

Deep carbon cycle through five reactions

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ABSTRACT

What are the key reactions driving the global carbon cycle in Earth, the only known habitable planet in the solar system? And how do chemical reactions govern the transformation and movement of carbon? The special collection “Earth in five reactions - A deep carbon perspective” features review articles synthesizing knowledge and findings on the role of carbon-related reactions in Earth's dynamics and evolution. These integrative studies identify gaps in our current understanding and establish new frontiers to motivate and guide future research in deep carbon science. The collection also includes original experimental and theoretical investigations

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23 of carbon-bearing phases and the impact of chemical and polymorphic reactions on Earth's deep
24 carbon cycle.

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26 **Background**

27 The Earth in Five Reactions (E5R) project was conceived in the fall of 2015 at the
28 University of Rhode Island, USA, where the Deep Carbon Observatory (DCO) held a synthesis-
29 planning meeting. DCO is a ten-year project supported by the Alfred P. Sloan Foundation with
30 an overarching goal to understand the quantities, movements, forms, and origins of Earth's deep
31 carbon. Members of the international DCO Science Network are addressing this goal through
32 investigations focused on four distinct and interconnected thematic disciplines – Deep Life (DL),
33 Deep Energy (DE), Reservoirs and Fluxes (RF), and Extreme Physics and Chemistry (EPC).
34 Since its launch in 2009, DCO has initiated and supported scientific campaigns to investigate
35 deep carbon, leading to numerous findings reported in more than 800 scholarly publications to
36 date, and created an international network of more than 1000 deep carbon scientists
37 (www.deepcarbon.net). At the synthesis-planning meeting, representatives of the four
38 communities brainstormed strategies for integrating the wealth of knowledge from this
39 multidisciplinary and international effort and creating a lasting legacy for the scientific
40 community and broader audiences.

41 “Serpentinization is the most important reaction in the universe!” This bold statement made
42 by a workshop participant provoked Jie Li, an EPC representative who had studied chemical and
43 polymorphic reactions for decades but thought little about serpentinization, to challenge the
44 assertion. Li argued that redox and melting reactions dictate global-scale differentiation and
45 therefore are far more important than serpentinization. This fundamental question about the key

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46 drivers in deep carbon science sparked a lively and spirited debate and revealed a general lack of
47 consensus. Jesse Ausubel, the Sloan Foundation's primary liaison to the DCO, watched this
48 exchange and asked "How about Earth in five reactions?"

49 The idea emerged as a promising framework for synthesis: Chemical reactions are
50 widespread and play important roles in Earth's carbon cycle. Viewing Earth processes through
51 the lens of reactions would highlight the chemical aspect of DCO science and could stimulate
52 dialogues across disciplines. Like math and music, chemical reactions are the same in the United
53 States, China, Italy or, France. The concept works internationally, even if people understand little
54 or no English, and therefore could be widely reported or easily translated.

55 The E5R project aimed to identify the five most important reactions governing the
56 transformation and movement of carbon in Earth, and then use these reactions as the central
57 themes for synthesizing and disseminating the findings of the Deep Carbon Observatory. This
58 thematic structure also provides a new and integrative perspective for understanding and
59 advancing deep carbon science as a new, multi-disciplinary scientific discipline.

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61 **Selecting top five carbon-related reactions**

62 We began by polling the DCO community to introduce the "Earth in Five Reactions"
63 initiative and seek input on the five most important carbon-related reactions on Earth. A survey
64 was launched just before the Third DCO International Science meeting in St. Andrews, Scotland
65 in the spring of 2017. The poll was distributed at the meeting and through newsletters of relevant
66 organizations, providing the opportunity for all members of the DCO science network and others
67 to weigh in. By the end of year, 120 submissions were received. Representatives from all four
68 DCO communities and researchers at various academic levels ranging from emeritus professors

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69 to undergraduate students completed the survey. More than half of the respondents not only
70 answered the multiple-choice questions but provided additional comments. We also received
71 about twenty very detailed answers with elaborate essays, illustrations, and references.

72 The first survey question is: What criteria should be used for selecting a handful of reactions
73 out of myriad chemical processes involving carbon in different host phases, variable valence
74 states, under a wide range of pressure and temperature conditions, and over a vast span of spatial
75 and temporal scales? A reaction may be considered important because it is essential to sustaining
76 life on Earth (e.g., photosynthesis that converts carbon dioxide and water into sugar and releases
77 oxygen). A top-ranking reaction may involve a component that is minor in quantity but is of
78 special economic and geological interest (e.g., diamond formation). Proposed as a potential
79 solution to the global warming problem, carbonation of mantle peridotite may be viewed as
80 potentially important. On a more fundamental level, crystallization of Earth's molten core to
81 concentrate carbon in the solid inner sphere could stand out because it may bear on the driving
82 power of the Earth's magnetic field. Some critical reactions may have stretched over an extended
83 time scale (e.g., inner core formation) whereas others may be widespread spatially and occur
84 under broad ranges of pressure and temperature (for example, redox reactions in solids and
85 liquids). Unique reactions that can be used as indicators, tracers, or diagnostic tools for carbon
86 cycling are other possible targets of interest.

87 The distribution of responses to the criteria question was not particularly clear cut, with
88 prevalence, timing, location, nature and impact all having an approximately similar number of
89 votes. The narrative comments were revealing as well, ranging from one individual stating that
90 most significant was "importance in terms of the fluxes of carbon they process and their impacts
91 on the habitable planet" to another who felt criteria should be based on how the reactions

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92 “change the oxidation state of carbon—reduced, neutral, oxidized, with the product of the
93 reaction having very different transport properties”. Most telling was a third commenter, who
94 stated “I feel very strongly about this”, which was reassuring given the effort that the team had
95 put in to enabling the whole exercise. A further provocative response suggested that “one way is
96 to ask what if the Earth could be made again, but with only five reactions, which five involving
97 carbon would make it look most like it does today?” Clearly there are many routes to discussing
98 the “importance” or interest in any particular reaction, or indeed what is meant by reaction –
99 whether the term should be restricted to chemical reactions or whether process or physical
100 reactions might also be included.

101 Survey respondents were then asked to pick their favorite reactions. Given the outcomes of
102 the suggested criteria, the outcomes of the reactions viewed most important were not particularly
103 surprising. The importance of photosynthesis to the development of life on Earth, and the
104 importance of life to the respondents, is an understandable priority. Other reactions that were
105 also mentioned at this stage include precipitation of calcite and dolomite in the sea,
106 mineralization to form limestone and dolomite (to sequester CO₂), silicate weathering to
107 carbonate, asthenosphere melting (to allow plate tectonics), dissolution of CO₂ gas into water,
108 respiration (reverse photosynthesis, to generate sugars), redox reactions of CO to C or CO₂,
109 redox melting, the Sabatier reaction: the passage from inorganic to organic geochemistry, the
110 burning of fossil fuels, and the polymerization increase in C-bearing minerals inside the deep
111 Earth.

112 On the basis of the polling responses, we defined five broad categories of reactions for
113 further consideration, including serpentinization, respiration/photosynthesis,
114 degassing/decarbonation, extreme carbon sequestration, and diamond formation. These reactions

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115 encompass the four DCO communities and represent a diversity of reactants/products, pressure
116 (P), temperature (T), and catalyst conditions, and reaction mechanism, energetics, and kinetics.
117 Their importance may vary with depth as well as time in Earth's history.

118 In March 2018 we convened a two-day workshop to select the top five carbon-related
119 reactions on Earth and develop a plan for sharing advances in deep carbon science with the
120 scientific community and broader audiences using the E5R framework. The workshop was held
121 at the Carnegie Institution for Science in Washington DC, USA. About 50 participants from
122 seven countries on three continents represented the DCO community. The group was selected to
123 reflect the totality of the DCO in terms of interests and scientific expertise, and achieved balance
124 in terms of academic level, gender, and geographic distribution. Education and media experts,
125 along with several members of DCO's Executive Committee, Secretariat, and SG2019, rounded
126 out the attendees.

127 The participants were charged with choosing five discrete chemical reactions from among
128 hundreds that make Earth the only known habitable planet. They began by considering the
129 survey results and pondered what carbon-related reactions make Earth unique. The workshop
130 provided a trans-disciplinary forum for researchers to review the state of current knowledge and
131 to identify the critical mechanisms and processes that govern the movements of carbon through
132 Earth. The group discussed the role of the deep carbon cycle in plate tectonics and the
133 geodynamo, the development of an oxygen-rich atmosphere, how microbial life has persisted
134 throughout Earth's history giving rise to a diverse biosphere, various ways water has influenced
135 Earth's evolution, and the origin of diamonds. All attendees presented their perspectives and
136 shared their ideas on how we could use chemical reactions as a framework to understand and
137 advance deep carbon science. With keynote speakers, short-talk presenters, and panelists primed

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138 to argue for or against their chosen reactions, debates were passionate and sometimes
139 intellectually divisive. By the middle of day two, however, the group converged on a set of
140 reactions central to defining Earth.

141 Five reactions were selected through anonymous voting. Prior to voting, participants agreed
142 that a pair of forward and reverse reactions counted as one reaction, and that similar and closely
143 related reactions would be grouped into a reaction class. With this understanding in mind, eight
144 reaction classes made to the ballot (Table 1). Hydrogenation, carboxylation, carbonation, carbon
145 dioxide dissolution, and hydration emerged as winners.

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147 **Implications**

148 **Understanding deep carbon cycle through key reactions**

149 The quest to identify the five most important reactions in deep carbon science has
150 demonstrated that chemical reactions can provide a unique and effective framework for
151 synthesizing deep carbon research. Looking at a particular reaction such as serpentinization has
152 stimulated dialogue across DCO communities and led to deeper appreciation of its role in Earth's
153 volatile cycles. Mafic and ultramafic rocks react with water to form serpentinite. The geological
154 process of serpentinization significantly affects the reservoirs and fluxes of carbon at subduction
155 zones. In the presence of iron, serpentinization may produce hydrogen and form methane, thus
156 profoundly influencing deep life on Earth, and maybe even life's origins. Chemical reactions
157 also can be used as "threads" to connect disparate findings into coherent and meaningful pictures.
158 For example, redox reactions are prevalent in geological and biological processes and often
159 involve carbon-bearing species with variable valence state. They are of interest to all
160 communities within DCO: Redox reactions have been found to influence volcanism, diamond

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161 formation, the abiogenic production of hydrocarbons and are central to life's metabolism. By
162 comparing the mechanisms, conditions and energetics of these reactions and studying how they
163 vary spatially and through geological time, we may gain insights into the connections among the
164 deep carbon cycle, the “great oxidation event”, and the origins of life on Earth.

165 Why five reactions? The idea was inspired by the familiar "five types of chemical reactions"
166 in typical high-school chemistry textbooks. Mathematicians and physicists have had success with
167 celebrating “Five Equations that Changed the World”. We considered selecting five reactions in
168 each DCO community, in addition to the five that encompass all communities. However, for the
169 idea to work effectively, we chose to limit the number to five, much like the number of medals in
170 each Olympic sport is limited to three. In reality, "Five” is not a magic or required number as the
171 outcome, rather a gimmick to stimulate the DCO community to build its shared experiences. It
172 was exciting to find out what the outcome would be!

173 At the workshop, the top five reactions received comparable numbers of votes, suggesting
174 that the richness of DCO findings cannot be straightforwardly captured by a small number of
175 reactions and that there is a healthy diversity of equally important processes. The three deep
176 Earth reactions, including two reactions involved in diamond formation, did not make the final
177 five. These deep Earth reactions are undoubtedly important because at least 90% of Earth's
178 carbon is likely stored in the deep mantle and core. The voting results thus suggest a lack of
179 awareness and appreciation for this deep carbon, even among DCO researchers. It implies that
180 understanding extreme carbon remains at the frontier of future research and will require more
181 effort to bring public awareness.

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Following discussion of the E5R survey and selection of reactions, a group of scientists was motivated to review and integrate recent findings through the lens of the selected reactions. This special issue will feature articles using the selected reactions to integrate DCO science findings and illuminate the forms and flows of carbon in Earth. The collection as whole provides a big picture view of DCO discoveries in the preceding decade, what its four Science Communities have learned about the role of carbon in planetary function, and how the identified five reactions play an integral role in carbon storage and pathways in Earth.

The E5R synthesis project distilled the planet's essence into a set of key carbon-related reactions that make Earth special and then used the reactions to encapsulate much of deep carbon science. It has led to new insights to motivate and guide future research. We hope that the special collection will help establish new frontiers for scientific exploration and investigation to address the fundamental question of Earth's habitability.

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209 **TABLE 1.** The eight reaction classes on the ballot

Reaction class	Representative Reactions
Hydrogenation dehydrogenation	$\text{Fe} + \text{H}_2\text{O} = \text{H}_2 + \text{FeO}$
Carboxylation decarboxylation	$6\text{CO}_2 + 6\text{H}_2\text{O} = \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$
Carbonation decarbonation	$\text{CaSiO}_3 + \text{CO}_2 = \text{CaCO}_3 + \text{SiO}_2$
Carbon dioxide dissolution outgassing	$\text{CO}_2(\text{aq}) = \text{CO}_2(\text{g})$
Hydration dehydration	$2\text{Mg}_2\text{SiO}_4 + 3\text{H}_2\text{O} = \text{Mg}_3\text{Si}_2\text{O}_5(\text{OH})_2 + \text{Mg}(\text{OH})_2$
Redox freezing melting	$\text{MgCO}_3 + 2\text{Fe} = 3(\text{Fe}_{2/3}\text{Mg}_{1/3})\text{O} + \text{C}$
Metal silicate partitioning	$\text{C}(\text{alloy}) + 2\text{FeO}(\text{silicate}) = \text{CO}_2(\text{silicate}) + 2\text{Fe}(\text{alloy})$
Fe-C solidification melting	$\text{FeC}_x(\text{l}) = \text{FeC}_y(\text{l}) + \text{Fe}_7\text{C}_3$, where $y < x$

210 **FIGURE 1** E5R logo (left) and special collection theme figure (right) showing where the
 211 reactions likely occur during Earth's deep carbon cycle.
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