

X-RAY STUDIES OF SYNTHETIC COFFINITE, THORITE
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ABSTRACT

X-ray data are presented for synthetic coffinite, thorite, and several uranothorites. The cell constants obtained for coffinite are $a=6.981 \pm 0.004$ kX, $c=6.250 \pm 0.005$ kX; for thorite $a=7.128 \pm 0.004$ kX, $c=6.314 \pm 0.003$ kX. Intermediate constants determined for several uranothorites indicate a continuous solid solution between $USiO_4$ and $ThSiO_4$. Coffinite and thorite are isostructural with zircon; the space group is $D_{4h}^{19}-I_4/amd$. The oxygen positions for coffinite are $u=0.180 \pm 0.010$, $v=0.347 \pm 0.010$ and for thorite $u=0.166 \pm 0.010$, $v=0.347 \pm 0.010$. No changes were observed either in line intensities or in cell constants when (OH) was removed from the hydrothermal preparations.

The mineral coffinite, described as a uranous silicate with hydroxyl substitution, was identified on the basis of the similarity of its x-ray powder pattern to that of zircon ($ZrSiO_4$) or thorite ($ThSiO_4$) (Stieff, Stern and Sherwood, 1956). Pabst (1951) obtained single crystal patterns of a New Zealand detrital uranothorite (11.5 wt% UO_2) and determined the space group to be the same as given for zircon. A similar assignment was made by Bonatti and Gallitelli (1951) on detrital thorite crystals from Nettuno Rome.

Although coffinite is isostructural with zircon and thorite, the naturally occurring mineral is reported to exist as the hydroxyl substituted form of $USiO_4$ (Stieff, Stern and Sherwood, 1956). This consideration is based on the low silicon content which appears in the analyses of the mineral. However, our observations on a synthetic product (Hoekstra and Fuchs, 1956) indicate that neither cell dimensions nor line intensities appear to be modified when water is removed, suggesting therefore that the synthesized product may be $USiO_4$ without hydroxyl substitution. Infrared examination of heated samples made in this laboratory clearly indicate that water has been expelled (Hoekstra and Fuchs, to be published).

In addition to coffinite, we have synthesized thorite and some uranothorites. The thorite can be prepared by a hydrothermal process or from the melt (Fuchs, 1958). The latter process yields water-free samples, and again it does not seem possible to distinguish between hydroxyl-containing and hydroxyl-free samples on the basis of x-ray powder patterns.

Powder data for the synthesized coffinite and thorite are presented in Table I. Cell dimensions and x-ray densities for these samples are given

* Based on work performed under the auspices of the U. S. Atomic Energy Commission.

TABLE I. X-RAY DATA FOR COFFINITE AND THORITE. $\text{CuK}\alpha_{1,2} = 1.5386 \text{ kX}$

Coffinite					Thorite				
<i>hkl</i>	$\sin^2 \theta$ obs.	$\sin^2 \theta$ calc.	I_o	I_e	<i>hkl</i>	$\sin^2 \theta$ obs.	$\sin^2 \theta$ calc.	I_o	I_e
101	0.02758	0.02750	197	214	101	0.02669	0.02666	174	208
200	.04917	.04912	203	227	200	.04711	.04712	210	249
211	.07636	.07653	96	105	211	.07348	.07357	99	111
112	.08540	.08538	196	184	112	.08287	.08299	162	188
220	.09784	.09792	53	58	220	.09376	.09376	63	68
202	.1097	.1098	9	5	202	.1065	.1064	6	4
301	.1252	.1252	62	54	301	.1202	.1202	63	57
103	.1489	.1490	60	38	103	.1456	.1456	42	40
321	.1737	.1739	69	58	321	.1672	.1669	67	63
312	.1827	.1829	144	141	312	.1765	.1764	135	140
400	.1964	.1953	66	67	400	.1870	.1870	40	34
213		.1977			213	.1924	.1922	33	39
411	.2227	.2226	32	25	411	.2138	.2134	25	24
004		.2429	53	53	420	.2335	.2335	39	35
420	.2437	.2440			004		.2378	18	19
303		.2464			303	.2383	.2388		
402	—	.2559	0	<1	402	—	.2462	0	<1
332	.2802	.2802	47	37	332	.2695	.2694	38	39
204	.2913	.2916	48	30	204	.2847	.2843	36	41
323	.2948	.2951	10	12	323		.2854		
422	—	.3045	0	<1	422	—	.2926	0	0
501	0.3200	0.3199	34	29	501	0.3066	0.3064	27	30
431					431				
224	.3405	.3402	47	45	224	.3315	.3309	36	43
413	.3433	.3437			413		.3318		
314	—	.3646	0	<1					
521	.3684	.3685	15	10	521	.3534	.3529	10	11
					314	—	.3541	0	<1
512	.3777	.3774	36	38	512	.3627	.3624	38	39
440		.3898	18	14	440	.3732	.3728	8	8
105	.3906	.3914			105	.3829	.3828	5	5
404		.4375	56	52	600	.4193	.4193	12	11
600		.4383			404		.4239	26	29
213	.4390	.4400			433	.4245	.4248		
503		.4409			503		.4248		
433		.4409			215	.4292	.4293	14	12
611	.4649	.4655	13	9	611	.4466	.4457	11	9
532	.4753	.4745	37	30	532	.4557	.4553	27	31
424		.4860	60	59	620	.4657	.4658	17	16
620		.4868			424		.4703	33	42
305	.4870	.4886			523	.4705	.4713		
523		.4894			602	—	.4785	0	0
					541	.4928	.4922	7	7
602	—	.4988	0	<1					

TABLE 1. (continued)

Coffinite					Thorite							
<i>hkl</i>	$\sin^2 \theta$ obs.	$\sin^2 \theta$ calc.	I_a	I_c	<i>hkl</i>	$\sin^2 \theta$ obs.	$\sin^2 \theta$ calc.	I_a	I_c			
541	.5145	.5141	9	7	325	.5227	.5224	7	7			
325	.5380	.5372	10	8	622	—	.5245	0	<1			
622	—	.5474	0	<1	631	.5395	.5388	9	10			
514	—	.5589	0	<1	514	—	.5398	0	0			
631	.5628	.5626	14	10	116	.5573	.5576	10	14			
116	.5697	.5701	20	14	613	.5649	.5644	9	10			
415}	.5866	.5858	.5862	17	18	415	.5682	.5690	7	9		
613}	.5866	.5865				206	—	.5809	0	<1		
206	—	.5944	0	<1	701	.5863	.5854	3	3			
701	.6122	.6111	6	4	640	.6051	.6055	12	13			
444}	.6340	.6317	.6330	29	34	444}	.6099	.6094	.6099	15	21	
640}		.6324				543}	.6103	5				6
543}		.6350				721	.6320					
534	—	.6560	0	<1	534	—	.6329	0	<1			
721	.6601	.6596	14	9	552}	.6411	.6408	20	22			
316}	.6684	.6672	.6681	56	56					712	.6424	.6439
552}		.6685				552}	.6424	.6439				
712}		.6685				712}						
604}	.6827	.6802	.6828	28	27	316 α_1	.6501	.6501	.6564	14	14	
505}		.6829				604}	.6559					
435}		.6829				633	.6566	.6568				13
633}	.6836	505}										
642	—	.6928	0	<1	642	—	.6641	0	<1			
624}	.7304	.7287	.7307	32	39	435}	.7023	.7024	28	27		
525}		.7314				624 α_1					.7034	
703}		.7320				703						.7055
406	—	.7400	0	<1	624 α_2	.7059						
107}	.7564	.7548	.7558	10	10	523 α_1	.7090	.7080	6	6		
651}		.7567				406	—	.7206	0	<1		
336}	.7660	.7643	.7650	33	38	651 α_1	.7247	.7243	.7257	8	7	
732}		.7656				107	.7272	15				17
800}		.7779				732 α_1	.7339					
723}	.7803	.7806	.7792	14	15	732 α_2	.7375	.7374	10	11		
426	—	.7885	0	<1	336}	.7444	.7431	.7437	10	11		
217}	.0837	.8025	.8033	21	18						800}	.7443
811}		.8036				800}	.7498	8	5			
741}		.8036				723 α_1	.7498			0	0	
217}	.8068	.8053	.8068	17	21	426	—	.7671				
811}		.8075				811}	.7715	.7708	8	9		
741}		.8075				741}	.7746	.7746	8	9		
820}	.8281	.8265	.8275	17	21	217 α_1	.7847	.7846	6	6		
615}	.8281	.8285	.8275	17	21	820 α_1	.7915	.7908	8	8		

TABLE 1. (continued)

Coffinite					Tho-rite				
<i>hkl</i>	$\sin^2 \theta$ obs.	$\sin^2 \theta$ calc.	I_o	I_e	<i>hkl</i>	$\sin^2 \theta$ obs.	$\sin^2 \theta$ calc.	I_o	I_e
802	—	.8385	0	<1					
714	—	.8502	0	<1					
307	.8517	.8518	8	5	820 α_2	.7951	.7948		
516 α_1	.8610	.8501	36	38	615 α_1	.8021	.8010	5	5
					802	—	.8046		
					307	—	.8319	0	
516 α_2	.8646	.8639			516	.8363	.8361	.8368	<1
					660		.8374		
					714	—	.8189		
644	—	.8744			516	.8421	.8393	.8404	
660	.8769	.8751	.8761		660		.8415		
545		.8771				644	.8420		.8424
653	—	.8778			653	.8421	.8429	.8424	
822	—	.8871							
327	—	.9004			644	.8467	.8457	.8463	
831	.9018	.9024	.9014		653		.8468		
635	—	.9257			545 α_1	.8467	.8475		
743	.9258	.9264	.9262		822	—	.8501		
813		.9264				545 α_2	.8525	.8510	
734	—	.9474			831 α_1	.8641	.8638		
417	.9480	.9490			831 α_2	.8682	.8680		
536	—	.9586			327 α_1	.8773	.8777		
752	.9585	.9600	.9593		327 α_2	.8815	.8807		
840	.9729	.9723			743	.8893	.8894		
705	—	.9743			813				
606	—	.9831			743	.8939	.8935		
					813				
					635 α_1	.8939	.8946		
					734	—	.9120		
					752 α_1	.9202	.9199		
					752 α_2	—	.9243		
					417 α_1	.9217	.9242		
					536	.9292	.9292	.9298	
					840		.9304		
					536	.9345	.9328	.9339	
					840		.9350		
					705 α_1	.9402	.9405		
					008 α_1	.9485	.9489		
					606	—	.9535		
					901 α_1	.9574	.9569		
					901 α_2	.9622	.9616		
					804	.9818	.9815	.9820	
					833		.9824		
					801	.9870	.9860	.9866	
					833		.9871		

Observed intensities not resolved.

Observed intensities not resolved.

Observed intensities not resolved.

TABLE II. CELL DIMENSIONS AND DENSITIES OF COFFINITE, THORITE AND URANOTHORITES

Material	<i>a</i>	<i>c</i>	<i>a/c</i>	ρ gm./cm. ³ calc.
USiO ₄ *	6.981 ± .004 kX	6.250 ± .005 kX	1.1170	7.15 ± 0.02
3 USiO ₄ · ThSiO ₄	7.007 ± .005	6.275 ± .003	1.1167	7.04 ± .02
USiO ₄ · ThSiO ₄	7.039 ± .003	6.294 ± .002	1.1184	6.91 ± .01
USiO ₄ · 3 ThSiO ₄	7.071 ± .002	6.314 ± .003	1.1199	6.80 ± .01
ThSiO ₄	7.128 ± .004	6.314 ± .003	1.1289	6.67 ± 0.01

* A value previously reported for USiO₄ (Hoekstra and Fuchs, 1956), varies slightly from the value given here, since the earlier value was based on an incomplete indexing of the powder pattern.

in Table II. We also include in this table *x*-ray data for the several uranothorite samples which were prepared by the hydrothermal process developed for coffinite. The only variation in technique involves the preparation of thorium and uranium tetrachloride solutions in the desired concentrations. Reference to Table II shows that the cell dimension changes are uniform throughout the composition range.

We have also attempted to determine the oxygen positions in coffinite and thorite from powder patterns. Although the oxygen scattering is very small, the assigned positions are necessary in order to obtain reasonable agreement between observed and calculated intensities.

The atomic positions are assumed to be those for zircon, D_{4h}^{19} — I_4/a *md* and are as follows:

4 Th or 4 U in (*a*)

4 Si in (*b*)

16 O in (*h*) (International Tables)

The only variables are *u* and *v* for oxygen and these were determined by trial methods. The resulting values for the oxygen positions are given in Table III. The extent of agreement between observed and calculated intensities is shown in Table I. The observed intensities were deduced from microphotometer tracings and the calculated intensities were obtained from the formula

$$I \sim F^2 p \frac{1 + \cos^2 2\theta}{\sin^2 \theta \cos \theta}$$

TABLE III. OXYGEN POSITIONS

	<i>u</i>	<i>v</i>
USiO ₄	.180 ± 0.010	.347 ± 0.010
ThSiO ₄	.166 ± 0.010	.347 ± 0.010

TABLE IV. BOND LENGTHS

USiO ₄	U—4 O	2.32 Å ± .08 Å
	U—4 O	2.52 Å ± .09 Å
	Si—4 O	1.58 Å ± .09 Å
ThSiO ₄	Th—4 O	2.46 Å ± .08 Å
	Th—4 O	2.50 Å ± .09 Å
	Si—4 O	1.55 Å ± .09 Å
ZrSiO ₄	Zr—4 O	2.05 Å
	Zr—4 O	2.41 Å
	Si—4 O	1.62 Å

where p is the multiplicity factor and the other quantities have their usual significance.

The bond distances which result are shown in Table IV. The bond lengths in zircon (Wyckoff and Hendricks, 1927) are given for comparative purposes.

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