## Evolution of local electronic structure in alabandite and niningerite solid solutions [(Mn,Fe)S, (Mg,Mn)S, (Mg,Fe)S] using sulfur *K*- and *L*-edge XANES spectroscopy

## SHANNON P. FARRELL,<sup>1</sup> MICHAEL E. FLEET,<sup>1</sup> ILIA E. STEKHIN,<sup>2</sup> ANTONINA KRAVTSOVA,<sup>2</sup> Alexander V. Soldatov,<sup>2</sup> and Xiaoyang Liu<sup>1</sup>

<sup>1</sup>Department of Earth Sciences, University of Western Ontario, London, Ontario N6A 5B7, Canada <sup>2</sup>Faculty of Physics, Rostov State University, 344090, Rostov-on-Don, Russia

## ABSTRACT

Synchrotron radiation S *K*- and *L*-edge X-ray absorption near-edge structure (XANES) spectra are reported for the cubic, rocksalt (B1) structure sulfides niningerite (MgS), alabandite (MnS), and oldhamite (CaS), and for their solid solutions (Mn,Fe)S and (Mg,Mn)S, and S *L*-edge XANES spectra are reported also for (Mg,Fe)S solid solutions. Pre-edge features at the S *K*-edge are attributed to transition of S 1*s* electrons to the lowest available unoccupied S 3*p*  $\sigma^*$  antibonding states hybridized with metal 3*d*(*e<sub>g</sub>*) states, and at the S *L*-edge to transition of S 2*p* electrons to unoccupied S 3*s*  $\sigma^*$ , and 3*d* antibonding states hybridized with metal 3*d*(*e<sub>g</sub>*) states.

The S *K*-edge XANES spectra for the solid solutions show a progressive participation of 3*d* orbitals in metal-S bonding with increase in substitution by Fe in (Mn,Fe)S and (Mg,Fe)S and Mn in (Mg,Mn)S through progressive increase in the area of the pre-edge feature. However, the pre-peak area does not increase linearly in each solid solution series showing that a real change in bulk electronic properties has occurred. Increase in pre-peak area reflects an increase in overall attainability of metal 3*d* states for hybridization with S 3*p*  $\sigma^*$  antibonding states as proportionally more metal 3*d* orbitals become available. The S *L*-edge XANES spectra show progressive evolution of pre-edge features at the  $L_3$ - and  $L_2$ -edges (a<sub>1</sub> and a<sub>2</sub>, respectively). Only a<sub>2</sub> is present in the S *L*-edge XANES spectrum of FeS (troilite), and with progressive decrease in Fe content in (Mn,Fe)S and (Mg,Fe)S solid solutions, a<sub>1</sub> first appears, then becomes dominant. Since a<sub>1</sub> is attributed to transition of S  $2p_{3/2}$  electrons to S 3*s*  $\sigma^*$  states hybridized with metal 3*d*( $e_g$ ) and 3*d*( $t_{2g}$ ) states, this appears to represent an increased contribution from metal-S  $\pi$ -bonding. The results show that the size and position of the pre-edge features to the S *K*- and *L*-edges are controlled more by the DOS of hybridized 3*d*( $e_g^\beta$ ) and 3*d*( $t_{2g}^\beta$ ) states and nearest-neighbor coordination of the metal atoms than by the precise coordination of S and the extended structure of the sulfide.

The full multiple scattering approach has been applied to the calculation of the S *K*-edge XANES spectra of MgS, MnS, and CaS. Results are consistent with experimental XANES spectra, especially for the pre-edge features, which are often neglected in such calculations.