In situ structure determination of the high-pressure phase of Fe₃O₄

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ABSTRACT

The crystal structure of a high-pressure Fe₃O₄ phase was determined by in situ X-ray diffraction measurements at high pressure and temperature, using an imaging plate detector and monochromatic synchrotron X-radiation. The high-pressure phase has the $Pbcm$ space group (CaMn₂O₄-type structure) with cell parameters $a = 2.7992(3)$ Å, $b = 9.4097(15)$ Å, and $c = 9.4832(9)$ Å at 23.96 GPa and 823 K. Fe³⁺ occupies an octahedral site and Fe²⁺ is in an eightfold-coordinated site described as a bicapped trigonal prism. The high-pressure CaMn₂O₄-type Fe₃O₄ phase is about 6.5% more dense than the spinel form at 24 GPa.

INTRODUCTION

Magnetite (Fe₃O₄) is a mixed valence iron oxide with some Fe³⁺ ions occupying the tetrahedral sites and the Fe²⁺ and the remaining Fe³⁺ ions occupying the octahedral sites in the spinel structure (space group $Fd\overline{3}m$ (e.g., Fleet 1981). Its high-pressure behavior has been extensively studied by X-ray diffraction (Mao et al. 1974; Huang and Bassett 1986; Nakagiri et al. 1986; Finger et al. 1986; Pasternak et al. 1994), Mössbauer spectroscopy (Mao et al. 1977; Pasternak et al. 1994), and electrical resistivity measurements (Samara 1968; Ramasesha et al. 1994; Rozenberg et al. 1996; Morris and Williams 1997) because of its geophysical importance and its interesting magnetic properties at high pressure. Mao et al. (1974) demonstrated that magnetite transforms to a high-pressure phase at about 25 GPa. The X-ray diffraction pattern for the high-pressure phase, collected using a diamond cell diffraction camera (Bassett et al. 1967) and conventional X-ray source, was indexed on a monoclinic cell. Because relatively few diffraction lines were observed, the structure assignment was tentative. The high-pressure phase transition was confirmed by Huang and Bassett (1986) who determined its temperature dependence. The structural transition is also accompanied with changes in magnetic properties (Mao et al. 1977; Pasternak et al. 1994) and electrical resistivity (Morris and Williams 1997). Without knowledge of the crystal structure of the high-pressure phase, ambiguity exists in interpreting the Mössbauer spectra of Fe₃O₄ at high pressures.

X-ray diffraction data for the high-pressure phase of Fe₃O₄ collected in previous studies suffer from either low resolution or broad diffraction peaks that sometimes overlap those of magnetite due to the sluggish transition at room temperature. This study presents new diffraction data for the high-pressure phase collected on an imaging plate using monochromatic synchrotron X-ray radiation. The diffraction data contain accurate structural information with high resolution. The high quality data allow us to solve the crystal structure of the high-pressure phase of Fe₃O₄.

EXPERIMENTAL PROCEDURE

The starting material used in this study is synthetic magnetite with enriched $^{57}$Fe, the same starting material used in the Mössbauer spectroscopic study by Mao et al. (1977). High-pressure–high-temperature experiments were conducted using an externally heated high-temperature diamond-anvil cell. The high-temperature cell is capable of achieving pressures greater than 125 GPa at temperatures up to 900 °C (Fei and Mao 1994; Fei 1996). A detailed description of this cell was given by Fei (1996). Here, we used 500 μm flat diamond anvils. The powdered magnetite sample was compacted in a sample chamber, 200 μm in diameter by 49 μm in thickness, drilled in a preindented rhenium gasket. A layer of NaCl powder was placed on the top of the magnetite sample to serve as a pressure-transmitting medium (especially at high temperatures) and as a pressure calibrant. A small piece (<10 μm) of thin gold foil was placed in only one quarter of the sample chamber as a pressure calibrant to obtain diffraction data with or without gold diffraction peaks. One small ruby grain (~5 μm) was also placed in the sample chamber for initial pressure determination at room temperature. At high T, sample temperature was measured by a Pt/Pt-10%Rh thermocouple placed near the sample chamber, whereas pressure was determined by measuring the lattice parameters of an internal standard (Au or NaCl), based on its $P-V-T$ equation of state (Anderson et al. 1989; Birch 1978).

We initially compressed the sample to $P = 34.45$ GPa at room temperature, based on the ruby pressure scale (Mao et al. 1978). At this pressure, magnetite should have transformed to the high-pressure phase completely (Huang and Bassett 1986). We then took the diamond cell to the European Synchrotron Radiation Facility (ESRF) for in situ X-ray diffraction measurements. The experimental setup at ESRF (beamline ID30) was described by

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Häusermann and Hanfland (1996). A monochromatic beam (Si 111 monochromator) with wavelength of 0.4253 Å was used in the experiments. The beam was focused down to a <15 μm spot by two multilayer mirrors. The diffraction data were recorded on an imaging plate and read with a scanner. The details of the imaging plate technique have been described by Amemiya (1995) and Nelmes and McMahon (1994). The sample-plate distance was 495.62 mm, determined by measuring the lattice parameter of gold at a known pressure determined from ruby fluorescence at room temperature. The two-dimensional imaging plate data were reduced to an intensity vs. 2θ plot using the FIT2D program (Hammersley et al. 1996).

RESULTS AND DISCUSSION
In situ X-ray diffraction measurements of Fe₃O₄ were made at 34.45 GPa and 300 K, 26.41 GPa and 723 K, 23.96 GPa and 823 K, and 9.04 GPa and 923 K. Figure 1 shows the pressure-temperature path of our experiments and Figure 2 shows the X-ray diffraction patterns. From the phase boundary determined by Huang and Bassett (1986), three of our experiments (Fig. 1) are within the stability field of the high-pressure phase of Fe₃O₄. The diffraction data corresponding to these conditions confirm the existence of the high-pressure phase with a rather complicated diffraction pattern (Fig. 2). The diffraction peaks at room temperature are relatively broad. With increasing temperature, the peaks sharpen because of completion of the transformation and decrease of the deviatoric stress in the sample chamber. The broad peak at 2θ = 10° in the 300 K pattern, indicative of the low-pressure phase magnetite, disappears at about 623 K, indicating complete conversion of magnetite to the high-pressure phase. The strongest peak at 2θ = 9.5° was assigned as a single peak in the early study by Mao et al. (1974) but in fact is a triplet as evident in the diffraction patterns at 823 K and 23.96 GPa (Fig. 2). At 923 K, the sample pressure was decreased to 9.04 GPa, and the observed diffraction pattern was indexed on a cubic cell of magnetite (Fig. 2). The refined unit-cell parameters at 923 K and 9.04 GPa for magnetite and NaCl (B1 structure) are 8.3348(1) Å and 5.2954(2) Å, respectively. The transformation from the high-pressure phase to magnetite with decreasing pressure at high temperature (Fig. 1) is consistent with the phase boundary determined by Huang and Bassett (1986). The high-pressure Fe₃O₄ phase is non-quenchable, and its structure can be determined only by in situ X-ray diffraction measurements at high pressure and temperature.

The imaging plate data collected at ESRF are of high resolution and contain accurate intensity information. Its resolution is comparable to that of a conventional powder diffractometer, Δd/d of 0.001 at 2 Å, and about an order of magnitude higher than that of energy-dispersive technique, Δd/d of 0.01 at 2 Å. High angular resolution can be further achieved by increasing the sample-plate distance. The high-quality data are good enough for Rietveld structure refinement, providing the space group of the phase is known (Hammersley et al. 1996). The observed diffraction data for the high-pressure phase of Fe₃O₄ cannot be indexed with the monoclinic cell proposed by Mao et al. (1974). Using systematics of high-pressure post-spinel transformations in materials such as MgAl₂O₄ (Irifune et al. 1991) and CaAl₂O₄ (Reid and Ringwood 1969), we initially tried to index the observed pattern on an orthorhombic cell of the CaFe₂O₄-type structure. All the observed d-spacings can indeed be explained by an orthorhombic cell, but the observed intensities are not consistent with the space group Pnam (CaFe₂O₄-type structure). Further structure refinements indicate that the high-pressure phase of Fe₃O₄ has the space group Pbcm (CaMn₂O₄-type structure) with cell parameters, a = 2.799(3) Å, b = 9.4097(15) Å, and c = 9.4832(9) Å. Figure 3 shows the Rietveld
refinement results for the orthorhombic high-pressure phase of FeO₄ using the program package GSAS (Larson and Von Dreele 1986). Both B1 and B2 structures of NaCl were observed at this pressure-temperature condition. The refined unit-cell parameters for B1 and B2 are 4.9696(9) Å and 3.0879(8) Å, respectively. The structure refinement parameters for the high-pressure phase of FeO₄ are listed in Table 1. The refinement yielded a good agreement between the observed and calculated X-ray diffraction patterns.

The CaMnO₄-type structure, closely related to the CaFeO₄-type structure, is one of the densest AB₂O₄ structures. The trivalent ions occupy the octahedral sites while the divalent ions occupy the eightfold-coordinated site. Reid and Ringwood (1969) observed that MnO₂ and CaAl₂O₄ transform to the CaMnO₄-type and the CaFeO₄-type structure, respectively, at about 10 GPa. Recent high-pressure experiments showed that MgAl₂O₄ spinel also transforms to the CaFeO₄-type structure at about 25 GPa (Irfune et al. 1991). It is not surprising that FeO₄ takes the CaMnO₄-type structure at high pressure, forming a dense high-pressure phase. The high-pressure CaMnO₄-type FeO₄ is about 6.5% more dense than the spinel form at 24 GPa. The structure refinements indicated that the octahedral and the eightfold-coordinated sites, occupied by the Fe³⁺ and the Fe²⁺ ions, respectively, are rather distorted. The Fe³⁺-O²⁻ bond lengths for the octahedral sites and the Fe³⁺-O²⁻ bond lengths for the eightfold-coordinated sites at 24 GPa range from 1.715 to 2.589 Å and from 1.775 to 2.719 Å, respectively (Table 1). The average Fe³⁺-O²⁻ bond length for the sixfold-coordinated site at 24 GPa is about 1.99 Å, slightly smaller than the 1-atm value of 2.06 Å because of the pressure effect. The shortest Fe³⁺-O²⁻ bond length for the eightfold-coordinated site is about 1.775 Å, which appears to be unusually small. However, the average Fe³⁺-O²⁻ bond length (2.288 Å) is consistent with the expected value for the eightfold-coordinated site at high pressure (cf.

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**Table 1.** Refined unit-cell and atomic positional parameters and selected interatomic distances (Å) for the high-pressure phase of FeO₄, at 23.96 GPa and 823 K.

**Notes:** The final discrepancy indices (Larson and Von Dreele 1986) are Rwp = 0.019, Rp = 0.013, R(F) = 0.17, and reduced χ² = 9.4. Space group: Pbcm; Z = 4, a = 2.7992(3) Å, b = 9.4097(15) Å, and c = 9.4832(9) Å.

**Figure 3.** Observed (crosses) and calculated (solid line) X-ray diffraction pattern for the orthorhombic high-pressure phase of FeO₄ at 23.96 GPa and 823 K. The experimental data were collected with wavelength of 0.4253 Å. Tick marks for NaCl (B2), NaCl (B1), and the high-pressure FeO₄ phase are shown in descending order below the pattern. The difference curve is shown at the bottom. The refinement is based on the space group Pbcm with cell parameters, a = 2.7992(3) Å, b = 9.4097(15) Å, and c = 9.4832(9) Å.

The structure determination of the high-pressure FeO₄ phase will allow us to understand the magnetic and electrical properties of FeO₄ at high pressures. The observed two quadrupole doublets for the high-pressure FeO₄ phase by Mössbauer measurements (Mao et al. 1977; Pasternak et al. 1994) are consistent with the current structure assignment in which the Fe³⁺ and the Fe²⁺ ions occupy two distinct crystallographic sites. The orthorhombic high-pressure phase is not magnetically ordered on the basis of Mössbauer measurements (Mao et al. 1977; Pasternak et al. 1994) whereas magnetite is the best known example of ferrimagnetic materials. The electrical resistivity measurements of FeO₄ at high pressures (Morris and Williams 1997) showed that the high-pressure phase is not metallic, consistent with the atomic bond distances determined here. Thus the magnetic transition in FeO₄, corresponding with the structural transformation, is best described by the change from the ferrimagnetic to the paramagnetic state (Mao et al. 1977).

**Acknowledgments**

This research was supported by the National Science Foundation Center for High Pressure Research and the Carnegie Institution of Washington. We acknowledge ESRF for providing us with beam time. We thank L.W. Finger for helpful discussions, W.A. Bassett and R.T. Downs for useful comments, and M. Kunz for experiment assistance.

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MANUSCRIPT RECEIVED AUGUST 7, 1998
MANUSCRIPT ACCEPTED OCTOBER 2, 1998
PAPER HANDLED BY JAMES W. DOWNS