

Solubility mechanisms of H₂O in silicate melts at high pressures and temperatures: a Raman spectroscopic study: reply

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Freund (1982) makes two comments on a recent paper by Mysen *et al.* (1980). (1) Substitution of D₂O for H₂O may not cause the shift of the 980 cm⁻¹ band, as suggested by Mysen *et al.* (1980). (2) Arguments for the absence of H₂ made by Mysen *et al.* (1980) are inadequate. These comments will be dealt with in order.

As correctly pointed out by Freund (1982), the expected frequency shift of the 980 cm⁻¹ Si-OH band depends wholly on whether or not the vibration is one of O-D domination or Si-O domination. Mysen *et al.* (1980) did not apply the D₂O substitution for H₂O as a critical test of whether or not there were OH-groups in the melt. This was simply an additional test on the influence of H₂O on aluminosilicate melts, conducted because of the poor spectroscopic resolution in this frequency region. The conclusions by Mysen *et al.* (1980) are not dependent on this assignment. The interpretation of the 980 cm⁻¹ band is, therefore, unaffected, whether or not the suggestion by Mysen *et al.* (1980) for the effect of D₂O is correct.

Freund (1982) suggests that the 880 cm⁻¹ band may be due to Al-OH. Mysen *et al.* (1980) considered this possibility and concluded that because the position of this band is independent of whether or not D₂O or H₂O is the solute and because of other data on the subject (as discussed on p. 904 in the original paper), vibration involving H or D together with Al were considered unlikely. More importantly, however, Mysen *et al.* (1980) on p. 905 pointed out that the frequency of the band near 880 cm⁻¹ depends on H₂O content of the melt and on the Al-content of the solvent. If this band were due to Al-H, Si-H, Al-OH or Si-OH, the relationship noted by Mysen *et al.* (1980) is not likely to be observed. In addition, it should be reiterated that the frequency changes in this region coincide with frequency changes due to altered (Si,Al)-coupling in the frequency range between 1000 and 1200 cm⁻¹, as also discussed through the paper by Mysen *et al.* (1980). The latter effects would not be likely if both Al-OH and Si-OH were formed in equal proportion to their

original abundance. The D₂O solution experiments have little bearing on this conclusion.

Freund (1982) finally suggests that Mysen *et al.* (1980) did not document a case for absence of H₂ in the quenched glasses. It does not matter which reference is chosen in this context. Studies by, for example, Van der Steen and Van den Bloom (1977) involving both H₂ and D₂, and by Faile and Roy (1970) and by Hartwig (1977) in samples that demonstrably contained H₂ or D₂, or both, prove the existence of the main H₂ band near 4100 cm⁻¹, and the analogous D₂ band near 2900 cm⁻¹. Mysen *et al.* (1980) observed neither.

In summary, the H₂O-solution model derived from interpretation of spectroscopic data by Mysen *et al.* (1980) and by references therein is internally consistent. The model is also consistent with published thermodynamic, phase equilibrium and other chemical data on H₂O-bearing systems in the upper mantle (for summary of such data, see Mysen, 1977; and references therein).

References

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*Manuscript received, August 18, 1981;
accepted for publication, August 19, 1981.*