# NEW MINERAL NAMES

## Lembergite

TOSIO SUDO, On some low temperature hydrous silicates found in Japan. Bull. Chem. Soc. Japan, 18, 281-329 (1943).

This name, for J. Lemberg, who described similar material in 1877, is given to a dark green to brownish-green mineral found in several Japanese deposits of iron-bearing sandstones. These ores contain magnetite, augite, feldspar, quartz, altered volcanic rocks, etc., cemented by the fine-grained green mineral (grains 0.02 mm. or less). The indices of refraction recorded are somewhat variable, 1.56–1.58; double refraction 0.015–0.030. Material was purified by suspension in water and by magnetic separation. The mineral is readily decomposed by hot dilute HCl, and 9 analyses are given (only in the form of molecular ratios) of the soluble portion. These analyses gave the ratios:

	Al <sub>2</sub> O <sub>3</sub>	FeO*	MgO	CaO	$\mathrm{H}_{2}\mathrm{O}+$	$\begin{cases} Fe_2O_3\\ FeO \end{cases}$
Average analysis	0.40	0.92	1.00	0.34	1.86	_
Extremes of analyses	0.36	0.67	1.00	0.15	1.70	0.76
	to 0.47	to 1.43	to 1.00	to 0.70	to 2.12	to 3.40

\* All iron calculated as FeO.

X-ray powder photographs of the green mineral showed a very close similarity to those of nontronite and montmorillonite, and some similarity to those of garnierite and genthite. It is concluded that the mineral belongs to the montmorillonite group.

DISCUSSION: The data do not seem to justify a new name for what appears to be an intermediate member of the montmorillonite group. Furthermore, the name lembergite has already been used, having been given in 1895 to a synthetic hydrated sodium aluminum silicate, see *Dana's System*, 6th Edition, Appendix I, p. 42.

MICHAEL FLEISCHER

#### Aluminocopiapite

L. G. BERRY, Composition and optics of copiapite. Univ. Toronto Studies, Geol. Ser., No. 51, 21-34 (1947).

The general structural formula of copiapite is shown to be X (OH)<sub>2</sub>  $R_4'''(SO_4)_6 \cdot nH_2O$ , where X is one oxygen equivalent of one or more of the following elements: Na, K, Cu, Fe'', Mn, Mg, Zn, Ca, Al, Fe''', and *n* equals 20. R''' is ferric iron and sometimes in part aluminum. The name aluminocopiapite is proposed for the variety in which X is predominantly aluminum. The names ferricopiapite, magnesiocopiapite, ferrocopiapite, and cuprocopiapite had been proposed previously.

M.F.

#### Frohbergite

R. M. THOMPSON, Frohbergite, FeTe<sub>2</sub>: a new member of the marcasite group. Univ. Toronto Studies, Geol. Ser., No. 51, 35-40 (1947).

See abstract in Am. Mineral., 32, 210 (1947).

M.F.

## NEW MINERAL NAMES

### Monteponite

ERNEST E. FAIRBANKS, The punched card identification of ore minerals. *Econ. Geol.*, **41**, 761-768 (1946). The name monteponite is suggested for the mineral cadmium oxide (CdO), reported by Wittich and Neumann in 1901 from near Monte Ponti, Sardinia. (See *Dana's System*, 7th Ed., pp. 502-503.)

M.F.

## NEW DATA

#### Heazlewoodite

M. A. PEACOCK, On heazlewoodite and the artificial compound Ni<sub>3</sub>S<sub>2</sub>. Univ. Toronto Studies, Geol. Ser., No. 51, 59–69 (1947).

Heazlewoodite was described in 1896 by Petterd as a nickel sulfide from Heazlewood, Tasmania. The description was sketchy, and the mineral has been regarded as being probably a variety of pentlandite. Restudy now shows it to be a valid species.

CHEMICAL PROPERTIES: Composition Ni<sub>3</sub>S<sub>2</sub>, Analysis by I. H. Milne on 0.1 g. gave: Ni 72.13, S 25.96, Fe 0.55, insol. 0.59; sum 99.23%. Etch tests are given.

CRYSTALLOGRAPHIC AND X-RAY DATA: X-ray powder photographs of the natural material were identical with those of the material synthesized by fusing Ni and S in the proportions 3:2 in an evacuated silica glass tube. The spacings agreed closely with those of Westgren (1938) on artificial Ni<sub>3</sub>S<sub>2</sub>. The mineral is rhombohedral with a 5.730, c 7.125 kX; r 4.072 kX,  $\alpha 89^{\circ}25'$ . The space group is  $D_3^7 - R 32$  and the rhombohedral unit cell contains Ni<sub>3</sub>S<sub>2</sub>. X-ray powder data are given.

PHYSICAL PROPERTIES: Luster metallic, color pale bronze, streak light bronze. The mineral is non-magnetic.

Hardness 4, Talmage hardness C+, G. (pycnometer) 5.82 on natural material, 5.87 on synthetic material, 5.87 calculated from x-ray data. In polished section, the mineral is yellowish cream-colored, with strong anisotropism and polarization colors brown to bluish gray, or even lilac to green. A mosaic structure was present (twinning?)

OCCURRENCE: Occurs intergrown with magnetite in a band in serpentine. A very small amount of pentlandite was present. The surface was coated with zaratite.

M.F.

#### Sjögrenite

P. QUENSEL, Om sjögrenite som mineralnamn. Geol. För. Förh., 68, 110-111 (1945).

Quensel points out that the name sjögrenite, proposed by Frondel (Am. Mineral., 26, 309 (1941)) for the hexagonal dimorph of pyroaurite, had previously been given by Krenner (Proc. Geol. Congress Stockholm, Vol. 1, 129 (1910)) to a basic ferric phosphate. This mineral, with the formula  $5Fe_2O_3 \cdot 3P_2O_5 \cdot 8H_2O$ , was first described, without a name, by Kinch and Butler (Mineralog. Mag., 7, 65 (1886)). The phosphate has priority.

DISCUSSION: The phosphate should be re-studied. If it is a valid species, the dimorph of pyroaurite will have to be renamed.

M.F.

#### Wiserite

W. EPPRECHT, Die Eisen- und Manganerze des Gonzen. Beitr. Geol. Schweiz, Geolechn. Ser. Lief. 24, 128 pp. (1946).

Wiserite, first described from Gonzen, Switzerland, in 1842, and considered since 1868 to be altered pyrochroite, was re-examined. The mineral occurs as fibrous, asbestiform masses in fissures on hausmannite, pyrochroite or rhodochrosite. The luster is silky, color white to brownish or reddish. Optically uniaxial, negative,  $n \ge 1.66-1.67$ , nO about 1.74, weakly pleochroic, with E bright yellow-orange, O colorless. Fresh pyrochroite had  $n \ge 1.681$ , nO 1.723.

X-ray powder photographs were distinct from those of pyrochroite, manganite, and synthetic  $\alpha$ —MnOOH and  $\beta$ —MnOOH. (Also seem to be distinct from that of groutite—Abstractor.) Dehydration study showed that most of the water is lost at 200–300°, hausmannite and  $\beta$ —Mn<sub>2</sub>O<sub>3</sub> being among the products formed. Pyrochroite gave a very similar dehydration curve.

Analysis by J. Jakob gave:  $Mn_2O_3$  64.23,  $Al_2O_3$  none,  $Fe_2O_3$  0.13, MnO 9.38, MgO 3.09, CaO,  $K_2O$ ,  $Na_2O$  none,  $H_2O+$  (120°) 16.17,  $H_2O-$  1.10,  $CO_2$  5.26, Cl, SO<sub>3</sub> none; sum 100.00%. The active oxygen content was not determined, but  $Mn_2O_3$  was calculated after assigning the analytical deficiency to active O. This gives roughly  $4Mn_2O_3 \cdot (Mn, Mg)_2 \cdot (CO_3)(OH)_2 \cdot 8H_2O$ .

M.F.

#### DISCREDITED MINERALS

#### Stiepelmannite = Florencite

ERIK R. YGBERG, Svanbergite from Horrsjöberg. Arkiv Kemi, Mineral. Geol., 20A, No. 4, 17 pp. (1945).

The mineral stiepelmannite was first described by Ramdohr and Thilo (see Am. Mineral., 25, 626 (1940)) as a member of the alunite group analogous to the cerium mineral florencite, but with yttrium and ytterbium predominant. Ygberg states (p. 13), "According to a personal communication from Professor Ramdohr, it has later been found that stiepelmannite is a variety of florencite containing Ce, La, and Pr."

M.F.

## Berthonite = Bournonite

R. M. THOMPSON, Berthonite identical with bournonite. Univ. Toronto Studies, Geol. Ser. No. 51, 81-83 (1947).

X-ray powder photographs of berthonite from the type locality Slata Mine, Tunisia, are identical with that of bournonite. This agrees with the work of Harcourt, Am. *Mineral.*, **27**, 109 (1942).

M.F.