

Integral molar absorptivities of OH in muscovite at 20 to 650 °C by in-situ high-temperature IR microspectroscopy

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ABSTRACT

The change with temperature of IR absorption bands in OH in muscovite was studied using unpolarized in-situ high-temperature infrared microspectroscopy. The molar absorption coefficient ϵ at 3628 cm^{-1} for OH in muscovite at room temperature ($20\text{ }^{\circ}\text{C}$) has been determined to be $127 \pm 6\text{ L/mol}\cdot\text{cm}$. Using the orientation factor $\gamma = 0.47$ for the angle between OH vector and c^* axis ($75 \pm 5^{\circ}$), the true molar absorption coefficient ϵ at 3628 cm^{-1} is determined to be $270 \pm 10\text{ L/mol}\cdot\text{cm}$ at $20\text{ }^{\circ}\text{C}$. Integral molar absorptivities are also determined from 20 to $650\text{ }^{\circ}\text{C}$ showing no weight loss. The value decreases from $7060 \pm 190\text{ L/mol}\cdot\text{cm}^2$ at $20\text{ }^{\circ}\text{C}$ to $5190 \pm 270\text{ L/mol}\cdot\text{cm}^2$ at $650\text{ }^{\circ}\text{C}$ (26% decrease). OH orientation of muscovite sample measured by polarized IR microspectroscopy at $20\text{ }^{\circ}\text{C}$ showed that the angle between the projection of OH vector to (001) plane (the OH' vector) and b axis is 30.5° at $20\text{ }^{\circ}\text{C}$ and did not change greatly at higher temperatures until $650\text{ }^{\circ}\text{C}$. The tilting of OH dipoles toward the c^* axis from 75° to about 43° while keeping the same angles along b axis would explain the observed decrease in integral molar absorptivities at higher temperatures, but further studies are needed to clarify the OH behavior at high temperatures.

Keywords: Muscovite, OH, in-situ high-temperature IR microspectroscopy, integral molar absorptivity, OH orientation