

*American Mineralogist, Volume 93, pages 1191–1194, 2008*

LETTER

**A  $P_{\text{H}_2\text{O}}$ -dependent structural phase transition in the zeolite natrolite**

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ABSTRACT

The crystal structures of natrolite and its dehydrated high-temperature phases ( $\alpha$ 1- and  $\alpha$ 2-metanatrolite) have been determined from powder X-ray diffraction measurements as a function of temperature and partial pressure of water ( $P_{\text{H}_2\text{O}}$ ) to characterize the phase transition behavior. The evolution of crystal structure as a function of temperature shows two different phase transitions, depending on the  $P_{\text{H}_2\text{O}}$ , with  $\alpha$ 1-metanatrolite occurring at elevated  $P_{\text{H}_2\text{O}}$  and  $\alpha$ 2-metanatrolite occurring at low  $P_{\text{H}_2\text{O}}$ . Our discovery of  $\alpha$ 2-metanatrolite implies the existence of more than one transition mechanism, which we correlate with the migration of  $\text{Na}^+$  ions and the *rate* of evolution of  $\text{H}_2\text{O}$  molecules. The transition behavior is rationalized in terms of two cooperating mechanisms: (1) dehydration-induced processes, which determine the phase transition temperature; and (2) thermally induced processes, which determine how the framework and its extraframework cations are modified.

**Keywords:** Phase transition, natrolite, crystal structure, Rietveld refinement, TGA