

## **The crystal structures of grossular and spessartine between 100 and 600 K and the crystal chemistry of grossular-spessartine solid solutions**

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### **ABSTRACT**

Spessartine ( $\text{Mn}_3\text{Al}_2\text{Si}_3\text{O}_{12}$ )-grossular ( $\text{Ca}_3\text{Al}_2\text{Si}_3\text{O}_{12}$ ) solid solutions were synthesized at high pressures and temperatures. Compositionally homogeneous garnets are obtained by crystallizing solid-solution glasses prepared from oxides. The unit-cell parameter,  $a$ , for the different solid solutions was determined by X-ray powder diffraction methods and the results give positive deviations from ideal volumes of mixing that can be described with a symmetric mixing model with  $W^v = 0.80$  ( $\pm 0.04$ )  $\text{cm}^3/\text{mol}$ . The degree of non-ideality is a function of the difference in size between the  $\text{Ca}^{2+}$  and  $\text{Mn}^{2+}$  cations and is consistent with the range of those observed for the other aluminosilicate garnet binary joins. The crystal structures of synthetic grossular and spessartine were collected at 50 K intervals between 103 K and 498/648 K using single-crystal X-ray diffraction methods. The rotation of the rigid  $\text{SiO}_4$  tetrahedra changes slightly by  $0.3(1)^\circ$  for grossular and  $0.2(1)^\circ$  for spessartine between 103 and 648 K. The volume expansions of the polyhedra were calculated and their distortions in grossular and spessartine were analyzed as a function of temperature. The linear thermal expansion coefficients of the Al-O and two X-O bond were also calculated for almandine, pyrope, grossular, and spessartine. The thermal expansion of spessartine is similar to that of grossular. In terms of polyhedral distortion and bond-valence values, spessartine has the most ideal structure of the aluminosilicate garnets. This could explain its large  $P$ - $T$  stability field and the ease of synthesis at low pressures.