## Zinconigerite-2N1S ZnSn<sub>2</sub>Al<sub>12</sub>O<sub>22</sub>(OH)<sub>2</sub> and zinconigerite-6N6S Zn<sub>3</sub>Sn<sub>2</sub>Al<sub>16</sub>O<sub>30</sub>(OH)<sub>2</sub>, two new minerals of the nolanite-spinel polysomatic series from the Xianghualing skarn, Hunan Province, China

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## ABSTRACT

Zinconigerite-2N1S ZnSn<sub>2</sub>Al<sub>12</sub>O<sub>22</sub>(OH)<sub>2</sub> and zinconigerite-6N6S Zn<sub>3</sub>Sn<sub>2</sub>Al<sub>16</sub>O<sub>30</sub>(OH)<sub>2</sub> are two new minerals with different numbers and ratios of nolanite (N) and spinel (S) modules. Both phases have been discovered in the Xianghualing skarn, Hunan Province, China. Zinconigerite-2N1S (zn-2N1S) and zinconigerite-6N6S (zn-6N6S) are named for their chemical composition, number, and ratios of N-S modules, according to the nomenclature of the nolanite-spinel polysomatic series of Armbruster (2002). Both phases occur as aggregates, sub-to-euhedral crystals, with maximal dimensions up to 100 µm, within fluorite aggregates, and they are closely associated with phlogopite, chrysoberyl, magnetite, cassiterite, margarite, and nigerite-taaffeite group minerals. They do not show fluorescence in long- or short-wave ultraviolet light. The calculated densities are  $4.456 \text{ g/cm}^3$  for zn-2N1S and 4.438g/cm<sup>3</sup> for zn-6N6S. Optically, zn-2N1S is uniaxial (+) with  $\omega = 1.83(1)$ ,  $\varepsilon = 1.84(2)$ ; zn-6N6S is uniaxial (+) with  $\omega = 1.85(1)$ ,  $\varepsilon = 1.87(2)$  ( $\lambda = 589$  nm). Their chemical compositions by electron-microprobe analyses give the empirical formulas  $(Zn_{0.734}Mn_{0.204}Na_{0.122}Ca_{0.063}Mg_{0.044})_{\Sigma1,166}(Sn_{1.941}Zn_{0.053}Ti_{0.007})_{\Sigma2}$  $(Al_{11.018}Fe_{0.690}^{3+}Zn_{0.200}Si_{0.092})_{12}O_{22}(OH)_2 \text{ for } zn-2N1S \text{ and } (Zn_{1.689}Mn_{0.576}Mg_{0.328}Fe_{0.407}^{3+})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882}Zn_{0.047})_{\Sigma3}(Sn_{1.882$  $Ti_{0.071}$   $\Sigma_{2}(Al_{14.675}Fe_{1.88}^{+}Na_{0.13}Ca_{0.086}Si_{0.017})$   $\Sigma_{15.996}O_{30}(OH)_{2}$  for zn-6N6S. Both phases have trigonal symmetry; the unit-cell parameters of  $zn-2N1S(P\overline{3}m1)$  and  $zn-6N6S(R\overline{3}m)$ , refined from single-crystal X-ray diffraction data, are, a = 5.7191(2) and 5.7241(2) Å, c = 13.8380(6) and 55.5393(16) Å, V = 391.98(3)and 1575.96(12) Å<sup>3</sup>, and Z = 1 and 3, respectively. The structure of zn-2*N*1S is characterized by the alternating  $O-T_1-O-T_2-O-T_1$  layers stacked along the *c*-axis, showing the connectivity of *N-S-N*. The polyhedral stacking sequence of zn-6N6S is  $3 \times (O-T_1-O-T_2-O-T_2-O-T_1)$ , reflecting a N-S-S-N-N-S-S-N-N-S-S-N connectivity of the polysomatic structure. By contrast, the structure of zn-2N1S shows the elemental replacements of Al  $\rightarrow$  Sn and Al  $\rightarrow$  Zn, suggesting the substitution mechanism of 2Al  $\rightarrow$  Zn + Sn. The complex substitution of Zn by multiple elements (Al, Fe<sup>3+</sup>, Mn, Mg) in the structure of zn-6N6S, is coupled with the low occupancy of Al5-octahedra. Fe<sup>3+</sup>  $\rightarrow$  Al substitution occurs in All-tetrahedra of both zn-2N1S and zn-6N6S. The new polysomes, zn-2N1S and zn-6N6S, likely crystallized under F-rich conditions during the late stages of the Xianghualing skarn formation. The discovery of zn-2N1S and zn-6N6S provides new insights into the crystal chemistry of the N-S polysomatic series and its origin.

**Keywords:** Zinconigerite-2*N*1*S*, zinconigerite-6*N*6*S*, nolanite module, spinel module, polysomatic series, Xianghualing skarn