

## Rapidcreekite in the sulfuric acid weathering environment of Diana Cave, Romania

BOGDAN P. ONAC,<sup>1,2,\*</sup> HERTA S. EFFENBERGER,<sup>3</sup> JONATHAN G. WYNN,<sup>1</sup> AND IOAN POVARĂ<sup>4</sup>

<sup>1</sup>Department of Geology, University of South Florida, 4202 E. Fowler Avenue, SCA 528, Tampa, Florida 33620, U.S.A.

<sup>2</sup>Emil Racoviță Institute of Speleology/Department of Geology, Babeş-Bolyai University, Kogălniceanu 1, 400084, Cluj-Napoca, Romania

<sup>3</sup>Institut für Mineralogie und Kristallographie, Universität Wien, Althanstrasse 14, A-1090 Wien, Austria

<sup>4</sup>Emil Racoviță Institute of Speleology, Frumoasă 31, 010986 Bucharest, Romania

### ABSTRACT

The Diana Cave in SW Romania develops along a fault line and hosts a spring of hot ( $T_{\text{avg}} = 51\text{ }^{\circ}\text{C}$ ), sulfate-rich, sodium-calcium-chloride bearing water of near-neutral pH. Abundant steam and  $\text{H}_2\text{S}$  rises from the thermal water to condensate on the walls and ceiling of the cave. The sulfuric acid produced by  $\text{H}_2\text{S}$  oxidation/hydrolysis causes a strong acid-sulfate weathering of the cave bedrock generating a sulfate-dominated mineral assemblage that includes rapidcreekite,  $\text{Ca}_2(\text{SO}_4)(\text{CO}_3)\cdot 4\text{H}_2\text{O}$  closely associated with gypsum and halotrichite group minerals. Rapidcreekite forms bundles of colorless tabular orthorhombic crystals elongated along [001] and reaching up to 1.5 mm in length. For verifying the hydrogen bond scheme and obtaining crystal-chemical details of the carbonate group a single-crystal structure refinement of rapidcreekite was performed. Its unit-cell parameters are:  $a = 15.524(2)$ ,  $b = 19.218(3)$ ,  $c = 6.161(1)$  Å;  $V = 1838.1(5)$  Å<sup>3</sup>,  $Z = 8$ , space group *Pcnb*. Chemical composition (wt%): CaO 35.65, SO<sub>3</sub> 24.97, CO<sub>2</sub> 13.7, H<sub>2</sub>O 23.9, Na<sub>2</sub>O 0.291, MgO 0.173, Al<sub>2</sub>O<sub>3</sub> 0.07, total 98.75%. The empirical formula, based on 7 non-water O atoms pfu, is:  $\text{Ca}_{1.98}\text{Na}_{0.029}\text{Mg}_{0.013}\text{Al}_{0.004}(\text{S}_{0.971}\text{O}_4)(\text{C}_{0.97}\text{O}_3)\cdot 4.13\text{H}_2\text{O}$ . The  $\delta^{34}\text{S}$  and  $\delta^{18}\text{O}$  values of rapidcreekite and other cave sulfates range from 18 to 19.5‰ CDT and from -9.7 to 7.8‰ SMOW, respectively, indicating that the source of sulfur is a marine evaporite and that during hydration of the minerals it has been an abundant <sup>18</sup>O exchange with percolating water but almost no oxygen is derived from O<sub>2(aq)</sub>. This is the first description of rapidcreekite from a cave environment and one of the very few natural occurrences worldwide. We also report on the mineral stability and solubility, parameters considered critical to understand the co-precipitation of carbonates and sulfates, a process that has wide applications in cement industry and scaling prevention.

**Keywords:** Rapidcreekite, acid-sulfate weathering, hydrogen bond scheme, carbonate group,  $\delta^{34}\text{S}$ - $\delta^{18}\text{O}$  values, Diana Cave, Romania