

Cranswickite $\text{MgSO}_4 \cdot 4\text{H}_2\text{O}$, a new mineral from Calingasta, Argentina

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ABSTRACT

Cranswickite is a newly recognized mineral of composition $\text{MgSO}_4 \cdot 4\text{H}_2\text{O}$ from Calingasta, San Juan Province, Argentina (IMA2010-016). Cranswickite is monoclinic, space group $C2/c$, $a = 11.9236(3)$, $b = 5.1736(1)$, $c = 12.1958(3)$ Å, $\beta = 117.548(2)^\circ$, $V = 667.0(1)$ Å³, $Z = 4$, $d_{\text{obs}} = 1.917$ g/cm³, and $d_{\text{calc}} = 1.918$ g/cm³. The mineral occurs as a soft white vein filling in a metasedimentary rock. The atomic structure has been determined by direct methods and refined by Rietveld analysis of powder diffraction data. The atomic structure consists of chains of corner-sharing magnesium-containing octahedra and sulfate tetrahedra similar to the structure of pentahydrate. All the water molecules directly coordinate magnesium in the structure. The five strongest lines in the powder X-ray diffraction data are [d_{obs} in angstroms (I) (hkl): 5.259 (100) (200), 3.927 (46) (11 $\bar{2}$), 3.168 (45) (11 $\bar{3}$), 4.603 (29) (11 $\bar{1}$), 2.570 (23) (311)]. Infrared and Raman spectra are very similar to the spectra measured from starkeyite. The chemical composition of cranswickite is the same as starkeyite $\text{MgSO}_4 \cdot 4\text{H}_2\text{O}$, but starkeyite has an atomic structure where two sulfate tetrahedra and two $\text{Mg}(\text{H}_2\text{O})_6$ octahedra share corners to form a four-membered ring and not a chain as in cranswickite. The new mineral is named in honor of Lachlan M.D. Cranswick (1968–2010), an Australian crystallographer who helped to develop and maintain the Collaborative Computational Project No. 14 in Powder and Small Molecule Single Crystal Diffraction (CCP14).

Keywords: Cranswickite, starkeyite, pentahydrate, epsomite, meridianiite, chalcantite, powder diffraction, atomic structure, magnesium sulfate, Argentina