

Supplementary Information for the manuscript

**Natrolite is not a “soda-stone” anymore:
Structural study of alkali (Li^+), alkaline-earth (Ca^{2+} , Sr^{2+} , Ba^{2+}) and heavy metal (Cd^{2+} , Pb^{2+} , Ag^+) cation-exchanged natrolites**

Yongjae Lee,* Donghoon Seoung, Yongmoon Lee

Department of Earth System Sciences, Yonsei University, Seoul 120-749, Korea

(office) +82-2-2123-5667, (e-mail) YongjaeLee@yonsei.ac.kr

Supplementary Table1. EDS chemical composition of the fully cation-exchanged natrolites.^a

Elements	Atomic percent (%) ^c					Composition	390 391 392
	1	2	3	4	5		
Ca-NAT	Ca	5.63	5.93	4.45	5.69	6.64	393
	K	0.00	0.00	0.00	0.00	0.00	394 395
	Al	11.94	12.01	11.13	12.43	12.43	396
Sr-NAT	Sr	5.93	5.62	6.36	5.96	5.76	396 397
	K	0.00	0.00	0.00	0.00	0.00	398
	Al	12.41	12.49	12.95	11.96	11.37	399
Ba-NAT	Ba	5.74	6.26	6.17	6.48	6.74	400
	K	0.00	0.00	0.00	0.00	0.00	401
	Al	12.59	13.08	13.3	12.87	12.07	402
Ag-NAT	Ag	12.78	12.31	12.32	12.32	13.05	403
	K	0.00	0.00	0.00	0.00	0.00	404
	Al	12.20	12.24	12.3	12.64	11.7	405
Cd-NAT	Cd	6.39	6.19	6.38	6.3	6.52	406
	K	0.00	0.00	0.00	0.08	0.14	407 408
	Al	12.07	12.9	12.42	11.99	12.72	409
Pb-NAT	Pb	6.63	6.11	6.58	6.28	6.24	410
	K	0.00	0.00	0.00	0.00	0.01	411
	Al	12.74	11.47	12.54	12.85	12.24	412
Li-NAT	K	0.00	0.07	0.00	0.00	0.07	413 ^b
	Al	13.57	14.39	13.41	12.97	12.54	414

^aValues are normalized based on 16 aluminum atoms per unit cell.^bLithium contents are estimated based on the measured potassium contents.^cInstrumental detection limit down to the third decimal point.

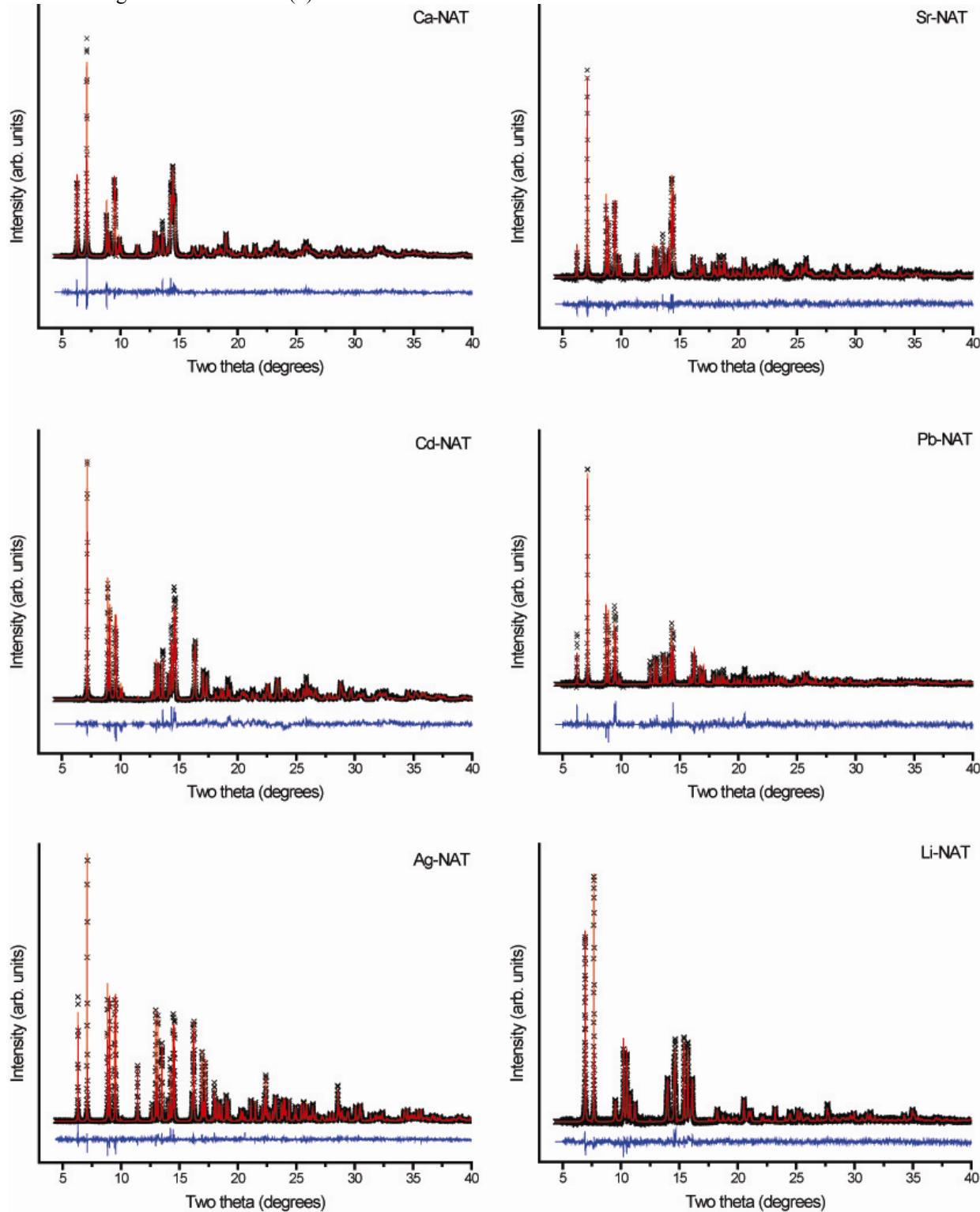
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420Supplementary Table2. Strcutural comparison between the Ca-NAT model and that of natural scolecite by Comodi et al.
(2002).

Scolecite				Ca-NAT				Difference			
	x	y	z		x	y	z		x	y	z
a	6.533(2)			a	6.5242(3)			a	0.0088(10)		
b	19.030(3)			b	18.9885(7)			b	0.0415(15)		
c	9.830(3)			c	9.8466(5)			c	-0.0166(15)		
alpha	90			alpha	90			alpha	0		
beta	109.95(3)			beta	109.70(1)			beta	0.25(2)		
gamma	90			gamma	90			gamma	0		
volume	1148.76(4)			volume	1148.4(1)			volume	0.36(5)		
	x	y	z		x	y	z		x	y	z
Ca	0.1613(1)	0.14323(2)	0.0521(1)	Ca	0.157(1)	0.1412(3)	0.0562(7)	Ca	0.0043(1)	0.00203(2)	-0.0041(4)
Si1	0.5	0.37057(3)	0	Si1	0.503(1)	0.3690(4)	0.0001(8)	Si1	-0.0032(5)	0.00157(20)	-0.0001(4)
Si2	0.2298(1)	0.33184(3)	0.2001(1)	Si2	0.229(1)	0.3335(3)	0.1997(7)	Si2	0.0008(5)	-0.00166(15)	0.0004(4)
Si3	0.5400(1)	0.08257(3)	0.3312(1)	Si3	0.549(1)	0.0826(4)	0.3337(7)	Si3	-0.009(5)	0.00003(2)	-0.0025(4)
Al1	0.9344(1)	0.46209(3)	0.0998(1)	Al1	0.933(1)	0.4628(4)	0.0989(7)	Al1	0.0014(5)	-0.00071(2)	0.0009(4)
Al2	0.3555(1)	0.21662(3)	0.4338(1)	Al2	0.356(1)	0.2186(4)	0.4391(7)	Al2	-0.0005(5)	-0.002(1)	-0.0053(4)
O1	0.5420(3)	0.0315(1)	0.4608(2)	O1	0.544(2)	0.0276(5)	0.459(1)	O1	-0.0020(10)	0.0039(3)	0.0018(5)
O2	0.4472(3)	0.0460(1)	0.1744(2)	O2	0.447(2)	0.0453(5)	0.1776(9)	O2	0.0002(10)	0.0007(3)	-0.0032(5)
O3	0.3836(3)	0.1513(1)	0.3156(2)	O3	0.401(2)	0.1520(5)	0.330(1)	O3	-0.0174(10)	-0.0007(3)	-0.0144(5)
O4	0.1154(3)	0.1998(1)	0.4681(2)	O4	0.116(2)	0.2038(5)	0.477(1)	O4	-0.0006(10)	-0.0040(3)	-0.0089(5)
O5	0.3535(3)	0.2994(1)	0.3852(2)	O5	0.342(2)	0.3008(5)	0.3595(9)	O5	0.0115(10)	-0.0014(3)	0.0257(5)
O6	0.0868(3)	0.2712(1)	0.0905(2)	O6	0.085(1)	0.2730(5)	0.0937(9)	O6	0.0018(5)	-0.0018(3)	-0.0032(5)
O7	0.4143(3)	0.3587(1)	0.1345(2)	O7	0.425(2)	0.3573(6)	0.139(1)	O7	-0.0107(10)	0.0014(3)	-0.0045(5)
O8	0.0767(3)	0.3956(1)	0.2148(2)	O8	0.086(2)	0.3995(5)	0.221(1)	O8	-0.0093(10)	-0.0039(3)	-0.0062(5)
O9	0.7894(3)	0.1101(1)	0.3565(2)	O9	0.797(1)	0.1104(6)	0.357(1)	O9	-0.0076(5)	-0.0003(3)	-0.0005(5)
O10	0.6602(3)	0.4369(1)	0.0342(2)	O10	0.658(1)	0.4385(5)	0.037(1)	O10	0.0022(2)	-0.0016(3)	-0.0028(5)
Wat1	0.8908(4)	0.0803(1)	0.1083(3)	Wat1	0.874(3)	0.0755(8)	0.086(2)	Wat1	0.0168(15)	0.0048(4)	0.0223(10)
Wat2	0.9085(5)	0.326(1)	0.4404(4)	Wat2	0.903(3)	0.3256(8)	0.429(2)	Wat2	0.0055(15)	0.0004(4)	0.0114(10)
Wat3	0.5782(4)	0.4454(1)	0.3742(2)	Wat3	0.568(3)	0.4537(7)	0.354(2)	Wat3	0.0102(15)	-0.0083(4)	0.0202(10)

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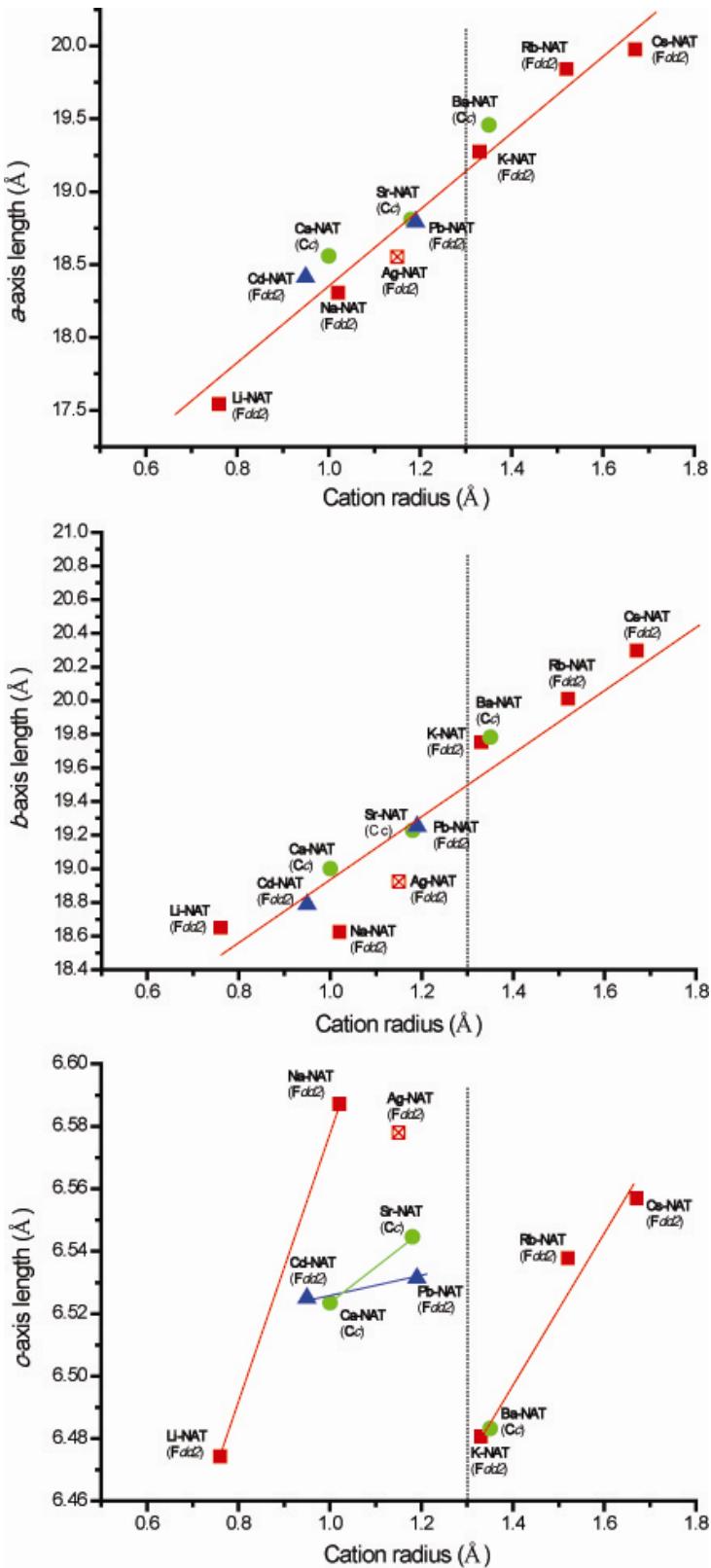
Supplementary Figure 1. The Rietveld refinement fits of the structural models of Ca-NAT, Sr-NAT, Cd-NAT, Pb-NAT, Ag-NAT, and Li-NAT to the synchrotron X-ray powder diffraction data measured at room temperature. Backgrounds were subtracted from the data. Points represent observed data. The continuous lines are the calculated profiles. The lower curves represent the differences between the observed and calculated profiles ($I_{\text{obs}} - I_{\text{calc}}$) plotted on the same scale as the observed data. The wavelength used was 0.7297(1) Å.



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Supplementary Figure 2. The refined unit cell lengths of the caion-exchanged natrolites, normalized to the orthorhombic unit cell. Data for Na-NAT are from the work of Baur et al. (1990), and K-, Rb-, and Cs-NAT from Lee et al. (2010). The lines are guides to the eyes. The dotted line indicates a possible threshold for the order-disorder transition.



433 Supplementary Figure 3. Changes in the degree of orthorhombicity, defined as $2(b-a)/(b+a)$, of the caion-exchanged natrolites
434 plotted as a function of the cation radius.
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