NOMINATIONS FOR OFFICERS OF THE MINERALOGICAL SOCIETY FOR 1927

The unanimous nominations of the Council for officers of the Mineralogical Society of America for 1927 are as follows:

President, Austin F. Rogers, Stanford University, California.

Vice-president, George L. English, Rochester, New York.

Secretary, Frank R. Van Horn, Case School of Applied Science, Cleveland, Ohio. Treasurer, Alexander H. Phillips, Princeton University, Princeton, New Jersey. Editor, Walter F. Hunt, University of Michigan, Ann Arbor, Michigan.

Councilor, 1926-30, Alexander N. Winchell, University of Wisconsin, Madison, Wisconsin.

The seventh annual meeting of the Society will be held December 27–29,1926 at the University of Wisconsin, Madison, Wisconsin. It is planned to publish in the December issue of the Journal a *preliminary* list of titles of papers to be presented before the Society at its annual meeting. In order to appear on the advance program titles of papers should be in the hands of the Secretary by *November 10*.

Frank R. Van Horn, Secretary.

REDEFINITION OF SPECIES

Jalpaite

GEORG KALB AND MAXIMILANE BENDIG: Erzmikroskopische Untersuchung der Mineralreihe Silberglanz-Kupferglanz. (Chalcographic Examination of the mineral Series Argentite-Chalcocite.) Centr. Min. Geol., p. 516 (1924).

An analysis of jalpaite from Schlangenberg, Altai, gave Ag 71.73, Cu 14.10 S 16.33, corresponding closely to the analysis of the mineral from Jalpa, Mexico, and the theoretical composition 3Ag₂S.Cu₂S. Cleavage pseudo-octahedral. Between crossed nicols anisotropic. Apparently distinct from argentite with a composition of 3Ag₂S.Cu₂S.

W. F. F.

Kornelite

JOZSEF KRENNER: Kornelite. *Math és Termés. Ert.*, **42**, pp. 1–3 (1926). Also Jozsef Loczka: Mineral analysen. *Ibid.*, pp. 6–17, pp. 20–21.

Chemical Properties: A ferric sulphate of iron; Fe₂O₃. 3 SO₃, $7\frac{1}{2}$ H₂O. Analysis: SO₃ 44.55, Fe₂O₃ 30.17, CaO 0.06, Na₂O 0.11, K₂O 0.09, (NH₄)₂O 0.03, FeO, CuO, P₂O₅ traces, H₂O 24.92. Soluble in water.

Crystallographic Properties: Orthorhombic. $(110):(1\overline{10}) = 52^{\circ}2'$. No terminal faces present.

Physical and Optical Properties: Color pale rose or violet. Luster silky, streak white. Plane of the optic axes parallel to the cleavage b(010). Acute bisectrix parallel to a. $2E=84^{\circ}12'$. Dispersion $\rho > \nu$.

OCCURRENCE. Found with voltaite and coquimbite in the pyrite mines of Szomolonok, Co Szepes, Hungary. W. F. F.