

Incorporation of water in iron-free ringwoodite: A first-principles study

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ABSTRACT

The structures, infrared active OH stretching modes, and relative energies of OH-defects in ringwoodite (γ -Mg₂SiO₄) have been studied by first-principles calculations based on density functional theory (DFT). Two types of fully protonated cationic defects in normal spinel were considered at 0 and 20 GPa, i.e., $[V_{\text{Mg}}(\text{OH})_2]^\times$, $[V_{\text{Si}}(\text{OH})_4]^\times$ defects. In addition, two defects associated with the partial inversion of the spinel structure have been investigated. The first one corresponds to two protons compensating a Mg substituted for Si in tetrahedral site, $[\text{Mg}_{\text{Si}}(\text{OH})_2]^\times$, whereas the second defect corresponds to a Mg vacancy located nearby a Mg-Si substitution, $[V_{\text{Mg}}(\text{OH})_2\text{Mg}_{\text{Si}}\text{Si}_{\text{Mg}}]^\times$. The infrared spectrum and evolution with pressure of these OH-defects make it possible to interpret the major IR absorption bands experimentally observed. The main absorption band at $\sim 3150\text{ cm}^{-1}$ corresponds to protons located between the O-O pairs shared by 16c and 16d octahedra, instead of OH along the tetrahedral edges as usually proposed in the literature. The large width of this band is most likely related to the association of OH defects with the various cationic configurations related to the partial inversion of a vacancy-bearing spinel structure. The less intense band at $\sim 3675\text{ cm}^{-1}$ is assigned to hydrogarnet-type defects with a protonation of the tetrahedral edges. This interpretation is consistent with an Mg/Si ratio lower than 2 and its weak variation as a function of water concentration, as experimentally observed. These results emphasize the importance of taking into account the structural relaxation experienced by defects, instead of using empirical correlation, to assign OH stretching bands to specific O-O pairs of the structure.

Keywords: Ringwoodite, Mg₂SiO₄ spinel, hydrogen, DFT, infrared spectroscopy