MINERALOGICAL NOTES

supported in part by grants from the National Aeronautics and Space administration (No. NsG317-63), the National Science Foundation (No. G-12325), and the Atomic Energy Commission (AT (30-1)-2623).

References

BERRY, L. G. AND D. A. MODDLE (1941) Studies of mineral sulpho-salts: V-Meneghinite from Ontario and Tuscany. Univ. Toronto Studies, Geol. Ser. 46, 5-17.

PALACHE, C., W. E. RICHMOND AND H. WINCHELL (1938) Crystallographic studies of sulpho-salts: baumhauerite, meneghinite, jordanite, diaphorite, freieslebenite. Am. Mineral. 23, 821–836.

RAMDOHR, P. (1960) Die Erzmineralien und Ihre Verwachsungen. Akademie-Verlag, Berlin, 714-715.

VON RATH, G. (1867) Mineralogische Mitteilungen. Ann. Phys. Chem. 132, 372-404.

ZIEBOLD, T. O. AND R. E. OGILVIE (1963) Quantitative analysis with the electron microanalyzer. Anal. Chem. 35(6), 621-627.

THE AMERICAN MINERALOGIST, VOL. 49, SEPTEMBER-OCTOBER, 1964

SYNTHESIS OF THORIANITE CRYSTALS FROM BISMUTH OXIDE-LEAD FLUORIDE MELTS

A. B. CHASE AND JUDITH A. OSMER, Aerospace Corporation, El Segundo, California.

Oxides that can be readily grown as single crystals, and into which rareearth ions can be incorporated, are of research interest and may be of value for certain electronic applications. Uranium and the lanthanides substitute for thorium in natural occurring thorianite (Palache *et al.*, 1944). The growth of thorianite by long continued fusion of ThO₂ in Borax has been previously described (Mellor, 1960). However, the synthesis of thorianite by this method results in either minute crystals or trellis-like aggregates of crystals. The purpose of this report is to describe a method for the growth of large thorianite crystals.

It has been found that ThO₂ crystals can be readily synthesized from the PbO-PbF₂, Bi₂O₃-PbF₂, and PbF₂ flux systems. The melts from which the largest ThO₂ crystals were grown contained 7 mole per cent ThO₂, 15 mole per cent Bi₂O₃, and 78 mole per cent PbF₂. Melts containing 10 mole per cent ThO₂ and 90 per cent PpF₂ also produced large crystals. The materials employed were 99.99 per cent pure ThO₂, reagent grade Bi₂O₃, and a purified grade of PbF₂. The powders were mechanically mixed (generally in 100 gram lots) and fused in tightly covered 50 ml platinum crucibles.

The crucibles were placed in two Super Kanthal heated horizontal muffle furnaces. The melts were held at 1250° C. for 4.5 to 8 hr and cooled

MINERALOGICAL NOTES

at a uniform rate (2 to 4° C/hr) to 1000° C. At this temperature the crucibles were removed from the furnace and allowed to cool to room temperature. The vertical temperature gradients in these furnaces are small (less than 3° C/in.) and probably have little effect on crystal growth. Between 35 and 50 per cent of the melt is lost by evaporation during a normal run, despite care in capping the crucibles. The major portion of this weight loss is probably PbF₂, which has an appreciable vapor pressure at these temperatures. The crystals were recovered by

Run No.	Mole %			Low	High	Soak Time,	Cooling Rate,	Wt %	Yield	Comments
	ThO ₂	Bi ₂ O ₂	PbF ₂	° C.	° C,	hr	° C ₄ /hr	Loss	%	
1	3	30	67	1000	1250	8.5	3.5	1	71.5	Colorless crystals up to 3 mm grew on [111] den- drites, {100} habit
2	5	30	65	1000	1250	19	3.5	38.9	94.0	Very clear subhedral crys- tals. 1-1.5 mm, {100 habit
3	5	15	80	950	1260	8	4.0	43.9	64.2	Large crystals, inclusions in larger ones, 3.5-5 mm
4	5	47	48	1000	1250	12	2.4	\rightarrow	65.0	Clear, colorless crystals up to 3 mm.
5	7	15	78	950	1260	8	4.0	38.7	67.0	Large, clear crystals 3.5- mm. Largest have inclu sions. BEST RUN
6	10	15	75	1015	1250	14.5	2.4	57.5	98.0	Good quality crystals 0.1- 1.5 mm in size
7	15	15	70	1000	1260	7	2.4	53.9	68.0	Good quality. Cubic {100}. average size 1- 1.5 mm
8	10	575	90	1015	1240	4.5	4.0	41.3	25.8	Very good quality. Cubi {100}. 2-5 mm

TABLE I. THORIANITE GROWTH DATA

digesting the contents of the crucibles in hot 50 per cent acetic acid and were identified by microscopic examination and confirmed by x-ray powder patterns. The average yield was approximately 70 per cent, based on the original content of ThO₂ in the melt. The results from a number of different runs are listed in Table 1. Runs 3, 5 and 8 gave the best results as far as crystal size is concerned. It is felt that this is due to decreased nucleation because of reduced losses of PbF₂ during the comparatively shorter soaking times and more rapid cooling rates.

The crystals obtained from a typical melt ranged from 2 to 6 mm (Fig. 1), with the largest crystals occurring on the bottom of the crucible. The crystals also occurred on the top surface of the melt and attached to the walls of the crucible. The crystals were generally subhedral, with the

1470



FIG. 1. Thorianite crystals grown from a typical melt $(4 \times)$.

{100} being the dominant habit. Microscopic examination revealed the outline of dendrites within some of the crystals. The arms of the dendrites extended along [111] directions. The material outlining the dendrites is finely divided and appears to be flux that was trapped during growth. During later growth the ends of the dendritic arms come in contact with each other, trapping melt in the regions between them. Further growth of the crystals is fairly uniform, and the exterior regions are free of occlusions and macroscopic defects. There are a few crystals from each batch that have not occluded melt during growth. These crystals are colorless and appear to be of good optical quality.

It has been found that the crystals grown by this technique are easily doped with the rare-earth sesquioxides and UO_2 . Preliminary investigations indicate that some of the rare-earth ions fluoresce strongly in single crystals of thorianite.

The authors wish to express appreciation to G. M. Wolten who identified the crystals by x-ray techniques.

References

MELLOR, S. W. (1960) A Comprehensive Treatise on Inorganic and Theoretical Chemistry. Longmans, Green and Co., London, vol. VII, p. 220.

PALACHE, C., H. BERMAN AND C. FRONDEL (1944) Dana's System of Mineralogy, John Wiley and Sons, Inc., New York, 7th Ed., vol. 1, p. 621.