

LETTER

**An isosymmetric phase transition of orthopyroxene found by high-temperature X-ray diffraction**

**SHUGO OHI,<sup>1,\*</sup> AKIRA MIYAKE,<sup>1</sup> NORIMASA SHIMOBAYASHI,<sup>1</sup> MASATOMO YASHIMA,<sup>2</sup> AND MASAO KITAMURA<sup>1</sup>**

<sup>1</sup>Department of Geology and Mineralogy, Division of Earth and Planetary Sciences, Graduate School of Science, Kyoto University, Kyoto 606-8502, Japan

<sup>2</sup>Department of Materials Science and Engineering, Interdisciplinary Graduate School of Science and Engineering, Tokyo Institute of Technology, Yokohama 226-8502, Japan

ABSTRACT

High-temperature synchrotron X-ray powder diffraction experiments for the composition of  $(\text{Ca}_{0.06}\text{Mg}_{1.94})\text{Si}_2\text{O}_6$  have been carried out in the present study to clarify whether orthopyroxene has a transition between low- and high-temperature phases. Our results show that discontinuous changes of unit-cell dimensions and volume occur at 1170 °C during both heating and cooling processes and that the space group of *Pbca* does not change during this reversible phase transition. These facts indicate a first-order and isosymmetric phase transition. This high-temperature phase is thermodynamically distinct from the low-temperature phase, i.e., orthoenstatite in the Mg-rich portion of  $\text{Mg}_2\text{Si}_2\text{O}_6$ - $\text{CaMgSi}_2\text{O}_6$  phase diagram, although they have the same space group.

**Keywords:** Orthopyroxene, isosymmetric phase transition, in-situ X-ray experiments, enstatite-diopside system