DIOPTAS (Prescher and Prakapenka 2015). Integration and correction for 97 geometric distortion were conducted with DIOPTAS (Prescher and Prakapenka 2015). 98 Diffraction from chenningite is prominent in some locations but generally overlaps with 99 xieite, host chromite, and ulvöspinel. A pattern from the border of a melt pocket (Fig. 2) that 100 contains no signal from xieite and comparatively weak diffraction from chromite was used 101 for structure modeling. 102 103 **RESULTS** 104 Occurrence, appearance, physical and optical properties 105 In Tissint, type chenmingite occurs in the interior of two chromite grains exposed to 106 melt pockets (Fig. 1). Xieite (FeCr<sub>2</sub>O<sub>4</sub> in a CaTi<sub>2</sub>O<sub>4</sub>-type (CT) structure; e.g., Chen et al. 107 2008) and Fe,Cr-rich ulvöspinel occur in the same precursor chromite grains but these phases 108 are in direct contact with shock melt pockets (Fig. 1), whereas chenmingite lamellae only 109 occur within the precursor chromite, a few micrometers away from the melt pockets. The 110 shock melt pockets, surrounded by olivine (Fo42-73), pyroxene and 'maskelynite' (An64-111 66), contain abundant ringwoodite-ahrensite, bridgmanite and wüstite transformed from 112 olivine, and tissintite in maskelynite, and clinopyroxene from relict pyroxene or the melt. Chenmingite occurs as lamellae,  $< 1 \mu m$  in width and up to 4  $\mu m$  in length, set in 113 114 arrays within the host chromite (Fig. 1c). It is optically not distinguishable from chromite. Higher scratching hardness than chromite is evidenced by disruption of polishing streaks as 115 116 they cross chenmingite lamellae. Streak, tenacity, cleavage, fracture and reflection 117 pleochroism could not be determined empirically because of the small grain size. The 118 density, calculated from its crystal structure and the empirical formula, as described below, is  $5.27 \pm 0.02 \text{ g/cm}^3$ . 119 120 Chemical composition and crystallography 121 The composition (Table 1) of type chenmingite corresponds to an empirical formula based on 3 cations and 4 oxygens apfu of (Fe<sup>2+0.75</sup>Mg<sub>0.23</sub>Mn<sub>0.02</sub>) 122  $(Cr_{1.60}Al_{0.29}Fe^{3+}_{0.06}Fe^{2+}_{0.04}Ti_{0.02})_{\Sigma_{2.01}O_4}$ , where  $Fe^{2+}/Fe^{3+}$  was obtained by assuming an M<sub>3</sub>O<sub>4</sub> 123 stoichiometry. Within 1- $\sigma$  uncertainties, the composition of chenmingite is the same as that 124 125 of host chromite as well as that of xieite from the same host grain (Table 1). Chenmingite has 126 a general formula of (Fe,Mg)(Cr,Al)<sub>2</sub>O<sub>4</sub> and an end-member formula of FeCr<sub>2</sub>O<sub>4</sub>. The Fe,Crrich ulvöspinel, as indicated by EBSD and synchrotron diffraction, has an empirical formula of  $(Fe^{2+}_{0.86}Mg_{0.12}Mn_{0.02})(Cr_{0.57}Fe^{3+}_{0.31}Al_{0.23}Ti_{0.44}Fe^{2+}_{0.45})O_4$  with ~ 44 mole% ulvöspinel (Fe<sub>2</sub>TiO<sub>4</sub>), 29% FeCr<sub>2</sub>O<sub>4</sub>, 15% Fe<sub>3</sub>O<sub>4</sub>, and 12% MgAl<sub>2</sub>O<sub>4</sub>.

130 In diffraction patterns, the occurrence of chenmingite is marked by disintegration of 131 the chromite grain into a highly textured aggregate of two spinel phases (based on lattice 132 parameters, chromite and an Fe-, Cr-rich ulvöspinel) with a grain size of roughly 50 nm. 133 Chenmingite diffraction implies a similar grain size but it exhibits less broad profiles than 134 nearby chromite and ulvöspinel and marked orientation. For structure refinement, we chose a 135 pattern from the border of the chromite grain near a shock melt pocket (Fig. 2). In this pattern, chenmingite exhibits continuous Debye fringes with intensity variation along the 136 137 azimuthal angle, which we fitted with both March-Dollase and spherical harmonics. 138 Diffraction from surrounding chromite (around and underneath the lamella) could not be 139 completely avoided and this resulted in spotty broad Bragg reflections, which were initially 140 fitted by modeling a preferred orientation, followed by a Le Bail fit. |F(hkl)| of overlapping 141 reflections of chenmingite may have been compromised but this could not be avoided. For 142 chenmingite initially, the xieite structure was initially used as a model because we assumed 143 the diffraction patterns were generated by this phase. After Le Bail extraction of apparent structure factor moduli, a reversed Monte Carlo optimization in space group P1 was 144 145 performed. The optimized structure was compatible with space group Pnma but not Cmcm 146 (xieite) and exhibited a marked shift of one oxygen atom, consistent with a difference in octahedral linkages relative to the xieite structure. The resulting structure was of the 147 148 CaFe<sub>2</sub>O<sub>4</sub>-type rather than the CaTi<sub>2</sub>O<sub>4</sub>-typeof xieite (Chen et al. 2003b, 2008). We obtained 149 an R<sub>F</sub> of 28% for the initial xieite model and 11% for an optimized model in the CaFe<sub>2</sub>O<sub>4</sub> 150 harmunite-type structure. We also checked the modified CF-type structure (mCF) that had 151 been observed by Ishii et al. (2014) and found that R<sub>F</sub> was nearly as high as that for xieite. 152 We repeated the Le Bail extraction based on a cell in space group Pnma and conducted a 153 second reversed Monte Carlo optimization using a CaFe<sub>2</sub>O<sub>4</sub>-type model structure. This 154 second optimization resulted in an  $R_F$  of 9.1%. Le Bail-extraction converged to  $R_p = 4.3\%$ 155 (Kraus and Nolze 1996). The difference between R<sub>F</sub> and R<sub>p</sub> reflects the noise level of the data 156 and overlap of chenmingite reflections with coexisting chromite. Rietveld refinement (von 157 Dreele and Larson 2004; Kraus and Nolze 1996) was then conducted based on the structure 158 model from the rMC optimization, and by either initially refining the preferred orientation

with the March-Dollase approach along 302 and 020 or by modeling texture with spherical harmonics to  $2^{nd}$  order. Both approaches converged to the same orientation along the same main directions. Subsequently, atomic positions and isotropic thermal displacement factors were refined. The final wR<sub>p</sub> of the Rietveld-refinement was 5.7% with a  $\chi^2 = 43.9$  for 2062 observations. Pseudovoigt peak profiles were used with Gaussian terms U = 858, V = 231, W = 17.1 and Lorentzian terms L<sub>x</sub> = 18.9 and L<sub>y</sub> = 16.2 (with GSAS).

We used the extracted structure factor moduli to obtain the average electron density at 165 166 each of the three cation sites. Site occupancies were assigned based on EPMA results and the 167 assessed relative average electron density of the cation sites. The relative electron density of 168 the three sites were assessed by rMC, using the Le Bail extracted structure factor moduli (the 169 multiplicity is taken into account by the rMC method and splitting of |F(hkl)| between strictly 170 overlapping hkl does not need to be considered here). rMC cannot directly assess partial 171 occupancies but electron density on distinct sites. Hence, we modeled a site with, for 172 example, ~ 20 electrons as 'Ca' in the rMC optimization (although there is no significant 173 concentration of Ca in the specimen). In reality, 20 electrons correspond to some occupancy 174 by the actually present species (e.g.,  $0.82 \cdot \text{Cr} + 0.18 \cdot \text{Al} \text{ or } 0.69 \cdot \text{Fe} + 0.16 \cdot \text{Al} + 0.15 \cdot \text{Mg}$ , etc). 175 Using bulk composition as a constraint, we obtained the range of possible occupancies given 176 in Table 2. We repeated the procedure using different background- and orientation-fits prior 177 to Le Bail extraction. Resulting differences are within the given uncertainties. No potentials 178 were used in the rMC optimizations. We vary electron density at different sites by integers 179 changing the apparent cation (Ca = 20, Sc = 21 etc.). We interpolated if the best  $R_F$  was 180 obtained for an electron density intermediate between two integers. The sequence of electron density for the three sites was reproducibly assessed as Site  $3 > \text{Site} 2 \ge \text{Site} 1$ . Form factors in 181 rMC are spherical. This is a sufficient approximation for the given data (with best  $R_F = 9\%$ . 182 Differences between spherical and higher order Bessel-function-based form factors are 183 184 generally not significant for  $R_F > 2\%$ ). Isotropic thermal displacement factors of cation sites 185 were refined individually, and those of O were arbitrarily set equal to each other. Synchrotron-diffraction data, which were obtained for regions free of xieite, reveal 186

that chenningite has an orthorhombic *Pnma* CF-type structure with unit cell: a = 9.715 (6) Å, b = 2.87 (1) Å, c = 9.49 (7) Å, V = 264.6 (4) Å, and Z = 4. Atom coordinates are given in Table 2 and X-ray powder diffraction data are listed in Tables S1-S2. Chenningite is isotypic

190	with harmunite (CaFe <sub>2</sub> O <sub>4</sub> ). 50-60 vol% of the selected sample region were from chenmingite
191	and 40 - 50% from chromite and ulvöspinel (Fig. 2).
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193	DISCUSSION
194	Phase relations for FeCr <sub>2</sub> O <sub>4</sub> (Ishii et al. 2014) show that chromite, the stable ambient
195	pressure phase in the spinel structure, breaks down to form ludwigite-type Fe <sub>2</sub> Cr <sub>2</sub> O <sub>5</sub> plus
196	eskolaite (Cr <sub>2</sub> O <sub>3</sub> ) with increasing pressure (12-16 GPa, decreasing with increasing
197	temperature; see Fig. 3). At higher pressures (16-18 GPa), this phase assemblage yields to
198	single phase fields of chenmingite (CF-structured FeCr <sub>2</sub> O <sub>4</sub> ) at lower temperature and xieite
199	(CT-structured $FeCr_2O_4$ ) at higher temperature. CF-type endmember $FeCr_2O_4$ undergoes a
200	transition upon release to ambient conditions (mCF; Ishii et al. 2014) whereas xieite can be
201	recovered to ambient conditions. Type chenmingite contains Mg and Al as minor components
202	and the formula may be broken down as approximately 75 mole% chenmingite, 15%
203	MgAl <sub>2</sub> O <sub>4</sub> , and 7% MgCr <sub>2</sub> O <sub>4</sub> . In the MgCr <sub>2</sub> O <sub>4</sub> system, the CT structure is stable relative to CF
204	to low temperatures (< 1100°C; Ishii et al. 2015). For MgAl <sub>2</sub> O <sub>4</sub> , the CF-structure is stable
205	only above 23 GPa and 1600 °C (Kojitani et al. 2007) up to at least 2000°C whereas the CT-
206	structured phase appears only above $\sim 40$ GPa (Ono et al. 2006; Ishii et al. 2015).
207	Disregarding real mixing volumes, an increasing FeCr <sub>2</sub> O <sub>4</sub> -component is expected to reduce
208	the pressure of the spinel-to-postspinel transformation, whereas the $MgAl_2O_4$ component
209	appears to stabilize the CF-over the modified CFstructue in the given (Fe,Mg)(Cr,Al) <sub>2</sub> O <sub>4</sub>
210	solid solution.
211	Comparing all examined end-member phase diagrams, the CF-structure (i.e.,
212	chenmingite) is the low-temperature, high-pressure form and the CT-structure (xieite) is the
213	high-temperature, high-pressure form. This experimental finding is consistent with the
214	observation of chenmingite as dominant over xieite in shocked chromite further away from
215	melt pockets.
216	The ratio of the sum of the lengths of the two long axes over that of the short axis
217	(a+c)/b (in standard cell setting) of 6.67 of type chenmingite is within the range of other CF-
218	type phases reported in the literature: 6.64 to 6.70 (Irifune et al.1991; Lazic et al. 2006;
219	Arevalo-Lopez et al. 2010; Ishii et al. 2018) including mCF-type FeCr <sub>2</sub> O <sub>4</sub> (6.66; Ishii et al.
220	2014). Thus, the relative contraction of the c-axis of chenmingite is compensated by
221	expansion of the a- and b-axis. An apparent relation of the ratio of the effective cation radii

with this axial ratio is well matched by chenmingite. On the other hand, the apparent bond valences are low, with values between 4 and 5 rather than 6. This observation suggests a high degree of disorder of the anion lattice (because the relation between cation radii and cellparameters are well within the expected range). Such disorder is not unexpected in a structure of limited metastability.

227 The EPMA analyses show no significant difference in chemistry between host 228 chromite and either xieite or chenmingite, which suggests that both chenmingite and 229 xieite formed by solid-state transformation from precursor chromite under high pressures and temperatures during the Tissint impact event on Mars. These features are consistent 230 231 with expectations based on phase relations of the FeCr<sub>2</sub>O<sub>4</sub> system (Fig. 3). Upon release 232 breakdown to eskolaite and Fe-Cr-ludwigite is not observed, consistent with rapid 233 temperature-release at still high dynamic stresses, which also allow for the conservation 234 of bridgmanite (Tschauner et al. 2014). Bridgmanite has been observed in shock melt 235 pockets of Tissint (Ma et al. 2016). It is important to note, in this context, that xieite 236 regions (aggregates of sub-micrometer-sized crystals) are always in contact with melt 237 pockets, whereas chenmingite lamellae only occur within the precursor chromite, a few 238 micrometers away from the melt pockets. This spatial correlation suggests that 239 chenningite formed in the same pressure regime as xieite but at lower temperatures. A 240 similar textural relationship between xieite and chenmingite has been reported for the 241 Suizhou S6-L6-chondrite (Chen et al. 2003a).

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## **IMPLICATIONS**

244 In meteorites and, possibly, in terrestrial impactites, chenmingite and xietite document 245 the effect of high-pressure, high-temperature shock conditions on precursor chromite. Based 246 on the FeCr<sub>2</sub>O<sub>4</sub> system, the formation of chenmingite from chromite requires pressures 247 exceeding 16-18 GPa and temperatures below ~1350°C. The formation of xieite requires 248 comparable pressures but higher temperatures. Peak shock pressures in Tissint have been estimated to ~ 30 GPa (Baziotis et al. 2013; Walton et al. 2014; Ma et al. 2016). Generally, 249 250 kinetic phase boundaries are shifted relative to the thermodynamic ones to higher pressure at 251 lower temperature as evidenced by the zoned paragenesis strained olivine  $\rightarrow$  ringwoodite or 252 ahrensite  $\rightarrow$  wüstite + bridgmanite around Tissint shock-melt pockets (Ma et al. 2016). 253 Equivalently, chenmigite represents a cooler regime within a sequence of kinetically shifted

254 phase boundaries along a temperature gradient of a hotspot: Chromite with deformation twinning  $\rightarrow$  chenmingite  $\rightarrow$  xieite. Hence, chenmingite and xieite are potential markers for 255 256 S4 to S6 level shock metamorphism depending on the approximate temperature regime of 257 formation. That is within or at some distance from a shock melt pocket or vein. 258 Chromite is commonly observed as an exsolution phase from former high-pressure 259 silicates in diamonds (e.g., Brenker et al. 2002). In the Earth's transition zone, Cr and Ti are easily dissolved in majoritic garnet (Kessel et al. 2005) and may not be major constituents of 260 261 accessory minerals such as in spinel peridotite. On the other hand, an ilmenite-inclusion at a 262 remnant pressure of 10-12 GPa has been found in diamond (Tschauner et al. 2018). Hence, 263 the occurrence of chenmingite and xieite in adamantiferous metasomatized mantle appears 264 possible. Within such an environment, chenmingite or xieite could operate as carriers of high 265 field strength elements. 266 267 **ACKNOWLEDGEMENTS** 268 SEM, EBSD and EPMA analyses were carried out at the Caltech GPS Division 269 Analytical Facility, which is supported, in part, by NSF Grants EAR-0318518 and DMR-270 0080065. OT acknowledges support by NSF EAR-1838330. GSECARS is supported through 271 DOE Award DESC0005278, and NSF awards EAR-1128799,-0318518, DE-FG02-272 94ER14466, and DMR-0080065. The Advanced Photon Source, a DOE Office of Science 273 User Facility is operated by Argonne National Laboratory under Contract No. DE-AC02-274 06CH11357. We thank Ross Angel and four anonymous reviewers for their constructive 275 reviews. 276 277 REFERENCES 278 Arevalo-Lopez, A.M., Dos santos-Garcia, A.J., Castillo-Martinez, E., Duran, A., and Alario-279 Franco, M.A. (2010) Spinel to CaFe<sub>2</sub>O<sub>4</sub> transformation: Mechanism and properties of β-280 CdCr<sub>2</sub>O<sub>4</sub>. Inorganic Chemistry, 49, 2827–2833. 281 Armstrong, J.T. (1995) CITZAF: a package of correction programs for the quantitative 282 electron microbeam X-ray analysis of thick polished materials, thin films, and particles. Microbeam Analysis, 4, 177–200. 283

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315	phases in MgV <sub>2</sub> O <sub>4</sub> , FeV <sub>2</sub> O <sub>4</sub> , and MnCr <sub>2</sub> O <sub>4</sub> : Crystal chemistry of AB <sub>2</sub> O <sub>4</sub> postspinel
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Constituent	chenmingite	xieite	chromite	Fe,Cr-rich ulvöspinel
wt%	n=15 <sup>a</sup>	n=7	n=12	n=11
Cr <sub>2</sub> O <sub>3</sub>	57.5 (0.5) <sup>b</sup>	55.9 (0.6)	57.7 (0.4)	20 (2)
Al <sub>2</sub> O <sub>3</sub>	7.1 (0.1)	6.79 (0.05)	7.2 (0.2)	5.3 (0.4)
TiO <sub>2</sub>	0.70 (0.05)	0.8 (0.1)	0.65 (0.03)	15.9 (0.6)
FeO	29 (1)	30.1 (0.7)	27.4 (0.9)	53 (2)
MgO	4 (1)	3.2 (0.2)	5.2 (0.6)	2.2 (0.2)
MnO	0.62 (0.03)	0.62 (0.01)	0.63 (0.03)	0.76 (0.03)
Total	98.9	97.4	98.8	97.2
No. O atoms	4	4	4	4
Cr	1.60	1.59	1.60	0.57
Al	0.29	0.29	0.30	0.23
Ti <sup>4+</sup>	0.02	0.02	0.02	0.44
Fe <sup>3+ c</sup>	0.06	0.08	0.07	0.31
Fe <sup>2+ c</sup>	0.79	0.83	0.73	1.31
Mg	0.23	0.17	0.27	0.12
Mn	0.02	0.02	0.02	0.02
Sum cations	3.01	3.00	3.01	3.00

#### 362 **Table 1**. EPMA data for type chenmingite and associated phases. 363

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<sup>a</sup>n = number of analyses. <sup>b</sup> Errors given inside parentheses are one standard deviation of the mean based on all of the 365 analyses. <sup>c</sup> Fe<sup>3+</sup> and Fe<sup>2+</sup> were calculated from total Fe to produce the ideal  $M_3O_4$  stoichiometry. 366

367

369	<b>Table 2</b> . Fractional atom coordinates, site fractional occupancies, and isotropic thermal
370	displacement factors of chenmingite. Site occupancies of O are fixed. All atoms reside on
371	Wyckoff sites 4c. Isotropic displacement parameters are in $Å^2$ .

2	72	
5	12	

Atom	X	у	Z	SOF	Uiso
Cr1	0.754(4)	0.25	0.347(2)	0.80(4)	0.040(8)
Al+Mg	0.754(4)	0.25	0.347(2)	0.20(4)	0.040(8)
Cr2	0.933(2)	0.25	0.102(9)	0.67(18)	0.011(8)
Al+Mg	0.933(2)	0.25	0.102(9)	0.29(7)	0.011(8)
Fe1	0.92(4)	0.25	0.61(2)	0.87(10)	0.048(1)
Cr3	0.92(4)	0.25	0.61(2)	0.13(10)	0.048(1)
01	0.21(1)	0.25	0.84(2)	1	0.03(2)
O2	0.115(4)	0.25	0.52(1)	1	0.03(2)
03	0.52(1)	0.25	0.21(7)	1	0.03(2)
O4	0.42(1)	0.25	0.58(1)	1	0.03(2)





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- **Figure 1**. (a) Backscatter electron (BSE) image showing the chenningite-bearing chromite grains next to shock melt pockets in Tissint. (b)(c) Enlarged BSE images of areas in panel a revealing chenningite in two chromite grains with xieite and Fe,Cr-rich ulvöspinel. (d)
- Enlarged BSE image of area in panel c showing chenmingite lamellae in chromite.
- 385









**Figure 3**. Phase diagram of  $FeCr_2O_4$  based on Ishii et al. (2014). Sp: spinel-FeCr<sub>2</sub>O<sub>4</sub>

398 (chromite); mLd: modified ludwigite type  $Fe_2Cr_2O_5$ ; Es: eskolaite; CF: CF-structured 399 FeCr<sub>2</sub>O<sub>4</sub> (chenmingite); CT: CT-structured FeCr<sub>2</sub>O<sub>4</sub> (xieite).