Probing the transformation paths from aluminum (oxy)hydroxides (boehmite, bayerite, and gibbsite) to metastable alumina: A view from high-resolution ²⁷Al MAS NMR

HYO-IM KIM^{1,}[†] AND SUNG KEUN LEE^{1,2,*}

¹School of Earth and Environmental Sciences, Seoul National University, Seoul, Korea ²Institute of Applied Physics, Seoul National University, Seoul, Korea

ABSTRACT

Detailed knowledge of the structural evolution of diverse aluminum (oxy)hydroxides with varying temperatures up to ~ 300 °C provides insights into the dehydration processes involving transitions among metastable phases on Earth's surface and in the crust. Here, we report the high-resolution solid-state ²⁷Al NMR spectra for three different types of aluminum (oxy)hydroxides (i.e., boehmite, bayerite, and gibbsite) with varying annealing temperatures up to 300 °C, revealing the effect of distinct precursor minerals on the stability of metastable alumina. ²⁷Al MAS NMR results allow us to obtain the quantitative fractions and NMR parameters for each phase during transformation. The results demonstrate that each aluminum (oxy)hydroxide phase follows a unique transformation path to metastable alumina. The ²⁷Al MAS and 3QMAS NMR spectra of boehmite show that a minor but observable ^[4]Al signal (~2%) is detected at ~50 °C, and the ^[4]Al fraction gradually increases up to 300 °C (~16%), indicating that the phase transformation from boehmite to γ - $/\eta$ -Al₂O₃ occurs at a temperature as low as ~50 °C, significantly lower than earlier estimations based on XRD. Together with the [4]Al fraction, the [5]Al fraction increases from <1% at 50 °C to $\sim2.3\%$ at 300 °C, whereas the NMR results of bayerite and gibbsite do not show the presence of ^[5]Al. In addition, the ²⁷Al 3QMAS NMR spectra resolved the ^[6]Al site in boehmite and that in γ -/ η -Al₂O₃, which could not be uniquely determined from 1D NMR spectra. The population of bayerite abruptly decreases from 100% (at 150 °C), through ~47% (at 200 °C), to 0% (at 250 °C), indicating that the phase transition from bayerite to boehmite $+\gamma$ -/ η -Al₂O₃ occurs within a narrow temperature range. As for gibbsite, while ^[4]Al is not observed in the spectra up to 200 °C, the [4]Al fraction of ~2% is observed in the spectra for gibbsite annealed at 250 °C, and the ^[4]Al fraction increases rapidly to $\sim 15\%$ as the annealing temperature increases to 300 °C, suggesting that the phase transformation into γ -/ η -Al₂O₃ occurs at ~250 °C. The results confirm that the phase transformation paths (gradual vs. dramatic) depend on the type of precursor minerals. Particularly, the onset temperature of the phase transformation from boehmite to metastable alumina (\sim 50 °C) is lower than those from other precursor minerals (>150 $^{\circ}$ C). Furthermore, the phase transformation from boehmite to γ -/ η -Al₂O₃ occurs gradually within broad temperature ranges from ~50 °C. This is due to their configurational disorder as evidenced by the presence of ^[5]Al. The observed structural evolution in aluminum (oxy)hydroxides in the low-temperature range facilitates our understanding of the nature of phase transformation and dehydration of oxides and hydroxides in the Earth's surface environments.

Keywords: Metastable (transition) alumina, aluminum (oxy)hydroxide, solid-state NMR, γ -/ η -Al₂O₃, phase transformation, extent of disorder