## Redetermination of high-temperature heat capacity of Mg<sub>2</sub>SiO<sub>4</sub> ringwoodite: Measurement and lattice vibrational model calculation

## HIROSHI KOJITANI,<sup>1,\*</sup> MADOKA OOHATA,<sup>1</sup> TORU INOUE,<sup>2</sup> AND MASAKI AKAOGI<sup>1</sup>

<sup>1</sup>Department of Chemistry, Faculty of Science, Gakushuin University, 1-5-1 Mejiro, Toshima-ku, Tokyo 171-8588, Japan <sup>2</sup>Geodynamics Research Center, Ehime University, 2-5 Bunkyo-cho, Matsuyama 790-8577, Japan

## ABSTRACT

Isobaric heat capacities ( $C_P$ ) of Mg<sub>2</sub>SiO<sub>4</sub> forsterite and ringwoodite were measured by differential scanning calorimetry in the temperature range of 306–833 K. The measured  $C_P$  of Mg<sub>2</sub>SiO<sub>4</sub> forsterite was consistent with those reported by previous studies. On the other hand, the present  $C_P$  of Mg<sub>2</sub>SiO<sub>4</sub> ringwoodite was about 3–5% larger than those measured by previous researchers. The calorimetric data of Mg<sub>2</sub>SiO<sub>4</sub> ringwoodite were extrapolated to 2500 K using a lattice vibrational model calculation, which well reproduced the low-temperature  $C_P$  data measured by thermal relaxation method. The calculated  $C_P$  shows good agreement with the present calorimetric data. The obtained  $C_P$  was expressed by the polynomial of temperature:  $C_P = 164.30 + 1.0216 \times 10^{-2}T + 7.6665 \times 10^{3}T^{-1} - 1.1595 \times 10^{7}T^{-2} + 1.3807 \times 10^{9}T^{-3}$  [J/(mol·K)] in the range of 250–2500 K.

**Keywords:** Mg<sub>2</sub>SiO<sub>4</sub>, ringwoodite, heat capacity, DSC, Kieffer model calculation, thermodynamic property, mantle transition zone