X-Ray Investigation of Single Crystals of β -Ca₂SiO₄ (Larnite) at High Temperatures

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Abstract

New data on the polymorphism of Ca₂SiO₄ have been obtained by single crystal heating experiments. Verneuil-grown crystals of β -Ca₂SiO₄ (= larnite) were heated up to 1450°C to examine the transformation mechanism. The transformation series $\beta \leftrightarrow \alpha'_{\rm L} \leftrightarrow \alpha'_{\rm H} \leftrightarrow \alpha$ is displacive and related to a twin geometry. In the $\alpha'_{\rm L}$ -range a superstructure along the *c* axis could be observed. A tilt mechanism of the SiO₄-tetrahedra may account for the formation of this superstructure similar to that found in the Ca₂SiO₄-Ca₃(PO₄)₂ solid solution series.

Introduction

The polymorphism of Ca₂SiO₄ has been studied extensively in recent years. Lehmann, Niesel, and Thormann (1969) have given a comprehensive summary of previous work. An orthorhombic phase, γ -Ca₂SiO₄, (Pmnb) is stable at room temperature but on heating changes to another orthorhombic phase (Pmnn) denoted as α' -Ca₂SiO₄. Niesel and Thormann (1967) and Regourd et al (1968) demonstrated the existence of two α forms: $\alpha'_{\rm L}$ (low *Pcmn*) $\leftrightarrow \alpha'_{\rm H}$ (high *Pcmn*) with transition at 1160°C. Further heating to 1450°C produces hexagonal α -Ca₂SiO₄ (P6₃mc). On cooling, monoclinic β -Ca₂SiO₄ (larnite, P2₁/n) appears at 670°C and is considered a metastable phase. Cooling to room temperature may result in conversion to the stable γ -Ca₂SiO₄. This conversion can be prevented by small amounts of so-called "stabilizers."

Extensive investigations on the effect of impurities were carried out by Newman and Wells (1946). Moreover Saalfeld (1967) was able to show that the presence of water vapor during heating also stabilizes the β -Ca₂SiO₄ structure. X-ray investigations of single crystals revealed a twin geometry suggesting the displacive character of the transformation series $\beta \leftrightarrow \alpha' \leftrightarrow \alpha$.

In order to investigate the transformation mechanism, it is necessary to hold single crystals at different temperatures during the X-ray diffraction experiments.

Experimental

Crystals large enough for the investigation were grown by the Verneuil method, and experimental details have been published previously (Saalfeld, 1971). To prevent the $\beta \rightarrow \gamma$ transition, 1-2 mole percent of Ca₃(PO₄)₂ was added. Thin sections of the polycrystalline Verneuil bowl were prepared for electron probe studies. Fragments 0.5 mm long were selected from these thin sections and mounted on top of a thermocouple. This thermocouple simultaneously heated the crystal and measured the temperature. The crystals were investigated by oscillation and Weissenberg techniques at temperatures up to 1450°C.

Results

Chemical Results

Wet-chemical analysis of the bowl material revealed an average content of 0.81 wt percent P_2O_5 , corresponding to 1.77 wt percent $Ca_3(PO_4)_2$. The P_2O_5 content is not uniformly distributed in the Verneuil bowl, varying between \pm 0.5 wt percent. From the results (Table 1) of two electron probe analyses carried out in the region investigated by X-rays, it can be concluded that 3-4 percent of all tetrahedra are occupied by P.

X-ray Results

In Table 2 crystallographic data for the series $\beta \leftrightarrow \alpha'_{\rm L} \leftrightarrow \alpha'_{\rm H}$ are presented. The lattice parameters measured from oscillation and Weissenberg photographs are not of very high accuracy but in general conform with X-ray powder data published by other authors (Regourd *et al*, 1968). At room temperature single crystals of β -Ca₂SiO₄ exhibit hexagonal and monoclinic twin-geometry due to cooling history. The twin geometry can be described by two

TABLE 1. Electron Probe Analyses of Two Crystals of β -Ca₂SiO₄

wt p er cent	1	2	
Ca0	64.60	64.68	
Si0 ₂	34.16	34.27	
P205	1.29	1.86	
Total	100.05	100.81	
Atomic	Proportions on	Basis of	16 Oxygen:
Ca	7,90	7.82	
Si	3.90	3.87	
Р	0.12	0.18	

twin operations: (a) reflection of the monoclinic cell of β -Ca₂SiO₄ in (010), (b) 60°-rotations around [001]_{β}. The complete twin symmetry is then 6/*mmm*.

Figure 1a shows an hk0 oscillation photograph taken at 650°C. Most reflections belong to β -Ca₂SiO₄. But weak α' -reflections can be detected between β -twin reflections in the layer lines as indicated between the twin reflections 222_{β} .

With increasing temperature the monoclinic twin reflections approach a limiting value. This can be explained by reduction of the monoclinic angle γ from 94° to 93°. On further heating the monoclinic reflections abruptly disappear and only the reflections of the $\alpha'_{\rm L}$ phase are present (Fig. 1b). This suggests a sudden structure change caused by geometrical distortions so that orthorhombic symmetry of $\alpha'_{\rm L}$ is attained. In the X-ray photographs of the $\alpha'_{\rm L}$ phase, superstructure reflections are distinctly visible (Fig. 2) between the 121-122 and the 221-222 reflections of $\alpha'_{\rm L}$. The superstructure reflections are distributed along the c^* axis so that the unit cell of $\alpha'_{\rm L}$ (space group: *Pcmn*) has a *c* axis three times as long as that of the basic cell.

Above 1160°C the superstructure reflections disappear, and the $\alpha'_{\rm H}$ phase exhibits the basic orthorhom-

TABLE 2. X-Ray Data of Ca₂SiO₄ Measured at Different Temperatures*

		1 0111 P			
Phase	<u>T</u> ^o C	<u>a</u> (Å)	<u>b</u> (Å)	<u>c</u> (Å)	Space Group
β	20	9.34	5.51	6,76	<u>P112,/n</u>
			γ = 94	, 3 ⁰	-
α'L	800	9.41	5.53	6.81x3	Pcmn
a'L	1000	9.48	5.59	6.83x3	Pcmn
α'L	1200	9.49	5.59	6.85	Pcmn
*Settin	g of the	axes as	given	by Eysel	(1971).



FIG. 1a. Oscillation photograph (*hk*0) of β -Ca₂SiO₄ at 650°C. Between twin reflections of the β phase reflections from the α'_L phase appear.

bic cell with the same space group *Pcmn*. Further transition to the α -modification takes place above 1400°C. At these temperatures there is considerable Pt evaporation from the thermocouple so that the photographs are of inferior quality and for that reason reliable lattice parameters could not be measured. Upon cooling, α -Ca₂SiO₄ transforms back to the α' and β modifications having the previously mentioned twin geometry.

Discussion

These experiments show that the transformations under consideration are displacive and are related to a twin geometry. Ca₂SiO₄ is a compound containing isolated SiO₄-tetrahedra. In the high-temperature α form the tetrahedra have a hexagonal arrangement with all apices oriented in the same direction (Eysel, 1971). In the α' -modification the tetrahedra have their apices oriented alternately up and down. In the β -modification the tetrahedra are similarly arranged but are tilted with respect to c. The transformations $\beta \leftrightarrow \alpha' \leftrightarrow \alpha$ require appreciable rotation of the tetrahedra, and Eysel and Hahn (1970) consider them



FIG. 1b. Oscillation photograph of the same crystal heated at 900°C. Only reflections of the α'_{L} phase are present.

to be "semi-reconstructive." Eysel (1971) has pointed out that such transformations can involve tetrahedral rotations up to 70°, and that the transformation $\alpha \leftrightarrow \alpha'$ must involve such rotations. The single crystal investigation described in this paper confirms the more displacive character of this transformation.

X-ray photographs taken at high temperatures reveal superstructure reflections in the α'_{L} -phase that disappear in the $\alpha'_{\rm H}$ -phase. The transformation effects of $\alpha'_{\rm L} \leftrightarrow \alpha'_{\rm H}$ are very weak and can be detected by sensitive methods only, as Niesel and Thormann (1967) have shown by DTA measurements. Regourd et al (1968) indexed a powder diagram of a specimen of $\alpha'_{\rm L}$ that had been heated to 1000°C and proposed a doubling of both a and b axes because otherwise five weak lines could not be indexed. The photographs of heated single crystals, however, show clearly that there is a superstructure in the c direction only. It seems unlikely that this superstructure develops in response to the presence of a small amount of P_2O_5 , because the P_2O_5 content does not correspond to a stoichiometry consistent with the observed superstructure. One can assume that pure β -Ca₂SiO₄ on heating will also show a reversible transformation



FIG. 2. Enlarged section of an oscillation photograph $(0kl)\alpha'_{L}$ taken at 1000°C. Arrows show the superstructure reflections.

involving a similar superstructure. It is not difficult to index at least four of the five powder lines on the basis of the observed superstructure as is demonstrated in Table 3. The very weak reflection 122 respresents an unexplained exception.

The phase diagram of the system Ca₂SiO₄-Ca₃(PO₄)₂, revised by Fix, Heymann, and Heinke (1969), shows that with increasing amounts of Ca₃(PO₄)₂ the high-temperature forms α' and α of Ca₂SiO₄ can be stabilized at room temperature. X-ray investigations on single crystals of different compositions reveal that the superstructures change with composition (Saalfeld, 1971). Crystal structure determination of the 2:1 compound has been carried out recently in our laboratory and has shown that the superstructure in the *c* direction is due to tilting of all tetrahedra. It seems likely that the superstructures in Ca₂SiO₄ are also caused by a tilting mechanism.

TABLE 3. a'L-Ca2SiO4 Indexing for Two Different Unit Cells

Regourd (1968) (2a, 2b, c)		This work (a, b, 3c)	
hkl	$d(\mathbf{\hat{A}})$	hkl	$d(\hat{A})$
231	3.565	211	3.561
122	3.088		
341	2.699	022	2.697
350	2.658	$\begin{cases} 121 \\ 312 \end{cases}$	2.658
652	1.502	038	1.508

Further structure determinations of Ca_2SiO_4 -type compounds with superstructures are presently being initiated.

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