1 REVISION 1

2	New insights into the zircon-reidite phase transition
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14	ABSTRACT
15	The structure, the elastic properties and the Raman frequencies of the zircon and reidite polymorphs
16	of ZrSiO ₄ were calculated as a function of hydrostatic pressure up to 30 GPa using HF/DFT ab-initio
17	calculations at static equilibrium (0 K). The softening of a silent (B _{1u}) mode of zircon leads to a phase
18	transition to a "high-pressure - low-symmetry" (HPLS) ZrSiO ₄ polymorph with space group $I\overline{4}2d$
19	and cell parameters a=6.4512 Å c=5.9121Å V = 246.05Å ³ (at 20 GPa). The primary coordination of
20	SiO ₄ and ZrO ₈ groups in the structure of zircon is maintained in the high-pressure phase, and the new
21	phase deviates from that of zircon by the rotation of SiO ₄ tetrahedra and small distortions of the ZrO_8
22	dodecahedra. The new polymorph is stable with respect to zircon at 20 GPa and remains a
23	dynamically-stable structure up to at least 30 GPa. On pressure release the new phase reverts back to
24	the zircon structure, and therefore cannot be quenched in experiments. In contrast, the transformation
25	from zircon to reidite is reconstructive in nature and results in a first-order transition with a volume
26	and density change of about 9%. The calculated energies from the DFT simulations yield an
27	equilibrium transition pressure of 9.13(1) GPa at 0 K. Simulations of the Raman spectra of the three

polymorphs at 20 GPa show how they can be distinguished. In particular, the peak due to the lowestenergy A_1 mode with a calculated wavenumber of 94 cm⁻¹ is diagnostic of the HPLS phase because it does not overlap with any of the peaks of zircon or reidite.

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32 Keywords: zircon, reidite, phase transition, Raman spectroscopy, high pressure

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INTRODUCTION

Zircon (ZrSiO₄, space group $I4_1/amd$) is an extremely stiff material and upon meteorite bombardment 34 shock microstructures form within it at pressures of 20 GPa or more (Leroux et al. 1999). These 35 microstructures remarkably survive subsequent metamorphism without being obliterated and, 36 therefore, they can provide diagnostic criteria to identify impact structures (French 1998; French and 37 Koeberl 2010, Wittmann et al. 2006). During shocks zircon may transform into the high-pressure 38 scheelite-type polymorph reidite (space group $I4_1/a$). Naturally-occurring reidite was first identified 39 using X-ray diffraction (XRD) in zircon grains from Eocene ejecta associated with the ~90 km 40 diameter, ca. 35.7 Ma Chesapeake Bay impact structure (Glass and Liu 2001; Glass et al. 2002) and 41 42 subsequently in several other impact structures (Gucsik et al. 2004; Wittmann et al. 2006, Wittmann 43 and Reimold 2009, Chen et al. 2013, Cavosie et al. 2015, Singleton et al. 2015, Reddy et al. 2015, Cavosie et al. 2018). Therefore, in order to study the origin of impact structures and to further 44 45 constrain the impact conditions it is crucial to understand the stability relationships between zircon and its high-pressure polymorph reidite. 46

This phase transition from zircon to reidite has a volume decrease of about 9% (e.g. Reid and 47 Ringwood 1969; Ono et al. 2004a) which suggests that it is thermodynamically first-order. Moreover, 48 the topological differences in the atomic linkages indicate that the zircon-reidite phase transition is 49 reconstructive, which is supported by ab initio calculations (e.g. Marqués et al. 2006, Dutta and 50 Mandal 2012a). Because of the reconstructive nature of the zircon-reidite transformation, the 51 equilibrium transition pressure is not well constrained. At elevated temperatures of 1000-1500 K the 52 pressure-induced zircon-to-reidite phase transition occurs near 8 GPa (Reid and Ringwood 1969, Ono 53 et al. 2004b). Several static loading experiments in diamond-anvil pressure cells at room temperature 54

have shown the presence of reidite when the pressure is increased above 18-23 GPa (Knittle and 55 Williams 1993; Van Westrenen et al. 2004, Morozova 2015). However, there seems to be a 56 discrepancy between the detection of reidite by Raman spectroscopy in the samples compressed to 57 58 ca. 20 GPa and higher pressures at room temperature, and the results of X-ray diffraction which show 59 evidence only of zircon (van Westrenen et al. 2004). On the other hand, shock wave studies found this transformation to occur at much higher pressures (30-50 GPa) (Liu 1979, Kusaba et al. 1986, 60 Gucsik et al. 2004) and very quickly, at microsecond time scales. The latter observation favors a 61 martensitic transformation mechanism for impact-produced reidite, since other solid-state 62 mechanisms seem unlikely to operate on the nanosecond to seconds timescale of shock events 63 because diffusion rates are too slow (Langenhorst and Deutsch 2012). 64

Therefore, to elucidate the pressure-induced structural transformations of zircon we have performed *ab initio* simulations of zircon and reidite to determine their structures, elastic behavior, phonon frequencies and the energy curves as a function of pressure (up to 25 GPa and 17 GPa respectively) at 0 K, which allow us to resolve the ambiguities in the experimental data. We also provide calculated Raman spectra which will aid in the identification of the polymorphs of zircon.

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METHODS

72 Ab initio HF/DFT simulations were performed with the two most recent versions of the CRYSTAL code, CRYSTAL14 (Dovesi et al., 2014) and CRYSTAL17 (Dovesi et al. 2018), employing the 73 hybrid Hamiltonian WC1LYP which is particularly suitable for the correct reproduction of the elastic 74 and vibrational properties of crystals (e.g. Prencipe 2012; Aliatis et al. 2015; Stangarone et al. 2016, 75 2017). In this Hamiltonian the Wu-Cohen DFT-GGA exchange contribution (Wu and Cohen 2005) 76 is corrected by mixing it with a percentage (16%) of the exact non-local Hartree-Fock exchange. The 77 correlation is described by the LYP functional (Lee et al. 1988). The localized contracted atomic 78 basis sets used were Si 86-311G(1d) (Pascale et al. 2004) and 8-411G(2d) (Valenzano et al. 2006) 79 for Si and O, respectively, and the Hay-Wadt (HAYW) small-core (SC) pseudopotential basis 80 SC HAYWSC-3111(32111df) (Sophia et al. 2013) for Zr. The basis sets were taken from the 81

repository at the address http://www.crystal.unito.it/basis-sets.php. In CRYSTAL, the level of 82 accuracy in evaluating the Coulomb and Hartree-Fock exchange series is controlled by five 83 parameters (Dovesi et al. 2018), specified by the TOLINTEG keyword in CRYSTAL. These 84 85 parameters were set to 8, 8, 8, 8, and 18. Reciprocal space was sampled using a regular sublattice 86 with a shrinking factor of 4, corresponding to 24 independent k vectors in the irreducible part of the Brillouin zone. The exchange-correlation DFT contributions to the Fock matrix were evaluated by 87 numerical integration over the unit cell volume. Radial and angular points for the integration grid 88 were generated through Gauss-Legendre radial quadrature and Lebedev two-dimensional angular 89 point distributions. In the present work, a pruned grid with 75 radial and 974 angular points was used 90 (see XLGRID keyword in the CRYSTAL manual). Such parameters defined grids of 17604 points in 91 the unit cell for zircon, and 32094 for reidite. A measure of the high numerical accuracy reached with 92 the grids is the evaluation of the total number of electrons in the unit cell, by the numerical integration 93 of the electron density over the cell volume. For zircon we obtained 116.00003 electrons and for 94 reidite 116.00013, both out of 116, for the reference static volume (0 K, with no zero-point vibrational 95 96 effects included) at 0 GPa. The crystal structures were optimized on the basis of analytical energy 97 gradients with respect to fractional atomic coordinates and unit cell parameters (Doll 2001, Civalleri et al. 2001). Default values were chosen for convergence of gradient components as well as nuclear 98 99 displacements. For the optimized crystallographic structure, the calculation of all phonon modes in 100 zircon was carried out at static equilibrium. Vibrational wavenumbers and normal modes were calculated within the limit of the harmonic approximation, by diagonalizing a mass-weighted Hessian 101 matrix, whose elements are the second derivatives of the full potential of the crystal with respect to 102 mass weighted atomic displacements. The first derivatives of the energy with respect to the atomic 103 positions are calculated analytically (Doll et al. 2001), whereas the second derivatives are numerically 104 calculated by setting the nuclear displacements to 0.003 Å from the equilibrium positions. The 105 phonon frequencies and the structural data (atom coordinates and unit-cell parameters) are reported 106 107 in the deposited *crystallographic information files (CIFs)*. Relative Raman intensities were computed 108 using a fully analytical approach (Maschio et al. 2013) implemented in the CRYSTAL14 program. It

109	combines analytical gradients (Doll et al. 2001) with solutions of first-order and second-order coupled
110	perturbed Hartree-Fock/Kohn-Sham equations (Ferrero et al. 2008) for the linear and quadratic
111	orbital responses to electric fields in the different Cartesian directions. The thresholds for
112	convergence of the coupled perturbed equations were set to the default values (Dovesi et al. 2014).
113	Simulations were performed at 10 pressures from 0 to 25 GPa for zircon and 8 pressures to 17 GPa
114	for reidite. The <i>P</i> - <i>V</i> equations of state have been computed from these data with both CRYSTAL17,
115	by using the fully-automated scheme (option EOS), or with CRYSTAL14 by fitting the unit-cell
116	volumes and pressures from the simulations with EosFit7c (Angel et al. 2014). Both methods yield
117	the same results within numerical rounding errors.
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119	RESULTS AND DISCUSSION
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121	Zircon Structure
122	The structure of zircon contains Si in tetrahedral coordination by oxygen, and Zr in 8-fold
123	coordination by oxygen in the form of a triangular-faced dodecahedron (sometimes called a sub-
124	disphenoid). Four of the 18 edges of each of the ZrO ₈ polyhedra are shared with neighboring ZrO ₈
125	polyhedra to form a 3-dimensional framework (Figure 1a). A further two edges are shared with two
126	SiO ₄ tetrahedra to form chains of alternating edge-shared SiO ₄ and ZrO ₈ polyhedra that run along the
127	c-axis. The remaining four edges of the SiO ₄ tetrahedra are unshared and the SiO ₄ tetrahedra are
128	isolated from one another.
129	The optimized structure of zircon obtained from the DFT simulation at static conditions (0 K, with
130	no zero point vibrational effects included) and zero pressure exhibits unit-cell parameters that are
131	about 0.6% larger than recent experimental values at room conditions (Finch et al., 2001, Kolesov et
132	al. 2001, Finch and Hanchar 2003) and the Si-O bonds and the shorter Zr-O bonds (around 2.15 Å)
133	in the simulation are both about 0.7-0.8% longer than the experimentally-determined values. This
134	behavior is expected when GGA functionals are employed, and is probably due to the self-interaction

error (Cremer 2001) as the overestimation of the volume of the core region where the electron

correlation is large (due to the high electron density) is not perfectly corrected by the DFT functionals. 136 However, since the core region plays no role in determining chemical properties of the system, this 137 is not a problem. What is important for studying the phase transformation of zircon to reidite, which 138 139 have the same primary coordination of Si and Zr, is that the local structural distortions in zircon are 140 well-reproduced by the DFT simulation. Thus, the Zr-O bonds to the shared dodecahedral O-O edges with the SiO₄ tetrahedra are 0.13 Å longer than those to O not involved in edge-sharing with the SiO₄, 141 and the O-Si-O angle to the same shared edge is only 96.5° (Figure 2), which are within 2 esd's of the 142 experimental values at room conditions (Finch et al., 2001, Kolesov et al. 2001). 143

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145 Zircon under hydrostatic pressure

The simulation at 5 GPa, and the changes from the simulation at zero pressure (both at 0 K), show 146 the same approximately uniform compression of the structure reported in the only (to the authors' 147 148 knowledge) experimental high-pressure structural study of zircon (Hazen and Finger 1979) which 149 was limited to a maximum pressure of 4.8 GPa. The higher precision of the DFT simulations, together with the simulations to the highest pressure of 25 GPa, allow the static compression mechanisms to 150 be defined in more detail. The O-Si-O angles change by less than 1° over this entire pressure range, 151 meaning that the distortion of the SiO₄ tetrahedra remains almost unchanged. However, the Si-O bond 152 lengths shorten considerably, and the bulk modulus of the tetrahedra is 351(2) GPa. In contrast, the 153 shorter Zr-O bonds are significantly softer than the longer Zr-O bonds, because the longer Zr-O bonds 154 are to the oxygen atoms involved in the shared polyhedral edge with the SiO₄ tetrahedra, and are 155 therefore stiffened by Zr-Si repulsion (Hazen and Finger 1979). The overall bulk modulus of the ZrO8 156 polyhedra in zircon calculated from the DFT simulations is 209(1) GPa, so they are significantly 157 158 softer than the SiO₄ tetrahedra.

The isothermal Reuss bulk modulus of zircon obtained from fitting a third-order Birch-Murnaghan equation of state (BM3 EoS) to the unit-cell volumes of the static DFT simulations (i.e. without zeropoint vibrational effects) at high pressures is 220.3(5) GPa, indicating that the compression of the

 ZrO_8 polyhedra dominate the bulk properties of zircon, with a $K_0' = 4.71(4)$. The bulk modulus is 162 lower than the value of 227(2) GPa at 0 K that can be derived from a complete analysis of all available 163 experimental data (Zaffiro et al. 2018). The pressure derivative of the bulk modulus, K₀', cannot be 164 165 reliably determined from currently-available experimental data, and has been estimated to lie in the 166 range 5 to 6.5 at 300 K (Zaffiro et al. 2018). The DFT simulations reproduce the correct sense of elastic anisotropy (Table 1) known from experimental data (Ozkan and Jamieson 1978, Zaffiro et al. 167 2018) with the *c*-axis being significantly stiffer under compression than the *a*-axis as a consequence 168 of the chains of edge-sharing of the SiO_4 and ZrO_8 polyhedra that run parallel to the *c*-axis. 169

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171 Raman vibrational frequencies of zircon

The experimental Raman shifts at 70 K reported by Syme et al. (1977) show a good agreement with 172 the Raman active modes calculated at zero pressure and 0 K (Table 2). The calculated trends of the 173 Raman shifts with pressure are in good agreement with recent experimental data up to 10 GPa 174 obtained from non-metamict single crystals under hydrostatic pressure (Pina-Binvignat et al. 2018). 175 176 That experimental study also confirmed that zircon does not undergo any phase transition under 177 hydrostatic pressures up to 10 GPa at room temperature. The wavenumbers of most of the Raman modes are predicted to increase with increasing pressure, meaning that $\partial \omega / \partial P$ is positive (Table 2), 178 179 in agreement with the experimental data. As expected, bond lengths decrease as pressure (evaluated at the static level) is imposed on the crystal and this also changes the energetics of the crystal: the 180 shortening of the interatomic distances increases the energy of the structure at the static equilibrium 181 and also the restoring forces on the atoms, and therefore the vibrational frequencies. However, two 182 Raman-active modes of zircon (E_{g1}(1) at $\omega_{DFT} = 197$ cm⁻¹ and B_{2g} at $\omega_{DFT} = 250$ cm⁻¹), show the 183 opposite behavior, with softening of the vibrational frequency with increasing pressure. This is also 184 in agreement with the experiments as shown in Figure 3 and Table 2. 185

As displayed in the animated sketches of the vibrations of these two modes shown in Figures S1 and S2 (see also Sheremetyeva et al. 2018), the specific atom displacements involved in these two modes are mainly *rigid* movements of the polyhedra which do not significantly modify the bond lengths. In

the B_{2g} mode the Zr⁴⁺ ions do not move appreciably, while the SiO₄ units undergo a shearing motion 189 190 that does not significantly modify the bond lengths. Similarly the $E_g(1)$ mode consists of a rigid rotation around the *c*-axis of the SiO₄ without changes in the Si–O bond lengths. In both cases, the 191 192 application of external hydrostatic pressure slightly reduces the force constants, resulting in a 193 lowering of the normal mode wavenumber with a negative $\partial \omega / \partial P$ coefficient. In other words, the increase of the repulsion among the nuclei due to the volume contraction is dynamically partially 194 195 compensated by the low-frequency SiO₄-rotation modes, which facilitate the reduction of the Si-O-Si bond angles and the increase of internuclear distances of the static equilibrium configuration (see 196 197 the discussion in Prencipe et al. (2011)). As a consequence, the potential wells become shallower and the corresponding vibrational frequencies decrease as pressure increases. 198

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In Table 2 we also report $\partial \omega / \partial P$ from our simulations up to 25 GPa, which shows that $\partial \omega / \partial P$ becomes 200 more negative as pressure increases. However, none of the calculated frequencies of the Raman-active 201 modes goes to zero, so this softening does not imply that the structure becomes dynamically unstable. 202 203 Nonetheless our calculations show that the lowest-frequency mode at 131 cm⁻¹, which is a silent mode 204 with B_{1u} symmetry and thus not observable with conventional experimental Raman spectroscopy, softens completely to zero at around 20 GPa (Figure 4). Therefore, according to our calculations, if 205 206 hydrostatic pressure conditions are maintained up to 20 GPa at 0 K zircon undergoes a displacive phase transition. This phenomenon was already reported by Smirnov et al. (2008) in DFT simulations 207 of zircon up to 70 GPa, but the issue was not further discussed. 208

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210 High-Pressure – low-symmetry new polymorph

Following the procedure used in Prencipe et al. (2011), the soft mode was analyzed by initially shifting the atoms of the unit cell along the eigenvector associated with the B_{1u} mode: $I\bar{4}2d$ space group symmetry resulted. By optimizing the geometry in this space group and keeping the cell volume fixed at the value corresponding to the softening (P = 20 GPa and V = 246.05Å³), the optimised

structure has a static energy lower than the one resulting from the optimization at the same cell 215 volumes but in the I41/amd space group. Therefore, the new polymorph can be described as a high-216 pressure – low symmetry zircon (HPLS ZrSiO₄), with a space group $I\overline{4}2d$ that is a direct subgroup 217 of the space group $I4_1/amd$ of zircon. After re-optimization, the structure does not show any drastic 218 structural reconstruction but subtle structural differences which break the symmetry of zircon. The 219 new phase has very similar unit-cell parameters, a = 6.399 Å, c = 5.881Å at P = 25 GPa, compared 220 to values for zircon constrained to the higher symmetry at the same pressure, of a = 6.426 Å, c =221 5.890 Å. Because the transition is driven by a soft mode and is displacive in character, the structural 222 topology of the new polymorph is the same as that of zircon, containing SiO₄ tetrahedra and ZrO₈ 223 224 dodecahedra, linked in the same way, so it is almost indistinguishable from zircon when viewed in 225 the same orientation as zircon in Figure 1. Compared to zircon, all of the atoms of the high-pressure phase occupy Wyckoff positions of lower symmetry, and the $\overline{4}$ symmetry of the Si position which is 226 $\overline{4}$ m2 in zircon allows the tetrahedra to be rotated compared to the fixed orientation in zircon. This is 227 most obvious when the two structures are viewed down the c-axis (Figure 5). The oxygen atom 228 therefore is no longer required to occupy a position on a {100} mirror plane (as in zircon) but occupies 229 an unconstrained general equivalent position within the new polymorph. Comparison of the structures 230 of zircon (constrained to I41/amd symmetry) and the relaxed structure of the new polymorph obtained 231 from DFT simulations at the same pressures above the transition show that the individual bond 232 lengths, O-Si-O angles, and O-O distances within the SiO₄ tetrahedra are identical within the 233 uncertainties in the simulations. Therefore, the mechanism of the transition is one of rotation of the 234 235 SiO_4 tetrahedra around the *c*-axis of the structure, which is the same distortion that is generated by 236 the soft B_{1u} mode within the zircon structure (Figure 5). This rotation alone, without any other distortion, leads to a reduction of the a and b cell parameters, a slight reduction in volume, and no 237 change in the *c* cell parameter. The simulated structures of the new polymorph however also show a 238 small reduction in the c cell parameter compared to zircon, which is due to small changes in the 239 distortion of the ZrO₈ dodecahedra. At 25 GPa, the new phase is 1% denser than zircon. 240

In summary, this phase transition is allowed by symmetry to be continuous in nature, and above 20 241 GPa the spontaneous volume change (the volume difference between the HPLS phase and zircon at 242 the same pressure) as well as the cosine of the tetrahedral rotation angle (Figure 5) evolve linearly 243 244 with pressure, which would be consistent with the transition being second-order in character (in the 245 Landau sense e.g. Carpenter et al. 1998). The wavenumber of the B_{1u} mode in zircon extrapolates to zero at ca. 19.5 GPa, and at 20 GPa it becomes imaginary, showing that the zircon structure becomes 246 dynamically unstable between 19.5 and 20 GPa (Figure 4). A simulation of the HPLS polymorph at 247 20 GPa confirms this; the tetrahedral rotation angle is 5.2° (clearly visible in Figure 5), and at higher 248 pressures the tetrahedral rotation angle continues to increase. The phonon frequencies of the new 249 structure under hydrostatic pressures up to 30 GPa remain positive, showing that it remains 250 dynamically stable and does not undergo any further displacive-type phase transitions in this pressure 251 range (Figure 4). A similar behavior is seen for the Raman-active mode at 40 cm⁻¹ in the high-252 symmetry phase of Pb₃(PO4)₂ under high pressure (Mihailova et al. 2015). However, linear 253 extrapolation of the volume strain of the HPLS phase to zero would suggest instead that the transition 254 255 occurs not at 19.5 GPa but at a static pressure of 16.5 - 17 GPa in our simulations, which then appears 256 to indicate that the transition has a small ΔV step of 0.4% between 19.5 and 20 GPa. This apparent discrepancy with the character of the phase transition may be due to the difficulties within DFT of 257 simulating small distortions of the structure near to the phase transition. We also note that the 258 simulations are performed at static pressures and the true character and the transition pressure may 259 be slightly different in zircon at finite temperatures. Only experimental data could resolve this 260 question. 261

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264 **Reidite structure and elasticity**

The only published experimental structure of reidite was based on a low-resolution powder diffraction 265 measurement sample recovered from a high-pressure experiment (Kusaba et al. 1986); several 266 computer simulations have subsequently confirmed the stability and structure type (Marqués et al. 267 268 2006, Smirnov et al. 2010, Dutta and Mandal 2012b, Du et al. 2012). Reidite has the scheelite 269 structure type (Kusaba et al. 1986), and contains SiO₄ tetrahedra linked only by corners to the 3dimensional framework of edge-sharing ZrO₈ dodecahedra. In our DFT simulation at static 270 equilibrium, the SiO₄ tetrahedra have significantly longer bond lengths than in zircon (1.660 Å 271 compared to 1.634 Å), and the tetrahedra are flattened along the c-axis compared to those in zircon, 272 where the elongation along the c-axis is due to the sharing of the tetrahedral edges with ZrO_8 (see 273 Figure 1b and Figure 2). 274

The simulations show that the SiO₄ tetrahedra in reidite have a bulk modulus of 436 GPa, slightly 275 stiffer than in zircon up to 17 GPa, and that the ZrO₈ polyhedra are very similar in bulk 276 compressibility to those in zircon (as for zircon, the results for reidite are from fits to static DFT 277 simulations without zero-point vibrational effects). In contrast to zircon, the *longer* of the two 278 279 symmetry-independent Zr-O bonds in reidite compresses significantly more than the shorter Zr-O 280 bond, which is a more normal behaviour, and confirms the inferred influence of the shared O-O edges between the ZrO₈ and SiO₄ groups and the consequent Zr-Si repulsion on the anisotropy of 281 compression of zircon (Hazen and Finger 1979). The direction of the elastic anisotropy is reversed in 282 reidite with respect to zircon, with the compressional modulus of the c-axis of reidite (501(1) GPa) 283 being similar to the a-axis of zircon (551(2) GPa). The bulk modulus for reidite is determined to be 284 241(5) GPa (with $K_0' = 4.7(6)$, Table 1), which is about 9% stiffer than zircon and consistent with the 285 9% density increase from zircon to reidite, as also estimated in previous DFT simulations (Marqués 286 et al. 2006, Smirnov et al. 2008 Du et al. 2012, Dutta and Mandal 2012b). The experimental bulk 287 modulus of reidite was reported to be 301.4 ± 11.5 GPa (Scott et al. 2002) and 392 GPa (Ono et al. 288 2004a), but the latter is based on only six data points without a measurement of the recovered sample. 289

The calculated wavenumbers of the Raman active modes (Table 3) are in very good agreement with the data reported by Gucsik (2007). The $\partial \omega / \partial P$ slopes are all positive and thus all of the mode frequencies remain positive in the investigated pressure range, meaning that reidite remains dynamically stable. The modes that are more sensitive to pressure are those at higher wavenumbers, which are related to the symmetric and antisymmetric stretching of the SiO₄. Moreover, the DFT simulations show that the doublet B_g (3) and E_g (3) that is present at P = 0 converges to a singlet at ca. 7 GPa.

298 Zircon-Reidite transition

The space group of reidite, $I4_1/a$, which is obviously a sub-group of the space group of zircon, 299 $I4_1/amd$, when referred to the same unit-cell and orientation led several authors to assume that reidite 300 (and scheelite structures in general) is a distorted form of the zircon structure and can be obtained by 301 a displacive-type distortion of the zircon structure (e.g. Kusaba et al. 1986, Knittle and Williams 302 1993, Liu 1979; Glass and Liu 2001). This is not correct because the orientation of the symmetry 303 304 elements with respect to the structural elements is different in the two phases. Structurally, the unit cell of reidite is rotated 45° around the *c*-axis with respect to that of zircon (Figure 1), and the 305 relationship between the unit-cell parameters is $a_r \sim a_z \sqrt{2}$. This means that the directions of 306 translation of the *a* glides in reidite and zircon are also at 45° to one another. The space group of 307 zircon, I_{4_1}/amd , does not contain *a*-glides with translations in the direction of the glides in reidite. 308 Similarly, the (very approximate) doubling of the *c*-lattice parameter on going from zircon to reidite 309 means that the elements of the 4₁ symmetry operator in reidite are not present in zircon. The 310 consequence of this symmetry relationship is that the topology of the polyhedral linkages in the 311 structures of zircon and reidite are different, and this difference is the major cause for the large density 312 and volume contrast between the two phases. Because the SiO₄ tetrahedra in zircon share two edges 313 with the ZrO₈ dodecahedra, each silicon atom has six zircon atoms in its secondary coordination 314 (Figure 2). In the room pressure simulation, two Zr are at a distance of 3.01 Å from Si across the 315 shared edges, and four at a distance of 3.65 Å. The average Si-Zr distance is 3.44 Å. In reidite, there 316

is only corner-sharing between the SiO₄ and ZrO₈ polyhedra, with each oxygen bonded to one Si, and two Zr. Each Si atom in the structure of reidite therefore has 8 Zr neighbours, four at 3.37 Å and four at 3.56 Å at room pressure. The average Si-Zr distance in reidite, 3.46A is thus almost the same as in zircon, but the density of reidite is much greater because the secondary (Si to Zr) coordination has increased from 6 to 8 (see Figure 1).

Consequently, zircon undergoes a significant structural rearrangement when it transforms to its high-322 323 pressure polymorph reidite. The transformation is reconstructive and will occur only when a substantial energy barrier is overcome, either by high-temperature or significant over-stepping of the 324 boundary in pressure, or a combination of the two. Calculation of the enthalpies H = U + PV of the 325 two polymorphs by combining the internal energy U, pressure P and volume V from the DFT 326 simulations, shows that reidite becomes energetically more stable than zircon above a pressure of 327 9.13(1) GPa (Figure 6) at 0 K. This is not dissimilar to experiments. The experimental Clapeyron 328 329 slope (Ono et al. 2004b) $\partial P/\partial T$ is zero or slightly positive, and extrapolation of the experimental results to 0 K would place the phase transition pressure between 6 and 8 GPa. 330

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IMPLICATIONS

The DFT simulations can now explain the various experimental data on zircon and its transformations 333 at high pressure. The calculated *P*-*V* curve from DFT is in good agreement with the data reported by 334 Ono et al. (2004a) on guenched synthetic reidite samples as well as the results from the in-situ high-335 pressure study of zircon by van Westrenen et al. (2004) (Figure 7). The deviation to lower volumes 336 seen above 19.7 GPa in the data of van Westrenen et al. (2004) are now seen to be the consequence 337 of the displacive phase transition to the new HPLS zircon phase that we have discovered in this DFT 338 339 study. Raman spectra of the recovered sample from this experiment unequivocally show reidite so this suggests that, due to the sluggish character of the zircon to reidite transition, at high pressures the 340 341 sample actually contained HPLS zircon plus an amount of reidite too small to be detected by 342 diffraction. This shows that compression at room temperature and above 20 GPa is sufficient to

overcome the activation barrier for the zircon to reidite transformation, and that at high pressures the
sample will consist of a mixture of the HPLS zircon phase which has partially transformed to reidite.
On pressure release the HPLS zircon will transform back to normal zircon, while the reidite may not,
so that recovered samples consist of a mixture of reidite and zircon.

347 The DFT simulations of the Raman spectra of all three polymorphs allow us to give some guidelines as to which peaks in experimental Raman spectra are diagnostic for the different phases of ZrSiO₄. 348 The simulation of zircon at high pressure (e.g. 19.5 GP) shows that peaks shifts toward higher 349 wavenumbers (Figure 8), while the mode $E_g(1)$ shifts to lower wavenumbers under pressure, as we 350 discussed in detail above. Above ca. 20 GPa zircon is no longer stable and the symmetry is broken, 351 resulting in the HPLS phase. At 20 GPa, according to the irreducible representations of point groups 352 (for the Γ point) the B_{1u} mode which is silent for the 4/*mmm* point group of zircon, turns into an A₁ 353 symmetry mode for the $\overline{4}2m$ point group symmetry of the HPLS phase, which is Raman active. This 354 mode has a high intensity even with random polarization as shown in Figure 8 and occurs at 94 cm⁻¹ 355 at P = 20 GPa. Tables S1 – S3 list the relative intensities of the Raman peaks of all three polymorphs 356 simulated for single crystals and polycrystalline powders. Therefore, the HPLS phase can be 357 distinguished from zircon in-situ in high-pressure experiments by making Raman measurements in 358 the low wavenumbers range (below 200 cm⁻¹), where the peak at 94 cm⁻¹ does not overlap with any 359 peaks of reidite or zircon. 360

361

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525

Tables

526 Table 1. Calculated zircon and reidite EoS parameters

	V ₀ (Å ³)	K₀ (GPa)	K ₀ '	a₀ (Å)	M _{a0} (GPa)	M _{a0} '	c₀ (Å)	M₀ (GPa)	M _{c0} '
zircon	266.49 (11)	220.3 (5)	4.71 (4)	6.6514 (2)	551 (2)	15.8 (2)	6.0238 (2)	1050 (1)	5.3 (2)
reidite	239.57 (6)	241 (5)	4.7 (6)	4.76078 (3)	921 (2)	10.1 (2)	10.5700 (2)	501 (1)	16.1 (2)
527									

528 Note: Parameters are for Birch-Murnaghan 3rd-order EoS, obtained by fitting the unit-cell volumes

and cell parameters from the static DFT simulations up to 25 GPa for zircon, and to 17 GPa for reidite.

	ω (cm	1 ⁻¹)	$\partial \omega / \partial P$ (cm ⁻¹ /GPa)					
	Experimental ^a	calculated	experin	nental ^b	calculated			
mode	T = 70 K	0 K P = 0	T= 298 K P= 0	up to 10 GPa	up to 10 GPa	up to 25 GPa		
A _{1g} (1)	441.5	439	439.6	1.44	1.50	1.37		
A _{1g} (2)	978	970	975.4	4.97	4.84	4.28		
B _{1g} (1)	216.5	216	215.1	1.27	1.37	1.12		
B _{1g} (2)	395	387	394.0	1.86	2.08	1.86		
B _{1g} (3)	642	636	_	_	2.05	1.82		
B _{1g} (4)	1012.5	1015	1008.6	5.54	5.23	4.61		
B _{2g}	265	250	_	_	-1.33	-1.46		
E _g (1)	202.5	197	202.6	-0.56	-0.56	-0.81		
E _g (2)	225.5	225	224.5	0.30	0.05	0.07		
E _g (3)	360	341	357.9	4.27	4.57	3.92		
E _g (4)	547	542	_	_	0.71	0.63		
E _g (5)	927	922	_	_	4.89	4.32		

Table 2. Wavenumbers (ω) and their pressure derivatives ($\partial \omega / \partial P$) of Raman-active modes of zircon

^aSyme et al. 1977

⁵³³ ^bPinvignat et al. 2018

Table 3. Wavenumbers (ω) and their pressure derivatives ($\partial \omega \partial P$) of Raman-active modes of reidite

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mode	Experimental ^a	calculated	
-	Т=298 К	Т=0 К	<i>∂ω/∂P</i>
Ag (1)	327	326	1.05
Ag (2)	406	409	0.53
Ag (3)	-	861	1.29
Bg (1)	238	242	0.71
Bg (2)	353	350	2.70
Bg (3)	464	465	1.80
Bg (4)	610	608	2.19
Bg (5)	847	852	1.64
Eg (1)	205	209	2.05
Eg (2)	297	300	1.90
Eg (3)	-	458	4.18
Eg (4)	558	558	3.42
Eg (5)	887	891	4.51

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Notes: ^a Gucsik 2007. Calculated slopes are from simulations up to a maximum pressure of 17 GPa

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Figures

- 543 Figure 1. Polyhedral representations of the crystal structures of (a) zircon and (b) reidite drawn with
- 544 CrystalMaker® (2018). The shared O-O edges between the ZrO₈ and the SiO₄ polyhedra in zircon
- 545 are highlighted in red.



Figure 2. Interatomic distances (Å) and angles of zircon and reidite in the simulated structures at zero
pressure. Only one-half of each SiO₄ tetrahedron is shown for each polymorph. Note that in zircon
two oxygens are shared by one Si and one Zr and form a shared polyhedral edge.



Figure 3. Calculated (lines) and experimental Raman shifts of zircon (Pina-Binvignat et al., 2018, symbols) as a function of pressure. Some of the calculated Raman active modes are not observed in experiments because of their low intensities. The inset shows the softening of the B_{2g} and $E_g(1)$ modes with pressure.



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Figure 4. Variation of the wavenumber of the silent (i.e. neither Raman nor IR active) B_{1u} mode with pressure. Up to ca. 20 GPa the mode frequency is positive, but softens with pressure, and then becomes imaginary (plotted here as negative). In HPLS ZrSiO₄ this mode becomes an A₁ mode (circles).



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Figure 5. Variation with pressure of the cosine of the tetrahedral rotation angle, with polyhedral representations of (left) zircon and (right) HPLS zircon viewed down their c-axes. The tetrahedra are not rotated in zircon but become rotated in the HPLS phase.

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- 568 Figure 6. Variation of the enthalpy change ΔH for the zircon to reidite transition with pressure
- calculated from DFT at 0 K, showing that the transition pressure is 9.13(1) GPa.



570

Figure 7. *P-V* curves of the three polymorphs calculated by DFT at 0 K (lines) and experimental data
at 300 K (red symbols Ono et al., 2004a; black symbols Van Westrenen, 2004). The deflection above
19 GPa in the experimental data for zircon is in agreement with the smaller volume of the HPLS
phase.



Figure 8. Calculated Raman spectra of the three polymorphs: the spectra are simulated for a polycrystalline powder without any polarization. The Raman intensities are calculated at 298 K and with a wavelength excitation of 532 nm. Calculated intensities for oriented single crystals in different scattering geometries are given in the supplementary materials Tables S1, S2 and S3.

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