

The structures of becquerelite and Sr-exchanged becquerelite

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ABSTRACT

The crystal structures of becquerelite, $\text{Ca}[(\text{UO}_2)_6\text{O}_4(\text{OH})_6](\text{H}_2\text{O})_8$, and Sr-exchanged becquerelite obtained by ion exchange, $\text{Sr}_{1.27}[(\text{UO}_2)_3\text{O}_{3.54}(\text{OH})_{1.46}](\text{H}_2\text{O})_3$, have been refined using diffraction data collected with $\text{MoK}\alpha$ X-rays and a CCD-based detector. The structure of becquerelite, orthorhombic, space group $Pn2_1a$, $a = 13.8527(5)$, $b = 12.3929(4)$, $c = 14.9297(5)$ Å, $V = 2563.2(1)$ Å³, has been refined on the basis of F^2 for 4875 unique reflections to $R1 = 3.39\%$, calculated using 4581 unique observed reflections ($|F| \geq 4\sigma_F$), and a goodness-of-fit (S) of 1.04. Sr-exchanged becquerelite was obtained by placing single crystals of synthetic becquerelite in 2.5 M SrCl_2 solution for 60 h at 160 °C. The structure of Sr-exchanged becquerelite is trigonal, space group $P3$, $a = 7.020(4)$, $c = 6.992(6)$ Å, $V = 298.4(3)$ Å³, and has been refined on the basis of F^2 for 683 unique reflections to $R1 = 4.26\%$, calculated using the 564 unique observed reflections ($|F| \geq 4\sigma_F$), and an S of 1.01. The results for becquerelite confirm the cation polyhedra and structural connectivity reported previously, but collection of data for a high-quality crystal using a CCD-based detector has substantially improved the precision of the atom positions. The structure contains α - U_3O_8 -type sheets of uranyl pentagonal bipyramids, with a single symmetrically distinct Ca cation and eight symmetrically distinct H_2O groups located in the interlayer. The structure of Sr-exchanged becquerelite also contains α - U_3O_8 -type sheets of uranyl pentagonal bipyramids, although the amount of H in the sheets is lower than for becquerelite. The interlayer contains two symmetrically distinct Sr cations and a single H_2O group. Ion exchange of Sr into the interlayer of becquerelite without destruction of the crystals has potential important implications for the mobility of ^{90}Sr in contaminated areas, and for the geologic disposal of nuclear waste.