

The crystal structure of jonesite, $\text{Ba}_2(\text{K},\text{Na})[\text{Ti}_2(\text{Si}_5\text{Al})\text{O}_{18}(\text{H}_2\text{O})](\text{H}_2\text{O})_n$: A first example of titanosilicate with porous double layers

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ABSTRACT

The crystal structure of jonesite, $\text{Ba}_2(\text{K},\text{Na})[\text{Ti}_2(\text{Si}_5\text{Al})\text{O}_{18}(\text{H}_2\text{O})](\text{H}_2\text{O})_n$ (monoclinic, $P2_1/m$, $a = 10.618(2)$, $b = 25.918(4)$, $c = 8.6945(14)$ Å, $\beta = 127.633(3)^\circ$, $V = 1894.8(6)$ Å³, $Z = 4$) from the Benitoite Gem Mine, San Benito County, California has been solved by direct methods from a crystal twinned on (001) and refined to $R_1 = 0.045$ ($wR = 0.119$, $S = 1.028$) using 3308 unique observed reflections ($|F_o| \geq 4\sigma_f$). The structure is based upon porous double layers of distorted $\text{Ti}\Phi_6$ octahedra ($\Phi = \text{O}, \text{H}_2\text{O}$) and TO_4 tetrahedra ($\text{T} = \text{Si}, \text{Al}$) parallel to (010). The layers consist of two sheets of corner-sharing $\text{Ti}\Phi_6$ octahedra and Si_2O_7 groups each. The two adjacent sheets are linked along b by T_4O_{12} tetrahedral rings that are disordered over two positions. The double layer has an open structure characterized by eight-membered tetrahedral rings with apertures (free diameters) of 3.37×3.37 and 3.33×3.33 Å², for two symmetrically non-equivalent rings. K^+ cations and H_2O molecules are located in the pores of the double layers. Ba^{2+} cations are between the double layers and provide their linkage into three-dimensional structure. Jonesite is a first example of titanosilicate with porous double-layer structure.