Neutron diffraction in gemology: Single-crystal diffraction study of brazilianite, NaAl₃(PO₄)₂(OH)₄

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ABSTRACT

The chemical composition and the crystal structure of a gem-quality brazilianite from the Telírio pegmatite, near Linópolis, Minas Gerais, Brazil, $[NaAl_3(PO_4), (OH)_4, a = 11.2448(5) \text{ Å}, b = 10.1539(6)$ Å, c = 7.1031(3) Å, $\beta = 97.351(4)^{\circ}$, V = 804.36(7) Å³, space group $P2_1/n$, Z = 4], have been reinvestigated by means of electron microprobe analysis in wavelength-dispersive mode, single-crystal X-ray and neutron diffraction. The chemical analysis shows that brazilianite from Telírio Claim approaches almost ideal composition. The neutron anisotropic structural refinement was performed with final agreement index $R_1 = 0.0290$ for 211 refined parameters and 2844 unique reflections with $F_0 > 4\sigma(F_0)$. the X-ray refinement led to $R_1 = 0.0325$ for 169 refined parameters and 2430 unique reflections with $F_0 > 4\sigma(F_0)$. The building-block units of the brazilianite structure consist of chains of edge-sharing AlO₄(OH)₂ and AlO₃(OH)₃ octahedra. Chains are connected, via corner-sharing, by P-tetrahedra to form a three-dimensional framework, with Na atoms located in distorted cavities running along [100]. Five independent H sites were located, here labeled as H(1), H(2a), H(2b), H(3), and H(4). The configuration of the OH groups, along with the complex hydrogen-bonding scheme, are now well defined. The O-H distances corrected for "riding motion" range between ~0.992 and ~1.010 Å, the O…O distances between ~2.67 and ~2.93 Å, and the O-H···O angles between ~151° and ~174°. The H(2a) and H(2b) are only ~1.37 Å apart and mutually exclusive (both with site occupancy factor of 50%). The differences between the crystal structure of brazilianite and wardite [ideally NaAl₃(PO₄)₂(OH)₄·2H₂O] are discussed. This work fulfills the need for accurate crystal-chemical data for this gem mineral.

Keywords: Brazilianite, crystal chemistry, single-crystal neutron diffraction, hydrogen bonding