

## Tsilaisite, $\text{NaMn}_3\text{Al}_6(\text{Si}_6\text{O}_{18})(\text{BO}_3)_3(\text{OH})_3\text{OH}$ , a new mineral species of the tourmaline supergroup from Grotta d'Oggi, San Pietro in Campo, island of Elba, Italy

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### ABSTRACT

Tsilaisite,  $\text{NaMn}_3\text{Al}_6(\text{Si}_6\text{O}_{18})(\text{BO}_3)_3(\text{OH})_3\text{OH}$ , is a long-expected new mineral of the tourmaline supergroup. It occurs in an aplitic dike of a LCT-type pegmatite body from Grotta d'Oggi, San Pietro in Campo, island of Elba, Italy, in association with quartz, K-feldspar, plagioclase, elbaite, and schorl. Crystals are greenish yellow with a vitreous luster, a white streak, and show no fluorescence. Tsilaisite has a Mohs hardness of approximately 7; it is brittle with a sub-conchoidal fracture, and has a calculated density of 3.133 g/cm<sup>3</sup>. In plane-polarized light, tsilaisite is pleochroic, O = pale greenish yellow, E = very pale greenish yellow; it is uniaxial negative,  $\omega = 1.645(5)$ ,  $\epsilon = 1.625(5)$ . Tsilaisite is rhombohedral, space group  $R\bar{3}m$ ,  $a = 15.9461(5)$ ,  $c = 7.1380(3)$  Å,  $V = 1571.9(1)$  Å<sup>3</sup>,  $Z = 3$ . The strongest eight X-ray-diffraction lines in the powder pattern [ $d$  in Å( $hkl$ )] are: 3.974(100)(220), 2.942(94)(122), 2.570(79)(051), 2.034(49)(152), 4.205(41)(211), 6.329(22)(101), 2.377(21)(003), and 1.592(21)(550). Analysis by a combination of electron microprobe, secondary ion mass spectrometry, and optical absorption spectroscopy gives SiO<sub>2</sub> = 36.10(3), TiO<sub>2</sub> = 0.32(4), Al<sub>2</sub>O<sub>3</sub> = 37.10(5), MnO = 9.60(10), CaO = 0.09(4), Na<sub>2</sub>O = 2.11(7), K<sub>2</sub>O = 0.03(1), F = 0.79(3), B<sub>2</sub>O<sub>3</sub> = 10.2(6), Li<sub>2</sub>O = 0.8(1), H<sub>2</sub>O = 3.1(2), sum 99.95 wt%. The unit formula is  $^{\text{X}}(\text{Na}_{0.67}\square_{0.30}\text{Ca}_{0.02}\text{K}_{0.01})^{\text{Y}}(\text{Mn}_{1.34}\text{Al}_{1.14}\text{Li}_{0.54}\text{Ti}_{0.04})^{\text{Z}}\text{Al}_6^{\text{T}}(\text{Si}_{5.94}\text{Al}_{0.06})\text{B}_{2.91}\text{O}_{27}^{\text{V}}(\text{OH})_3^{\text{W}}(\text{OH}_{0.39}\text{F}_{0.41}\text{O}_{0.20})$ . The structure, refined also taking into account the positional disorder of the O1 and O2 anions, converged to statistical indices  $R1$  for all reflections of about 2%. The resulting site populations indicate that the Z site is occupied by Al and that the Y site is dominated by Mn<sup>2+</sup>. Aluminum is incorporated at Y through two types of substitutions:  $^{\text{Y}}\text{Al}^{\text{+W}}\text{O}^{2-} \rightarrow ^{\text{Y}}\text{Mn}^{2+\text{+W}}\text{OH}$ , which has the result of replacing OH at the W site by O<sup>2-</sup>, and  $^{\text{Y}}(\text{Al}+\text{Li})^{\text{+W}}\text{F} \rightarrow 2^{\text{Y}}\text{Mn}^{2+\text{+W}}\text{OH}$ , which relates fluor-elbaite to the tsilaisite component. Infrared absorption spectra measured in the principal OH-stretching region were interpreted on the basis of local arrangements consistent with the short-range bond-valence requirements. A compositional trend from fluor-elbaite to tsilaisite is observed in samples from Elba Island. The occurrence of tsilaisite is very rare in nature, as a consequence of both the requirement of extraordinary petrogenetic conditions and limited structural stability.

**Keywords:** Tsilaisite, tourmaline, new mineral species, crystal-structure refinement