

Crystal chemistry of trioctahedral micas- $2M_1$ from Bunyaruguru kamafugite (southwest Uganda)

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ABSTRACT

The crystal chemistry of $2M_1$ micas from Bunyaruguru kamafugite (southwest Uganda) was studied by electron probe microanalysis, single-crystal X-ray diffraction, Mössbauer and Fourier transform infrared spectroscopy. Chemical analyses showed that the studied crystals are Ti-rich, F-poor phlogopites with an annitic component, $Fe_{tot}/(Fe_{tot} + Mg)$, ranging from 0.15 to 0.22. Unit-cell parameters from single-crystal X-ray data are in the range: $5.3252(1) \leq a \leq 5.3307(1)$, $9.2231(3) \leq b \leq 9.2315(3)$, $20.1550(6) \leq c \leq 20.1964(8)$ Å, and $94.994(2) \leq \beta \leq 95.131(2)^\circ$.

Anisotropic structure refinements, in the space group $C2/c$, converged to $2.77 \leq R_1 \leq 3.52\%$ and $2.91 \leq wR_2 \leq 4.02\%$. Mössbauer spectroscopy showed that the studied sample has: $^{VI}Fe^{2+} = 60(1)\%$, $^{VI}Fe^{3+} = 24(1)\%$, and $^{IV}Fe^{3+} = 16(1)\%$. FTIR investigations pointed to the occurrence of Fe^{3+} -oxy substitutions and ruled out the presence of vacancy mechanisms. The overall crystal-chemical features are consistent with the following substitutions: tetraferriphlogopite [$^{IV}Fe^{3+} \leftrightarrow ^{IV}Al$]; Ti-oxy [$^{VI}M^{2+} + 2(OH)^- \leftrightarrow ^{VI}Ti^{4+} + 2(O^{2-}) + H_2\uparrow$] and Al, Fe^{3+} , Cr-oxy [$^{VI}M^{2+} + (OH)^- \leftrightarrow ^{VI}M^{3+} + O^{2-} + \frac{1}{2}(H_2)\uparrow$]; Al, Fe^{3+} -Tschermak [$^{VI}M^{2+} + ^{IV}Si^{4+} \leftrightarrow ^{VI}M^{3+} + ^{IV}Al$]; kinoshitalite [$^{XII}K + ^{IV}Si^{4+} \leftrightarrow ^{XII}Ba^{2+} + ^{IV}Al$] and [$^{XII}K^+ + ^{IV}Al^{3+} \leftrightarrow ^{IV}Si^{4+} + ^{XII}\square$].

The estimation of the OH^- content for Ugandan mica- $2M_1$ was obtained, for the first time, from the linear regression equation $c = 0.20(2) \times OH^- (\text{gpfu}) + 19.93(2)$ derived from literature data of $2M_1$ -samples with known OH^- content. The orientation of the O-H vector with respect to c^* was found in the range from 2.0 to 6.9°.

Keywords: Kamafugitic $2M_1$ -phlogopites, crystal chemistry, substitution mechanisms, Mössbauer, FTIR