

Stability of Al-bearing superhydrous phase B at the mantle transition zone and the uppermost lower mantle

SHO KAKIZAWA^{1,*}, TORU INOUE^{1,2,3}, HIDETO NAKANO¹, MINAMI KURODA⁴, NAOYA SAKAMOTO⁵, AND HISAYOSHI YURIMOTO^{4,5,6}

¹Geodynamics Research Center, Ehime University, Matsuyama, Ehime 790-5877, Japan

²Department of Earth and Planetary Systems Science, Hiroshima University, Higashi-Hiroshima, Hiroshima 739-8526, Japan

³Hiroshima Institute of Plate Convergence Region Research (HiPeR), Hiroshima University, Higashi-Hiroshima, Hiroshima 739-8526, Japan

⁴Department of Natural History Sciences, Hokkaido University, Sapporo, Hokkaido 060-0810, Japan

⁵Isotope Imaging Laboratory, Hokkaido University, Sapporo, Hokkaido 001-0021, Japan

⁶Institute of Space and Astronautical Science, JAXA, Sagami-hara, Kanagawa 252-5120, Japan

ABSTRACT

We determined the stability and chemical composition of Al-bearing superhydrous phase B at 20–24 GPa and 1400–2000 °C to discuss the mechanism of water transport in the mantle transition zone and uppermost lower mantle at temperatures close to the mantle geotherm. Superhydrous phase B contained significant amounts of Al₂O₃, from 14 to 32 wt%, and Al-bearing superhydrous phase B remained stable, even at 2000 °C and pressures of approximately 20–24 GPa. Moreover, two types of superhydrous phase B with different chemical compositions coexisted at 20–24 GPa and 1600 °C. The Al₂O₃ and H₂O contents increased, and the MgO and SiO₂ contents decreased as the pressure and temperature increased up to 1600 °C. Above 1600 °C, the MgO and Al₂O₃ contents increased, and the SiO₂ and H₂O contents decreased as the temperature increased. We found two substitution mechanisms: (1) $2\text{Mg}^{2+} + \text{Si}^{4+} \rightleftharpoons 2\text{Al}^{3+} + 2\text{H}^+ + \square_{\text{Mg}}$ (Mg site vacancy) (2) $2\text{Mg}^{2+} = \text{Al}^{3+} + \text{H}^+ + \square_{\text{Mg}}$: $(\text{Si}^{4+} = \text{Al}^{3+} + \text{H}^+) = 1:1$, (2) $\text{Si}^{4+} + 16\text{H}^+ \rightleftharpoons 4\text{Mg}^{2+} + 4\text{Al}^{3+}$. The maximum H₂O content of Al-bearing superhydrous phase B is 11.1(3) wt%, which is ~1.9 times larger than that of the Mg-end-member. The crystal structures of the two coexisting superhydrous phase B values are expected to be slightly different from each other. The present results indicate that Al-bearing superhydrous phase B can be stable in a subducted slab with a high Al content compared to pyrolite (e.g., chlorite) at temperatures typical of the mantle transition zone and the lower mantle. Thus, water can be transported to the lower mantle by Al-bearing superhydrous phase B in the subducting slab, even at the typical mantle geotherm.

Keywords: Superhydrous phase B, dense hydrous magnesium silicates, stability, substitution mechanism, hydrous mineral; Physics and Chemistry of Earth's Deep Mantle and Core