## A crystal-chemical investigation of clinozoisite synthesized along the join Ca<sub>2</sub>Al<sub>3</sub>Si<sub>3</sub>O<sub>12</sub>(OH)-Ca<sub>2</sub>Al<sub>2</sub>CrSi<sub>3</sub>O<sub>12</sub>(OH)

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## ABSTRACT

Cr<sup>3+</sup>-bearing clinozoisite along the join Ca<sub>2</sub>Al<sub>3</sub>Si<sub>3</sub>O<sub>12</sub>(OH)-Ca<sub>2</sub>Al<sub>2</sub>Cr<sup>3+</sup>Si<sub>3</sub>O<sub>12</sub>(OH) was synthesized using cold-seal pressure vessels at  $P_{\rm H_{2O}} = 0.35$  to 0.40 GPa and T = 500 °C and a piston-cylinder apparatus at  $P_{\rm H_{2O}} = 0.8$  to 1.5 GPa and T = 500 to 800 °C. Gel-starting materials of Ca<sub>2</sub>Al<sub>3-q</sub>Cr<sup>3+</sup>Si<sub>3</sub>O<sub>12.5</sub> composition with q = 1.00, 0.75, 0.50, and 0.25 were employed to maximize the yields of clinozoisite. Mass fractions of clinozoisite in the experimental products with q = 0.50, 0.75, and 1 were about 70 to 90% along with lesser amounts of eskolaite, garnet, and quartz. Clinozoisite crystallized from the gel with q = 0.25 was associated only with zoisite. The crystal structures of clinozoisite in four runs, containing 0.28, 0.49, 0.50, and 0.62 Cr apfu were refined using X-ray powder diffraction data and the Rietveld method. The amount of Cr<sup>3+</sup> at the octahedral M3 and M1 sites ranged from 0.37(1)–0.16(1) to 0.25(1)–0.12(1) apfu, respectively. Corresponding  $K_{\rm D} = ({\rm Cr}^{3+}/{\rm Al})^{M1}/({\rm Cr}^{3+}/{\rm Al})^{M3}$  values range between 0.57 and 0.73. The M2 site contained only Al. The  $K_{\rm D}$  values, and published results for intracrystal-line partitioning in epidote and piemontite, show that the preference of Cr<sup>3+</sup> for M1 is stronger than that of Fe<sup>3+</sup> and Mn<sup>3+</sup> in spite of the fact that most Cr<sup>3+</sup> is partitioned into M3. Unit-cell parameters of clinozoisite increase with increasing Cr<sup>3+</sup>. Variations in macroscopic unit-cell parameters can be related to variations in the local M3-O*i* and M1-O*i* distances.

Keywords: Clinozoisite, zoisite, chromium, synthesis, crystal chemistry, Rietveld refinement