

Supplementary Materials

Precipitation of low-temperature disordered dolomite induced by extracellular polymeric substances of methanogenic Archaea *Methanosarcina barkeri*: Implications for sedimentary dolomite formation

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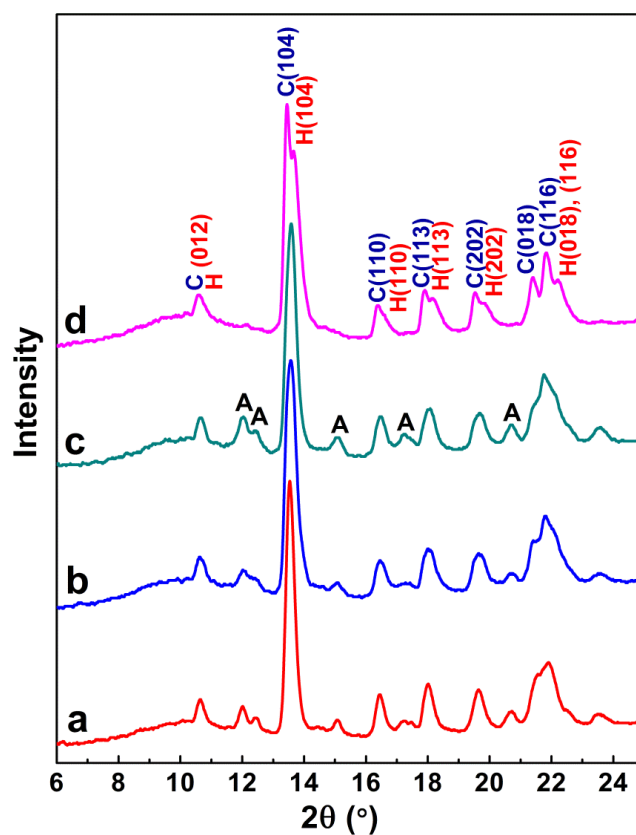


Fig. S1 Typical XRD patterns of synthetic HMC from control experiments. The initial Mg:Ca ratio in solutions where carbonate precipitated was 3:1 (a), 4:1 (b), 5:1 (c), and 8:1 (d), respectively.

Peaks correspond to: A: aragonite; C: calcite seeds; H: HMC.

(a): HMC ($d_{104} = 3.0128 \text{ \AA}$, 8.4 mol% MgCO_3) and a small amount of aragonite.

(b): HMC ($d_{104} = 3.0078 \text{ \AA}$, 9.5 mol% MgCO_3) and a small amount of aragonite.

(c): HMC ($d_{104} = 3.0027 \text{ \AA}$, 11.5 mol% MgCO_3) and a small amount of aragonite.

(d): HMC ($d_{104} = 2.9831 \text{ \AA}$, 18.5 mol% MgCO_3) and a small amount of aragonite.

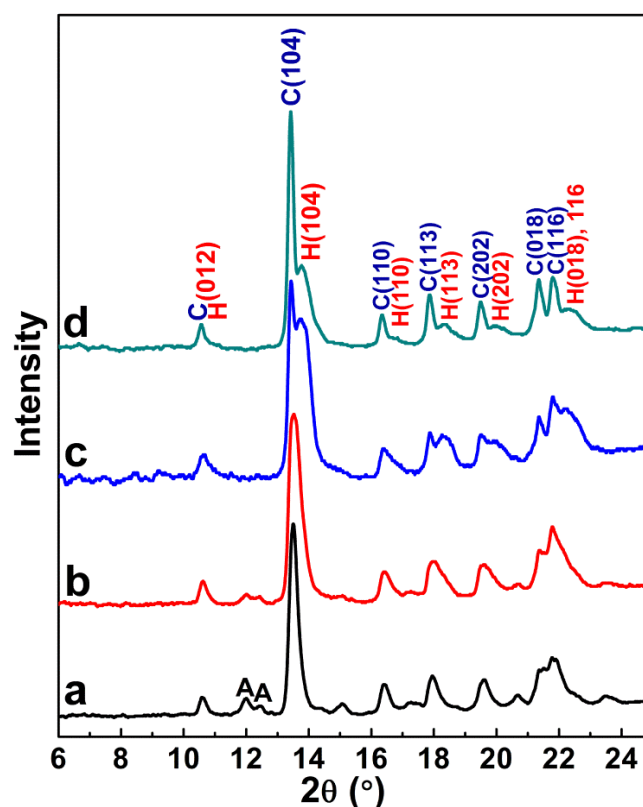


Fig. S2 Typical XRD patterns of synthetic Ca-Mg carbonates induced by non-metabolizing biomass of *M. barkeri* (65 mg L⁻¹). The initial Mg:Ca ratio in experimental solutions where carbonate precipitated was 3:1 (a), 4:1 (b), 5:1 (c), and 8:1 (d), respectively. Peaks correspond to: A: aragonite; C: calcite seeds; H: HMC.

(a): HMC ($d_{104} = 3.0215$ Å, 4.7 mol% of MgCO₃) and a small amount of aragonite.

(b): HMC ($d_{104} = 3.0004$ Å, 12.5 mol% of MgCO₃) and a small amount of aragonite.

(c): HMC ($d_{104} = 2.9590$ Å, 28.6 mol% of MgCO₃).

(d): HMC ($d_{104} = 2.9568$ Å, 30.0 mol% of MgCO₃).

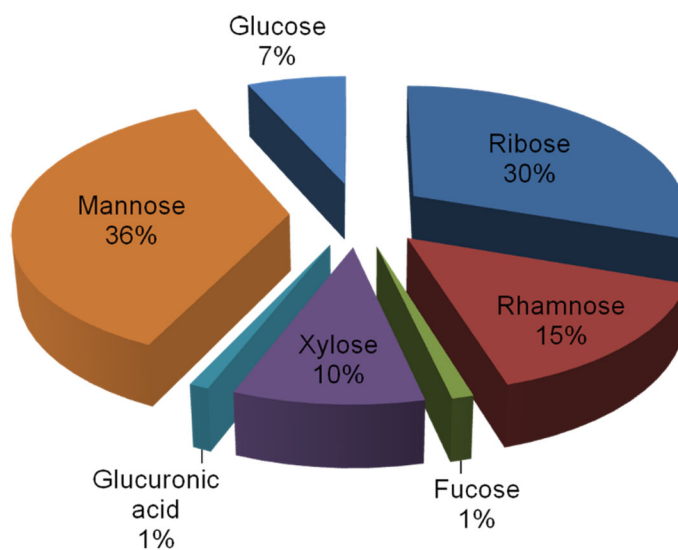


Fig. S3. The monosaccharide composition of the polysaccharides in EPS as analyzed by GC/MS.