ERRATUM

Thermodynamic studies of zeolites: Heulandite by G. K. Johnson, H. E. Flotow, P.A.G. O'Hare, and W. S. Wise (v. 70, p. 1065–1071). In Table 5, ΔH for Reaction 11 should be $-(1096.16 \pm 0.15)$ kJ/mol. As a result, ΔH_f^0 and ΔG_f^0 of heulandite at 298.15 K should be $-(10594.6 \pm 10.2)$ and $-(9779.1 \pm 10.2)$ kJ/mol, respectively, and each entry for $\Delta H_f^0(T)$ and $\Delta G_f^0(T)$ in Table 7 should be more negative by 103.5 kJ/mol.