

## Metasideronatrite: Crystal structure and its relation with sideronatrite

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### ABSTRACT

Metasideronatrite was obtained as the first dehydration product of sideronatrite,  $\text{Na}_2\text{Fe}(\text{SO}_4)_2(\text{OH})\cdot 3\text{H}_2\text{O}$ , from Sierra Gorda, Chile. The crystal structure of metasideronatrite was solved by direct methods and refined by full-matrix least-squares to  $R = 0.039$ , using 574 independent reflections with  $I > 3.0\sigma(I)$ . It is orthorhombic, space group  $Pbnm$ , with  $a = 7.3959(8)$ ,  $b = 16.0979(15)$ ,  $c = 7.1607(8)$  Å,  $V = 852.5(2)$  Å<sup>3</sup>,  $Z = 4$ . The crystal-chemical formula derived from this structural study is  $\text{Na}_2\text{Fe}(\text{SO}_4)_2(\text{OH})\cdot\text{H}_2\text{O}$ . The backbone of the structure is the same as that in sideronatrite: infinite  $[\text{Fe}^{3+}(\text{SO}_4)_2(\text{OH})]^{2-}$  chains of interconnected octahedra and tetrahedra parallel to the  $c$  axis. These chains are linked primarily by Na atoms to build a 3-dimensional network of strong (Fe-O-S) and weak (Na-O) bonds. Another prominent feature of the structure is the arrangement of distorted  $(\text{NaO}_5\text{H}_2\text{O})$  octahedra, which alternately share one edge and one face to form columns parallel to the  $[\text{Fe}^{3+}(\text{SO}_4)_2(\text{OH})]$  chains. Subsidiary intra-chain bonds are provided by H atoms belonging to  $\text{OH}^-$  groups shared by adjacent Fe octahedra, and to the unique water molecule shared between two adjacent  $(\text{NaO}_5\text{H}_2\text{O})$  octahedra. At normal conditions of relative humidity (RH) and temperature (i.e.,  $\text{RH} > 60\%$  and  $T < 40$  °C), metasideronatrite rehydrates rapidly to sideronatrite.

The structure solution has allowed us to: (1) investigate the strong relation between sideronatrite and metasideronatrite; (2) elucidate the mechanism involved in the transformation of metasideronatrite into the order/disorder (OD) structure of sideronatrite; and (3) get insight into the stability of this mineral from the valence-matching principle applied to the main structural unit  $[\text{Fe}^{3+}(\text{SO}_4)_2(\text{OH})]^{2-}$  and  $\text{Na}^+$  interstitial species. The weak hydrogen bonds and the particular arrangement of the face-sharing adjacent  $[\text{NaO}_5(\text{H}_2\text{O})]$  octahedra are the main factors affecting the stability of metasideronatrite.

**Keywords:** Metasideronatrite, structure, crystal chemistry, dehydration, sideronatrite