Electron paramagnetic resonance spectroscopic study of synthetic fluorapatite: Part I. Local structural environment and substitution mechanism of Gd³⁺ at the Ca2 site

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ABSTRACT

Gd-doped fluorapatite $(1.2 \pm 0.2 \text{ wt}\% \text{ Gd}_2\text{O}_3)$, synthesized from CaF₂-rich melts, has been investigated as single crystals and powder samples by using X-band (~9.4 GHz) electron paramagnetic resonance (EPR) spectroscopy at ~295 and 120 K. The well-resolved X-band EPR spectra yielded a previously unreported type of Gd³⁺ center "a" (S = 7/2) and also suggested the possible presence of a second and partly resolved type of Gd³⁺ center "b." In particular, the single-crystal X-band EPR spectra of center "a" from three orthogonal-rotation planes allowed determination of the spin-Hamiltonian parameters, including the spin terms of type BS (matrix **g**) and S² (matrix **D**) and the parameters has been confirmed by agreement between observed and simulated EPR spectra for both single crystals and powder samples.

The principal values of the matrices **g** and **D** indicate that the local symmetry of center "a" in the X-band EPR spectra is rhombic. The principal axis directions of the **D** suggest that this Gd³⁺ center arises from a substitution of Gd³⁺ ion into the Ca2 type of site. This assignment is supported by the results of a pseudo-symmetry analysis using the S⁴ parameters, e.g., the calculated twofold pseudo-symmetry axis coincides with the twofold rotoinversion axis of the Ca2 site. The local structural environment of this Gd³⁺ ion suggests that the ion is incorporated via the mechanism Gd³⁺ + O²⁻ \leftrightarrow Ca²⁺ + F⁻.