| 26 | Layers stacking disorder in Mg-Fe chlorites based on powder X-ray diffraction data  |
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| 33 | Abstract  |
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| 35 | The primary aim of this study was the accurate determination of unit cell parameters and  |
| 36 | description of disorder in chlorites with semi-random stacking using common X-ray diffraction   |
| 37 | (XRD) data for bulk powder samples.   |
| 38 | In the case of ordered chlorite structures, comprehensive crystallographic information can  |
| 39 | be obtained based on powder XRD data. Problems arise for samples with semi-random stacking,   |
| 40 | where due to strong broadening of <i>hkl</i> peaks with $k \neq 3n$ , the determination of unit cell parameters                           |
| 41 | is demanding. In this study a complete set of information about the stacking sequences in   |
| 42 | chlorite structures was determined based on XRD pattern simulation, which included  |
| 43 | determining a fraction of layers shifted by $\pm 1/3\mathbf{b}$ , interstratification with different polytypes and                        |
| 44 | 2:1 layer rotations.  |
| 45 | A carefully selected series of pure Mg-Fe tri-trioctahedral chlorites with iron content in  |
| 46 | the range from 0.1 to 3.9 atoms per half formulae unit cell was used in the study. In addition,   |

46 the range from 0.1 to 3.9 atoms per half formulae unit cen was used in the study. In addition, 47 powder XRD patterns were carefully investigated in terms of the broadening of the odd-number 48 basal reflections to determine interstratification of 14Å and 7Å layers, which interstratification 49 finally was not found. This result was also confirmed by the XRD pattern simulations, assuming 50 interstratification with R0 ordering.

Based on *h0l* XRD reflections, all the studied chlorites were found to be the IIbb polytype with a monoclinic-shaped unit cell ( $\beta \approx 97^\circ$ ). For three samples, the *hkl* reflections with  $k \neq 3n$  were partially resolvable; therefore, a conventional indexing procedure was applied. Two of the chlorites were found to have a monoclinic cell (with  $\alpha$ ,  $\gamma = 90^\circ$ ). Nevertheless, among all the samples, the more general triclinic (pseudomonoclinic) crystal system with symmetry C-1 was assumed, to calculate unit cell parameters using La Bail fitting.

A detailed study of semi-random stacking sequences shows that simple consideration of 57 proportion of IIb-2 and IIb-4/6 polytypes, assuming equal content of IIb-4 and IIb-6 is not 58 59 sufficient to fully model the stacking structure in chlorites. Several, more general, possible models were therefore considered. In the first approach, a parameter describing a shift into one 60 of the  $\pm 1/3b$  directions (thus, the proportion of IIb-4 and IIb-6 polytypes) was refined. In the 61 second approach, for samples with slightly distinguishable *hkl* reflections with  $k \neq 3n$ , some kind 62 of segregation of individual polytypes (IIb-2/4/6) was considered. In the third approach, a model 63 with rotations of 2:1 layers about 0°, 120°, 240° were shown to have the lowest number of 64 parameters to be optimized and therefore, giving the most reliable fits. In all of the studied 65 samples, interstratification of different polytypes was revealed with the fraction of polytypes 66 being different than IIbb ranging from 5% to even 19%, as confirmed by fitting of hol XRD 67 reflections. 68

Keyworlds: chlorites, polytype interstratification, stacking disorder, powder X-ray
 diffraction

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# Introduction

Interest in the chlorite structure has been growing since it was first described by Pauling (1930) who showed that chlorites are built of two main units: the 2:1 and brucite-type layers, arranged in an alternating manner. The negative charge generated in the 2:1 layer –  $(R^{2+}, R^{3+})_3(Si_{4-x}Al_x)O_{10}(OH)_2$  – is compensated by the octahedral sheet –  $(R^{2+}, R^{3+}, \Box)_3(OH)_6$  –

present in the interlayer; where  $R^{2+} - Mg$ , Fe and  $R^{3+} - Al$ , Fe, rarely Cr, Mn;  $\Box$  – vacancy. In the 77 2:1 layer, one octahedral sheet is sandwiched between two tetrahedral sheets, thus for one 78 crystallographic unit cell, six octahedral and eight tetrahedral sites are available. In the 79 interlayer, a maximum of six octahedral sites can be occupied. Depending on the occupancy of 80 available octahedral sites, four sub-groups of chlorites can be distinguished (tri-trioctahedral; di-81 trioctahedral; tri-dioctahedral; di-dioctahedral). Tri-trioctahedral chlorites, where virtually all six 82 octahedral sites (in both,the2:1 layer and interlayer sheets) are occupied by divalent cations, are 83 the most abundant class. 84

Independently of octahedral occupancy, Brown and Bailey (1962) proposed a chlorite polytype classification based on a mutual arrangement of the interlayer sheet and the 2:1 layer and involving a shift by  $\mathbf{a/3}$  (Ia, IIa, Ib, IIb). These four structural units (2:1 layer + interlayer sheet) can be shifted with respect to each other by  $\pm 1/3\mathbf{b}$ . The shift results in aperiodic stacking– a so-called semi-random structure - manifesting in diffuse *hkl* reflections with  $k \neq 3n$  due to incoherent scattering (Brindley et al. 1950).

For regular-stacking chlorites, twelve unique periodic arrangements of 2:1 layers (with respect to the position of the adjacent 2:1 layer) occur, resulting in twelve polytypes with different cell-shapes and symmetries. In the case of semi-random stacking structures, they are reduced to six arrangements based on two types of unit cell shapes: (1) orthorhombic-shaped with  $\alpha=\beta=\gamma=90^{\circ}$ , and (2) monoclinic-shaped with  $\beta\approx97^{\circ}$ . In X-ray diffraction (XRD) patterns, an identification of polytypes as well as the unit cell shape can be performed based on *hol* reflections, even for samples with semi-random stacking (Brown and Bailey (1962)).

In order to differentiate between polytypes with the same structural units and the same unit cell shape, for example IIb-2, IIb-4, IIb-6, which differ by the position of subsequent interlayer sheets sandwiching the 2:1 layer, accurate information about *hkl* intensities is needed. Based on powder diffraction patterns, finding the symmetry of such polytypes and distinguishing between them is not possible even for regular-stacking specimens. For example, in C2/m, F(*hkl*)

and  $F(h\bar{k}l)$  are equal; however, in powder diffraction patterns the reflections overlap and individual intensities cannot be distinguished (Aja et al. 2015). This situation becomes more complicated for semi-random stacking structures, characterized by an additional shift  $\pm 1/3b$ .

The polytype IIb-4 is often reported as the most abundant (e.g. Brown and Brindley 106 107 (1962); Joswig et al. 1980; Zheng and Bailey (1989); Aja et al. 2015; Beaufort et al. 2015). Such an assertion can be supported by records available in crystallographic databases; for example, in 108 the Crystallographic Open Database (COD), thirty-two trioctahedral chlorite structures were 109 found (Grazulis et al. 2009). Eighteen of the structures become modified under different 110 pressures (Welch and Marshall (2001); Zanazzi et al. 2006, 2007) and temperature conditions 111 (Guggenheim and Zhan (1999); Zanazzi et al. 2009). Another sixteen structures belong to 112 untreated, raw chlorites. Among them, two are Ib-1 (B=90°) chlorites (Shirozu and Bailey 113 (1965)) and one is the Ia-4 polytype (Bailey (1986)). Nine IIb structures were found in triclinic 114 (IIb-4) and four in monoclinic (IIb-2) crystal systems. Except for one (Walker and Bish (1992)), 115 all deposited structures were refined based onsingle-crystal X-ray or neutron (Joswig et al 1980) 116 117 diffraction data. These data show that despite great progress in powder diffraction in recent decades, especially in the implementation of the Rietveld method, the refinement of a chlorite 118 structure based on powder XRD data is still challenging. In addition, the XRD-based refinement 119 120 of semi-random stacking chlorites has been tested without obtaining satisfactory fits (Walker and Bish (1992)). 121

An invaluable technique that allows for investigating polytypes and disordered stacking sequences in chlorites is high-resolution transmission electron microscopy (HRTEM). This method is often supported by simulations of powder XRD patterns (Kogure et al. 2006; Kameda et al. 2007) and is to-date the only alternative for single crystal diffraction method in studying stacking sequences in chlorites.

127 This study represents an attempt to refine the stacking patterns of semi-random chlorite 128 structures based only on the simulation of powder XRD data. Here we show that - despite

difficulties - reliable results involving the magnitude of the random shift, 2:1 layers rotations andinterstratification of polytypes can be obtained using a simple approach.

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## Samples

Finding pure chlorite, not interstratified with vermiculite (Herbillion and Makumbi (1975)), serpentine (Ahn and Peacour (1985); Ryan and Reynolds (1996); Xu and Veblen (1996); Inoue and Kogure (2016)), or kaolinite (Hillier and Velde (1997)) is challenging, especially for Fe-rich chlorites. Therefore, a large set of chlorites was pre-tested to select the purest chlorite material for further analyses.

Seven tri-trioctahedral chlorites from a common Mg-Fe-series were used in the study. 138 Sptb (from Spitsbergen, Norway), SG7 (Strzegom, Poland), and MtBl (Plan de l'Aguille, Massif 139 du Mont-Blanc, France) were gently ground in a mortar to pass through a <100µm sieve; POST, 140 CCC, CCa-2 (all three from Flagstaff Hill, El Dorado County, CA, USA; Source: Clay Project of 141 the Clay Minerals Society, Post and Plummer, 1972) and Mal (Malacachetta, Brazil) were 142 ground with hexane in a McCrone micronizing mill for 5 minutes. MtBl, Sptb, and CCC 143 chlorites were characterized in work of Lempart et. al. (2018). An identical preparation 144 145 procedure for all the samples was not possible because some samples were obtained thanks to the courtesy of collaborating laboratories and in a few cases, these were received after grinding. 146

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# Methods

**Chemical composition.** Chemical analyses of chlorites were performed using a JEOLJXA-8230 electron microprobe (EPMA). Grains with a 50–100  $\mu$ m diameter were fixed in an epoxy resin and polished with a diamond paste to about half their thickness; powder samples were prepared with randomly oriented crystallites. Individual crystals of chlorites were analyzed in the wavelength-dispersion (WDS) mode with an accelerating voltage of 15 kV, probe current of 15 nA, and beam diameter of 3-5  $\mu$ m. The counting time was 20 s for the peak and 10 s for both

background positions. A high homogeneity of chemical composition amongchlorite crystals was foundduring observations usinghigh-contrast, back-scattered electron (BSE) images. For each chlorite sample, 25 measurement points were found to be sufficient to obtain a reliable and repeatable analysis. Structural formulas(per formula unit) were calculated based on 14 oxygen atoms.

160 The Fe(II)/Fe(III) ratio of the natural samples studied was determined via Mössbauer 161 transmission measurements using an MsAa-3 spectrometer. For a 14.41-keV resonant transition 162 in  ${}^{57}$ Fe a single line commercial  ${}^{57}$ Co(Rh) source kept at room temperature was applied. 163 Collection of data lasted about 24 hours for each spectrum at room temperature. The 164 Fe(II)/Fe(III) ratio was calculated using transmission integral approximation. Spectral shifts were 165 reported versus natural  $\alpha$ -Fe at room temperature.

Structural analysis. In order to minimize effects of texture and preferred orientation in XRD, 166 capillary measurements (diameter: 0.3 mm) were performed. Bruker D8 Advance diffractometer 167 (Karlsruhe, Germany) working in Debye-Scherrer geometry with a CoKa X-ray tube (35 kV, 168 40 mA) was used. The X-ray beam was monochromatized and formed using a Göbel mirror, 169 0.2 mm fixed slit, 2.5° Soller slits, and a beam knife. For the secondary beam, a VANTEC 170 detector equipped with radial and 2.5° Soller slits was used. The scan range was set from 5 to 171 110°20 with a step size of 0.007°20. Qualitative analyses were performed using DIFFRAC.EVA 172 software ver. 4.2.0.31 equipped with a Crystallographic Open Database (COD) (Grazulis et al. 173 2009). Unit cell parameter determination was performed using TOPAS software, version. 5 with 174 the indexing algorithm based on iterative use of least squares, following Coelho (2003). 175

A simulation of XRD patterns was performed using Sybilla3D and 2D softwares (Chevron ETC proprietary). Parameters describing: (1) the content of the IIb-2 polytype ( $W_{IIbb,y=0b}$ or $W_{IIb-2}$ ), (2) distribution of Fe between the 2:1 layer and the interlayer sheet ( $R_{int}$ ), and (3) the percentage of layers shifted only in one direction, +1/3**b** or -1/3**b** ( $W_{IIbb,y=+1/3b}$  or  $W_{IIb-4}$ and  $W_{IIbb,y=-1/3b}$  or  $W_{IIb-6}$ ), thus the proportion of IIb-4 and IIb-6 polytypes ( $R_{shift}$ ), were

introduced. All the above modifications were considered for R0 (random interstratification) and R1 (interstratification described by  $W_j$  and  $P_{ij}$  probabilities with  $W_j \neq P_{ij}$ ) types of stacking order. In addition, the interstratification of different polytypes were analysed; thus, parameters describing the probability of finding of Ibb, Iaa,Iab, Ibb, IIab, and IIaa polytypes fragments were added.

Separately, the possible disorder of orientation of successive 2:1 layers with rotations by 186 0, 120 and 240 degrees were considered along with three translations in y for each case  $(0, +1/3\mathbf{b})$ 187 and -1/3b), which gives structure with 9 types of layers. For both above cases (polytypes and 2:1) 188 layer rotations), only R0 ordering was assumed, which means that the P<sub>ii</sub>- the probability of 189 finding j-type layer after i-type layer in layers succession, was equal to W<sub>i</sub> - probability of the 190 presence of a given polytype or rotated fragment in the structure:  $P_{ii} = W_i$ . Trigonal rotational 191 axes perpendicular to the interlayer plane, imposes that after rotation by 120° and 240° it is 192 indistinguishable from the original one. Therefore, only rotations of 2:1 layer were considered. 193 In this model it was assumed that:  $W_{rot=0} = W_{rot=120} = W_{rot=240}$ , and  $W_{rot=X, y=+1/3b} = W_{rot=X, y=-1/3b}$ 194 (where:  $X=0^{\circ}$ ,  $120^{\circ}$ ,  $240^{\circ}$ ). The only optimized parameter was percent of layers not shifted along 195 196 b.

Also, the simulation of interstratification with 7Å mineral were performed assuming only
R0 ordering (Sybilla 2D). Detailed information about the parameters used in the individual
calculationsis given in Table 1.

Unit cell parameters for all the samples were fitted using the Le Bail approach (Le Bail et al. 1988) implemented in Jana2006 software (Petricek et al. 2014). However, in three cases (MAL, CCa-2 and Mtbl) in which *hkl* reflections with  $k \neq 3n$  were detectable, a conventional indexing procedure was also used.

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### **Results and discussion**

208 Chemical typology of the studied chlorites. The studied chlorites represent a uniform Mg-Fe series where total Fe content ranged from 0.1 in clinochlore (POST) to 3.9 in chamosite (SG7) 209 and proportional Mg content ranged from 4.56 to 0.45 atoms per half formula unit (Table 2) 210 (Bayliss (1975), Guggenheim et.al., (1996)). All Si was assigned to the tetrahedral position and 211 completed to four with Al; the remaining Al was assigned to the octahedral position along with 212 Mg, Fe, Mn, Cr, and Ni cations. In order to fill six octahedral positions, the number of vacancies 213 were calculated; however, it cannot be ruled out that this is an artifact from the EMPA analytical 214 215 error.No tetrahedral Fe was found using either Mössbauer spectroscopy or formula calculation in any of the studied samples. 216

General classification, admixtures, and the interstratification of 14Å and 7Å layers. 217 Detailed analysis of XRD patterns obtained for disoriented specimens including FWHM (B) 218 measurements of 00l reflections excludes the possibility of R0 interstratification of 14Å and 7Å 219 layers (cf. Reynolds (1988); Reynolds et al. 1992; Drits et al. 2001; Inoue and Kogure, 2016). A 220 00l peak width ( $\beta$ \*cos( $\theta$ )) vs. peak diffraction order for 001 to 005 is shown in Figure 2. As 221 observed in all the studied samples, there is no tendency for line broadening of odd-order peaks 222 (Figure 2), which is typical for 7Å-interstratified R0 chlorites (Reynolds et al. 1992). A slight 223 224 increase in reflection width together with  $2\theta$  is an expected outcome related to the strain effect and the  $K_{\alpha 1}$ - $K_{\alpha 2}$  separation (the calculation of FWHMs with stripped  $K_{\alpha 2}$  are presented in Figure 225 SI 1 in the Supporting Information electronic appendix). To confirm our statement, the 226 simulation assuming R0 interstratification of 14Å and 7Å layers was performed. The serpentinite 227  $d_{001}$  was assumed as 7.09Å according to crystallographic data presented by J.S. Slack et al. 1992. 228 Results show that even an assumption of 2% of 7Å mineral interstratification significantly 229 changed the calculated diffraction patterns leading to broadening of the odd number reflections. 230 In Figure SI 2, the FWHM's for diffraction patterns simulated with 2%, 5%, 10% and 30% 231 serpentinite layers content, are presented. 232

233 Based on powder diffraction patterns, all the studied samples may be classified as nearly pure trioctahedral IIb-even (IIbb) chlorites (Brown and Brindley (1962)). Some of the samples 234 contained small admixtures of other minerals - total below 3%: rutile in POST and guartz in 235 Mal, CCC, Sptb, Mtbl and SG7. In all the studied samples, semi-random stacking manifests itself 236 as decreased intensity of peaks in the region between 20 and 28  $^{\circ}2\theta$  (CoK $\alpha$ ) where 021 and 111 237 (1-11) reflections are present. The diffraction patterns of the 'end-members' in respect of Fe 238 content of the studied chlorite series are shown in Figure 3. See Figure SI. 3for all other XRD 239 patterns. 240

241 Determination of unit cell parameters - chlorite indexing. In the XRD patterns of the studied chlorites, the observed first twenty low 20 angle reflections belong to either 00l or h0l 242 overlapping with hkl (k=3n); thus, the calculation of unit cell parameters is difficult due to the 243 indeterminacy of the b dimension as well as  $\alpha$  and  $\gamma$  angles. Nevertheless, for Mal, CCa-2 and 244 Mtbl chlorites, in the range 20-28 °20 (CoK $\alpha$ ), *hkl* reflections with k $\neq$ 3n were slightly 245 distinguishable (Figure SI 4), which suggested better structural ordering, offering a chance for 246 appropriate indexing. For all the above chlorites, the calculated unit cell parameters (using 247 TOPAS software) were close to expected for a monoclinic-shaped cell ( $\beta$ =97°) and in two cases 248 - CCa-2 and Mtbl – the  $\alpha$  and  $\gamma$  angles were found to be 90° (Table 3). The indexing of the Mal 249 250 sample resulted in a primitive (P) unit cell, thus the transformation into a C-centered cell was performed using the transformation matrix shown below: 251

$$\begin{bmatrix} a \\ b \\ c \end{bmatrix} \begin{bmatrix} 1 & -1 & 0 \\ 1 & 1 & 0 \\ 0 & 0 & 1 \end{bmatrix} = \begin{bmatrix} a' \\ b' \\ c' \end{bmatrix}$$

As shown in Table 3, the obtained unit cell shape is not perfectly monoclinic-like. The output sheets for these solutions are presented in the Supporting Information section.

A structural model with a monoclinic-shaped cell and C-1 symmetry, as determined by Zanazzi et al. (2009), was used for La Bail fitting. This choice was made considering the fact that IIb-4 is the polytype most frequently observed in nature in regular-stacking samples as well as based on the conclusion by Brown and Bailey (1962) that all chlorites with semi-random stacking should have triclinic symmetry as a result of averaging of two potential triclinic and one monoclinic layer symmetries. In the three chlorites analysed, the background, zero shift, unit cell, and profile parameters including asymmetry were fitted simultaneously. As expected, *hkl* reflections with  $k \neq 3n$  were not fitted to a satisfying level; however, this lack of proper fit may be accepted since its impact on the determination of unit cell parameters was negligible. Calculated cell parameters are provided in Table 4. The known dependency,  $a=b/\sqrt{3}$ , is preserved for calculated parameters.

The obtained results, from both the conventional indexing and La Bail fitting, can be considered as equally reliable, as long as, no conclusion about symmetry is drawn based on these results.

**Layer stacking disorder consideringR0 ordering**. In the three studied samples, where *hkl* reflections with  $k \neq 3n$  were distinguishable, suggesting a higher ordering of these structures, their stacking pattern could be determined based on XRD features in the range 20-28 °20 (CoK $\alpha$ ). Therefore, by a comparison between experimental and simulated diffraction patterns, the semirandom stacking sequence can be identified (cf. Kogure et al. 2006; Kameda et al. 2007).

Sets of parameters provided in Table 1 were optimized using Sybilla3D software according to two different protocols (S1, S2) and assuming R0 ordering, as shown in Table 5. In all performed simulations, the total Fe content and the percentage of vacancies were fixed based on chemical composition (Table 2). The vacancies were assumed to be present only in the interlayer sheet, for the sake of simplification. The sigma star parameter was optimized in order to reflect increasing crystallite orientation in Fe-rich chlorite specimens.

In the first step of each simulation (Protocol S1, Table 5), all the parameters excluding the total Fe content and percentage of vacancies were fitted simultaneously. The obtained b and **c**\* parameter values were in good agreement with the values obtained using La Bail fitting (comparison in Table SI 1 in the electronic appendix). In the S1 protocol, the percentage of layers shifted at - 1/3**b** was assumed equal to that shifted about +1/3**b**, thus the R<sub>shift</sub> was fixed at 0.5 (W<sub>IIb-4</sub>=W<sub>IIb-6</sub>). Several simulated XRD patterns for different values of W<sub>IIbb,y=0b</sub>, thus

283 different IIb-2 content, are shown in Figure 4 for two chlorites with different shapes of 284 diffraction pattern in the range 20-28 °2 $\theta$  (CoK $\alpha$ )and different Fe contents. Simulations for all 285 other samples are shown in Figures SI 5-9.

In all the studied chlorites, the distribution of Fe between the interlayer sheet and 2:1 layer had a strong impact on simulated XRD patterns. The modelled crystallite orientation (sigma star parameter) strongly correlated with the results of Fe distribution. The general tendency found as a result of the simulations indicates that in clinochlorites, Fe occurs mostly in the interlayer sheet; with increasing Fe content, the distribution between the octahedral sheets of the interlayer and the 2:1 layer becomes uniform(close to 50:50).

As observed in Protocol S1 results, the change in the  $W_{IIbb,y=0b}$  parameter (with the assumption that  $R_{shiff}=0.5$ , thus  $W_{IIb-4}=W_{IIb-6}$ ) has a significant impact on the diffraction patterns in the range 20-28 °20, offering an opportunity to investigate disordered layer stacking in chlorites. In four cases: POST, CCC, Sptb and SG7,  $W_{IIbb,y=0b}$  was found to be between 0.2 and 0.4, indicating that 20% to 40% of the layers are not shifted at ±1/3b. In turn, in three other cases (Mal, CCa-2, Mtbl), this simple approach turned out to be insufficient for obtaining a good fit (Figures SI 5, SI 8 and Figure 4b, respectively).

In the second step (Protocol S2, Table 5), the fit and parameter values obtained in step S1 299 300 were used as inputs along with the parameter expressed as the percentage of layers shifted only in one direction (R<sub>shift</sub>), +1/3b or -1/3b. In the case of POST, CCC, Sptb and SG7 chlorites, no 301 significant improvements of fit were found, mostly due to the fact that R<sub>shift</sub> had changed slightly 302 in comparison to the value 0.5 fixed in Protocol S1, and optimized values were in the range 0.5-303 0.6. However, using the S2 simulation protocol for CCa-2 and Mtbl, the obtained fit was better 304 than that using the S1 protocol, but still not perfect (Figures5a and 5b, respectively). 305 Nevertheless, the results showed that in these two cases, roughly 80% to 90% of layers were 306 shifted in one direction (R<sub>shift</sub>= 0.8-0.9), suggesting some kind of ordering manifested in a one-307 direction shift of layers in the b crystallographic direction (predominance of one polytype IIb-4 308

or IIb-6). During the S2 simulation, the input parameters obtained from S1 were refined,however, the values remained practically identical.

Due to the persistently poor fit of CCa-2 and Mtbl patterns, the presence of an additional, highly ordered chlorite phase was postulated: the  $W_{IIbb,y=0b}$  and  $R_{shift}$  parameters for these additional structures were refined and yielded 0.7 and 0.8, respectively. Other parameters were fixed at the same values as in the dominant chlorite phase (details in Table SI 2). This approach led to a significant improvement of fit, which is clearly visible in Figure 6.

Summarizing, in the case of CCa-2 and Mtbl (assuming R0 ordering), two effects are responsible for the unique features of XRD patterns in the range 20-28 °20 (Figure 6). The first effect is related to a preferential shift along b in one direction (predominance of IIb-4 or IIb-6), and the second effect was due to the presence of an additional, highly ordered phase, where almost 70% of the layers are not shifted along b (predominance of IIb-2). We emphasize that the presence of both these phases is required to obtain an acceptable fit for R0 ordering.

In the case of chlorite Mal, assuming only the presence of a second, highly ordered chlorite population was insufficient to obtain a good fit. In order to obtain a matching simulation, an assumption of the presence of three phases was required, including one with a much lower value of the b parameter (Figure 7). The presence of three phases may explain difficulties with the indexing and obtaining of a deformed unit cell for the chlorite Mal (Table 3). Details of the optimization for chlorite Mal are presented in the Supporting Information section (Table SI 3).

Interstratification of different polytypes assuming R0 ordering. Because different polytypes (Figure SI 10)can occur not only as different crystallite populations, but also as part of a stacking sequence, parameters describing the interstratification of different polytypes were also considered during the third step of refinement (Protocol S3, Table 5). In an analogy to the previous step, the output parameters from step S2 were used as the inputs in S3. In order to limit the complexity of the model, besides IIbb, for which probabilities of all shifts along **b** were considered (0,  $\pm 1/3$ **b**), all the other polytypes' structures were refined, assuming that the shift

335 along b equals zero. As it turned out, adding interstratification had only a minor impact on the range 20-28 °20; however, in the range 30-60 °20, a significant improvement of fit was observed 336 (Figure 8, Figures, SI 11-16). The interstratification of polytypes other than IIbb (Iaa, IIab and 337 IIaa) was found in all the studied chlorite samples. The obtained fractions of other, interstratified 338 polytypes ranged from 5% to 19% and were not correlated with Fe content in the samples. 339 Similar results were observed by Kogure and Banfield (1998); Inoue and Kogure (2016) used 340 HRTEM to show the presence of several chlorite polytypic sequences including IIab, Iab, Ibb, 341 and Iaa. In contrast to these studies, the IIaa phase was also found among the stacking sequences 342 of the studied chlorite samples (TableSI4). Fits in full 20 range are presented in Figure SI 17-343 23A. 344

Layer stacking disorder considering rotations of 2:1 layer assuming R0 ordering. The possible rotations of 2:1 layers by 0, 120 and 240 degrees together with shifts along **b** (0, +1/3**b** and -1/3**b**) were considered giving nine types of layers (Protocol S4, Table 5). Results show that, the following structures turned out to be nearly equivalent:(1) (rot0, y=1/3**b**)  $\equiv$  (rot0, y=-1/3**b**)  $\equiv$ (rot120, y=0**b**)  $\equiv$  (rot 120, y=-1/3**b**)  $\equiv$  (rot240, y=0**b**)  $\equiv$  (rot 240, y=+1/3**b**) (2) (rot0, y=0**b**)  $\equiv$ (rot120, y=1/3**b**)  $\equiv$  (rot240, y=-1/3**b**). The reason for this similarity for IIbb chlorite is presented in Supplementary Materials (Figure SI 24-27).

In spite of this correspondence and thus very similar XRD patterns of structures in each of the 352 two groups, stacking faults give different results for rotation defects and structures with shifts 353 along **b** (Figure 9). In the case of structure containing only rotations in crystallite there are clear 354 seven peaks in the range  $20 - 28^{\circ} 2\theta$  corresponding to *hkl* reflections with k≠3n. Relative 355 proportions between intensities of these peaks depends on proportions between rot0, y=0b, 356 rot120, y=0b and rot240, y=0b. There is no significant broadening of reflections in the mixed 357 layer crystallite. On the other hand, in the case of a crystallite composed of layers with shifts 358 along **b** peak broadening of *hkl* reflections with  $k \neq 3n$  is significantly more pronounced (Figure 359 9). 360

Calculated diffraction patterns for mixed layer structures that considers rotations and shifts along b have very good correspondence with experimental patterns of chlorites that have distinguishable *hkl* reflections with  $k \neq 3n$ : CCa-2 and Mtbl (Figure 10). In both cases, the best fits were obtained for equal content of 2:1 rotations ( $W_{rot0}=W_{rot120}=W_{rot240}=0.3333$ ). For each 2:1 rotation,  $R_{shift}$  was set as 0.5 ( $W_{rot=X, y=+1/3b}=W_{rot=X, y=-1/3b}$ ; where:  $X=0^{\circ}$ , 120°, 240°) and the percent of layer not shifted along b (sum of  $W_{rot=0, y=0b}, W_{rot=120, y=0b}$  and  $W_{rot=240, y=0b}$ ) was established as 75%, and 64%, respectively for Mtbl and CCa-2(Table SI 5).

Layer stacking disorder considering R1 ordering. The output parameters from the protocol S2 368 were used as the input data for the simulation assuming R1 ordering. All parameters were fixed 369 (the value obtained from S2) and only the junction probabilities of the stacking sequences were 370 refined (protocol S5, Table 5). In the first step, the probability P<sub>IIb-2 IIb-2</sub> was optimized. As it 371 turned out, for chlorite POST, CCC, Sptb and SG7, the best results were obtained, when the 372  $P_{IIb-2}IIb-2=W_{IIb-2}$  (thus  $P_{IIb-2}IIb-2$  is equal to the content of IIb-2). This result is identical to those 373 obtained previously for R0 ordering. In turn, for chlorite CCa-2 and Mtbl, to obtain a reliable fit 374 (Figure 11), the  $P_{IIb-2 IIb-2}$  was evaluated as 0.5, thus higher than for R0 ordering ( $W_{IIb-2} = 0.25$ 375 and 0.23, respectively). All parameters are presented in Table SI 6. 376

This suggests some kind of segregation and presence in the structure domain with predominance of IIb-2 and IIb-4/-6 polytypes. It should be highlighted that this result does not contradict the fits with assumed R0 ordering. In this case, two populations of chlorites that differ in content of IIb-2 and IIb-4 or IIb-6 was assumed to obtain satisfactory fits, which is also kind of segregation. Fits are presented in Figure SI 17-23B.

382 Obtaining reliable fits for MAL was impossible, since the presence of an additional phase 383 with a lower  $d_{010}$  was probably needed and in case of R1 ordering, the model became too 384 complicated. Selection of the most probable model. There are several structural models that offer good agreement between experimental and calculated patterns of the studied chlorites for *hkl* reflections with  $k \neq 3n$ :

- Assumption of more than one chlorite phases and shifts only along **b** (Figure 6 and 7),

- Consideration of rotations and shifts along **b** (Figure 9 and 10),

- Consideration of R1 ordering and shifts only along **b** (Figure 11),

The second model is the most elegant as it contains the lowest number of parameters to be 391 optimized. In the case of several phases in the system, there is much more parameters to be 392 optimized and therefore there is a possibility to optimize a larger variability of experimental 393 patterns, if assuming contrasting patterns as components. On the other hand, the third model 394 (with R1 ordering) does not have an intuitive explanation for why certain P<sub>ii</sub> parameters were 395 chosen. Ordering for certain shift along b can, however, correspond to some extent to the 396 rotation, which was shown in Supplementary Materials. Therefore, the second model can be 397 chosen as the most probable. Moreover, this conclusion can be supported by HRTEM results as 398 rotations by 120 and 240° were experimentally found (Kogure et al., 2017). 399

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## Implications

Chlorite polytypes are conventionally interpreted in terms of geothermometry for 402 diagenesis, metamorphism, and hydrothermal conditions (Walker 1993, Spoetl et al. 1994; 403 Beaufort et al. 2015). In this study we have shown how a careful simulation of chlorite powder 404 XRD patterns allows for a comprehensive determination of the layers' stacking sequence with a 405 combination of shifts not only in the crystallographic a direction (= conventional chlorite 406 polytypism) but also in the b direction, which also significantly improves the interpretation of *hkl* 407 reflections with  $k \neq 3n$ . Even in semi-random stacking chlorites, there is a certain kind of ordering 408 409 manifesting itself as a predominant shift in one of the  $\pm 1/3b$  directions. The new capability of

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| 410 | refining both dimensions of the chlorite stacking sequence - using common XRD methods -          |
|-----|--|
| 411 | offers new interpretations of chlorite formation conditions (cf. Mata et al. 2004).              |
| 412 | Up to now, information about chlorite polytype interstratification was obtained mostly           |
| 413 | using HRTEM methodology on selected spots (Kogure and Banfield (1998); Mata et al. 2004;         |
| 414 | Inoue and Kogure (2016)), but not on bulk samples. Powder XRD-based determination of             |
| 415 | polytype interstratification allows for the examination of chlorite polytype geothermometry as a |
| 416 | continuous temperature-structure relationship, rather than thermal stability ranges of discrete  |
| 417 | polytypes (Walker (1993); Mata et al. 2004; Spoetl et al. 1994).                                 |
| 418 |  |
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### References

- Ahn, H.J., and Peacor, D.R. (1985) Transmission electron microscope study of diagenetic
  chlorite in Gulf Coast argillaceous sediments. Clays and Clay Minerals, 33, 228-236.
- 439 Aja, S., Omotoso, O., Bertoldi, C., Dasch E., and Benisek, A. (2015) The structure and
- thermochemistry of three Fe-Mg chlorites. Clay and Clay Minerals, 63, 351-367.
- Bailey, S.W. (1986) Re-evaluation of ordering and local charge balance in Ia chlorite. Canadian
  Mineralogist, 24, 649-654.
- Bayliss, P. (1975) Nomenclature of the trioctahedral chlorites, Canadian Mineralogist, 13, 178180.
- Beaufort, D., Rigault, C., Billon, S., Billault, V., Inoue, A., Inoue, S. and Patrier, P., 2015.
- 446 Chlorite and chloritization processes through mixed-layer mineral series in low-temperature
- 447 geological systems–a review. Clay Minerals, 50(4), 497-523.
- Brindley, G.W., Oughton B.M., and Robinson K. (1950) Polymorphism of the chlorite. I.
  ordered structures. Acta Crystallographica, 3, 408-416.
- Brown, B.E., and Bailey, S.W. (1962) Chlorite polytypism: 1. Regular and semi-random 1 layer
  structures. American Mineralogist, 47, 819-850.
- 452 Coelho, A.A (2003) Indexing of powder diffraction patterns by iterative use of singular value
  453 decomposition, Applied Crystallography, 36, 85-96.
- 454 Drits V.A., Ivanovskaya T. A., Sakharov B. A., Gor'kova N. V., Karpova G. V., Pokrovskaya E.
- 455 V. (2001) Pseudomorphous Replacement of Globular Glauconite by Mixed-Layer Chlorite-
- 456 Berthierine in the Outer Contact of Dike: Evidence from the Lower Riphean Ust'-Il'ya
- 457 Formation, Anabar Uplift, Lithology and Mineral Resources, 36, 337.
- 458 Guggenheim, S., Alietti, A., Bain, D.C., Drits, V.A, Formoso, M.M.L., Galan, E., Hudnall, W.,
- 459 Köster, H.M., Paquet, H., Watanabe, T. (1996) Clay Minerals, 32, 493-495.

- 460 Grazulis, S., Chateigner, D., Downs, R.T., Yokochi, A.T., Quirós, M., Lutterotti L., Manakova
- 461 E., Butkus J., Moeck P., and Le Bail A. (2009) Crystallography Open Database an open-access
- 462 collection of crystal structures. Journal of Applied Crystallography, 42, 726-729.
- Herbillion, A., and Makumbi, M. (1975) Weathering of chlorite in a soil delivered from a
  chloritoschist under humid tropical conditions. Geoderma, 13, 89-104.
- 465 Hillier, S., and Velde, B. (1997) Octahedral occupancy and the chemical-composition of
  466 diagenetic (low-temperature) chlorites. Clay Mineralogy, 26(2),149–168.
- 467 Inoue, S., and Kogure, T. (2016) High resolution transmission electron microscopy (HRTEM)
- 468 study of stacking irregularity in Fe- rich chlorite from selected hydrothermal ore deposits. Clay
- 469 and Clay minerals, 64, 131-144.
- Joswig, W., Fuess, H., Rothbauer, R., Takeuchi, Y., and Mason, S.A. (1980) A neutron
  diffraction study of a one-layer triclinic chlorite (penninite). American Mineralogist, 65, 349352.
- +72 JJ2.
- Kameda, J., Miyawaki, R., Kitagawa, R., and Kogure, T. (2007) XRD and HRTEM analyses of
  stacking structures in sudoite, di-trioctahedral chlorite. American Mineralogist, 92, 1586-1592.
- Kogure, T., and Banfield, J.F. (1998) Direct identification of the six polytypes of chlorite
  characterized by semi-random stacking. American Mineralogist, 83,925-930.
- Kogure, T., Kameda, J., Matsui, T., and Miyawaki, R. (2006) Stacking structure in disordered
  talc: Interpretation of its X-ray diffraction pattern by using pattern simulation and highresolution transmission electron microscopy. American Mineralogist, 91, 1363-1370.
- Kogure, T., Ishii, T., Kikuchi, R., Miyawaki, R., Yuguchi, T. (2017) Two types of chlorite
  transformed from biotite by hydrothermal alteration of granite, 16th International Clay
  Conference (Granada, Spain, July 20, 2017), 418.
- 483 Le Bail A., Duroy, H., and Fourquet, J.L. (1988) Ab-initio structure determination of LiSbWO<sub>6</sub>
- 484 by X-ray powder diffraction. Materials Research Bulletin, 23, 447–452.

- 485 Lempart, M., Derkowski, A., Luberda-Durnaś, K., Błachowski, A., and Skiba, M. (2018)
- 486 Dehydrogenation and dehydroxylation as drivers of the thermal decomposition of Fe-chlorites.
- 487 American Mineralogist, 103(11), 1837-1850.
- 488 Mata, M.P., Peacor, D.R. and López-Aguayo, F. (2004) Polytypism of cookeite in low-grade
- 489 metapelites of the Cameros Basin, Spain: Lack of correlation of well-ordered polytypes with
- 490 pressure. American Mineralogist, 89(10), 1510-1515.
- 491 Pauling L. (1930) The structure of the chlorites. Proceedings of the National Academy of
  492 Science of USA, 16, 578-582.
- 493 Petricek, V., Dusek, M., and Palatinus, L. (2014) Crystallographic Computing System Jana2006:
- 494 general features. Zeitschrift für Kristallographie, 229(5), 345-352.
- 495 Post J., and Plummer C. (1972) The chlorite series of Flagstaff Hill area, California: A
  496 preliminary investigation. Clays and Clay Minerals, 20, 271-283.
- 497 Reynolds, R.C. (1986) The Lorenz-Polarization factor and preferred orientation in oriented clay
  498 aggregates. Clay and Clay minerals, 34, 359-367.
- 499 Reynolds, R. C., Jr. (1988) Mixed layer chlorite minerals: in Hydrous Phyllosilicates Exclusive
- of Micas, S. W. Bailey, ed., Reviews in Mineralogy, Vol. 19, Mineralogical Society of America,
  Washington, D.C., 601–629.
- 502 Reynolds, R.C. Jr., DiStefano, M.P., and Lahann, R.W. (1992) Randomly interstratified
- serpentine/chlorite: its detection and quantification by powder X-ray diffraction methods. Clay
- 504 and Clay Minerals, 40, 262-167.
- 505 Ryan, P.C, and Reynolds R.C. Jr. (1996) The origin and diagenesis of grain-coating serpentine-
- chlorite in Tuscaloosa formation sandstone. American Mineralogist, 81, 213-225.
- 507 Shirozu, H., and Bailey, S.W. (1965) Chlorite polytypism. III. Crystal structure of an 508 orthohexagonal iron chlorite. American Mineralogist, 50, 868-885.

- 509 Slack, J.F., Jiang, W.-T., Peacor, D. R., Okita, P.M. (1992) Hydrothermal and metamorphic
- 510 berthierine from the Kidd Kreek volcanogenic massive sulfide deposit, Timmins, Ontario.
- 511 Canadian Mineralogist, 30, 1127-1142.
- 512 Spoetl, C., Houseknecht, D.W., and Longstaffe, F.J., 1994. Authigenic chlorites in sandstones as
- 513 indicators of high-temperature diagenesis, Arkoma foreland basin, USA. Journal of Sedimentary
- 514 Research, 64(3a), pp.553-566.
- Walker J. R., and Bish D. L. (1992) Application of Rietveld refinement techniques to a
  disordered IIb Mg-chamosite. Clay and Clay Minerals, 40, 319-322.
- 517 Walker, J.R., 1993. Chlorite polytype geothermometry. Clays and Clay minerals, 41(2), pp.260518 267.
- 519 Welch, M. D.; and Marshall, W. G. (2001) High-pressure behaviour of clinochlore Sample: P = 0
- 520 GPa. American Mineralogist, 86, 1380-1386.
- 521 Zanazzi, P. F., Comodi, P., Nazzareni, S., and Andreozzi, G. B. (2009) Thermal behaviour of
- 522 chlorite: an in situ single-crystal and powder diffraction study Locality: Val Malenco, Italy Note:
- 523  $T = 25, 301^{\circ}$ C. European Journal of Mineralogy, 21, 581-589.
- 524 Zheng, H., and Bailey S.W. (1989) Structures of intergrown triclinic and monoclinic IIb chlorites
- from Kenya. Clays and Clay Minerals, 37, 308-316.
- 526 Xu, H., and Veblen, D.R. (1996) Interstratification and other reaction microstructures in the
- 527 chlorite-berthierine series, Contribution to Mineral Petrology, 124, 291-301.
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## 535 List of figure captions:

- 536 Figure 1. Composition of natural chlorites used in our study projected onto the chemographic coordinates.
- **Figure 2**. Calculated peak widths multiplied by  $cos(\theta)$  for all investigated chlorites vs. diffraction order for the 001
- 538 to the 005 reflections for disoriented specimens. No removal of instrumental broadening effects or stripping of the

539 K $\alpha$ 2 signal was performed.

- 540 Figure 3. Diffraction patterns and their inserts with the region between 20 to 28 °2θ (CoKα) of (a) POST chlorite
- 541 (low Fe content) and (b) SG7 chlorite (high Fe content). Note large differences in peak intensity in the inserts. The
- 542 largest peaks corresponding to admixtures are shown using red lines (R-rutile, Q-quartz).
- 543 Figure 4. Comparison of experimental and simulated XRD patterns of semi random-stacking chlorites in the range
- 544 20-40 °20 for (a) POST chlorite (low Fe content) and (b) Mtbl chlorite (high Fe content). W<sub>IIbb,y=0b</sub> was calculated
- assuming that the number of layers shifted by  $+1/3\mathbf{b}$  is equal to the number of layers shifted by  $-1/3\mathbf{b}$ , i.e.  $R_{shift} = 0.5$ .
- 546 Figure 5. Experimental and simulated data for(a) CCa-2, (b) Mtbl for fits where R<sub>shift</sub> was fixed at the value 0.5 (red
- 547 line) and  $R_{shift}$  was refined ( $W_{IIbb,y=\pm 1/3b}=1-W_{IIbb,y=0b}$ ).
- 548 Figure 6. Experimental and simulated data for (a) CCa-2, (b) Mtbl for models considering two chlorite phases.
- 549 Figure 7. Experimental and simulated data for Mal assuming the presence of three phases.
- 550 Figure 8. Comparison of fit for Mtbl models (a) without and (b) with consideration of interstratified polytypes other
- than IIbb. The error of calculation of probability was evaluated as  $(\pm 2\%)$ .
- 552 Figure 9. Comparison of experimental and simulated XRD patterns for Mtbl chlorite in the range 20-28 °2θ,
- 553 considering various percentages of different rotations and shifts along **b**.
- Figure 10. Experimental and simulated data for (a) CCa-2, (b) Mtbl for models considering rotations of 2:1 layer
  and shifts along b.
- 556 Figure 11. Experimental and simulated data for (a) CCa-2, (b) Mtbl assuming R1 ordering.
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## **Table 1**. Detailed information on the parameters used in the simulation.

| Parameter   | Description  |  |  |  |  |  |  |
|---|--|--|--|--|--|--|--|
| General parameters (parameters used in all simulations) |  |  |  |  |  |  |  |
| Sigma star (SS) $\sigma$                                | Parameter describing degree of preferred orientation (Reynolds, 1986)  |  |  |  |  |  |  |
| $N_{mean}$  | Mean number of crystallites along c* direction   |  |  |  |  |  |  |
| Particle Radius (PR)*                                   | Mean a×b plane radius  |  |  |  |  |  |  |
| d   | 001 distance for chlorite  |  |  |  |  |  |  |
| b*  | b unit cell parameter  |  |  |  |  |  |  |
| Fe <sub>tot</sub>                                       | Total Fe content per unit cell   |  |  |  |  |  |  |
| Interlayer content (IC)                                 | Octahedral sheet occupancy in the interlayer   |  |  |  |  |  |  |
| R <sub>Fe</sub>   | Percentage of Fe in the interlayer relative to total Fe content, $R_{Fe}$ =Fe <sub>int</sub> /Fe <sub>tot</sub>                          |  |  |  |  |  |  |
| $W_{IIbb,y=0\boldsymbol{b}} \boldsymbol{*}$             | Probability of occurrence of IIbb layers without a shift along b (IIb-2 content)   |  |  |  |  |  |  |
| R <sub>shift</sub> *                                    | Probability of finding the 1/3b shift in one direction   |  |  |  |  |  |  |
|   | Simulation of 14Å and 7Å layer interstratification   |  |  |  |  |  |  |
| d   | 001 distance for serpentinite  |  |  |  |  |  |  |
| SerR <sub>Fe</sub>                                      | Octahedral iron content in serpentinite  |  |  |  |  |  |  |
| Wi  | Probability of founding 7Å layers  |  |  |  |  |  |  |
|   | Simulation of polytype interstratification   |  |  |  |  |  |  |
| W <sub>IIbb</sub>                                       | Probability of occurrence of the IIbb polytype in the sample(y=0b, +1/3b, -1/3b)   |  |  |  |  |  |  |
| W <sub>Iaa</sub>  | Probability of occurrence of the Iaa polytype in crystals (y=0b)   |  |  |  |  |  |  |
| W <sub>Iab</sub>  | Probability of occurrence of the Iab polytype in crystals (y=0b)   |  |  |  |  |  |  |
| W <sub>Ibb</sub>  | Probability of occurrence of the Ibb polytype in crystals (y=0b)   |  |  |  |  |  |  |
| W <sub>IIab</sub>                                       | Probability of occurrence of the IIab polytypecrystals(y=0b)   |  |  |  |  |  |  |
| W <sub>IIaa</sub>                                       | Probability of occurrence of the IIaa polytype in crystals (y=0b)  |  |  |  |  |  |  |
|   | Simulation of rotations of 2:1 layer   |  |  |  |  |  |  |
| W <sub>rot=0</sub>                                      | Probability of occurrence of 2:1 layer rotated about degree $0^{\circ}$  |  |  |  |  |  |  |
| W <sub>rot=120</sub>                                    | Probability of occurrence of 2:1 layer rotated about degree 120°   |  |  |  |  |  |  |
| W <sub>rot=240</sub>                                    | Probability of occurrence of 2:1 layer rotated about degree 240°   |  |  |  |  |  |  |
| $W_{rot=0, y=0b}$                                       | Probability of occurrence of layers rotated about $0^{\circ}$ without a shift along <b>b</b>   |  |  |  |  |  |  |
| $W_{rot=0, y=\pm 1/3b}$                                 | Probability of occurrence of layers rotated about 0° with a shift along $b = \pm 1/3 \mathbf{b}$   |  |  |  |  |  |  |
| $W_{rot=120, y=0b}$                                     | Probability of occurrence of layers rotated about 120° without a shift along <b>b</b>  |  |  |  |  |  |  |
| $W_{rot=120, y=\pm 1/3b}$                               | Probability of occurrence of layers rotated about $120^{\circ}$ with a shift along $b = \pm 1/3$ <b>b</b>                                |  |  |  |  |  |  |
| $W_{rot=240, y=0b}$                                     | Probability of occurrence of layers rotated about $240^{\circ}$ without a shift along <b>b</b>   |  |  |  |  |  |  |
| W <sub>rot=240, y=±1/3b</sub>                           | Probability of occurrence of layers rotated about $240^{\circ}$ with a shift along $b = \pm 1/3$ b                                       |  |  |  |  |  |  |
| Simulation assuming R1 ordering                         |  |  |  |  |  |  |  |
| P <sub>ij</sub>   | Junction probabilities of finding j-type of layer after i-type of layer (in the case of 3-layer type there are 9 junction probabilities) |  |  |  |  |  |  |

## **571 Table 2**. Chemical composition of chlorite samples.

|                                | POST              | Mal               | CCC               | Sptb                  | CCa2                       | MtBl             | SG7             | Analytical techniques |
|--------------------------------|-------------------|-------------------|-------------------|-----------------------|----------------------------|------------------|-----------------|-----------------------|
|                                |                   |                   |                   | wt%                   |                            |                  |                 |                       |
| SiO                            | 30 77±0 57        | 29 93±0 33        | 30 23±0 426       | 28 035±0 205          | 26 42±0 52                 | 23 79±0 25       | 23 31±0 39      | а                     |
| $Al_2O_2$                      | $22.48 \pm 1.00$  | $20.75\pm0.33$    | $19.79\pm 0.493$  | $20.39\pm0.200$       | 20.12 = 0.02<br>21 17±0 76 | $20.49\pm 0.29$  | $20.28\pm0.56$  | a                     |
| FeO                            | $1.38\pm0.85$     | 4.05±0.15         | $5.495 \pm 0.22$  | $16.00\pm0.24$        | 20.656±0.63                | $36.29 \pm 0.52$ | 39.60±0.82      | a.b                   |
| Fe <sub>2</sub> O <sub>3</sub> | $0.03 \pm 0.02$   | 1.20±0.045        | $1.62 \pm 0.06$   | 1.13±0.17             | $2.00\pm0.06$              | 2.12±0.03        | 2.81±0.17       | a,b                   |
| $Cr_2O_3$                      | b.d.l.            | $0.20\pm0.02$     | 0.13±0.04         | b.d.l.                | $0.04{\pm}0.03$            | b.d.l.           | b.d.l           | a                     |
| MgO                            | 33.25±0.89        | $31.58 \pm 0.40$  | $31.01 \pm 0.49$  | 22.63±0.24            | 18.52±0.625                | $5.586 \pm 0.16$ | $2.70 \pm 0.63$ | а                     |
| MnO                            | b.d.l.            | b.d.l.            | $0.12 \pm 0.04$   | $0.12 \pm 0.05$       | $0.085 \pm 0.045$          | $0.822\pm 0.08$  | $0.83 \pm 0.21$ | а                     |
| NiO                            | b.d.l.            | $0.25 \pm 0.03$   | 0.13±0.03         | b.d.l.                | $0.09 \pm 0.05$            | b.d.l.           | b.d.l           | а                     |
| TiO <sub>2</sub>               | $0.02 \pm 0.01$   | $0.09 \pm 0.01$   | $0.05 \pm 0.01$   | $0.03 \pm 0.01$       | $0.07 \pm 0.03$            | $0.025 \pm 0.01$ | $0.03 \pm 0.01$ | а                     |
| $H_2O(+)$                      | 12.91             | 12.66             | 12.56             | 12.00                 | 12.73                      | 10.87            | 10.57           | а                     |
|                                |                   |                   | Numbe             | r of atoms per half f | formula unit               |                  |                 |                       |
| Si                             | 2.83±0.04         | 2.810±0.015       | 2.85±0.03         | 2.79±0.02             | 2.68±0.05                  | 2.64±0.02        | 2.64±0.03       |                       |
| Al <sup>IV</sup>               | $1.17 \pm 0.11$   | 1.190±0.024       | 1.15±0.05         | 1.21±0.02             | $1.32 \pm 0.09$            | $1.35 \pm 0.03$  | 1.36±0.06       |                       |
| $Al^{VI}$                      | $1.27 \pm 0.11$   | 1.105±0.024       | $1.04{\pm}0.05$   | 1.17±0.02             | $1.22 \pm 0.09$            | 1.33±0.03        | $1.34 \pm 0.06$ |                       |
| Cr <sup>3+</sup>               | -                 | $0.020 \pm 0.003$ | $0.010 \pm 0.004$ | b.d.l.                | b.d.l.                     | b.d.l.           | b.d.l.          |                       |
| Fe <sup>3+</sup>               | $0.002 \pm 0.001$ | $0.085 \pm 0.003$ | 0.115±0.005       | $0.085 \pm 0.01$      | $0.150 \pm 0.005$          | $0.18 \pm 0.002$ | $0.24{\pm}0.02$ |                       |
| Fe <sup>2+</sup>               | $0.11 \pm 0.07$   | $0.32 \pm 0.013$  | $0.43 \pm 0.02$   | $1.33 \pm 0.02$       | $1.755 \pm 0.06$           | 3.37±0.04        | $3.75 \pm 0.09$ |                       |
| Mg                             | 4.56±0.11         | $4.42 \pm 0.04$   | 4.35±0.06         | 3.35±0.03             | $2.80\pm0.09$              | $0.93 \pm 0.03$  | $0.45 \pm 0.10$ |                       |
| Mn                             | -                 | -                 | $0.010 \pm 0.004$ | $0.010 \pm 0.004$     | $0.007 \pm 0.004$          | $0.08 \pm 0.007$ | $0.08 \pm 0.02$ |                       |
| Ni                             | -                 | -                 | $0.010 \pm 0.002$ | b.d.l.                | $0.007 \pm 0.004$          | b.d.l.           | b.d.l.          |                       |
| Fe/(Fe+Mg)                     | 0.02              | 0.08              | 0.11              | 0.30                  | 0.405                      | 0.79             | 0.90            |                       |
| Total oct                      | 5.94              | 5.95              | 5.965             | 5.945                 | 5.94                       | 5.89             | 5.86            |                       |

572 Notes: Analytical techniques: (a) electron microprobe analysis, (b) Mössbauer analysis, b.d.l. = below detection limit

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Table 3. Unit cell parameters obtained from the indexing procedure implemented in TOPAS software.

|                         | Mal <sup>a</sup> | CCa-2   | Mtbl    |  |
|-------------------------|------------------|---------|---------|--|
| a [Å]                   | 5.7073 (5.5667)  | 5.3568  | 5.3890  |  |
| b [Å]                   | 5.3263 (9.4730)  | 9.2688  | 9.3371  |  |
| c [Å] 14.2559 (14.2560) |                  | 14.2608 | 14.2441 |  |
| α [°]                   | 86.4920 (89.50)  | 90.0000 | 90.0000 |  |
| β [°]                   | 92.4370 (95.76)  | 97.0550 | 97.3640 |  |
| γ [°]                   | 61.7230 (85.51)  | 90.0000 | 90.0000 |  |
| V [Å <sup>3</sup> ]     | 379.670 (759.24) | 702.701 | 710.817 |  |
| GOF <sup>b</sup> [%]    | 14.16            | 18.53   | 31.70   |  |

<sup>a</sup>The cell parameters in C-centered cell are present in the partentheses.

<sup>b</sup>GOF=  $\sqrt{\frac{w(y_0-y_c)^2}{M-P}}$ , w=1/(y\_0), y\_o - observed intensity, y\_c - calculated intensity, M-number of data points, P- number of parameters.

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Table 4. Unit cell parameters calculated using Le Bail fitting.

|                     | POST        | Mal         | CCC        | Sptb         | CCa-2       | Mtbl        | SG7        |
|---------------------|-------------|-------------|------------|--------------|-------------|-------------|------------|
| a [Å]               | 5.31209(12) | 5.32801(19) | 5.3318(2)  | 5.34355(17)  | 5.35619(13) | 5.38984(13) | 5.3968(5)  |
| b [Å]               | 9.1997(3)   | 9.2271(5)   | 9.2327(5)  | 9.2577(2)    | 9.2766(3)   | 9.3344(3)   | 9.3508(10) |
| c [Å]               | 14.3017(3)  | 14.3023(6)  | 14.3090(9) | 14.26647(16) | 14.2615(4)  | 14.2387(3)  | 14.2349(5) |
| α [°]               | 90.955(4)   | 90.277(14)  | 90.540(13) | 89.940(3)    | 90.393(3)   | 90.371(3)   | 90.687(13) |
| β [°]               | 97.157(2)   | 97.162(5)   | 97.124(4)  | 96.909(3)    | 97.234(2)   | 97.243(2)   | 97.294(8)  |
| γ [°]               | 90.458(5)   | 89.448(11)  | 90.110(13) | 90.122(3)    | 90.005(3)   | 89.990(3)   | 89.945(11) |
| V [Å <sup>3</sup> ] | 693.34(3)   | 697.61(5)   | 698.92(7)  | 700.62(3)    | 703.08(3)   | 710.48(4)   | 712.49(11) |

# 586 Table 5. Parameters that were fixed and optimized in the applied calculation protocols using modified Sybilla 3D

587

code.

| Parameter                              | Protocol S1 | Protocol S2 | Protocol S3 | Protocol S4 | Protocol S5 |
|--|-------------|-------------|-------------|-------------|-------------|
| Sigma star (SS) $\sigma^*$             | refined     | refined     | refined     | fixed       | fixed       |
| N <sub>mean</sub>                      | refined     | refined     | refined     | fixed       | fixed       |
| Particle Radius (PR)                   | refined     | refined     | refined     | fixed       | fixed       |
| d                                      | refined     | refined     | refined     | fixed       | fixed       |
| b                                      | refined     | refined     | refined     | fixed       | fixed       |
| Fe <sub>tot</sub>                      | fixed       | fixed       | fixed       | fixed       | fixed       |
| Interlayer content (IC)                | fixed       | fixed       | fixed       | fixed       | fixed       |
| R <sub>Fe</sub>                        | refined     | refined     | refined     | fixed       | fixed       |
| W <sub>IIbb</sub>                      | Fixed=1     | Fixed=1     | refined     | Fixed=1     | Fixed=1     |
| $W_{IIbb,y=0\mathbf{b}}$               | refined     | refined     | refined     | -           | fixed       |
| R <sub>shift</sub>                     | Fixed=0.5   | refined     | refined     | Fixed=0.5   | fixed       |
| W <sub>Iaa</sub>                       | Fixed=0     | Fixed=0     | refined     | -           | Fixed=0     |
| W <sub>Iab</sub>                       | Fixed=0     | Fixed=0     | refined     | -           | Fixed=0     |
| W <sub>Ibb</sub>                       | Fixed=0     | Fixed=0     | refined     | -           | Fixed=0     |
| $W_{\mathrm{IIab}}$                    | Fixed=0     | Fixed=0     | refined     | -           | Fixed=0     |
| W <sub>IIaa</sub>                      | Fixed=0     | Fixed=0     | refined     | -           | Fixed=0     |
| W <sub>rot=0</sub>                     | -           | -           | -           | refined     | -           |
| W <sub>rot=120</sub>                   | -           | -           | -           | refined     | -           |
| W <sub>rot=240</sub>                   | -           | -           | -           | refined     | -           |
| W <sub>rot=0, y=0b</sub>               | -           | -           | -           | refined     | -           |
| $W_{rot=0, y=\pm 1/3b}$                | -           | -           | -           | refined     | -           |
| $W_{rot=120, y=0b}$                    | -           | -           | -           | refined     | -           |
| $W_{rot=120,\;y=\pm1/3\boldsymbol{b}}$ | -           | -           | -           | refined     | -           |
| W <sub>rot=240, y=0b</sub>             | -           | -           | -           | refined     | -           |
| $W_{rot=240,\;y=\pm1/3\boldsymbol{b}}$ | -           | -           | -           | refined     | -           |
| P <sub>ij</sub>                        | -           | -           | -           | -           | refined     |

588 All abbreviations as in Table 1.

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**Figure 1**. Composition of natural chlorites used in our study projected onto the chemographic coordinates.

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Figure 2. Calculated peak widths multiplied by cos(θ) for all investigated chlorites vs. diffraction order for the 001
to the 005 reflections for disoriented specimens. No removal of instrumental broadening effects or stripping of the
Ka2 signal was performed.

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600Figure 3. Diffraction patterns and their inserts with the region between 20 to 28 °2θ (CoKα) of (a) POST chlorite601(low Fe content) and (b) SG7 chlorite (high Fe content). Note large differences in peak intensity in the inserts. The602largest peaks corresponding to admixtures are shown using red lines (R-rutile, Q-quartz).

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**Figure 4**. Comparison of experimental and simulated XRD patterns of semi random-stacking chlorites in the range 20-40 °20 for (a) POST chlorite (low Fe content) and (b) Mtbl chlorite (high Fe content).  $W_{IIbb,y=0b}$ was calculated assuming that the number of layers shifted by +1/3b is equal to the number of layers shifted by -1/3b, i.e.  $R_{shift} = 0.5$ .

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**Figure 5**. Experimental and simulated data for(**a**) CCa-2, (**b**) Mtbl for fits where  $R_{shift}$  was fixed at the value 0.5 (red line) and  $R_{shift}$  was refined ( $W_{IIbb,y=\pm 1/3b}=1-W_{IIbb,y=0b}$ ).

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624 Figure 7. Experimental and simulated data for Mal assuming the presence of three phases.

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**Figure 8**. Comparison of fit for Mtbl models (**a**) without and (**b**) with consideration of interstratified polytypes other

629 than IIbb. The error of calculation of probability was evaluated as  $(\pm 2\%)$ .

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Figure 9. Comparison of experimental and simulated XRD patterns for Mtbl chlorite in the range 20-40 °2θ,
considering various percentages of different rotations and shifts along b.



635

Figure 10.Experimental and simulated data for (a) CCa-2, (b) Mtbl for models considering rotations of 2:1 layerand shifts along b.





Figure 11. Experimental and simulated data for (a) CCa-2, (b) Mtbl assuming R1 ordering.